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(54) ACTIVE-SITE ENGINEERING OF

NUCLEOTIDYLYLTRANSFERASES AND GENERAL ENZYMATIC METHODS FOR THE SYNTHESIS OF NATURAL AND
"UNNATURAL" UDP- AND TDP-NUCLEOTIDE SUGARS
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## ABSTRACT

The present invention provides mutant nucleotidylyl-transferases, such as $\mathrm{E}_{p}$, having altered substrate specificity; methods for their production; and methods of producing nucleotide sugars, which utilize these nucleotidyly1-transferases. The present invention also provides methods of synthesizing desired nucleotide sugars using natural and/or modified Ep or other nucleotidyltransferases; and nucleotide sugars sythesized by the present methods. The present invention further provides new glycosyl phosphates, and methods for making them.

## 2 Claims, 32 Drawing Sheets

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## FIG. 1



FIG. 2
Substrate

## FIG. 3

(a)



FIG. 4


FIG. 5



amphotericin

 (antittungal)



FIG. 6


## FIG. 7


b


## FIG. 8 (a)



FIG. 8(b)

Ep


FIG. 9


## FIG. 10

a


FIG. 10 cont.


FIG. 11


Thymidylyltransferase Alignment


FIG. 12

## Ep with Gle-1-P Uridylyltransferases



## Ep with Glc-1-P Adenylyltransferases




FIG. 14

## Ep with Man-1.P Guanylyltransferases

ED
D83000. (G1c.dT)
BAA34807 (Man.G)
AAC39498 (man-G)



340

MLKO四
A.

Ep

Ep $083000 \cdot(G / 6 \cdot d T)$
D83000.(G1e-dT)
AA34807 (MAN-G)
BAA34B07 (Man-G)
AAC3949 (man-G)
AA 34807 (Man-G)
$\rightarrow \frac{1}{1}$

FIG. 15

Ep with NAcGlc-1-P Uridylyltransferases


FIG. 16

## Ep with Gle-1-P Cytidylyltransferases




FIG. 19 (A)

| Sequences producing high-scoring segment pairs: |  |  | Smallest Sum Probability P(N) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | gi\|1710100 | GLUCOSE-1-PHOSPHATE THYMTDYLYLTRANSFERAS.- | 1510 | 0 | 1. |
| 2 | gi\|9957817 | Glucose-1-phosphate thymidylyltransferas... | 1507 | 0 | 1 |
|  | gi\|9957822 | Glucose-1-phosphate thymidylyltransferas... | 1499 | 0 | 1 |
|  | gi\|9957847 | Glucose-1-phosphate thymidylyltransferas... | 1497 | 0 | 1 |
|  | gi\|9957866 | Glucose-1-phosphate thymidylyltransferas... | 1496 | 0 | 1 |
|  | gi\|9957852 | Glucose-1-phosphate thymidylyltransferas... | 1488 | 0 | 1 |
|  | gi\|9957857 | Glucose-1-phosphate thymidylyltransferas... | 1450 | 0 | 1 |
|  | gi\|9957836 | Glucose-1-phosphate thymidylyltransferas... | 1444 | 0 | 1 |
|  | gi\|1073702 | Rfba protein - Shigella flemneri (strain | 1440 | 0 | 1 |
|  | gi\|141362 | GLUCOSE-1-PHOSPHATE THYMIDYLYLTEANSFERAS | 1437 | 0 | 1 |
|  | gi\|9957831 | Glucose-1-phosphate thymidylyltransferas... | 1429 | 0 | 1 |
|  | gi\|9957841 | Glucose-1-phosphate thymidylyltransferas.- | 1424 | 0 | 1 |
|  | gi\|2507297 | GLUCOSE-1-PHOSPHATE THYMIDYLYLTRANSFERAS.. | 1408 | 0 | 1 |
|  | gi\|2121141 | Glucose-1-phosphate thymidylyltransferas..- | 1408 | 2.7e-178 | 1 |
|  | gi\|9957862 | Glucose-1-phosphate thymidylyltransferas... | 1359 | $2.7 e-178$ $7.1 e-178$ | 1 |
|  | gi\|9957827 | Glucose-1-phosphate thymidylyltransferas... | 1356 | 7.1e-178 2. $2 e-171$ | 1 |
|  | gi\|S85826 | GLUCOSE-1-PHOSPHATE THYMLDYLYLTRANSFERAS... | 1185 | $2.8 e-154$ | 1 |
|  | gi\|11348597 gi| 3135675 | Glucose-1-phosphate thymadylyltransferas... Putative glucosen 1 -phosphate thymidyltra.. | 1139 | $6.3 e-148$ | 1 |
|  | gi\|3608394 | Putative glucose-1-phosphate thymidyl tr... | 1112 | 3. $4 e-144$ | 1 |
|  | gi\|1666508 | Rfba [Leptospira interrogans] | 1103 | $5.9 e-143$ | 1 |
|  | gi 4234804 | Pola [Leptospira borgpeterseniil | 1092 | 1.9e-141 | 1 |
|  | gi\|1881544 | Glucose-1-phosphate thymidyl transferase.. | 1073 | 8.1e-139 | 1 |
|  | gi\|2500162 | GLUCOSE-1-PHOSPHATE THYMIDYLYLTRANSFERAS... | 1070 | $2.1 e$ | 1 |
|  | gi\|7471939 | Glucose-1-phosphate thymidylyltransEeras... | 1069 | 2.9 e | 1 |
|  | gi\|7434861 | Glucose-1-phosphate thymidylyltransferas.. |  | $1.8 e-136$ | 1 |
|  | gi 4200433 | Cps 2L [Streptococcus pneumoniae] | 1055 | 1.8e-136 | 1 |
|  | gi\|3320399 | Glucose-1-phosphate thymidyl transferase.. | 1051 | 8. 8 e- -136 | 1 |
|  | gi\|7592816 | D-glucose-l-phosphate thymidylyltransfer... | 1051 | $8.8 e-136$ $5.9 e-135$ | 1 |
|  | gi]5545318 | Glucose-1-phosphate thymidylyltransferas... | 1045 | $5.9 e-135$ $5.9 e-135$ | 1 |
|  | gi\|1944160 | Glucose-1-phosphate-thymidylyltransferas... | 1045 | $5.9 e-135$ $4 e-134$ | 1 |
|  | gi\{4406249 | Glucose-1-phosphate thymichlyl transfera... | 1039 | 4e-134 | 1 |
|  | gi\|3832506 | Glucose-1-phosphate thymidylyl transfera.. | 1036 | 4e-134 | 1 |
|  | Gi\|1710101 | GLUCOSE-1-PHOSPHATE THMMIDYLYLIRANSFERAS.. | 1033 | $2.7 e-133$ | 1 |
|  | gi\|3907610 | Glucose-1-phosphate thimidylyl transfera.. | 1033 1031 | $2.7 e-133$ $5.1 e-133$ | 1 |
|  | gi\|9977936 | GLUCOSE-1-PHOSPHATE THYMIDYLYLTRANSFERAS... | 1031 | 9.6e-133 | 1 |
|  | gi\|1098479 | Glucose-1-phosphate thymidyl transferase. | 1023 | $6.5 e-132$ | 1 |
|  | gi\|7434867 | Probable glucose-1-phosphate thymidylylt.. | 1023 | $6.5 e-132$ $2.3 e-131$ | 1 |
|  | gi\|9978667 | GLUCOSE-1-PHOSPHATE THMTHYLYLTRANSEERAS... | 1019 | $2.3 e-131$ $4 \mathrm{e}-130$ | 1 |
|  | gi\|585825 | GLUCOSE-1-PHOSPHATE THYTIDYLYLTRANSFERAS.. | 1010 | 4e-130 | 1 |
|  | Gi\|2500161 | PRORABLE GLUCOSE-1-PHOSPHATE THYMLDYLYLT... |  | $1 e^{-129}$ 8e-128 | 1 |
|  | gi\|2507298 | GLUCOSE-1-PHOSPHATE THYMTDYLYLTRANSFERAS. | 993 | $1.8 e-128$ |  |


|  |  |
| :---: | :---: |
| 4. gi\|59 | Glucose-1-phosphate thymi |
| 45. gi\|11261715 | Glucose-1-phosphate thymidylyltransEeras... |
| . gi\|5199111 | Glucose-1-phosphate thymidyl |
| 47. gi\|74348 |  |
| 8. gi\|1710029 | LUCOSE-1-PHOSPHATE THYMTDYLYLTRA |
| . gi\|1314579 | lucose-1-phosphate thymidy |
| 9. gi\|189 | 7 |
| 51. gi\|667 | tive glucose-1-phosphate thymidyl |
| 52. gi)6688595 | ala protein [Legionella preumophila] |
| 53. gi\|148192 | milar to Streptomyces griseus stro |
| 1421098 | pothetical protein o292 |
| 74 | 1-phosph |
| 56. gil92 | ative dTDP-1-glucose symmase; Akny [... |
| 57. gi\|9714084 | Glucose-1-phosphate thymidyltrans |
| 58. 9i\}6018314 | Putative drop-glucose synthase |
| 59. ฐi\|3789899 | Alpha-D-giucose-1-phosptate thy |
| 60. gi/7688728 | NowV [Streptor |
| 61. gi\|110952 | OTOP-glucose synthase; glucose-1-phospha. |
| 62. gi\|108 |  |
| 63. gi\|48 | omolog |
| \|592 | cose-1-phosphate thymid |
| 65. gi 4884 | -glucose synthase (Streptomy |
| 66. gi\|5579 | SpcK [Streptonyces flayopersicus] |
| 67. gi\|4033331 | DIDP-glucose symthase [Actinoplanes |
| 68. gi\|580705 | OAC3 [Azorhizobium caulinodansl |
| 69. gil1072851 | Probable glucose-1-phosphate thy |
| 70. gi\|2804683 | Glucose-1-phosphate thymidyl |
| 71. gij2804721 | Glucose-1-phosphate thymidyl |
| 72. gi\|2127 | Glucose-1-phosphate thymidyl |
| 73. gi\|19 | Gl |
| 74. gil45 | Glucose-1-phosphate thymidy |
| 75. gi\|73 | T poly |
| 76. gi\|101 | spore coat polysaccharide synthesis folu.. |
| 77. gi\|11279395 | Glucose-1-phosphate thymidylyltransfer |
| 78. gi\|7329194 | DTDP-D-glucose synthase \{Stre |
| 79. gi\|4731596 | Blmb [Streptonyces bluensis] |
| 80. gi 440 | Glucose-1-phosphate thymidyly |
| 81. gi/74 | Glucose-1-phosphate thymidylyltransferas... |
| 82. gi\| 74 | Hypothetical protein - Symechoc |
| 日3. gi\|11279397 | Probable dTOP-2-glucose synthase [imp |
| 84. gi\|280334 | StrD protein - Streptomyces griseus |
| 85. gi\|134 |  |
| 86. gi\|l1 | Glucose-1-phosphate thymidyl |
| B7. gi/3256058 | StrD [Streptomyces glaucescens] |
| 88. gi\|11497938 | Glucose-1-phosphate thymidylyltra |
| 89. gil975 | Glucose-1-phosphate thymidylyltran |
| 90. gi\|7 | lucose synthase - Strept |
| 91. gi\|7 | Glucose-1-phosphate thymidylyltransferas |
| 92. gi\|6002933 | DNDP-glucose symthetase [Streptomyces Er |
| 93. gi\|7448164 | Glucose-1-phosphate thymidylyltransfer |
| 4. gi\|7448 | ble glucose-1-phosphate thymidylyl |
| 95. gi\|4240 | hexose synthetase homolog [Streptomy... |
| 96. gi\|2209217 | Glucose-1-phosphate thymidyl transf |
| 7. gi\|6015646 | l-phosphate thymidylyl |
| 98. gi\|69 |  |
| i 4884958 |  |


| 998 | 1. Be-128 |  |
| :---: | :---: | :---: |
| 549 | $6.2 \mathrm{e}-12 \mathrm{c}$ | 2 |
| 984 | 1.6e-126 | 1 |
| 540 | 6.5e-125 | 2 |
| 966 | 4. $8 e-124$ | 1 |
| 948 | 1. $5 \mathrm{e}-121$ | 2 |
| 466 | 3. 5e-120 | 3 |
| 551 | $5.9 \mathrm{e}-120$ | 2 |
| 933 | 1.7e-119 |  |
| 933 | 1.7e-119 | 1 |
| 543 | 1e-118 | 2 |
| 536 | 9.3e-118 | 2 |
| 518 | 6.2e-117 | 2 |
| 907 | 6. $6 e-116$ | 1 |
| 511 | 6. $6 e-114$ | 2 |
| 883 | $1.4 e-112$ | 1 |
| 882 | $1.9 \in-112$ | 1 |
| 504 | 1. $4 \in-111$ | 2 |
| 499 | 9.6e-111 | 2 |
| 865 | 4.1e-110 | 1 |
| 853 | $7.8 e-110$ | 1 |
| 859 | 2.8 - 109 | 1 |
| 483 | $2.6 e-107$ | 2 |
| 470 | 3e-105 | 2 |
| 452 | 4. $4 e-102$ | 2 |
| 798 | $7.3 e-101$ | 1 |
| 798 | $7.3 e-101$ | 1 |
| 758 | 2. $4 \mathrm{e}-95$ | 1 |
| 737 | $1.9 \mathrm{e}-92$ | 1 |
| 466 | 1. $7 e-89$ | 3 |
| 652 | 1e-80 | 2 |
| 651 | 1. 4e-80 | 1 |
| 268 | 1. $3 \mathrm{e}-50$ | 4 |
| 261 | $6.1 e-44$ | 3 |
| 279 | $4.8 \mathrm{e}-42$ | 3 |
| 279 | 4.8e-42 | 3 |
| 175 | 1e-41 | 4 |
| 365 | $4.2 e-41$ | 1 |
| 250 | 5.9e-39 | 3 |
| 220 | 8.6e-39 | 3 |
| 181 | 1. 7e-3a | 4 |
| 156 | 1.1e-37 | 4 |
| 156 | $1.1 e-37$ | 4 |
| 153 | 2e-37 | 4 |
| 153 | $8.1 e-36$ | 4 |
| . 209 | $7.1 e-35$ | 4 |
| 160 | 7.1e-35 | 4 |
| 158 | 3.3e-34 | 4 |
| 181 | 4.5e-34 | 4 |
| 163 | 8. $3 \mathrm{e}-34$ | 4 |
| 131 | $1.1 e-33$ | 5 |
| 134 | 1.2e-32 | 5 |
| 156 | 1. 8 e-32 | 4 |
| 267 | 1. $4 e-27$ | 1 |
| 156 | 5. $4 e-27$ | 4 |
| 220 | 1. $8 e-26$ | 3 |
| 243 | 2. $6 \mathrm{e}-25$ | 2 |

FIG. 19 (C)

|  |  |  |
| :---: | :---: | :---: |
| 1. | 1346094 | GLUCOSE-1-PHOSPHATE URIDYLYLTRANSFE |
| 12. gi | 1095634 | PxO1-94 |
| 3. | 10176276 | UTP-glucose-1-phosphate uridyl |
| 104. |  |  |
| 105. | 7521719 | phosphate nucleotydyl |
| 106. | 6138 |  |
| 107. | 2501471 | TIVE UTP--GLUCOSE-1- |
| 108. | 105 |  |
| 109. | 6960274 | $96 \%$ identity with amino acids 1-24 |
| 0. | 17434850 | UTP--glucose-1-phosphate uridy |
| 111. | 250146 | TP--GLUCOSE-1-PHOSPHATE |
| 2. | 7434856 | UTP--glucose-1-phosphate uridy |
| 113. | 2501467 | UTP--GLUCOSE-1-PHOSPHATE |
| 4. | 1080270 | Glucose-1-P thymidyly |
| 5. | 10802777 | Glucose-1-phosphate thymidyly |
| 116. 9 | 7739964 | 㖪 |
| 7. | \|556004 | -1-phosphate uridyly |
| 118. | 7434852 | lucose 1 |
| 9. |  | riay |
| 120. | 3550619 | ucose-1-phosphate uridy |
| 121. | 1177038 | E |
| 22. | 10174923 | ucose-1-phosphate uridyly |
| 123. 9 | 3777501 | Putative GDP-mannose pyrophosphorylas |
| 124. | 3970895 | GDP-mannose pyrophosphorylase [Ca |
| 125. | 3777503 | ive GDP-mannose pyrophosphorylase |
| 126. | 7296813 | CG1129 gene product [alt 1] [Droso |
| 127. gi | 4240429 | NDP-hexose synthetase homolog (Streptorny... |
| 128. | 7448166 | Probable glucose-1 phosphate thymidylylt... |
| 129. | 2127932 | Glucose-1-phosphate thymidy |
| 130. | 323397 | osphate guanylyltransferase ... |
| 131. | 585225 | OSE-1-PHOSPHATE URIDYLYLTFANSFE. |
| 132. gi | 51 | sphate uridy |
| 133. | 10176341 | UTP-glucose-1-phosphate uridylyltransfer.. |
| 134. | 50 | UTP--GLUCOSE-1-PHOSPHATE URIDYL |
| 135. | 7497318 | protein |
| 136. | 10174033 | Mannose-1-phosphate guanyltransferase [B... |
| 137. | 58 | mannose pyrophosphorylase [Pichia an.. |
| 138. |  | Mannose-1-phosphate guanyltransferase, G... |
| 139. | 18 | Glucose-1-phosphate thymidylyltransferas... |
| 140.9 | 7448165 | Mannose-1-phosphate guanyltransferase PA.. |
| 141. 9 | 21 | Glucose-1-phosphate thymidy |
| 142. gi | 10 | Mannose-1-phosphate guanyl |
| 143. g |  | ate |
| 144. gi | 90 | DIDP-glucose synthase [Streptoryces |
| 145. | 10579656 | phosphate thymidylyltransferas... |
| 146. | 90 | GDP-mannose pyrophosphorylase \{Candida g... |
| 147. g |  | e nucleotide phosphorylase [Strep... |
| 148. g | 7448158 | Glucose-1-phosphate thymidylyltransferas. |
| 149. | 7448170 | Probable rmla 2 protein - Mycobacterium $t$. |
| 150. gi\| | 10880965 | Putative UTP-glucose-1-phosphate uridyly |
| 151. | 6015731 | ucose-1-phosphate thymidylyltransferas... |
| 152. gi\|7 | 7434855 | glucose-1-phosphate uridylyltransfe. |
| 153. | 11352828 | osphate uridylyltransfe |
| 154. gi\|4 | 4884956 | . |
| 155. gi\|4 | 4378170 | -glucose-1-phosphate uridyitransferas... |
| 156. gi | 7451544 |  |

UTP--GL

| 117 | $9.6 e-23$ | 5 |
| ---: | :--- | :--- |
| 124 | $3.3 e-19$ | 3 |
| 138 | $3.4 e-19$ | 3 |
| 126 | $6.5 e-19$ | 3 |
| 139 | $1 e-18$ | 3 |
| 140 | $2.5 e-18$ | 3 |
| 121 | $4.9 e-18$ | 3 |
| 130 | $9.2 e-18$ | 3 |
| 119 | $1.9 e-17$ | 4 |
| 130 | $3.1 e-17$ | 2 |
| 130 | $3.5 e-17$ | 3 |
| 132 | $3.8 e-17$ | 3 |
| 139 | $4.6 e-17$ | 3 |
| 131 | $4.6 e-17$ | 3 |
| 154 | $6.4 e-17$ | 4 |
| 150 | $8.7 e-17$ | 4 |
| 130 | $1.7 e-16$ | 3 |
| 132 | $1.7 e-16$ | 3 |
| 109 | $1.8 e-16$ | 3 |
| 133 | $2.4 e-16$ | 3 |
| 133 | $3.1 e-16$ | 3 |
| 123 | $3.3 e-16$ | 3 |
| 111 | $3.4 e-16$ | 3 |
| 150 | $6.5 e-16$ | 3 |
| 150 | $6.5 e-16$ | 3 |
| 150 | $6.5 e-16$ | 3 |
| 115 | $8.3 e-16$ | 2 |
| 112 | $8.3 e-16$ | 2 |
| 116 | $1 e-15$ | 5 |
| 143 | $1.3 e-15$ | 4 |
| 152 | $1.7 e-15$ | 4 |
| 137 | $1.8 e-15$ | 3 |
| 109 | $2.1 e-15$ | 3 |
| 133 | $3.4 e-15$ | 3 |
| 130 | $3.7 e-15$ | 3 |
| 171 | $5.4 e-15$ | 2 |
| 120 | $1.1 e-14$ | 4 |
| 135 | $1.9 e-14$ | 3 |
| 150 | $4.8 e-14$ | 3 |
| 106 | $6 e-14$. | 3 |
| 113 | $6.3 e-14$ | 4 |
| 168 | $6.6 e-14$ | 1 |
| 132 | $7 e-14$ | 2 |
| 148 | $8.8 e-14$ | 3 |
| 167 | $9 e-14$ | 1 |
| 148 | $1.1 e-13$ | 2 |
| 144 | $3 e-13$ | 3 |
| 86 | $5.5 e-13$ | 3 |
| 106 | $6.5 e-13$ | 5 |
| 102 | $8.4 e-13$ | 2 |
| 113 | $8.4 e-13$ | 3 |
| 115 | $9.4 e-13$ | 3 |
| 107 | $1.2 e-12$ | 3 |
| 106 | $1.4 e-12$ | 3 |
| 157 | $2.2 e-12$ | 1 |
| 116 | $2.8 e-12$ | 3 |
| 120 | $3.3 e-12$ | 3 |
|  |  |  |

FIG. 19 (D)

| $\begin{aligned} & \text { 157. gi } \\ & \text { 158. } \mathrm{gi} \end{aligned}$ | $\left\lvert\, \begin{aligned} & 2501468 \\ & 7492163 \end{aligned}\right.$ |
| :---: | :---: |
| 159. gi | 11261675 |
| 160. gi | 7448173 |
| 161. gi | 7434849 |
| 162. gi | 11261677 |
| 163. gi | 11261687 |
| 164. gi | 7381245 |
| 165. gi | 3372537 |
| 166. gi | 6015664 |
| 167. gi | 10579698 |
| 168. gi | 7448161 |
| 169. gi | 11261691 |
| 170. gi | 7448163 |
| 171. gi | 3559951 |
| 172. gi | 1169833 |
| 173. gi | 2117938 |
| 174. gi | 120929 |
| 175. gi | 541005 |
| 176. gi | 7649599 |
| 177. gi | 6066425 |
| 178. gi | 462035 |
| 179. gi | 7434854 |
| 180. gi | 10579655 |
| 181. gi | 4103324 |
| 182. gi | 4234784 |
| 183. gi | 5814301 |
| 184. gi | 7448169 |
| 185. gi | 3183009 |
| 186. gi | 7448154 |
| 187. gi | 11261685 |
| 188. gi | 3319929 |
| 189. gi | 116099 |
| 190. gi | 7521439 |
| 191. gi | 7269958 |
| 192. gi | 7448168 |
| 193. gi | 7447202 |
| 194. gi | 1360733 |
| 195. gi | 120926 |
| 196. gi | 96782 |
| 197. gi | 2501466 |
| 198. gi | 10803043 |
| 199. gi | 7448172 |
| 200. gi | 7434875 |
| 201. gi | 585168 |
| 202. gi | 2133477 |
| 203. gi | 3041673 |
| 204. gi\| | 7206597 |
| 205. gi | 11354079 |
| 206. gi | 10436247 |
| 207. gi | 2494312 |
| 208. gi | 9966779 |
| 209. gi | 7434857 |
| 210. gi | 11261689 |
| 211. gi | 1078827 |
| 212. gi | 11353788 |
| 213. gi\| | 529245 |

13. gi|629245

UTP--GLUCOSE-1-PHOSPHATE URTDYLYLTRANSFE.. Mannose-1-phosphate guanyltransferase - ... UTP--glucose-1-phosphate uridylyltransfe.. Probable glucose-1-phosphate thymidylylt... UrP--glucose-1-phosphate uridylyltransfe... UTP-glucose-1-phosphate uridylyltransfer... Probable UTP-glucose-1-phosphate uridyl.. UDPG-pyrophosphorylase [Acetobacter xyli... UTP-glucose-1-phosphate uridylyltransfer... UDP-glucose pyrophosphorylase [Sulfolobu.. Glucose-1-phosphate thymidylyltransferas... Probable mannose-1-phosphate guanylyltra.. UTP--glucose-1-phosphate uridylyltransfe... Glucose-1-phosphate thymidylyltransferas... UDP-glucose pyrophosphorylase [Pseudomon.. UTP--GLUCOSE-1-FHOSPHATE URIDYLYLTRANSFE.. UTP--glucose-1-phosphate uridylyltransfe.. UTP--GLUCOSE-1-PHOSPHATE URIDYLYLTTRANSFE.. Eẋon protein - Rhizobium meliloti Putative marnose-1-phosphate guanyltrans... Mannose-1-phosphate guanyltransferase [L... UTP--GLUCOSE-1-PHOSPHATE URIDYLYLTRANSFE.. Probable UTP--glucose-1-phosphate uridyl... Glucose-1-phosphate thymidylyltransferas... GDP-mannose pyrophosphorylase [Solanum t.... Unknown [Leptospira borgpetersenii] Unknown [Leptospira interrogans] Probable mannose-1-phosphate guanyltrans... UTP--GLUCOSE-1-PHOSPHATE URIDYLYLTRANSFE... Mannose-1-phosphate guanylyltransferase ... UTP--glucose-1-phosphate uridylyltransfe... GalU protein (Pectobacterium carotovorum.. UTP--GLUCOSE-1-PHOSPHATE URIDYLYLTRANSFE.. Probable sugar phosphate transferase A.PE. GDP-mannose pyrophosphorylase like prote... Mannose-1-phosphate guanyltransferase - ... Probable glucose-1-phosphate thymidylylt... UTP--glucose-1-phosphate uridylyltransfe... UTP--GLUCOSE-1-PHOSPHATE URIDYLYLTPANSFE.. UTP--glucose-1-phosphate uridylyltransfe... UTP--GLUCOSE-1-PHOSPHATE URIDYLYLTRANSFE.. UDP-glucose pyrophosphorylase [Haemophil... Probable glucose-1 phosphate transferase.. Glucose-1-phosphate adenylyltransferase ... UTP--GLUCOSE-1-pHOSPHATE URIDYLYLTRANSFE.. Pyrophosphorylase ppp-1 homolog - Caenor... pUTATIVE TRANSLATION INITLATION FACTOR E.. C. elegans (PPp-1) putative translation ... Probable sugar-phosphate nucleotidyl tra... Unnamed protein product [Homo sapiens] TRANSLATION INITIATION FACTOR EIF-2B GAM.. Eukaryotic translation initiation factor... UTP--glucose-1-phosphate uridylyltransfe.. UTP--glucose-1-phosphate uridylyltransfe.. Fyrophosphorylase 1 - Caenorhabditis bri... Mannose-1-phosphate guanyltransferase-re... Lambo protein - Streptomyces lincolnensis

| 107 | $6.2 \mathrm{e}-12$ | 3 |
| ---: | :--- | ---: |
| 136 | $1.6 \mathrm{e}-11$ | 3 |
| 101 | $1.7 \mathrm{e}-11$ | 3 |
| 83 | $1.8 \mathrm{e}-11$ | 4 |
| 126 | $2 \mathrm{e}-11$ | 2 |
| 106 | $4.4 \mathrm{e}-11$ | 3 |
| 107 | $5.7 \mathrm{e}-11$ | 3 |
| 103 | $5.7 \mathrm{e}-11$ | 3 |
| 115 | $6 \mathrm{e}-11$ | 3 |
| 122 | $9.4 \mathrm{e}-11$ | 4 |
| 126 | $9.9 \mathrm{e}-11$ | 3 |
| 136 | $1.3 \mathrm{e}-10$ | 3 |
| 107 | $1.4 \mathrm{e}-10$ | 3 |
| 79 | $1.5 \mathrm{e}-10$ | 4 |
| 106 | $2.4 \mathrm{e}-10$ | 3 |
| 117 | $2.7 \mathrm{e}-10$ | 3 |
| 108 | $3.4 \mathrm{e}-10$ | 3 |
| 129 | $4.7 \mathrm{e}-10$ | 2 |
| 112 | $7.1 \mathrm{e}-10$ | 3 |
| 105 | $7.2 \mathrm{e}-10$ | 3 |
| 130 | $8 \mathrm{e}-10$ | 3 |
| 112 | $8.1 \mathrm{e}-10$ | 3 |
| 138 | $9.1 \mathrm{e}-10$ | 1 |
| 90 | $2 \mathrm{e}-09$ | 3 |
| 115 | $2 \mathrm{e}-09$ | 3 |
| 93 | $2 \mathrm{e}-09$ | 3 |
| 95 | $2 \mathrm{e}-09$ | 3 |
| 115 | $3.7 \mathrm{e}-09$ | 2 |
| 125 | $5.7 \mathrm{e}-09$ | 3 |
| 104 | $6.5 \mathrm{e}-09$ | 3 |
| 93 | $7.3 \mathrm{e}-09$ | 3 |
| 131 | $8.4 \mathrm{e}-09$ | 1 |
| 103 | $9.4 \mathrm{e}-09$ | 3 |
| 129 | $1.6 \mathrm{e}-08$ | 1 |
| 107 | $3.1 \mathrm{e}-08$ | 2 |
| 89 | $4.7 \mathrm{e}-08$ | 2 |
| 84 | $7.6 \mathrm{e}-08$ | 2 |
| 122 | $8.7 \mathrm{e}-08$ | 2 |
| 122 | $8.8 \mathrm{e}-08$ | 2 |
| 122 | $8.8 \mathrm{e}-08$ | 2 |
| 122 | $8.8 \mathrm{e}-08$ | 2 |
| 113 | $1.6 \mathrm{e}-07$ | 2 |
| 80 | $1.9 \mathrm{e}-07$ | 3 |
| 93 | $2.1 \mathrm{e}-07$ | 2 |
| 119 | $2.2 \mathrm{e}-07$ | 2 |
| 120 | $2.8 \mathrm{e}-07$ | 1 |
| 120 | $2.8 \mathrm{e}-07$ | 1 |
| 120 | $2.8 \mathrm{e}-07$ | 1 |
| 95 | $3.6 \mathrm{e}-07$ | 2 |
| 118 | $5.2 \mathrm{e}-07$ | 1 |
| 118 | $5.2 \mathrm{e}-07$ | 1 |
| 118 | $5.2 \mathrm{e}-07$ | 1 |
| 88 | $7.9 \mathrm{e}-07$ | 2 |
| 93 | $9.4 \mathrm{e}-07$ | 2 |
| 116 | $9.9 \mathrm{e}-07$ | 1 |
| 91 | $1.2 \mathrm{e}-06$ | 2 |
| 84 | $1.2 \mathrm{e}-06$ | 2 |
|  |  |  |



| 271. gi | 40692 |
| :---: | :---: |
| 272. gi | 11261806 |
| 273. gi | 7671532 |
| 274. gi | 633874 |
| 275. gi | 485384 |
| 276. gi | 421276 |
| 277. gi | 10638192 |
| 278. gi | 10638156 |
| 279. gi | 10638153 |
| 280. gi | 10638168 |
| 281. gi | 2146023 |
| 282. gi | 2558972 |
| 283. gi | 1237080 |
| 284. gi | 7447201 |
| 285. gi | 7521163 |
| 286. gi | 121289 |
| 287. gi | 100675 |
| 288. gi | \|5917789 |
| 289. gi | 6593019 |
| 290. gi\| | 1575754 |
| 291. gi | 11347147 |
| 292. gil | 21403 |
| 293. gi\| | 100426 |
| 294. gi\| | 1633678 |
| 295. gi | 10638150 |
| 296. gi | 2130035 |
| 297. gi | 232172 |
| 298. gi | 7340287 |
| 299. gi\| | 1707939 |
| 300. gil | 1237082 |
| 301. gi | 1707943 |
| $302 . \mathrm{gi}$ | 1707940 |
| 303. gi | 3015514 |
| 304. gi\| | 1071859 |
| 305. gi | 1325984 |
| 306. gi | 1232164 |
| 307. gi | 7434881 |
| 308. gi\| | 1707930 |
| 309. gi | 17434879 |
| 310. gi | 12625084 |
| 311. gi | 7434871 |
| 312. gi | \|7434891 |
| 313. gi | \|1707928 |
| 314. gi | \|2149021 |
| 315. gi\| | 4586350 |
| 316. gi | 1707923 |
| 317. gi | 100580 |
| 318. gi | 5091608 |
| 319. gi | 7434885 |
| 320. gi | 7688095 |
| 321. gi\| | 7743886 |
| 322. gi | \|7488396 |
| 323. gi | \|6320417 |
| 324. gil | 7521184 |
| 325. gi | 1197640 |
| 326. gi | 154448 |
| 327. gil | 7447199 |

Homology to UDP PYrophosphorylase M76548... Probable glucose-1-phosphate adenylyltra.. Glucose-1-phosphate adenylyltransferase ... Alpha-D-glucose cytidylyltransferase; Ep.. Alpha-D-glucose-1-phosphate cytidylyltra.. Glucose-1-phosphate cytidylyltransferase... UTP-glucose-1-phosphate uridylyltransfer... UTP-glucose-1-phosphate uridylyltransfer... UTP-glucose-1-phosphate uridylyltransfer... UTP-glucose-1-phosphate uridylyltransfer... Lnbo protein - Streptomyces lincolnensis... DdhA [Vibrio anguillarum]
ADP-glucose pyrophosphorylase [Pisum sat... Glucose-1-phosphate cytidylyltransferase... Probable licC protein (licC) - syphilis ... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase small subu.. T22C5.13 [Arabidopsis thaliana]
ADP glucose pyrophosphorylase small subu.. Probable sugar nucleotidyltransferase Cj... ADP-glucose pyrophosphorylase; glucose-1... Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase [Spinacia ... UTP-glucose-1-phosphate uridylyltransfer... Glucose-1-phosphate adenylyltransferase. GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... Small subunit ADP glucose pyrophosphoryl... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... ADP-glucose pyrophosphorylase [Pisum sat... GLUCOSE-1-PHOSPHATE ADENYLYLITRANSEERASE --glucose-1-phosphate adenklyltransferase ... ADPG pyrophosphorylase small subunit [Ar... Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase small subu... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... Glucose-1-phosphate adenylyltransferase ... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSEERASE ... Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase small subu... Glucose-1-phosphate adenylyltransferase ... Glucose-1-phosphate adenylyltransferase ... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... ADPG pyrophosphorylase large subunit [Ar... Glucose-1-phosphate adenylyltransferase ..-GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... Glucose-1-phosphate adenylyltransferase ... Identical to gb|D50317 ADP glucose pyrop... Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase small subu... Glucose-1-phosphate adenylyltransferase ... Translation regulator GCD6 homolog T9A21... Translation initiation factor eIF-2B eps... Probable mannose-1-phosphate guanyltrans... Doth [Yersinia enterocolitica (type 0:8)] ADP-glucose pyrophosphorylase [Synechocy... Glucose-1-phosphate cytidylyltransferase..

| 87 | 0.0099 | 1 |
| :--- | :--- | :--- |
| 73 | 0.011 | 2 |
| 79 | 0.011 | 2 |
| 86 | 0.014 | 1 |
| 86 | 0.014 | 1 |
| 86 | 0.014 | 1 |
| 70 | 0.015 | 2 |
| 64 | 0.015 | 2 |
| 64 | 0.015 | 2 |
| 64 | 0.015 | 2 |
| 73 | 0.016 | 2 |
| 83 | 0.017 | 2 |
| 80 | 0.018 | 2 |
| 85 | 0.019 | 1 |
| 85 | 0.019 | 1 |
| 80 | 0.023 | 2 |
| 80 | 0.023 | 2 |
| 80 | 0.024 | 2 |
| 73 | 0.024 | 2 |
| 80 | 0.024 | 2 |
| 84 | 0.025 | 1 |
| 80 | 0.028 | 2 |
| 80 | 0.028 | 2 |
| 80 | 0.028 | 2 |
| 70 | 0.028 | 2 |
| 80 | 0.03 | 2 |
| 80 | 0.03 | 2 |
| 80 | 0.03 | 2 |
| 80 | 0.031 | 2 |
| 80 | 0.032 | 2 |
| 80 | 0.032 | 2 |
| 80 | 0.032 | 2 |
| 80 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 62 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 80 | 0.033 | 2 |
| 62 | 0.044 | 2 |
| 62 | 0.044 | 2 |
| 75 | 0.044 | 2 |
| 75 | 0.044 | 2 |
| 62 | 0.044 | 2 |
| 62 | 0.045 | 2 |
| 62 | 0.045 | 2.088 |
| 73 | 0.057 | 2.059 |
| 80 | 0.05 |  |
| 64 | 0.064 | 2.065 |
| 81 | 0.068 | 2 |
| 81 | 0.065 | 0.065 |
| 80 | 0.088 | 2 |
| 80 | 0.088 |  |
| 80 |  |  |


| 328. gi | \|1707944 |
| :---: | :---: |
| 329. gi | \|3023677 |
| 330. gi\| | \|121293 |
| 331. gi | \|7434883 |
| 332. gi\| | $\mid 97169$ |
| 333. gi | 77404390 |
| 334. gi | \|9757341 |
| 335. gi | 7434873 |
| 336. gi | 1707922 |
| 337. gi\| | \|3023676 |
| 338. gi | 1707932 |
| 339. gi | 1707929 |
| 340. gi\| | 7671230 |
| 341. gi | 7448160 |
| 342. gi | 11023507 |
| 343. gi | 1840114 |
| 344. gi | 5739461 |
| 345. gi\| | 2506458 |
| 346. gi | 7434870 |
| 347. gi | 1778436 |
| 348. gi | 7448167 |
| 349. gi | 7434893 |
| 350. gi | 5882732 |
| 351. gi\|6 | 6646773 |
| 352. gi | 7434888 |
| 353. gi | 7434889 |
| 354. gi\|5 | 5917791 |
| 355. gi\|7 | 7471938 |
| 356. gi\|7 | 7471937 |
| 357. gi\|1 | 11386853 |
| 358. gi\|7 | 7434869 |
| 359. gi\|7 | 7492700 |
| 360. gi\|21 | 2130037 |
| 361. gi\|21 | 2146810 |
| 362. gi\|7 | 7448100 |
| 363. gi\|1 | 135927 |
| 364. gi\|7 | 7522214 |
| 365. gi\|4 | 4544432 |
| 366. gi\|1 | 10639507 |
| 367. gi\|7 | 7447200 |
| 368. gi\|2 | 232166 |
| 369. gi\|2 | 2981290 |
| 370. gi\|5 | 5923897 |
| 371. gi\|7 | 7434884 |
| 372. gi\|6 | 6626264 |
| 373. gi\|l | 11359709 |
| 374. gi\|7 | 7434874 |
| 375. gi\|7 | 7543739 |
| 376. gi\|5 | 5701881 |
| 377. gi\|5 | 5852076 |
| 378. gi\|7 | 7331959 |
| 379. gi\|4 | 479426 |
| 380. gi\|4 | 476970 |
| 381. gi\|3 | 3211989 |
| 382. gi\|7 | 7671234 |

GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE .PROBABLE TRANSLATION INITLATION FACTOR E.. GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE .-Glucose-1-phosphate adenylyltransferase ... Lic-1 protein C - Haemophilus influenzae... LICC PROTEIN
Probable mannose-1-phosphate guanyltrans... Glucose-1-phosphate adenylyltransferase ... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... PROBABLE TRANSLATION INITIATION FACTOR E.. GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE .. ADP-glucose pyrophosphorylase catalytic ... Glucose-1-phosphate adenylyltransferase ... Putative glucose-1-p-cytidylyltransferas... ADP-glucose pyrophosphorylase large subu.. GalF [Escherichia coli]
GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase large subu.. Probable glucose-1-phosphate thymidylylt... Glucose-1-phosphate adenylyltransferase ... Similar to gb|AF135422 GDP-mannose pyrop... Putative GDP-mannose pyrophosphorylase: Glucose-1-phosphate adenylyltransferase .. Glucose-1-phosphate adenylyltransferase ... ADP-glucose pyrophosphorylase large subu.. Glucose-1-phosphate adenylyltransferase ... Glucose-1-phosphate adenylyltransferase ... PROBABLE GLUCOSE-1-PHOSPHATE ADENYLYLTRA. Glucose-1-phosphate adenylyltransferase ... Probable mannose-1-phosphate gaunyl tran.. Glucose-1-phosphate adenylyltransferase ... Glucose-1-phosphate adenylyltransferase ... UDP-N-acetylglucosamine pyrophosphorylas... UDP-N-ACETYLLLLUCOSAMINE PYROPHOSPHORYLAS. Glucose-l-phosphate adenylyltransferase ... Putative GDP-mannose pyrophosphorylase ... Mannose-1-phosphate guanyltransferase re.. Glucose-1-phosphate cytidylyltransferase.. GLUCOSE-1-PHOSPHATE ADENYLYLTRANSFERASE ... Ribosomal protein 54 homolog [Trypanosom.. ADP-glucose pyrophosphorylase large subu.. Glucose-1-phosphate adenylyltransferase ... GalF-like [Bradyrhizobium japonicum] Type 2C Protein Phosphatase related prot... Glucose-1-phosphate adenylyltransferase ... Hypothetical protein; 66083-64412 [Arabi... ADP-glucose pyrophosphorylase [Ipomoea b... ADP-glucose pyrophosphorylase [Ipomoea b... Contains similarity to Pfam families pFo... Fibronecin-binding protein - Streptococc... Mannose-1-phosphate guanylyltransferase ... ADP-glucose pyrophosphorylase large subu.. ADP-glucose pyrophosphorylase large subu..

| 80 | 0.088 | 1 |
| :--- | :--- | :--- |
| 77 | 0.09 | 2 |
| 62 | 0.1 | 2 |
| 73 | 0.1 | 2 |
| 79 | 0.12 | 1 |
| 79 | 0.12 | 1 |
| 66 | 0.13 | 2 |
| 72 | 0.14 | 2 |
| 72 | 0.14 | 2 |
| 78 | 0.16 | 1 |
| 71 | 0.17 | 2 |
| 69 | 0.18 | 2 |
| 69 | 0.18 | 2 |
| 62 | 0.21 | 2 |
| 77 | 0.21 | 1 |
| 69 | 0.24 | 2 |
| 64 | 0.27 | 3 |
| 76 | 0.28 | 1 |
| 72 | 0.29 | 2 |
| 69 | 0.31 | 2 |
| 75 | 0.36 | 1 |
| 60 | 0.39 | 2 |
| 74 | 0.46 | 1 |
| 74 | 0.46 | 1 |
| 57 | 0.49 | 2 |
| 57 | 0.49 | 2 |
| 69 | 0.5 | 2 |
| 71 | 0.53 | 2 |
| 63, | 0.59 | 3 |
| 66 | 0.6 | 2 |
| 62 | 0.6 | 2 |
| 57 | 0.62 | 2 |
| 72 | 0.69 | 1 |
| 72 | 0.69 | 1 |
| 48 | 0.75 | 4 |
| 43 | 0.79 | 5 |
| 66 | 0.84 | 2 |
| 70 | 0.89 | 1 |
| 70 | 0.89 | 1 |
| 70 | 0.89 | 1 |
| 63 | 0.93 | 2 |
| 54 | 0.94 | 2 |
| 63 | 0.95 | 2 |
| 63 | 0.95 | 2 |
| 69 | 0.95 | 1 |
| 69 | 0.95 | 1 |
| 59 | 0.98 | 2 |
| 68 | 0.98 | 1 |
| 63 | 0.99 | 2 |
| 62 | 1 | 2 |
| 54 | 1 | 1 |
| 66 | 1 | 1 |
| 66 | 1 | 1 |
| 61 | 1 | 1 |
| 58 | 1 | 2 |
|  |  | 2 |

## FIG. 20(a)





Determined: 579.10


$\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{15} \mathrm{P}_{2}{ }^{-}$
Exact Mass: 588.10
Determined: 588.04

Determined: 572.14


## FIG. 20(b)



## FIG. 21



FIG. 22

4

5

6

17


18

19

21

22
23

12


FIG. 23


## ACTIVE-SITE ENGINEERING OF NUCLEOTIDYLYLTRANSFERASES AND GENERAL ENZYMATIC METHODS FOR THE SYNTHESIS OF NATURAL AND "UNNATURAL" UDP- AND TDP-NUCLEOTIDE SUGARS

This application claims the benefit of U.S. application Ser. No. 60/254,927, filed 13 Dec. 2000.

## FIELD OF THE INVENTION

The present invention is directed to nucleotidylyl-transferases and mutant nucleotidylyltransferases having altered substrate specificity and methods for their production.

The present invention is also directed to methods of synthesizing desired nucleotide sugars using natural and/or mutant $\mathrm{E}_{p}$ or other nucleotidyltransferases, preferably $\mathrm{E}_{p}$ or other nucleotidylyltransferases modified by the present methods. Additionally, the present invention is directed to nucleotide sugars sythesized by the present methods.

The present invention is further directed to new glycosyl phosphates, and methods for making them.

## BACKGROUND

Many bioactive metabolites possess unusual carbohydrates required for molecular recognition. (See for example, Liu, H.-w.; Thorson, J. S. Ann. Rev. Microbiol., 1994, 48, 223-256; Weymouth-Wilson, A. C. Nat. Prod. Rep. 1997, 14, 99-110; In Macrolide Antibiotics, Chemistry, Biology and Practice; Omura, S. Ed., Academic Press: New York; 1984; Johnson, D. A.; Liu, H.-w. Curr. Opin. Chem. Biol. 1998, 2, 642-649; and Trefzer, A.; Salas, J. A.; Bechthold, A. Nat. Prod. Rep. 1999, 16, 283-299.) In fact, roughly 70\% of current lead compounds in modern drug discovery derive directly from natural products, many of which are glycosylated metabolites. (See Thorson, J. S. et al. Nature's Carbohydrate Chemists: The Enzymatic Glycosylation of Bioactive Bacterial Metabolites. Curr. Org. Chem. manuscript in press, (2000); and references therein and Weymouth-Wilson, A. C. The Role of Carbohydrates in Biologically Active Natural Products. Nat. Prod. Rep. 14, 99-110 (1997)). Examples of pharmaceutically important glycosylated metabolites include, for example, amphotericin, megalomicin/erythromycin, mithramycin, doxorubicin, vancomycin and calicheamicin, as shown in FIG. 5. While it is known that the sugar moieties of these pharmaceutically important metabolites often define their corresponding biological activity, (see Weymouth-Wilson, A. C., The Role of Carbohydrates in Biologically Active Natural Products, Nat. Prod. Rep. 14, 99-110 (1997)), efficient methods to systematically alter these essential carbohydrate ligands are still lacking.

In metabolite biosynthesis, glycosylation begins with the nucleotidylyltransferase-catalyzed activation of a sugar phosphate as a nucleotide diphosphosugar (NDP-sugar) donor. After activation, a number of enzymatic processing reactions often occur (e.g., deoxygenation, transamination, oxidation/reduction, epimerization, alkylation, and decarboxylation) prior to the culminating glycosyltransferasecatalyzed attachment to the aglycon. (Liu, H.-w. \& Thorson, J. S. Pathways and Mechanisms in the Biogenesis of Novel Deoxysugars by Bacteria. Ann. Rev. Microbiol. 48, 223-256 (1994); Kirschning, A., Bechtold, A. F-W. \& Rohr, J. Chemical and Biochemical Aspects of Deoxysugars and Deoxysugar Oligosaccharides. Top. Curr. Chem. 188, 1-84 (1997); Johnson, D. A. \& Liu, H.-w. Mechanisms and

Pathways from Recent Deoxysugar Biosynthesis Research. Curr. Opin. Chem. Biol. 2, 642-649 (1998); Hallis, T. M. \& Liu, H.-w. Learning Nature's Strategies for Making Deoxy Sugars: Pathways, Mechanisms, and Combinatorial Appli5 cations. Acc. Chem. Res. 32, 579-588 (1999); Johnson, D. A. \& Liu, H.-w. In Comprehensive Chemistry of Natural Product Chemistry (Barton, D.; Nakanishi; K.; Meth-Cohn, O. eds), Elsevier Science, Oxford, 311, (1999); Trefzer, A., Salas, J. \& Bechthold, A. Genes and Enzymes Involved in Deoxysugar Biosynthesis in Bacteria. Nat. Prod. Rep. 16, 283-299 (1999); and Bechthold, A. \& Rohr, J. In New Aspects of Bioorganic Chemistry (Diederichsen, U.; Lindhorst, T. K.; Wessjohann, L.; Westerman, B., eds.) WileyVCH, Weinheim, 313, (1999)).

The glycosyltransferases that incorporate these essential ligands are thought to rely almost exclusively upon UDPand TDP-nucleotide sugars; however some have demonstrated promiscuity towards the sugar donor, (e.g., Gal, D-galactose; Glc, D-glucose; Man, D-mannose; NTP, nucleotide triphosphate; pFPTC , pentafluorophenoxythiocarbonyl; TDP, thymidine diphosphate; TMP, thymidine monophosphate; TTP, thymidine triphosphate; UDP, uridine diphosphate.) Genetic experiments suggest that downstream glycosyltransferases in secondary metabolism are promiscuous with respect to their NDP-sugar donor, setting the stage for the expansion of "combinatorial biosynthesis" approaches to change metabolite glycosylation. (See Madduri, K. et al., Production of the antitumor drug epirubicin ( $4^{\prime}$-epidoxorubicin) and its precursor by a genetically engineered strain of Streptomyces peucetius Nat. Biotech. 16, 69-74 (1998); and Hutchinson, C. R. Combinatorial Biosynthesis for New Drug Discovery. Curr. Opin. Microbiol. 1, 319-329 (1998).) This information has led to the exploitation of the carbohydrate biosynthetic machinery to manipulate metabolite glycosylation, (Madduri, K.; Kennedy, J.; Rivola, G.; Inventi-Solari, A.; Filppini, S.; Sanuso, G.; Colombo, A. L.; Gewain, K. M.; Occi, J. L.; MacNeil, D. J.; Hutchinson, C. R. Nature Biotech. 1998, 16, 69-74; and Zhao, L.; Ahlert, J.; Xue, Y.; Thorson, J. S.; Sherman, D. H.; Liu, H.-w. J. Am. Chem. Soc., 1999, 121, 9881-9882 and references therein), revitalizing interest in methods to expand the repertoire of available UDP- and TDP-sugar nucleotides. (See Zhao, Y.; Thorson, J. S. J. Org. Chem. 1998, 63, 7568-7572; and Elhalabi, J. M.; Rice, K. G. Cur. Med. Chem. 1999, 6, 93-116.)
These in vivo methods are limited by both a particular host's biosynthetic machinery and the specific host's tolerance to each newly constructed metabolite. Further, in vitro progress in this area is limited by the availability of the required NDP-sugar substrates. (Solenberg, P. J. et al., Production of Hybrid Glycopeptide Antibiotics in vitro and in Streptomyces toyocaensis. Chem. \& Biol. 4, 195-202 (1997).)

Thus, there is a need for a greater variety of available NDP-sugar substrates.

Salmonella enterica LT2 $\alpha$-D-glucopyranosyl phosphate thymidylyltransferase ( $\mathrm{E}_{p}$ ) is a member of the prevalent nucleotidylyltransferase family responsible for the reversible conversion of $\alpha$-D-hexopyranosyl phosphate and NTP to the corresponding NDP-sugar nucleotide and pyrophosphate. Of the many nucleotidylyl-transferases studied, the NDP-sugar nucleotide-forming thymidylyltransferases have received the least attention in prior work. (See Lindquist, L.; Kaiser, R.; Reeves, P. R.; Lindberg, A. A. Eur. J. Biochem. 1993, 211, 763-770, and Gallo, M. A.; Ward J.; Hutchinson, C. R. Microbiol. 1996, 142, 269-275.) Even in $\mathrm{E}_{p}$, substrate specificity studies prior to the work of the present inventors
were limited to only a few available hexopyranosyl phosphates. (See Lindquist, L.; Kaiser, R.; Reeves, P. R.; Lindberg, A. A. Eur. J. Biochem. 1993, 211, 763-770.)

## SUMMARY OF THE INVENTION

The present invention is directed to methods of engineering or mutating nucleotidylyltransferases, such as $\mathrm{E}_{p}$, to vary their specificity in a directed manner. The invention is also directed to nucleotidylyl-transferases and mutated nucleotidyltransferases, preferably $\mathrm{E}_{p}$ or other nucleotidyltransferases modified by the present methods. The present invention is further directed to mutant $\mathrm{E}_{p}$ and other nucleotidyltransferases with altered substrate specificity, methods for their production, and methods of producing nucleotide sugars, which utilize these nucleotidyl-transferases.

The present invention is also directed to methods of synthesizing desired nucleotide sugars using natural and/or mutated $\mathrm{E}_{p}$ or other nucleotidylyltransferases, preferably $\mathrm{E}_{p}$ or other nucleotidylyltransferases mutated by the present methods. Additionally, the present invention is directed to nucleotide sugars sythesized by the present methods.

Examples of nucleotide sugars produced the present methods (that is, via the exploitation of the promiscuity of $\mathrm{E}_{p}$ ) include, but are not limited to Thymidine $5^{\prime}$-( $\alpha$-Dglucopyranosyl diphosphate) (58); Uridine $5^{5}$-( $\alpha$ - D-glucopyranosyl diphosphate) (59); Thymidine 5'-(2-deoxy- $\alpha$ -D-glucopyranosyl diphosphate) (60); Uridine $5^{2}$-(2-deoxy-$\alpha$-D-glucopyranosyl diphosphate) (61); Thymidine 5'-(3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (62); Uridine $5^{\prime}$ -(3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (63); Thymidine 5'-(4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (64); Uridine 5'-(4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (65); Thymidine 5'-(6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (66); Uridine $5^{\prime}$-(6-deoxy- $\alpha-D$-glucopyranosyl diphosphate) (67); Thymidine 5'-( $\alpha$-D-mannopyranosyl diphosphate) (68); Uridine 5'-( $\alpha$-D-mannopyranosyl diphosphate) (69); Thymidine $5^{\prime}$-( $\alpha$-D-galactopyranosyl diphosphate) (70); Uridine $5^{\prime}$-( $\alpha$-D-galactopyranosyl diphosphate) (71); Thymidine 5'-( $\alpha$-D-allopyranosyl diphosphate) (72); Uridine $5^{\prime}$-( $\alpha$-D-allopyranosyl diphosphate) (73); Thymidine $5^{\prime}$-( $\alpha$-D-altropyranosyl diphosphate) (74); Uridine $5^{\prime}-$ ( $\alpha$-D-altropyranosyl diphosphate) (75); Thymidine $5^{\prime}$-( $\alpha$-Dgulopyranosyl diphosphate) (76); Uridine $5^{\prime}$-( $\alpha$-Dgulopyranosyl diphosphate) (77); Thymidine 5'-( $\alpha$-Didopyranosyl diphosphate) (78); Uridine $5^{\prime}$-( $\alpha$-Didopyranosyl diphosphate) (79); Thymidine 5'-( $\alpha-\mathrm{D}-$ talopyranosyl diphosphate) (80); Uridine $5^{\prime}$-( $\alpha$-Dtalopyranosyl diphosphate) (81); Thymidine 5'-(6-amino-6-deoxy- $\alpha-\mathrm{D}$-glucopyranosyl diphosphate) (109); Uridine 5'-(6-amino-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (110); Thymidine $\quad$ 5'-(4-amino-4-deoxy- $\alpha$-D-lucopyranosyl diphosphate) (111); Uridine $5^{\prime}$-(4-amino-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (112); Thymidine 5'-(3-amino-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (113); Uridine $5^{\prime}$ -(3-amino-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (114); Thymidine $\quad 5$ '-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (115); Uridine 5'-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (116); Thymidine 5'-(6-aceta-mido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (117); Uridine $\quad 5^{\prime}$-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (118); Thymidine 5'-(4-acetamido-4-deoxy- $\alpha$ -D-glucopyranosyl diphosphate) (119); Uridine 5'-(4-aceta-mido-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (120); Thymidine 5'-(3-acetamido-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (121); Uridine 5'-(3-acetamido-3-deoxy- $\alpha$-D-
glucopyranosyl diphosphate) (122); Thymidine 5'-(2-aceta-mido-2-deoxy-a-D-glucopyranosyl diphosphate) (123); Uridine $\quad 5$ '-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (124); Thymidine $5^{\prime}$-(4-amino-4,6-dideoxy- $\alpha$ -D-glucopyranosyl diphosphate) (125); and Uridine 5'-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate) (126). Nucleotide sugars such as these, and methods for making them, are provided by the present invention.
Examples of nucleotide sugars according to the present invention, which may be produced by designed mutants of $\mathrm{E}_{p}$ include, but are not limited to, Thymidine 5'-(6-aceta-mido-6-deoxy-a-D-glucopyranosyl diphosphate) (117); Uridine $\quad 5$ '-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (118); Thymidine $5^{\prime}$-( $\alpha$-D-glucopyran-6uronic acid diphosphate) (130); Uridine $5^{\prime}$-( $\alpha$-D-glucopy-ran-6-uronic acid diphosphate) (131); Thymidine 5'-(2-ac-etamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (123); Uridine $\quad 5^{\prime}$-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (124); Thymidine 5 '-(4-amino-4,6-dideoxy- $\alpha$ -D-glucopyranosyl diphosphate) (125); Uridine 5'-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate) (126); Thymidine $5^{\prime}$-( $\alpha$-D-arabinopyranosyl diphosphate) (128); and Uridine $5^{\prime}$-( $\alpha-$-D-arabinopyranosyl diphosphate) (129). These nucleotide sugars, and methods for making them, are provided by the present invention.
The present invention is also directed to new glycosyl phosphates, and methods for making them. Examples of these new glycosyl phosphates and methods for synthesizing them are represented for example in FIG. $\mathbf{1}(b)$.

The present inventors have discovered that $\mathrm{E}_{p}$ is pliable in terms of its substrate specificity. The present inventors have also discovered the three dimensional structure of $\mathrm{E}_{p}$ and the molecular details of $\mathrm{E}_{p}$ substrate recognition.
In general, the present invention provides a very rapid method of converting sugar phosphates to nucleotide diphosphosugars.
The present invention will broadly impact efforts to understand and exploit the biosynthesis of glycosylated bioactive natural products, many of which are pharmacologically useful. (See Thorson, J. S.; Shen, B.; Whitwam, R. E.; Liu, W.; Li, Y.; Ahlert, J. Bioorg. Chem., 1999, 27, 172-188; Whitwam, R. E.; Ahlert, J.; Holman, T. R.; Ruppen, M.; Thorson, J. S. J. Am. Chem. Soc., 2000, 122, 1556-1557; Thorson, J. S.; Sievers, E. L.; Ahlert, J.; Shepard, E.; Whitwam, R. E.; Onwueme, K. C.; Ruppen, M. Cur. Pharm. Des., 2000, manuscript in press; and J. S. Thorson, T. J. Hosted Jr., J. Jiang, J. B. Biggins, J. Ahlert, M. Ruppen, Curr. Org. Chem. 2000.)

## BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1. FIG. $1(a)$ depicts a reaction according to the present invention, catalyzed by $\mathrm{E}_{p}$. In this reaction, the enzyme catalyzes the reversible conversion of an $\alpha$-Dhexopyranosyl phosphate (such as an $\alpha$-D-glucopyranosyl phosphate)(2) and NTP, such as TTP (1) to the corresponding NDP-sugar nucleotide (for example a TDP-sugar nucleotide, such as TDP-Glc) (3) and pyrophosphate (4). Glc1P (2) depicted in the reaction of FIG. $\mathbf{1}(a)$ is a commercially available $\alpha$-D-hexopyranosyl phosphate (although other $\alpha$-D-hexopyranosyl phosphates that may be used in accordance with the present invention may include those synthesized from free sugars.)
FIG. $\mathbf{1}(b)$ depicts the synthesis of $\alpha$-D-hexopyranosyl phosphates.

FIG. 2. $\mathrm{E}^{p}$-Catalyzed Conversion of Substrates wherein superscript indicators (a)-(h) are as follows: (a) Percent
conversion $=\left[\mathrm{A}_{p} /\left(\mathrm{A}_{P}+\mathrm{A}_{T}\right)\right] \times 100$, where $\mathrm{A}_{P}$ is the NDP-sugar product peak integration and $A_{T}$ represents the NTP peak integration. HRMS for all observed products reported in the supporting information. (b) Standard retention times: TDP, 4.5 mm ; TTP, 7.2 mm ; UDP, 4.0 mm ; UTP, 6.1 mm . (c) Commercially available. (d) Coelutes with commercially available standard. (e) Product hydrolysis observed (43, $7.6 \%$ TDP and $10.2 \%$ UDP). (f) Adjusted for the $2: 1 \alpha / \beta-28$. (g) In contrast to previously published studies (See Lindquist, L; Kaiser, R.; Reeves, P. R.; Lindberg, A. A., Eur J. Biochem, 1993, 211, 763-770). (h) No products observed

FIG. 3. FIG. $3(a)$ sets forth a reaction according to the present invention, catalyzed by $\mathrm{E}_{p}$. FIG. 3(b) shows an overview of the key steps in the described syntheses of $\mathrm{E}_{p}$ substrates analogs. The box highlights the point from which the aminodeoxy- $\alpha$-D-glucose phosphate series and N -acetyl-aminodeoxy- $\alpha$-D-glucose phosphate series diverge. The reaction conditions of the steps are as follows: (a) TMSSEt, $\mathrm{ZnI}_{2}$ ( $84.2 \%$ overall yield); (b) i) MeONa, ii) $\mathrm{NaH}, \mathrm{BnBr}(77.3 \%$ average overall yield, two steps) (c) i) $\mathrm{SnCl}_{2}, \mathrm{PhSH}, \mathrm{Et}_{3} \mathrm{~N}$, ii) $\mathrm{Ac}_{2} \mathrm{O}$, pyr ( $84.0 \%$ average overall yield, two steps); (d) i) $\mathrm{Tf}_{2} \mathrm{O}$, pyr, ii) $\mathrm{NaN}_{3}(87.7 \%$ average overall yield, two steps); (e) i) NaOMe , ii) $\mathrm{CH}_{3} \mathrm{CH}\left(\mathrm{OCH}_{3}\right)_{2}$ $\mathrm{CH}_{3}, \mathrm{TsOH}$, iii) $\mathrm{NaH}, \mathrm{BnBr}$, iv) $\left.\mathrm{HCl} / \mathrm{MeOH}, ~ v\right) ~ \mathrm{BzCl}$, DMAP, $\mathrm{Et}_{3} \mathrm{~N}$ (87.3\% average overall yield, five steps); final steps (not shown): i) phosphorylation, ii) reductive deprotection, iii) cation exchange to give the $\mathrm{Na}^{+}$salt ( $44.4 \%$ average overall yield).

FIG. 4. Percent conversion to product using substrates according to the present invention.

FIG. 5. Examples of pharmacologically important glycosylated metabolites. The general nucleotidylyl-transferasecatalyzed formation of NDP-sugars is highlighted in the box while the carbohydrate ligands of each metabolite are accentuated in red. Note the difference between erythromycin from S. erythrea and megalomicin from M. megalomicea is the addition of a third sugar megosamine (highlighted by the arrow).

FIG. 6. Representative region of the density-modified experimental electron density map showing the substrate binding pocket in the $\mathrm{E}_{p}$ UDP-Gle structure configured at 1.2 $\sigma$.

FIG. 7 Quaternary structure of $\mathrm{E}_{p}$ bound to UDP-Glc or dTTP. (a) Two 90 degree views of the $\mathrm{E}_{p}$ tetramer bound to four molecules of UDP-Glc. (b) The $\mathrm{E}_{p}$ tetramer bound to eight molecules of dTTP.

FIG. 8. Structures of the $\mathrm{E}_{p}$ monomer and structural homologs SpsA and GlmU. The $\beta$ strands and $\alpha$ helices of the $\alpha / \beta$ open sheet Rossmann fold are shown in red and green respectively, while variable regions are shown in yellow. (a) Two 90 degree views of the $\mathrm{E}_{p}$ monomer (upper) and the corresponding structures of of SpsA (lower left) and GlmU (lower right). (b) The folding topology of $\mathrm{E}_{p}, \mathrm{SpsA}$, and GlmU.

FIG. 9. Close up views of the $\mathrm{E}_{p}$ active site. Hydrogen bonds are depicted by green dashed lines. (a) Interactions between $\mathrm{E}_{p}$ and the dTTP substrate (left) and the UDP-Glc product (right). (b) Interactions between $\mathrm{E}_{p}$ and the glucose moiety in the sugar binding pocket. (c) Two different views of dTTP bound in the 'accessory' site at the monomer interface. The different chains of the tetramer are labeled either in blue (chain-A) or red (chain-B). The $\beta$-phosphate of dTTP hydrogen bonds with both His117 of chain-A and Gly221 of chain-B.

FIG. 10. (a) The proposed enzymatic mechanism based on the structures of substrate- and product-bound $\mathrm{E}_{p}$. (b) The determination of $\mathrm{E}_{p}$ steady state kinetic parameters. The
conditions for the $\mathrm{E}_{p}$ assay conditions and HPLC resolution of reactants and products were similar to those described in Jiang, J., Biggins, J. B. \& Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP- and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000). The Lineweaver-Burke plots of from assays (done in triplicate) varying dTTP concentration as a function of $\alpha$-D-glucose-1 -phosphate concentration (mM): $0.5(\square), 0.3 \mathrm{nM}(\mathrm{o}), 0.2(\diamond), 0.1(\Delta)$ and $0.05(\square)$. (c) Secondary plot from FIG. $6 b$ (dTTP $K_{m}=0.7 \pm 0.2$; $\mathrm{V}_{\text {max }}=0.03 \pm 0.01 \mathrm{mM} \mathrm{min}^{-1}$ ). (d) The Lineweaver-Burke plots of assays (done in triplicate) varying $\alpha$-D-glucose-1phosphate concentration as a function of dTTP concentration $(\mathrm{mM}): 0.25(\square), 0.15 \mathrm{nM}(0), 0.1(\diamond), 0.05(\Delta)$ and 0.02
(■). (e) Secondary plot from FIG. $6 d$ ( $\alpha$-D-glucose-1phosphate $\mathrm{K}_{m}=0.3 \pm 0.1 ; \mathrm{V}_{\max }=0.03 \pm 0.02 \mathrm{mM} \mathrm{min}^{-1}$ ).
FIG. 11. Percent conversion of sugar phosphates according to the present invention by wild-type and mutant enzymes. The alterations from native substrate (Glc-1-P,1) are highlighted in red. For the mutant pool, mutants Asp41Asn, Glu62Asp, Thr201A and Trp224His were pooled, concentrated and an aliquote constituting $60 \mu \mathrm{~g}$ of each mutant (corresponding to $3.5 \mathrm{U} \mathrm{E}_{p}$ ) was utilized for the assay.
Percent conversion was determined as described in Jiang, J., Biggins, J. B. \& Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDPand TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000).
$\S R e p r e s e n t s$ less than $5 \%$ conversion to product.
FIG. 12. Shows alignment of the Thymidylyltransferase sequence for Ep (SEQ ID NO:1) and D8300 (SEQ ID NO:2).

FIG. 13. Shows alignment of the Ep sequence (SEQ ID NO:1) with Glc-1-P Uridylyltransferases D8300 (SEQ ID NO:2), CAA06172 (SEQ ID NO:4), and GALU MYCGE (SEQ ID NO:3).
FIG. 14. Shows alignment of the Ep sequence (SEQ ID NO:1) with Glc-1-P Adenylyltransferases D8300 (SEQ ID NO:2), B72403 (SEQ ID NO:6), and GLGC BACSU (SEQ ID NO:5).
FIG. 15. Shows alignment of the Ep sequence (SEQ ID NO:1) with Man-1-P Guanylyltransferases D8300 (SEQ ID NO:2), BAA34807 (SEQ ID NO:8), and AAC39498 (SEQ ID NO:7).

FIG. 16. Shows alignment of the Ep sequence (SEQ ID NO:1) with NAcGlc-1-P Uridylyltransferases D8300 (SEQ ID NO:2), GCAD BACSU (SEQ ID NO:10), and E72229 (SEQ ID NO:9).

FIG. 17. Shows alignment of the Ep sequence (SEQ ID NO:1) with Glc-1-P Cytidylyltransferases D8300 (SEQ ID NO:2), RFBF SALTY (SEQ ID NO:12), and AAB31755 (SEQ ID NO:11).
FIG. 18. Shows general Nucleotidylyltransferase Alignment between: Ep (SEQ ID NO:1); D8300 (SEQ ID NO:2); GALU MYCGE (SEQ ID NO:3); CAA06172 (SEQ ID NO:4); GLGC BACSU (SEQ ID NO:5); B72403 (SEQ ID NO:6); AAC39498 (SEQ ID NO:7); BAA34807 (SEQ ID NO:8); E72229 (SEQ ID NO:9); GCAD BACSU (SEQ ID NO:10); AAB31755 (SEQ ID NO:11); and RFBF SALTY (SEQ ID NQ:12).
FIG. 19 (A-G). FIG 19 (A-G) is a BLAST analysis for $\mathrm{E}_{p}$ sequences, showing sequences producing high-scoring segment pairs. showing sequences producing high-scoring segment pairs.

FIG. 20. FIGS. $\mathbf{2 0}(a)$ and $\mathbf{2 0}(b)$ depict NDP-sugar nucleotides that may be prepared using nucleotydylyl-transferases as enzymes in accordance with the present invention.

FIG. 21. Interaction between Ep and the glucose moiety in the sugar binding pocket.

FIG. 22. Summary of sugar phosphate accepted by Ep and mutants

FIG. 23. One dimensional representation of FIG. 21 illustrating some of the important contacts and potential sites for engineering promiscuity of nucleotidyly-transferases.

## DETAILED DESCRIPTION OF THE INVENTION

The present inventors discovered that the Salmonella enterica LT2 rmlA-encoded $\alpha$-D-glucopyranosyl phosphate thymidylyltransferase ( $\mathrm{E}_{p}$ ), (also referred to as dTDP-glucose synthase, dTDP-glucose pyrophosphorylase, thymidine diphosphoglucose pyrophosphorylase and thymidine diphosphate glucose pyrophosphorylase), which catalyzes the conversion of $\alpha$-D-glucopyranosyl phosphate (Glc-1-P) and dTTP to dTDP- $\alpha$-D-glucose (TDP-Glc) and pyrophosphate ( $\mathrm{PP}_{i}$ ), displays unexpected promiscuity toward both its nucleotide triphosphate (NTP) and its sugar phosphate substrates. Through a substrate specificity reevaluation of Salmonella enterica LT2 $\alpha$-D-glucopyranosyl phosphate thymidylyltransferase ( $\mathrm{E}_{p}$ ), the present inventors made the surprising discovery that this enzyme can convert a wide variety of phosphates, including for example, $\alpha$-D-hexopyranosyl phosphates, including, but not limited to, deoxy- $\alpha-$ D-glucopyranosyl, aminodeoxy- $\alpha$-D-hexopyranosyl and acetamidodeoxy- $\alpha$-D-hexopyranosyl phosphates to their corresponding dTDP- and UDP-nucleotide sugars.

This discovery led to the invention by the present inventors of general chemo-enzymatic methods of rapidly generating nucleotide diphosphosugar reagents. These methods allow for the provision of a substrate set for developing in vitro glycosylation systems, which are useful for, inter alia, in vitro production of known bioactive metabolites and of 4 new bioactive metabolites.
$\alpha$-D-Hexopyranosyl Phosphates and Methods of Making the Same

An embodiment of the invention includes $\alpha$-D-hexopyranosyl phosphates, methods including combining these phosphates with NTP in the presence of nucleotidylyltransferase, which may be wild type or mutated, and nucleotide sugars produced by converting such hexopyranosyl phosphates using nucleotidylyl-transferases, such as $\mathrm{E}_{p}$.
$E_{p}$ is encoded by rmlA, which was previously known as rfbA (Reeves et al. Trends Microbiol. 1996, 4, 495-502). The rmlA-encoded $\mathrm{E}_{p}$ was overexpressed in E. coli to provide the desired $\mathrm{E}_{p}$ as $>5 \%$ of the total soluble protein. The corresponding $\mathrm{E}_{p}$ was purified to near homogeniety with a specific activity of $110 \mathrm{Umg}^{-1}$, a 2 -fold improvement over the previously reported values. (See Lindquist, L.; Kaiser, R.; Reeves, P. R.; Lindberg, A. A. Eur. J. Biochem., 1993, 211, 763-770.) An $\left(\mathrm{NH}_{4}\right) 2 \mathrm{SO}_{4}$ precipitate of E. coli-prfbA-C crude extracts was dialyzed against buffer B (20 mM Tris.HCl, 1 mM EDTA, pH 7.5 ) The dialysate was resolved by anion exchange (DE52, $3 \times 15 \mathrm{~cm}, 50 \mathrm{~mL}$ buffer B wash followed by a linear gradient of $0-500 \mathrm{mM} \mathrm{NaCl}$, $1.0 \mathrm{~mL} \mathrm{~min}^{-1}$ ) and the $\mathrm{E}_{p}$ fractions combined, concentrated and further resolved by FPLC gel filtration (S-200, $2 \times 70 \mathrm{~cm}$, 50 mM Tris. $\mathrm{HCl}, 200 \mathrm{mM} \mathrm{NaCl}, \mathrm{pH} 7.5$ ). The purified $\mathrm{E}_{p}$ was stored in aliquots ( $-80^{\circ} \mathrm{C}$.) until used.

Although $\alpha$-D-glucopyranosyl phosphate (2) (FIG. 2), $\alpha$-D-mannopyranosyl phosphate (compound 56) (FIG. 2) and $\alpha$-D-galactopyranosyl phosphate (57) (FIG. 2) were commercially available for examination as potential substrates for $\mathrm{E}_{p}$, most of the $\alpha$-D-hexopyranosyl phosphates examined were synthesized from free sugars.
For synthetically derived $\alpha$-D-hexopyranosyl phosphates, particularly glycosyl phosphates, a general phosphorylation strategy from the appropriately protected precursor relied upon
i) anomeric activation via the ethy 1 -thio- $\beta$-D-pyranoside [to form e.g., Ethyl 2,3,4-tri-O-benzoyl-6-deoxy-1-thio- $\beta$-D-glucopyranoside (9), Ethyl 2,3,6-tri-O-ben-zoyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (17), Ethyl 2,4,6-tri-O-benzoyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (25), Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- $\beta$-Dgulopyranoside (30), Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- $\beta$-D-allopyranoside (35) and Ethyl 3,4,6-tri-O-benzoyl-2-deoxy-1-thio- $\beta$-D-glucopyranoside (40) (The $\alpha / \beta-40$ mixture ( $1: 1.5$ ) was chromatographically resolved.) (FIG. $1(b)$ )],
ii) deprotection/reprotection [to form e.g., Ethyl 2,3,4-tri-O-benzyl-6-deoxy-1-thio- $\beta$-D-glucopyranoside (10), Ethyl 2,3,6-tri-O-benzyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (18), Ethyl 2,4,6-tri-O-benzyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (26), Ethyl 2,3,4,6-tetra-O-benzyl-1-thio- $\beta$-D-gulopyranoside (31), and Ethyl 2,3, 4,6-tetra-O-benzyl-1-thio- $\beta$-D-allopyranoside (36) (FIG. 1(b))],
iii) phosphorylation [to form e.g., Dibenzyl-(2,3,4-tri-O-benzyl-6-deoxy- $\alpha$-D-glucopyranosyl) phosphate (11), Dibenzyl-(2,3,6-tri-O-benzyl-4-deoxy- $\alpha$-D-glucopyranosyl) phosphate (19), Dibenzyl-(2,4,6-tri-O-benzyl-3-deoxy- $\alpha$-D-glucopyranosyl) phosphate (27), Dibenzyl-(2,3,4,6-tetra-O-benzyl- $\alpha$-D-gulopyranosyl) phosphate (32), Dibenzyl-(2,3,4,6-tetra-O-benzyl-o-D-allopyranosyl) phosphate (37), and Dibenzyl-(3,4,6-tri-O-ben-zoyl-2-deoxy- $\alpha$-D-glucopyranosyl) phosphate (41) (FIG. 1(b))], and
iv) complete deprotection [to form e.g., Disodium 6-deoxy- $\alpha$-D-glucopyranosyl phosphate (12), Disodium 4-deoxy- $\alpha$-D-glucopyranosyl phosphate (20), Disodium 3-deoxy- $\alpha$-D-glucopyranosyl phosphate (28), Disodium $\alpha$-D-gulopyranosyl phosphate (33), Disodium $\alpha$-D-allopyranosyl phosphate (38) and Disodium 2-deoxy- $\alpha$-D-glucopyranosyl phosphate (43) (FIG. 1(b))].
In FIG. 1(b): (a) $\mathrm{Ph}_{3} \mathrm{P}, \mathrm{CCl}_{4}$; (b) $\mathrm{Ac}_{2} \mathrm{O}$, pyr; (c) (i) $\mathrm{LiAlH}_{4}$, (ii) $\mathrm{AcOH} / \mathrm{HCl}$, (iii) BzCl , pyr; (d) BzCl , pyr; (e) $\mathrm{pFPTC}-\mathrm{Cl}$, DMAP; (f) (n-Bu) 3 SnH ; (g)(i) NaH , imidazole; (ii) $\mathrm{CS}_{2}$; (iii) $\mathrm{CH}_{3} \mathrm{I}$; (h) AIBN, $(\mathrm{n}-\mathrm{Bu})_{3} \mathrm{SnH}$; (i) (i) $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}$, (ii) BzCl, pyr; (j) EtS-TMS, $\mathrm{ZnI}_{2}$; (k) (i) NaOMe ; (ii) NaH ; (iii) BnBr ; (1) (i) ( BnO ) $2 \mathrm{P}(\mathrm{O}) \mathrm{OH}$, NIS; (m) $\mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C}$; (n) (i) HBr ; (ii) ( BnO ) $2 \mathrm{P}(\mathrm{O}) \mathrm{OH}$, silver triflate, 2,4,6-collidine; (O) NaOH ; (p) $\mathrm{AcOH} / \mathrm{HCl}$. In each case, cation exchange provided the $\mathrm{Na}+$ salt.

The overall yield of this four-step phosphorylation strategy ranged from $19 \%-28 \%$ including the final ion exchange. FIG. 1 (b) shows these glycosyl phosphates and methods for synthesizing them. These glycosyl phosphates, and methods for making them, are provided by the present invention.

The present method includes anomerically activating an ethyl 1-thio- $\beta$-D-pyranoside to form a compound having the formula

wherein $\mathrm{R}^{2}$ is $\mathrm{OCH}_{3}, \mathrm{OBz}$, or OH ,
$\mathrm{R}^{3}$ is $\mathrm{OH}, \mathrm{OAc}$, or OBz ,
$\mathrm{R}^{4}$ is $\mathrm{H}, \mathrm{OH}$, or a halogen atom, and $R^{5}, R^{6}, R^{7}, R^{8}$, and $R^{9}$ are each $O B z$,
and three or more of $\mathrm{R}^{3}, \mathrm{R}^{5}, \mathrm{R}^{6}, \mathrm{R}^{7}, \mathrm{R}^{8}$, and $\mathrm{R}^{9}$ are OBz substituents; deprotecting the OBz substituents to convert at least one such substituent to a OBn substituent; phosphorylating to form a compound of the formula

wherein $\mathrm{R}^{1}$ is $\mathrm{OCH}_{3}, \mathrm{OBz}, \mathrm{OAc}$ or OH ,
$\mathrm{R}^{2}$ is $\mathrm{OCH}_{3}, \mathrm{OBz}, \mathrm{OAc}$ or OH , $\mathrm{R}^{3}$ is OH , OAc , or OBz ,
$\mathrm{R}^{4}$ is $\mathrm{H}, \mathrm{OH}, \mathrm{OBz}, \mathrm{OAc}$ or a halogen atom, $\mathrm{R}^{5}, \mathrm{R}^{6}, \mathrm{R}^{7}, \mathrm{R}^{8}$, and $\mathrm{R}^{9}$ are each OBz or OAc , and $\mathrm{R}^{10}$ is OH , or OBn ,
wherein at least four of $\mathrm{R}^{3}, \mathrm{R}^{4}, \mathrm{R}^{5}, \mathrm{R}^{6}, \mathrm{R}^{7}, \mathrm{R}^{8}, \mathrm{R}^{9}$, and $\mathrm{R}^{10}$ are independently OBn or OBz substituents; and
deprotecting to convert any OBn substituents to OH substituents.

Preferably, the $\alpha$-D-hexopyranosyl phosphate is a glycosyl phosphate. Also included are $\alpha$-D-hexopyranosyl phosphates, preferably glycosyl phosphates synthesized by these methods. Preferably these $\alpha$-D-hexopyranosyl phosphates are selected from the group consisting of deoxy- $\alpha-\mathrm{D}-\mathrm{glu}-$ copyranosyl, aminodeoxy- $\alpha$-D-hexopyranosyl and acetami-dodeoxy- $\alpha$-D-hexopyranosyl phosphates.

The present invention also includes a method that includes providing isolated $\mathrm{E}_{p}$ having the formula

wherein $\mathrm{R}^{1}$ is $\mathrm{OCH}_{3}, \mathrm{OBz}$, OAc or OH ,
$\mathrm{R}^{2}$ is $\mathrm{OCH}_{3}, \mathrm{OBz}, \mathrm{OAc}$ or OH ,
$\mathrm{R}^{3}$ is $\mathrm{OH}, \mathrm{OAc}$, or OBz ,
$\mathrm{R}^{4}$ is $\mathrm{H}, \mathrm{OH}, \mathrm{OBz}, \mathrm{OAc}$ or a halogen atom,
$R^{5}, R^{6}, R^{7}, R^{8}$, and $\mathrm{R}^{9}$ are each OBz or OAc , and $\mathrm{R}^{10}$ is OH , or OBn ,
wherein at least four of $R^{3}, R^{4}, R^{5}, R^{6}, R^{7}, R^{8}, R^{9}$, and $R^{10}$ are independently OH or OBz substituents.
Alternatively, phosphorylation of Dibenzyl-(2,3,4,6-tetra-O-benzoyl- $\alpha$-D-altropyranosyl) phosphate (45), Dibenzyl5 (2,3,4,6-tetra-O-benzoyl- $\alpha$-D-idopyranosyl) phosphate (49) and Dibenzyl-(2,3,4,6-tetra-O-acetyl- $\alpha$-D-talopyranosyl) phosphate (53) (FIG. 1(b)) via the glycosyl halide followed by complete deprotection gave the glycosyl phosphates Disodium $\alpha$-D-altropyranosyl phosphate (47), Disodium $10 \alpha$-D-idopyranosyl phosphate (51) and Disodium $\alpha$-D-talopyranosyl phosphate (55) as depicted in FIG. $\mathbf{1}(b)$ in an overall yield ranging from $37 \%-47 \%$. The 6-deoxy precursor 1,2,3,4-tetra-O-benzoy1-6-deoxy- $\alpha, \beta$-D-glucopyranose (8) may be synthesized by $\mathrm{LiAlH}_{4}$ reduction and subsequent benzoylation of the halide Methyl 2,3,4-tri-O-acetyl-6-chloro-6-deoxy- $\alpha$-D-glucopyranoside (7). (See Anisuzzaman, A. K. M.; Whistler, R. L. Carbohydr. Res. 1978, 61, 511-518.). For the 4 -deoxy progenitor, deoxygenation at C-4 may be accomplished by selective benzoylation of methyl $\beta$-D-galactopyranoside Methyl $\beta$-D-galactopyranoside (13) (as depicted in FIG. $\mathbf{1}(b)$ ) to provide the desired tribenzolated Methyl 2,3,6-tri-O-benzoyl- $\beta$-D-galactopyranoside (14) (54\%) as well as the tetrabenzolated derivative (19\%). Subsequent C-4 activation Methyl 2,3,6-tri-O-ben-zoyl-4-O-pentafluorophenoxythiocarbonyl- $\beta$-D-galactopyranoside (15) and ( $\mathrm{n}-\mathrm{Bu})_{3} \mathrm{SnH}$ reductive 4-deoxygenation were accomplished as described in Kanie, O.; Crawley, S. C.; Palcic, M. M.; Hindsgaul, O. Carbohydr. Res. 1993, 243, 139-164 to give the desired 4-deoxy precursor Methyl ${ }^{30}$ 2,3,6-tri-O-benzoyl-4-deoxy- $\beta$-D-galactopyranoside (16). The 3-deoxy predecessor 1,2,4,6-tetra-O-benzoyl-3-deoxy-$\alpha$-D-glucofuranose (24) (FIG. 1(b)) was synthesized from 1,2:5,6-di-O-isopropylidene- $\alpha$-D-glucofuranose (21) by reduction of the previously reported furanose $1,2: 5,6-\mathrm{Di}-\mathrm{O}-$ 35 isopropylidene-3-O-(methylthio)thiocarbonyl- $\alpha$-D-glucofuranose (22) (See Zhiyuan, Z.; Magnusson, G. Carbohydr. Res. 1994, 262, 79-101), while the 2-deoxy precursor (39) derived from a commercial source.
Thus, another embodiment of the present invention 40 includes methods of making $\alpha$-D-hexopyranosyl phosphates, which include, but are not limited to, phosphorylating a phosphate selected from the group consisting of Dibenzyl-(2,3,4,6-tetra-O-benzyol- $\alpha$-D-altropyranosyl) phosphate, Dibenzyl-(2,3,4,6-tetra-O-benzyl- $\alpha$-D-idopyranoxyl)phosphate, and Dibenzyl-(2,3,4,6-tetra-O-acetyl- $\alpha$-Dtalopyranosyl) phosphate via a glycosyl halide; and deprotecting to form a glycosyl phosphate selected from the group conssiting of Disodium $\alpha$-D-altropyranosyl phosphate, Disodium $\alpha$-Didopyranosyl phosphate and Disodium $\alpha$-D-
50 talopyranosyl phosphate. The present invention also includes $\alpha$-D-hexopyranosyl phosphates prepared according to this method.

Nucleotide Sugars and Methods of Synthesizing the Same
The present invention includes methods of making nucleotide sugars, which include combining $\alpha$-D-hexopyranosyl phosphate and NTP in the presence of at least one mutated nucleotidylyltransferase. Other methods according to the present invention include combining $\alpha$-D-hexopyranosyl 60 phosphate and NTP other than TTP in the presence of at least one nucleotidylyltransferase, and combining NTP and $\alpha$-Dhexopyranosyl phosphate other than Glc1P in the presence of at least one nucleotidylyltransferase.
The present invention includes a method of synthesizing 65 nucleotide sugars that includes combining a nucleotidylyltransferase, $\alpha$-D-glucopyranosyl phosphate, $\mathrm{Mg}^{+2}$, NTP and inorganic pyrophosphatase, and incubating. Preferably, the
incubating is at a temperature of from about $30^{\circ} \mathrm{C}$. to about $45^{\circ} \mathrm{C}$., preferably about $33^{\circ} \mathrm{C}$. to about $42^{\circ} \mathrm{C}$., even more preferably about $37^{\circ} \mathrm{C}$. for about 20 to about 40 minutes, preferably about 25 to about 35 minutes, and even more preferably about 30 minutes. The nucleotydylyltransferase according to these methods may include one or more natural and/or mutated nucleotidylyltransferases, such as natural and/or mutated $\mathrm{E}_{p}$.

The present invention further includes nucleotide sugars made by the methods described herein.

Nucleotide sugars of the present invention may be selected from the group consisting of TDP-sugar, GDPsugar, CDP-sugar, UDP-sugar, and ADP-sugar and combinations thereof.

In embodiments utilizing or including mutated nucleotidylyltransferases, a preferred mutated nucleotidylyltransferase is $\mathrm{E}_{p}$ mutated at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177.

In embodiments utilizing or including mutated nucleotidylyl transferases, a preferred mutated nucleotidylyl transferases is $\mathrm{E}_{p}$ mutated at one or more amino acids in its active site, its divalent cation binding site, and/or its auxiliary site.

Likewise, other preferred mutated nucleotidylyl transferases include nucleotidylyl transferases mutated at one or more amino acids in their active sites, divalent cation binding sites, and/or their auxiliary sites.

To evaluate the synthetic utility of purified thymidylyltransferase, $\mathrm{E}_{p}, \alpha$-D-hexopyranosyl phosphate, $\mathrm{Mg}^{+2}$ and NTP were incubated at about $37^{\circ} \mathrm{C}$. for about 30 min and the extent of product formation determined by HPLC. The results of these assays are illustrated in FIG. 2. Confirmation of product formation was based upon HPLC co-elution with commercially available standards and/or HPLC isolation and high resolution mass spectroscopy of the product. (For select compounds, product peaks were lyopholized and submitted directly for HRMS (FAB) analysis.) As controls, little or no product formation was observed in the absence of $\mathrm{E}_{p}$, glycosyl phosphate, $\mathrm{Mg}^{+2}$, or NTP. A reaction containing 5 mM NTP, 10 mM sugar phosphate and $5.5 \mathrm{mM} \mathrm{MgCl}_{2}$ in a total volume of $50 \mu \mathrm{~L} 50 \mathrm{mM}$ potassium phosphate buffer, pH 7.5 at $37^{\circ} \mathrm{C}$. was initiated by the addition of $3.52 \mathrm{UE}_{p}$ ( $1 \mathrm{U}=$ the amount of protein needed to produce $1 \mu \mathrm{~mol}$ TDP-D-glucose $\mathrm{min}^{-1}$ ). The reaction was incubated with slow agitation for 30 min at $37^{\circ} \mathrm{C}$., quenched with MeOH ( $50 \mu \mathrm{~L}$ ), centrifuged ( $5 \mathrm{~min}, 14,000 \times \mathrm{g}$ ) and the supernatant was stored at $-20^{\circ} \mathrm{C}$. until analysis by HPLC. Samples (20 $\mu \mathrm{L}$ ) were resolved on a Sphereclone 5 u SAX column ( $250 \times 4.6 \mathrm{~mm}$ ) fitted with a guard column ( $30 \times 4.6 \mathrm{~mm}$ ) using a linear gradient ( $20-60 \mathrm{mM}$ potassium phosphate buffer, $\mathrm{pH} 5.0,1.5 \mathrm{~mL} \mathrm{~min}^{-1}, \mathrm{~A}_{275} \mathrm{~nm}$ ).

The following nucleotide sugars are non-limiting examples of nucleotide sugars according to the present invention, which may preferably be produced in accordance with one or more of the methods described herein, and in particular the reactions of FIG. 2: (58) Thymidine $5^{\prime}$-( $\alpha$-Dglucopyranosyl diphosphate) (HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 563.0705$; found $\mathrm{m} / \mathrm{z} 563.0679(\mathrm{M}+\mathrm{H})$ ); (59) Uridine $5^{\prime}$-( $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{17} \mathrm{~N}_{2} \mathrm{P}_{2} 565.0507$; found $\mathrm{m} / \mathrm{z}$ $565.0472(\mathrm{M}+\mathrm{H})$ ); (60) Thymidine $5^{\prime}$-(2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2}$ 547.0704; found $\mathrm{m} / \mathrm{z} 547.0714(\mathrm{M}+\mathrm{H})$ ); (61) Uridine $5^{\prime}$-( 2 -deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0510(\mathrm{M}+\mathrm{H})$ ); (62) Thymidine $5^{\prime}$-(3-deoxy- $\alpha$-D-glu-
copyranosyl diphosphate) (HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2} 547.0704$; found $\mathrm{m} / \mathrm{z} 547.0720(\mathrm{M}+\mathrm{H})$ ); (63) Uridine $5^{\prime}$-(3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0485(\mathrm{M}+\mathrm{H})$ ); (64) Thymidine $5^{\prime}$-(4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2} 547.0704$; found $\mathrm{m} / \mathrm{z} 547.0693(\mathrm{M}+\mathrm{H})$ ); (65) Uridine $5^{\prime}$-(4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0500(\mathrm{M}+\mathrm{H})$ ); (66) Thymidine 5 '-(6-deoxy- $\alpha-\mathrm{D}-\mathrm{glu}-$ copyranosyl diphosphate) (HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2} 547.0704$; found $\mathrm{m} / \mathrm{z} 547.0730(\mathrm{M}+\mathrm{H})$ ); (67) Uridine $5^{\prime}$-(6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0492(\mathrm{M}+\mathrm{H})$ ); (68) Thymidine $5^{\prime}$-( $\alpha$-D-mannopyranosyl diphosphate) (HRMS (FAB) calc 563.0705; found $\mathrm{m} / \mathrm{z}$ $563.0701(\mathrm{M}+\mathrm{H})$ ); (69) Uridine 5 '-( $\alpha$-D-mannopyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0503(\mathrm{M}+\mathrm{H})$ ); (70) Thymidine $5^{\prime}$-( $\alpha$ - D -galactopyranosyl diphosphate) (HRMS (FAB) calc 563.0705; found $\mathrm{m} / \mathrm{z}$ $563.0710(\mathrm{M}+\mathrm{H})$ ); (71) Uridine $5^{\prime}$-( $\alpha$-D-galactopyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0508(\mathrm{M}+\mathrm{H})$ ); (72) Thymidine $5^{7}$-( $\alpha$-D-allopyranosyl diphosphate) (HRMS (FAB) calc 563.0705; found $\mathrm{m} / \mathrm{z}$ $563.0715(\mathrm{M}+\mathrm{H})$ ); (73) Uridine $5^{5}$-( $\alpha$-D-allopyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0507(\mathrm{M}+\mathrm{H})$ ); (74) Thymidine $5^{\prime}$-( $\alpha$-D-altropyranosyl diphosphate) (HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z}$ $563.0699(\mathrm{M}+\mathrm{H})$ ); (75) Uridine $5^{\prime}$-( $\alpha$-D-altropyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0511(\mathrm{M}+\mathrm{H})$ ); (76) Thymidine $5^{\prime}$-( $\alpha$-D-gulopyranosyl diphosphate) (HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z}$ $563.00712(\mathrm{M}+\mathrm{H})$ ); (77) Uridine $5^{\prime}$-( $\alpha$-D-gulopyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0512(\mathrm{M}+\mathrm{H})$ ); (78) Thymidine 5 '-( $\alpha$-D-idopyranosyl diphosphate) (HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z}$ $563.0708(\mathrm{M}+\mathrm{H}))$; (79) Uridine 5 '-( $\alpha$-D-idopyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0507(\mathrm{M}+\mathrm{H})$ ); (80) Thymidine $5^{5}$-( $\alpha$-D-talopyranosyl diphosphate) (HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z}$ $563.0710(\mathrm{M}+\mathrm{H})$ ); and (81) Uridine $5^{5}$-( $\alpha$-D-talopyranosyl diphosphate) (HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z}$ $565.0499(\mathrm{M}+\mathrm{H})$ ); although data is not depicted for all products.

Other nucleotide sugars in accordance with the present invention include, but are not limited to, the following: (109) Thymidine $\quad 5^{\prime}$-(6-amino-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 562.0839; found $\mathrm{m} / \mathrm{z} 562.0837(\mathrm{M}+\mathrm{H})$ ); (110) Uridine $5^{\prime}$ -(6-amino-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 564.0632$; found $\mathrm{m} / \mathrm{z} 564.0640(\mathrm{M}+\mathrm{H})$ ); (111) Thymidine 5'-(4-amino-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2} 562.0839$; found $\mathrm{m} / \mathrm{z} 562.0848$ $(\mathrm{M}+\mathrm{H})$ ); (112) Uridine 5 '-(4-amino-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 564.0632$; found $\mathrm{m} / \mathrm{z} 564.0638(\mathrm{M}+\mathrm{H})$ ); (113) Thymidine 5 '-(3-amino-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 562.0839 ; found $\mathrm{m} / \mathrm{z} 562.0835(\mathrm{M}+\mathrm{H})$ ); (114) Uridine $5^{\prime}$ -(3-amino-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 564.0632$; found $\mathrm{m} / \mathrm{z} 564.0622(\mathrm{M}+\mathrm{H})$ ); (115) Thymidine 5'-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2} 562.0839$; found $\mathrm{m} / \mathrm{z} 562.0842$ -
$(\mathrm{M}+\mathrm{H})$ ); (116) Uridine 5'-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 564.0632$; found $\mathrm{m} / \mathrm{z} 564.0630(\mathrm{M}+\mathrm{H})$ ); (117) Thymidine $5^{\prime}$-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 604.0945; found $\mathrm{m} / \mathrm{z} 604.0953(\mathrm{M}+\mathrm{H})$ ); (118) Uridine $5^{\prime}$ -(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2} 606.0737$; found $\mathrm{m} / \mathrm{z} 606.0732(\mathrm{M}+\mathrm{H})$ ); (119) Thymidine 5'-(4-acetamido-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 604.0945$; found $\mathrm{m} / \mathrm{z} 604.0940$ $(\mathrm{M}+\mathrm{H})$ ); (120) Uridine $5^{\prime}$-(4-acetamido-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2} 606.0737$; found $\mathrm{m} / \mathrm{z} 606.0730(\mathrm{M}+\mathrm{H})$ ); (121) Thymidine 5'-(3-acetamido-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 604.0945; found $\mathrm{m} / \mathrm{z} 604.0947(\mathrm{M}+\mathrm{H})$ ); (122) Uridine $5^{\prime}$ -(3-acetamido-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2}$ 606.0737; found $\mathrm{m} / \mathrm{z} 606.0735(\mathrm{M}+\mathrm{H})$ ); (123) Thymidine 5'-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 604.0945$; found $\mathrm{m} / \mathrm{z} 604.0951$ $(\mathrm{M}+\mathrm{H})$ ); (124) Uridine $5^{5}$-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2} 606.0737$; found $\mathrm{m} / \mathrm{z} 606.0738(\mathrm{M}+\mathrm{H})$ ); (125) Thymidine 5 '-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{14} \mathrm{~N}_{3} \mathrm{P}_{2}$ 546.0889; found $\mathrm{m} / \mathrm{z} 546.0895(\mathrm{M}+\mathrm{H})$ ); and (126) Uridine 5'-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2} 548.0682$; found $\mathrm{m} / \mathrm{z} 548.0673(\mathrm{M}+\mathrm{H})$ ).

Further nucleotide sugars in accordance with the present invention include, but are not limited to, the following:




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-continued








FIG. 2 illustrates the utiltity of $\mathrm{E}_{p}$ as a catalyst/reagent to simplify the synthesis of useful nucleotide sugars - of the twelve glycosyl phosphates tested (which include all possible $\alpha$-D-hexoses and monodeoxy $\alpha$-D-glucoses), all produce product with both TTP and UTP under the conditions described. These yields might be further improved by using pyrophosphatase to drive the equilibrium of the reaction. An examination of accepted $\alpha$-D-hexopyranosyl phosphates with TTP suggests that $\mathrm{E}_{p}$ prefers pyranosyl phosphates, which are predicted to exist predominately as ${ }^{4} \mathrm{C}_{1}$ conformers [e.g., (12), (20), (28), (43), a-D-glucopyranosyl phosphate (2), $\alpha$-D-mannopyranosyl phosphate (56), and $\alpha$-Dgalactopyranosyl phosphate (57) (FIGS. 1 and 2)], while those predicted to not adopt the ${ }^{4} \mathrm{C}_{1}$ conformation [e.g., Ethyl 2,3,4,6-tetra-O-benzyl-1-thio- $\beta$-D-gulopyranoside (31), $\alpha$-D-allopyranosyl phosphate (38), $\alpha$-D-altropyranosyl phosphate (47), $\alpha$-D-idopyranosyl phosphate (51) and $\alpha$-Dtalopyranosyl phosphate (55) (FIGS. 1 and 2)] show less activity.

Regarding specific interactions required for conversion, analysis of the corresponding deoxy series [(12), (20), (28) and (43) (FIGS. 1 and 2)] implicates only a single required hydroxyl (C-2), the removal of which impairs the yield by $>70 \%$. A similar trend is observed in the UTP series with two exceptions, glycosides (28) and $\alpha$-D-mannopyranosyl phosphate (56) (FIGS. 1 and 2). Cumulatively, these results suggest that, while the C-2 hydroxyl is important for turnover, alterations at C-3 in the context of UTP result in adverse cooperativity.
Aminodeoxy- $\alpha$-D-hexapyranosyl phosphates and aceta-midodeoxy- $\alpha$-D-hexapyranosyl phosphates are each examples of $\alpha$-D-hexapyranosyl phosphates that may be used in accordance with the present invention. A direct comparison of the aminodeoxy- $\alpha$-D-glucopyranosyl phosphate series to their corresponding acetamidodeoxy analogs provides insight pertaining to the ability of the $\mathrm{E}_{p}$ active site to accommodate additional steric bulk.
Of the aminodeoxy- $\alpha$-D-glucopyranosyl phosphates examined, only two, 2-amino-2-deoxy- $\alpha$-D-glucopyranosyl phosphate (107) (FIG. 4) and 2-acetamido-2-deoxy- $\alpha$-Dglucopyranosyl phosphate (108) (FIG. 4), were commercially available. The syntheses of the remaining analogs diverged from the key intermediates Ethyl 6-azide-2,3',4-tri-O-benzyl-6-deoxy-1-thio- $\beta$-D-glucopyranoside (89), Ethyl 4-azide-2,3,6-tri-O-benzyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (94) and Ethyl 3-azide-2,4,6-tri-O-benzyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (100) (FIG. 3 (b)).

Thus, the present invention includes a method of making aminodeoxy- $\alpha$-D-glucopyranosyl phosphates comprising converting an intermediate selected from the group consisting of ethyl 6 -azide-2,3,4-tri-O-benzyl-6-deoxy-1-thio- $\beta$-Dglucopyranoside (89), ethyl 4-azide-2,3,6 tri-O-benzyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (94), and ethyl 3-azide-

2,4,6-tri-O-benzyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (100) to a corresponding amide.

Ethyl 1-thio- $\beta$-D-pyranosides (89) and (100) derived from previously reported glycosides (FIG. $\mathbf{3}(b)(87)$ ) (see V. Maunier, P. Boullanger, D. Lafont, Y. Chevalier, Carbohydr. Res. 1997, 299, 49-57) and FIG. 3 (b) (98) (W. A. Greenberg, E. S. Priestley, P. S. Sears, P. B. Alper, C. Rosenbohm, M. Hendrix, S.-C. Hung, C.-H. Wong, J. Am. Chem. Soc. 1999, 121, 6527-6541), respectively), while (94) was synthesized, from the previously reported compound (93) (FIG. 3 (b)) (P. J. Garegg, I. Kvarnstrom, A. Niklasson, G. Niklasson, S. C. T. Svensson, J. Carbohydr. Chem. 1993, 12, 933-953) in a manner similar to that of the deoxy- $\alpha$-D-glucopyranosyl phosphate syntheses described herein. Specifically, this strategy invoked a protection scheme to selectively expose the position of substitution followed by activation (via TsCl or $\mathrm{Tf}_{2} \mathrm{O}$ ) and $\mathrm{SN}^{2}$ displacement by sodium azide. From the divergent point (FIG. 3 (89), (94) and (100)), an efficient azide selective $\mathrm{SnCl}_{2}$ reduction followed by acetylation gave the desired ethyl 1-thio- $\beta$-D-pyranoside precursors (90), (95), and (101). Finally, the subsequent phosphorylation of FIG. 3(b) (89), (90), (94), (95), (100), and (101) was accomplished by reaction with dibenzyl phosphate as previously described where the culminating reductive deprotection also led to the conversion of the FIG. 3 (b) (89), (94), and (100) azides to the desired amines. As an aminodideoxy sugar representative, 4 -amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl phosphate (FIG. $3(b)(102)$ ) was also synthesized from peracetylated D-fucose (FIG. $\mathbf{3}(b)$ (103)) as illustrated in FIG. 3 using a similar strategy.

To evaluate the synthetic utility of thymidylyl-transferase, purified $\mathrm{E}_{p}, \alpha$-D-glucopyranosyl phosphate, $\mathrm{Mg}^{+2}$, NTP and inorganic pyrophosphatase were incubated at $37^{\circ} \mathrm{C}$. for 30 min and the extent of product formation determined by HPLC. The inorganic pyrophosphatase was included to drive the reaction forward. A reaction containing 2.5 mM NTP, 5.0 mM sugar phosphate, $5.5 \mathrm{mM} \mathrm{MgCl} \mathrm{I}_{2}$ and 10 U inorganic pyrophosphatase in a total volume of $50 \mu \mathrm{~L} 50$ mM potassium phosphate buffer, pH 7.5 at $37^{\circ} \mathrm{C}$. was initiated by the addition of $3.52 \mathrm{UE}_{p}(1 \mathrm{U}=$ the amount of protein needed to produce 1 mol TDP-D-glucose $\mathrm{min}^{-1}$ ). The reaction was incubated with slow agitation for 30 min at $37^{\circ} \mathrm{C}$., quenched with $\mathrm{MeOH}(50 \mu \mathrm{~L})$, centrifuged ( 5 min , $14,000 \times \mathrm{g}$ ) and the supernatant was stored at $-20^{\circ} \mathrm{C}$. until analysis by HPLC. Samples ( $30 \mu \mathrm{~L}$ ) were resolved on a Sphereclone 5 u SAX column ( $150 \times 4.6 \mathrm{~mm}$ ) fitted with a SecurityGuard ${ }^{\text {TM }}$ cartridge (Phenomenex; Torrance, Calif.) using a linear gradient ( $50-200 \mathrm{mM}$ potassium phosphate buffer, $\mathrm{pH} 5.0,1.5 \mathrm{~mL} \min ^{-1}, \mathrm{~A}_{275} \mathrm{~nm}$ ).

The results of these assays are illustrated in FIG. 4. For each assay, confirmation of product formation was based upon high resolution mass spectroscopy of HPLC-isolated products and, also in some cases, HPLC co-elution with commercially available standards. (Allosteric activation is common for the nucleotidylyltransferase family (for examples see: M. X. Wu, J. Preiss, Arch. Biochem. Biophys. 1998, 358, 182-188; and D. A. Bulik, P. van Ophem, J. M. Manning, Z. Shen, D. S. Newburg, E. L. Jarroll, J. Biol. Chem. 2000, 275, 14722-14728) although data is not yet available pertaining to the allosteric effectors of $\mathrm{E}_{p}$. As controls, no product formation was observed in the absence of $\mathrm{E}_{p}$, glucopyranosyl phosphate, $\mathrm{Mg}^{+2}$, or NTP.

The nucleotide sugars (109)-(126) set forth above are examples of nucleotide sugars of the present invention, which may be produced in accordance with the methods described herein, and in particular the reactions diagramed in FIG. 4. A comparison of the aminodeoxy- $\alpha$-D-glucopy-
ranosyl phosphate/dTTP assay results (FIG. 4 (85), (91), (96), and (107)) to the $\mathrm{E}_{p}$ native reaction (FIG. 4, (2)/dTTP) reveals that amino substitution has little or no effect on product formation, and, with the exception of compound (85) (FIG. 4), a similar phenomenon is observed in presence of UTP.

The divergence of compound (85) from this trend is consistent with UTP-dependent $\mathrm{E}_{p}$ "adverse cooperativity" in the presence of certain hexopyranosyl phosphates, as described herein. This phenomenon is perhaps attributable to allosteric activation by dTTP. Evaluation of the acetami-dodeoxy- $\alpha$-D-glucopyranosyl phosphate/dTTP assays (FIG. 4 compounds (86), (92), (97) and (108)), in comparison to their non-acetylated counterparts (FIG. 4 (85), (91), (96) and (107)), reveal that a bulky N -acetyl group at C-2 or C-3 (FIG. 4 (97) and (108)) is well-tolerated while the identical C-4 or C-6 substitution (FIG. 4 (92)) and (86)) results in less activity. Given that these effects most likely derive from unfavored steric interactions, it follows that the $\mathrm{E}_{p}$ active site is able to accommodate additional C-2/C-3 bulk while sterics limit the allowed C-4/C-6 substitutions.
Surprisingly, product formation from FIG. 4 (86)/UTP was markedly increased (8-fold) in comparison to (86)/ dTTP. This is the first example to contradict the typical adverse UTP-dependent effect upon yields observed, as illustrated by FIG. 4 compounds (85) and (97). Finally, a comparison of aminodideoxy-a-D-glucopyranosyl phosphate (FIG. 4 (102)(The product of this reaction, thymidine 5'-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate), is an important critical intermediate in the formation of the calicheamicin aryltetrasaccharide to that of (FIG. 4 (96)) reveals C-6 deoxygenation does not effect dTTP-dependent $\mathrm{E}_{p}$ catalysis but greatly diminishes UTP-dependent conversion. However, given independent deoxygenation at $\mathrm{C}-6$ or amino substitution at C-4 (FIG. 4 (91)) each has no effect on product yield, data from independent substitutions may not be reliable in predicting the effects of multiple substitutions on product yield.
FIG. 4 illustrates the utiltity of $\mathrm{E}_{p}$ as a catalyst/reagent to simplify the synthesis of useful nucleotide sugar pools of the nine substrate analogs tested, all provide product with dTTP and with dUTP under the conditions described. Further, seven with dTTP and four with UTP provide appreciable product ( $>50 \%$ conversion) under the conditions described.
Nucleotide sugars produced via the exploitation of the promiscuity of $\mathrm{E}_{p}$ include, but are not limited to, compounds (58)-(81), (109)-(126) and those set forth in FIGS. 20(a) and $\mathbf{2 0}(b)$.

## Nucleotidylyltransferases

## Structure-Based Engineering of $\mathrm{E}_{p}$

The present inventors have determined the first three dimensional structures of this unique enzyme in complex with the product UDP- $\alpha$-D-glucose (UDP-Glc) and with the substrate dTTP at $1.9 \AA$ and $2.0 \AA$ resolution, respectively. A three dimensional structure of $\mathrm{E}_{p}$ is depicted in FIG. 6. This discovery has facilitated the elucidation of the molecular details of $\mathrm{E}_{p}$ substrate recognition. The structures reveal the catalytic mechanism of thymidylyltransferases, which is further supported by new kinetic data. The present inventors have also used structure-based engineering or mutations of $\mathrm{E}_{p}$ to produce modified enzymes. These inventive enzymes are capable of utilizing "unnatural" sugars previously not accepted by wild-type $\mathrm{E}_{p}$.

Structure Determination
The $\mathrm{E}_{p}$-UDP-Gle structure was determined using seleno-methionine-labeled protein crystals and a data-set collected at a wavelength corresponding to the selenium absorption peak. A representative portion of the experimental electron density is shown in FIG. 6. The $\mathrm{E}_{p}$-dTTP structure was subsequently determined by molecular replacement using the $\mathrm{E}_{p}$-UDP-Gle monomer structure as a search model.

Overview of the $\mathrm{E}_{p}$ structure
The structure of the biologically active $\mathrm{E}_{p}$ tetramer is illustrated in FIG. 7. The present model is refined at $2.0 \AA$ resolution to an R factor of $18.3 \%$ with restrained temperature factors and good stereochemistry. FIG. $7 a$ shows $\mathrm{E}_{p}$ in complex with UDP-Gle and FIG. $7 b$ displays the $\mathrm{E}_{p}$-dTTP complex. The two tetrameric structures are very similar with r.m.s.d. for equivalent $C_{\alpha}$ positions $=1.0 \AA$. The enzyme has overall dimensions of about $80 \AA \times 80 \AA \times 60 \AA$ and a compact tertiary structure generated by four monomers packing tightly against each other along two two-fold axes of symmetry drawn on the leftmost panel of FIG. 7. The overall surface area buried during tetramer formation is approximately $10,000 \AA$, equivalent to the surface of one monomer. The monomer interactions are dominated by helix-helix packing of the four large helices in the center of the $\mathrm{E}_{p}$ tetramer and surrounding extensive loop-loop interactions involving multiple van der Waals contacts, hydrogen bonds, and salt bridges. The active site pockets of the monomers are located close to, but not overlapping with, the monomer interface.

The $\mathrm{E}_{p}$ monomer (FIG. 8 ) is a two-domain molecule with overall size of approximately $50 \AA \times 50 \AA \times 50 \AA$. The domain containing the active site is dominated by a large sevenstranded mixed central $\beta$-sheet, with an unusual left-handed twist, packed against three $\alpha$-helices on one side and another three $\alpha$-helices on the other. Its extensive hydrophobic core contains no cavities and is dominated by aromatic side chains.

This domain has overall resemblance, including the location of the active site in a large pocket on the top of the $\beta$-sheet, to other nucleotide binding proteins (see Vrielink, A., Ruger, W., Dreissen, H. P. C. \& Freemont, P. S. Crystal Structure of the DNA Modifying Enzyme $\beta$-Glucosyltransferase in the Presence and Absence of the Substrate Uridine Diphosphoglucose, EMBO J. 13, 3413-3422 (1994); Charnock, S. J. \& Davies, G. J. Structure of the Nucleotide-Diphospho-Sugar Transferase, SpsA from Bacillus subtilis, in Native and Nucleotide-Complexed Forms. Biochem. 38, 6380-6385 (1999); Gastinel, L. N. Cambillau, C. \& Bourne, Y. Crystal Structures of the Bovine 4Galatosyltransferase Catalytic Domain and Its Complex with Uridine Diphosphogalactose, EMBO J. 18, 3546-3557 (1999); Ha, S., Walker, D., Shi, Y. \& Walker, S. The 1.9 A Crystal Structure of Escherichia coli MurG, a Membrane-Associated Glycosyltransferase Involved in Peptidoglycan Biosynthesis. Prot. Sci. 9, 1045-1052 (2000); and Brown, K., Pompeo, F., Dixon, S., Mengin-Lecreulx, D., Cambillau, C. \& Bourne, Y. Crystal Structure of the Bifunctional N-Acetylglucosamine 1-phosphate uridylyltransferase from Escherichia coli: A Paradigm for the Related Pyrophosphorylase Superfamily. EMBO J. 18, 4096-4107 (1999)), containing the $\alpha / \beta$ opensheet Rossmann fold. (Rossmann, M. G., et al., Evolutionary and structural relationship among dehydrogenases, in The Enzymes, I. P. D. Boyyer, Editor. Academic Press: New York. p. 61-102 (1975); and Branden, C. \& Tooze, J. Introduction to Protein Structure. New York: Garlan Publishing, Inc. (1991).) The second $\mathrm{E}_{p}$ domain (represented by yellow in (FIG. 8), packing tightly to the side of the
active-site domain, contains four $\alpha$-helices and a twostranded $\beta$-sheet and is involved in the inter-monomer packing interactions forming the $\mathrm{E}_{p}$ tetramer.
Structural Homology to Glycosyltransferases and Uridylyltransferases

The present inventors' elucidation of the structure of $\mathrm{E}_{p}$ represents the first such elucidation of a structure of a thymidylyltransferase. Comparison of the structure with the contents of the FSSP database, (Holm, L. \& Sander, C. Touring Protein Fold Space with Dali/FSSP. Nucleic Acids Res., 26, 316-319 (1998)) revealed that the overall $\mathrm{E}_{p}$ fold is different from other previously determined structures. The closest structural homologs of $\mathrm{E}_{p}$ are the SpsA glycosyltransferase from Bacillus subtilis and the functionally related E. coli enzyme GlmU. GlmU is a bifunctional enzyme containing acetyltransferase and uridylyltransferase domains, respectively. FIG. 8 illustrates these three proteins, highlighting the structurally similar regions. As expected, the structural homology lies within the nucleotide-sugar binding domains. The active sites of the enzymes are located in pockets on top of the large $\beta$-sheet, although the precise positioning differs between glycosyltransferases and nucleotidyltransferases and involves secondary structure elements, which are not structurally equivalent. The threedimensional structures of two other sugar-phosphate transferring enzymes, $\alpha$-D-galactopyranosyl phosphate (Gal-1-P) uridylyltransferase and kanamycin nucleotidyltransferase are known, but do not activate sugars and both differ structurally and functionally from $\mathrm{E}_{p}$.

Active Site Interactions: Substrate and Product Binding
FIG. 8 shows two $90^{\circ}$ views of the $\mathrm{E}_{p}$ active site pocket. In both of the $\mathrm{E}_{p}$-dTTP and $\mathrm{E}_{p}$-UDP-Glc structures, the experimental electron density for the dTTP and UDP-Glc is excellent. $\mathrm{E}_{p}$ utilizes both dTTP and UTP, but not CTP, and FIG. $9 a$ illustrates the structural basis for this substrate specificity. Specifically, the exocyclic N3 and O4 ring atoms of both dT and U are hydrogen bonded to Gln83. In addition, O 4 hydrogen bonds to the main chain N of Gly 88 while O 2 is bound to the main chain N of Gly11. Finally, the 3'-hydroxyl group of the pentose forms a hydrogen bond with Gln27. The substrate dTTP also makes extensive van der Waals contacts with Leu9, Leu89 and Leu109, which form a hydrophobic bed for the nucleoside, and position 5 of the pyrimidine base is far enough from any protein atom to allow an easy fit for the methyl group of dT in the pocket. The phosphate groups of dTTP lie in an extended position firmly held in place by multiple interactions with the main chain nitrogen atoms of Ser13, Gly14, and Thr15, and with the catalytically important $\mathrm{Mg}^{+2}$ (see below). The $\gamma$-phosphate also makes a hydrogen bond with Thr 15 and both the $\alpha$ - and $\gamma$-phosphates bind Arg16. The nearby $\operatorname{Arg}$ 145, Lys163, and Arg 195 create a favorable electrostatic environment, but do not interact directly with dTTP.
The $\mathrm{E}_{p}$ product, UDP-Glc, is bound along the diameter of the surface pocket. The nucleoside sits in the active site in virtually the same conformation as the substrate dTTP, with the addition of a hydrogen bond between the 2 '-hydroxyl of the ribose and the main chain O of Gly11. In the glucosebinding pocket, as illustrated on (NAT) FIG. 5 $b$, the hydroxyl groups $\mathrm{O} 2, \mathrm{O} 3$ and O 4 of the glucose moiety are directly hydrogen-bonded to protein residues, while O6 is bound to $\mathrm{E}_{p}$ via a water molecule. Gln 162 binds both O 2 and O 3 , the main chain N of Gly 147 binds both O 3 and O 4 , and the main chain O of Val173 binds O4. The side chain of Thr201 is also close to both O 2 and O 3 . In addition, four well-ordered water molecules, shown on FIG. $5 b$, bridge $\mathrm{E}_{p}$ and the glucose moiety. Leu109, Leu89, and Ile200 make
van der Waals contacts with the underside of the hexose ring and Trp224 and Tyr146 close the glucose binding pocket which would prevent bulkier sugars, for example disaccharides, from binding. In the $\mathrm{E}_{p}$-UDP-Gle structure, the phosphate groups are now twisted away from their straight conformation in dTTP so that they can connect the nucleoside with the hexose - see also FIG. 10. The phosphates are also much more solvent exposed and do not interact with any main chain atoms, but instead, with the side chains of the positively charged Arg16, Lys163, and Arg 195, as well as with water molecules.

Divalent Cation Binding Site
The activity of nucleotidyltransferases is strictly dependent on a divalent cation involved in catalysis via stabilizing the leaving $\mathrm{PP}_{i}$ (See Kornfeld, S. \& Glaser, L. J. Biol. Chem. 236, 1791-1794 (1961)). Crystallographic data generated by the present inventors allow for the identification of the location of this cofactor and, in this region, $\mathrm{a} \mathrm{Mg}^{+2}$ electron density feature, larger than a water molecule and chemically ideal for a metal location, was modeled. Indeed the $\mathrm{Mg}^{+2}$ is $2.6 \AA$ away from the $\beta$-phosphate oxygen and is also coordinated by the side chain of Gln 26 , main chain O of Gly11, main chain nitrogens of Ser13 and Gly14, and a water molecule. This region (particularly Gly10 to Gly15) is mostly disordered in the $\mathrm{E}_{p}$-UDP-Gle structure, indicating that the $\mathrm{Mg}^{+2}$, in addition to electrostatically stabilizing the leaving group, also plays a structural role in folding the substrate-binding region of $\mathrm{E}_{p}$ around itself to fix the NTP at an optimal position for the catalytic event.

A Secondary dTTP-Binding Site and Possible Allosteric Control

The structure of $\mathrm{E}_{p}$-dTTP, disclosed herein (FIG. 7), indicates that the $\mathrm{E}_{p}$ tetramer binds eight molecules of dTTP-four in the active site pockets on top of the $\beta$-sheet, and four in an auxiliary sites at the interface between the subunits. FIG. 9 C shows a close-up of a dTTP molecule in the auxiliary site. There are fewer contacts between $\mathrm{E}_{p}$ and dTTP here than in the active site. As a result, CTP, which is not accepted by $\mathrm{E}_{p}$, could easily fit in the auxiliary site. The dTTP base and the ribose in the secondary site interact with one $\mathrm{E}_{p}$ monomer, including hydrogen bonds to the main chain N of Gly116 and Ser152, and van der Waals contacts with Leu46, Tyr115 and Ile249. The dTTP phosphates, on the other hand, interact primarily with residues of an adjacent $\mathrm{E}_{p}$ monomer, including Arg220 and Gly221.

Several other nucleotidylyltransferases are under allosteric control by metabolites distinct from their products or substrates. The presence of an auxilary site strongly suggests that $\mathrm{E}_{p}$ is also under allosteric control. Indeed, binding of an effector in this hydrophobic pocket at the monomer interface could alter the relative orientation of the $\mathrm{E}_{p}$ monomers, thus altering the conformation or the access to the active site. Given the non-specific nature of the observed interactions, and the fact that nucleotidylyl-transferase effectors are generally not substrates, the putative $\mathrm{E}_{p}$ allosteric effector is most likely not dTTP.

The $\mathrm{E}_{p}$ Catalytic Mechanism
Before the present experiments, two conflicting hypotheses for nucleotidylyltransferase catalysis were suggested. Lindquist and co-workers proposed a ping-pong bi-bi mechanism for $\mathrm{E}_{p}$, the necessary prerequisite for which is the formation of an enzyme-substrate covalent intermediate. (See Lindquist, L., Kaiser, R., Reeves, P. R. and Lindberg, A. A., Purification, Characterization and HPLC Assay of Salmonella Glucose-1-Phosphate Thymidylylphospherase from the cloned rfbA Gene, Eur. J. Biochem, 211, 763-770 (1993). Alternatively, in a related enzyme, Frey and co-
workers had previously presented evidence for inverted geometry about the $\alpha$-phosphate upon attack by Gle-1-P which led the authors to propose a single displacement mechanism for the entire nucleotidylytransferase family. (Sheu, K.-F. R., Richard, J. P. \& Frey, P. A. Stereochemical Courses of Nucleotidyl-transferase and Phosphotransferase Action. Uridine Diphosphate Glucose Pyrophosphorylase, Galactose-1-Phosphate Uridylyltransferase, Adenylate Kinase, and Nucleoside Diphosphate Kinase Biochem. 18, 5548-5556 (1979)).
In the present invention, a comparison of the topology of the $\mathrm{E}_{p}$-bound substrate ( dTTP ) to the $\mathrm{E}_{p}$-bound product (UDP-Glc) (FIG. 10a) suggests that the Glc-1-P oxygen nucleophile must directly attack the $\alpha$-phosphate of dTTP. In this reaction, the formation of a phosphodiester bond on one side of the $\alpha$-phosphate atom is simultaneous with the breaking of a phosphodiester bond on the opposing face (to give $\mathrm{PP}_{i}$ as the leaving group). Consistent with an $\mathrm{S}_{N} 2$ type mechanism, the bond undergoing formation in the structure disclosed herein is "in-line" or 180 degrees away from the leaving group and thus, the two oxygen atoms bonded to the phosphate invert their geometry upon bond formation. Although the $\alpha$-phosphate here is not chiral, both the reactant (substrate) and product topologies, as well as the architecture of the active site, clearly suggest that an inversion has occurred.
The present inventors evaluated the $\mathrm{E}_{p}$ steady state kinetics in order to further probe the enzymatic mechanism. The intersecting patterns observed in FIG. $\mathbf{1 0} b$ and FIG. $\mathbf{1 0} d$ are consistent with the structural data in support of a single displacement mechanism rather than the previously postulated ping-pong bi bi (double displacement) mechanism. Finally, the $\mathrm{E}_{p}$-dTTP crystals were soaked in a solution containing 2 mM of either Glc-1-P or D-Glc, in addition to the 2 mM dTTP and $\mathrm{Mg}^{+2}$ already present. The glucose soaks did not significantly alter the electron density in the active site. On the other hand, Glc-1-P soaks quickly caused deterioration of the crystal diffraction quality. Data collected with crystals soaked for 30 min revealed electron density in the active site that was an average of the density in our $\mathrm{E}_{p}$-dTTP and EP-UDP-Glc crystals. Therefore, the phosphate of Glc-1-P is necessary for binding by $\mathrm{E}_{p}$, and the lack of any observable $\mathrm{E}_{p}$-UMP covalent intermediate in these experiments further supports the single displacement mechanism.

## Active-Site Engineering

Two sugar phosphates not utilized by wild-type $\mathrm{E}_{p}$ and two additional sugar phosphates poorly utilized by the enzyme were selected to test rational engineering of $\mathrm{E}_{p}$ substrate promiscuity. Specifically, 6-acetamido-6-deoxy- $\alpha$ -D-glucopyranosyl phosphate (FIG. 11 (86)) is not wellaccepted and $\alpha$-D-glucopyranuronic acid 1-(dihydrogen phosphate) (FIG. 11 (127)) is not accepted by $\mathrm{E}_{p}$, and 2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl phosphate (FIG. 11 (108)) and $\alpha$-D-allopyranosyl phosphate (FIG. 11 (38)) lead to poor conversion. Because a representative "unnatural" sugar phosphate was believed to be efficiently converted only by wild-type $\mathrm{E}_{p}, 4$-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl phosphate (FIG. 11 (102)) was also tested with all mutants. Sugar phosphates not utilized by a wild-type enzyme, e.g., $\mathrm{E}_{p}$, and sugar phosphates poorly utilized by the enzyme may be referred to herein as "unnatural" with respect to that enzyme.
Structure-based modeling reveals steric and/or electrostatic infringements may be the limiting factor in the conversion of "unnatural" sugar phosphates. In an attempt to relieve these constraints, three mutants were constructed. In
particular, a Thr201Ala mutant and Glu162Asp were believed to decrease the steric interference at the sugar positions C-2 and/or C-3 for compounds (108) and (38), while a Trp224His substitution was designed to decrease steric constraints at $\mathrm{C}-6$ of the substrate (e.g. compound (86)). Furthermore, the glucuronic acid derivative (127) offers the unique challenge of engineering electrostatic balance and the $\operatorname{Trp} 224$ His variant was predicted to provide a partial positive charge to assist in (127)-binding in addition to steric relief. Alternatively, Asp-111 ( $6 \AA$ from the substrate C-6-OH) was predicted to result in the electrostatic repulsion of substrates containing a negative charge at the C-6 of the sugar phosphate. Thus, an additional mutant (Asp111Asn) was constructed to eliminate this effect.

As a rapid means to assay the entire pool of the four newly constructed mutants, the mutants were combined and the mixture directly tested for ability to convert compounds (2), (108), (89), (102), (38), and (127). FIG. 11 shows that the mutant pool was able to turn over all but one (5) of the sugar phosphates tested. Those substrates turned over include (86) and (127), the two sugar phosphates not accepted or poorly accepted by wild-type $\mathrm{E}_{p}$.

The following nucleotide sugars were produced by the reactions of FIG. 11: (117) Thymidine 5'-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 604.0945$; found $\mathrm{m} / \mathrm{z} 604.0953$ $(\mathrm{M}+\mathrm{H})$ ); (118) Uridine $5^{\prime}$-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2} 606.0737$; found $\mathrm{m} / \mathrm{z} 606.0732(\mathrm{M}+\mathrm{H})$ ); (130) Thymidine $5^{\prime}$-( $\alpha$-D-glucopyran- 6 -uronic acid diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{O}_{17} \mathrm{~N}_{2} \mathrm{P}_{2}$ 577.0472; found $\mathrm{m} / \mathrm{z} 577.0465(\mathrm{M}+\mathrm{H})$ ); (131) Uridine $5^{\prime}$-( $\alpha$-D-glu-copyran-6-uronic acid diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{O}_{18} \mathrm{~N}_{2} \mathrm{P}_{2} 579.2774$; found $\mathrm{m} / \mathrm{z} 579.2775(\mathrm{M}+\mathrm{H})$ ); (123) Thymidine 5 '-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 604.0945; found $\mathrm{m} / \mathrm{z} 604.0951(\mathrm{M}+\mathrm{H})$ ), (124) Uridine $5^{\prime}-$ (2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2}$ 606.0737; found $\mathrm{m} / \mathrm{z} 606.0738(\mathrm{M}+\mathrm{H})$ ); (125) Thymidine $5^{\prime}$-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{14} \mathrm{~N}_{3} \mathrm{P}_{2} 546.0889$; found $\mathrm{m} / \mathrm{z} 546.0895$ $(\mathrm{M}+\mathrm{H})$ ); (126) Uridine $5^{\prime}$-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2} 548.0682$; found $\mathrm{m} / \mathrm{z} 548.0673(\mathrm{M}+\mathrm{H})$ ) although data is not depicted for all products.

A deconvolution of the mutant pool, by individual mutant analysis, revealed the Trp224His mutation as responsible for converting both (86) and (127). Thr102Ala, on the other hand, was responsible for the 2 -fold increase in the conversion of (108). The remaining two mutants (Asp41Asn and Glu162Asp, not shown in FIG. 11) failed to enhance conversion, over wild-type $\mathrm{E}_{p}$, of any of the tested putative substrates. Yet, cumulatively, this small set of directed mutants was able to successfully turn over three of four targeted "unnatural" substrates. Of particular interest is the Trp224His mutant, which displays enhanced promiscuity without affecting wild-type traits. This $\mathrm{E}_{p}$ variant will serve as an excellent foundation for second generation double mutants. Finally, the demonstrated ability to test mutant sets via pooling will rapidly expedite the development of this methodology.

In $\mathrm{E}_{p}$, amino acids that make contacts or near contacts to the sugar in the active site include V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177. These amino acids may be mutated in order to alter the specificity of $\mathrm{E}_{p}$, as demon-
strated herein. Any mutation that alters the specificity may be made and tested, as taught herein, to determine its effect on the specificity of $\mathrm{E}_{p}$ for its substrate and the efficiency of conversion of substrate to product.
Thus, the present invention includes a nucleotidylyltransferase mutated at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177. Preferably the nucleotidylyltransferase is $\mathrm{E}_{p}$.
An embodiment of the present invention is directed to a nucleotidylyltransferase mutated such that it is capable of having a different substrate specificity than a non-mutated nucleotidylyltransferase. Examples include nucleotidylyltransferases having a substrate specificity for GTP, ATP, CTP, TTP or UTP. Further provided are methods of altering nucleotidylyltransferase substrate specificity comprising mutating the nucleotidylyltransferase at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177. Preferably the nucleoti-dylyl-transferase is $\mathrm{E}_{p}$. Also provided are nucleotidylyltransferases, so modified.
The present invention also includes purine or pyrimidine triphosphate type nucleotidylyltransferases set forth in FIG. 19, and purine or pyrimidine triphosphate type nucleotidy-lyl-transferases mutated at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177.
Further, sequence comparison reveals that many nucleotidyltransferases bear high degrees of sequence identity to $\mathrm{E}_{p}$. The substrate specificity of such enzymes may be altered, using methods described herein for $\mathrm{E}_{P}$, at amino acids that make contacts or near contacts to the sugar in the active site. These amino acids may be located via sequence comparison with $\mathrm{E}_{p}$-the contact sites will often be those at the same relative position as V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177 in $\mathrm{E}_{p}$. FIG. 19 provides a list of nucleotidyltransferases that bear high degrees of sequence identity to $\mathrm{E}_{p}$. FIGS. 12 to 18 show the alignment of the $\mathrm{E}_{p}$ sequence and those of other representative nucleotidyltransferases. Other nucleotidylyltransferases may also be mutated at one or more amino acids in their active sites, divalent cation binding sites and/or auxiliary sites.

Methods for mutating proteins are well-known in the art. For the present invention, it is preferable to perform sitedirected mutagenesis on the nucleotide encoding the enzyme of interest. In this manner, and using the guidance provided herein, one of skill in the art can make mutations to the codons encoding the amino acids at the sites of the enzyme desired to be changed. Likewise, the use of site directed mutagenesis allows the worker to ensure that each codon desired to be changed is changed to encode a different amino acid from the wild-type molecule. In contrast, the use of random mutagenesis might result in mutated codons encoding the same amino acids as the wild-type codons, due to the degeneracy of the genetic code. Methods for manipulation and mutation of nucleotides, as well as for the expression of recombinant peptides are well known in the art, as exemplified by Sambrook et al., Molecular Cloning: A Laboratory Manual, Second Edition, Cold Spring Harbor Laboratory Press, 1989

References for nucleotidyltransferases in these Figures, include: AAB31755-Glc-1-P Cytitdylyltransferase from Yersinia pseudotuberculosis. See Thorson J S, Lo S F, Ploux

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According to one embodiment of the present invention mutations at amino acid L89T were tested. Such mutations increased the yield of allo-, altro-, talo-, gulo- and idoderivatives. Wild-type and/or this mutant also led to the production of the new nucleotide sugar compounds set forth in FIGS. $20(a)$ and (b). Methods of production of such compounds and of the mutant nucleotidylyltransferase are as set forth herein with regard to other compounds and mutant nucleotidylyl-transferase. In particular, the compounds may be produced by synthesizing the corresponding sugar phosphate followed by $\mathrm{E}_{p}$ catalyzed conversion of the sugar phosphate to the new products.

The present invention includes the nucleotide sugars of FIGS. $\mathbf{2 0}(a)$ and $\mathbf{2 0}(b)$, their corresponding sugar phosphates and nucleotidylyltransferases mutated at L89T, which may convert such sugar phosphates to a nucleotide sugar.

Glycorandomization of Natural Product-Based Metabolites

The wild-type glycosyltransferases in secondary metabolism show significant flexibility with respect to their NDPsugar donors. Coupled with the presented $\mathrm{E}_{p}$-catalyzed production of NDP-sugar donor libraries and the appropriate aglycon, a diverse library of "glycorandomized" structures based upon a particular natural product scaffold can be rapidly generated.
Accordingly, the present invention is also directed to nucleotide sugar libraries including two or more of the nucleotide sugars described herein. More preferably the nucleotide sugars are nucleotide sugars made by the methods described herein, preferably using a natural or mutated nucleotidylyltransferase as a catalyst. The present invention also includes in vitro glycorandomization using such sugar libraries.
Exploiting the promiscuity of wild type $\mathrm{E}_{p}$ and utilizing the ability conferred by the methods of the present invention to rationally engineer variants able to utilize sugar phosphates not previously usable, libraries of previously unavailable nucleotide sugars may be generated. The ability to generate a set of $\mathrm{E}_{p}$ variants provides the subsequent ability to generate, in a simple one pot reaction, diverse libraries of NDP-sugars. Both sugars that were unknown prior to the present invention and those that could not be synthesized in vitro prior to the present invention may be synthesized using the methods of the present invention. Such libraries of NDP-sugars, in conjunction with downstream glycosyltransferases, form the basis for the in vitro manipulation of metabolite sugar ligands in a combinatorial fashion (or "glycorandomization").
For example, a diverse library of "glycorandomized" structures based upon the known antitumor agent mithramycin (FIG. 5) may be constructed. Beginning with a small pool of sugar phosphates, e.g., 25 different sugar phosphates, the anticipated library size would be the result of combining 25 different sugars at 5 different positions on mithramycin to give $25^{5}$, or $>9.7$ million, distinct mithramycin-based variants. Furthermore, as alterations of the carbohydrate ligands of biologically active metabolites can lead to drastically different pharmacological and/or biological properties, this approach has significant potential for drug discovery. As an example of alterations of the carbohydrate ligands of biologically active metabolites producing dramatically different pharmacological properties, the structure of 4-epidoxorubicin differs from that of doxorubicin, which is more toxic, only in carbohydrate ligands. As an example of alterations of the carbohydrate ligands of biologically active metabolites producing dramatically different biological properties, the structure of erythromycin, an antibiotic, differs from that of megalomicin, a compound with antiviral and antiparasitic activity, only in carbohydrate ligands.

An embodiment of the invention includes incubating a glycotransferase with one or more of the sugars of a nucleotide sugar library according to the present invention, and a molecule capable of being glycosylated.

The present inventors have discovered that $\mathrm{E}_{p}$ is pliable in terms of its substrate specificity. The present inventors have also discovered the three dimensional structure of $\mathrm{E}_{p}$ and the molecular details of $\mathrm{E}_{p}$ substrate recognition. The present inventors have invented methods of engineering or modifying $\mathrm{E}_{p}$ to vary its specificity in a directed manner, conferring the ability to rationally engineer variants able to utilize sugar phosphates not previously usable. The present inventors have also invented a method for the synthesis of desired nucleotide sugars using both natural and engineered $\mathrm{E}_{p}$. Thus, the present invention will likely broadly impact efforts to understand and exploit the biosynthesis of glycosylated
bioactive natural products, many of which are pharmacologically useful. The ability conferred by the methods of the present invention to alter nucleotidylyltransferase specificity by design allows the creation of promiscuous in vitro systems, which could provide large and diverse libraries of potentially new bioactive metabolites.

The present invention will now be illustrated by the following examples, which show how certain specific representative embodiments of the compounds and methods of the present invention, the compounds, intermediates, process steps, and the like being understood as examples that are intended to be illustrative only. In particular, the invention is not intended to be limited to the conditions, order of the steps and the like specifically recited herein. Rather, the Examples are intended to be illustrative only.

## EXAMPLES

## General Methods.

Infrared spectra were recorded on a Perkin Elmer 1600 series FTIR spectrophotometer. ${ }^{1} \mathrm{H}$ NMR spectra were obtained on a Bruker AMX $400(400 \mathrm{MHz})$ and are reported in parts per million ( $\delta$ ) relative to either tetramethylsilane ( 0.00 ppm ) or $\mathrm{CDCl}_{3}(7.25 \mathrm{ppm})$ for spectra run in $\mathrm{CDCl}_{3}$ and $\mathrm{D}_{2} \mathrm{O}(4.82 \mathrm{ppm})$ or $\mathrm{CD}_{3} \mathrm{OD}$ ( 3.35 ppm ) for spectra run in $\mathrm{D}_{2} \mathrm{O}$. Coupling constants (J) are reported in hertz. ${ }^{13} \mathrm{C}$ NMR are reported in $\delta$ relative to $\mathrm{CDCl}_{3}(77.00 \mathrm{ppm})$ or $\mathrm{CD}_{3} \mathrm{OD}(49.05 \mathrm{ppm})$ as an internal reference and ${ }^{31} \mathrm{P}$ NMR spectra are reported in $\delta$ relative to $\mathrm{H}_{3} \mathrm{PO}_{4}(0.00 \mathrm{ppm}$ in $\mathrm{D}_{2} \mathrm{O}$ ). Routine mass spectra were recorded on a PE SCIEX API 100 LC/MS mass spectrometer and HRMS was accomplished by the University of California, Riverside Mass Spectrometry Facility. Optical rotations were recorded on a Jasco DIP-370 polarimeter using a 1.0 or 0.5 dm cell at the room temperature ( $25^{\circ} \mathrm{C}$.) and the reported concentrations. Melting points were measured with Electrothermal 1A-9100 digital melting point instrument. Chemicals used were reagent grade and used as supplied except where noted. 'Analytical TLC was performed on either E. Merck silica gel $60 \mathrm{~F}_{254}$ plates ( 0.25 mm ) or Whatman AL Sil G/UV silica gel 60 plates. Compounds were visualized by spraying I/KI/ $\mathrm{H}_{2} \mathrm{SO}_{4}$ or by dipping the plates in a cerium sulfate-ammonium molybdate solution followed by heating. Liquid column chromatography was performed using forced flow of the indicated solvent on E. Merck silica gel $60(40-63 \mu \mathrm{~m})$ and high pressure liquid chromatography was performed on a RAININ Dynamax SD-200 controlled with Dynamax HPLC software.

Although the above methods of IR, NMR, Mass Spec, and HRMS were those used in the examples of the present invention, it would be known to those skilled in the art that other instruments, conditions, types of techniques, and the like may be used to visualize compounds, identify compounds and determine their concentrations and purity
Methyl 2,3,4-tri-O-acetyl-6-chloro-6-deoxy- $\alpha$-D-glucopy- 5 ranoside (7).

Compound 7 was prepared as previously described from methyl $\alpha$-D-glucopyranoside (5), ( $7.26 \mathrm{~g}, 27.7 \mathrm{mmol}$ ) in $82 \%$ yeild (Anisuzzaman, A. K. M.; Whistler, R. L. Carbohydr. Res. 1978, 61, 511-518). $\mathrm{R}_{\mathrm{f}}=0.34$ ( $2: 1$ hexane/EtOAc); $[\alpha]_{D}=147^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 5.46(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{J}=9.5 \mathrm{~Hz}), 4.98(\mathrm{~m}, 2 \mathrm{H}), 4.02(\mathrm{~m}, 1 \mathrm{H}), 3.82(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=12.0$, 2.5 Hz ), 3.73 (dd, 1H, J=12.0, 6.5 Hz ), $3.43(\mathrm{~s}, 3 \mathrm{H}), 2.06(\mathrm{~s}$, 3H), $2.04(\mathrm{~s}, 3 \mathrm{H}), 1.99(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 170.45$, 170.42, 169.96, 96.98, 77.67, 71.08, 70.46, 70.32, 69.13, $55.85,43.81,21.07,21.05$. MS: calcd for $\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{O}_{8} \mathrm{ClNa}$ 360.9 , found $\mathrm{m} / \mathrm{z} 360.9(\mathrm{M}+\mathrm{Na})$.

1,2,3,4-tetra-O-benzoyl-6-deoxy- $\alpha, \beta$-D-glucopyranose (8). Compound $7(2.9 \mathrm{~g}, 8.57 \mathrm{mmol})$ was dissolved in 100 mL dry THF and $1.0 \mathrm{~g} \mathrm{LiAlH}_{4}$ slowly added. The corresponding mixture was refluxed for 10 hr under argon and the reaction quenched with 10 mL MeOH and concentrated. The concentrate was then dissolved in a mixture of 40 mL acetic acid and 10 mL 1 N HCl and the reaction stirred at $95^{\circ} \mathrm{C}$. for 10 hrs. The reaction was neutralized with 1 N NaOH and the organics concentrated, dried over $\mathrm{MgSO}_{4}$ and purified by silica gel chromatography ( $4: 1 \mathrm{CHCl}_{3} / \mathrm{MeOH}$ ). The resulting product was dissolved in 50 mL dry pyridine, 8.0 mL benzoyl chloride ( 68.9 mmol ) was added and the reaction stirred overnight at room temperature. To the reaction mixture was added to 100 mL saturated $\mathrm{NaHCO}_{3}$ solution and the mixture extracted with $\mathrm{CHCl}_{3}(3 \times 100 \mathrm{~mL})$. The combined organics were washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$, brine ( 50 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated and purified by silica gel chromatography ( $2: 1$ hexane/EtOAc) to give 2.8 g ( $56.2 \%$ ) of the desired product $8(\alpha / \beta=3: 2)$. This mixture was utilized directly for the next step without further resolution. MS: calcd for $\mathrm{C}_{34} \mathrm{H}_{28} \mathrm{O}_{9} \mathrm{Na} 630.2$, found $\mathrm{m} / \mathrm{z} 630.2$ ( $\mathrm{M}+\mathrm{Na}$ ).
Methyl 2,3,6-tri-O-benzoyl- $\beta$-D-galactopyranoside (14).
Methyl $\beta$-D-galactopyranoside (13), $3.7 \mathrm{~g}, 19 \mathrm{mmol}$ ) gave the desired product $14(5.2 \mathrm{~g}, 54 \%)$ and $2.3 \mathrm{~g}(19 \%)$ of the corresponding tetra benzoylated derivative as described in Reist, E. J.; Spencer, R. R.; Calkins, D. F.; Baker, B. R.; Goodman, L. J. Org. Chem. 1965, 2312-2317. $[\alpha]_{D}=7.3^{\circ}$ $\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 8.18-7.92 (m, 6H), $7.52-7.38(\mathrm{~m}, 8 \mathrm{H}), 5.77$ (dd, 1H, J=8.0, 10.4 Hz ), 5.37 (dd, $1 \mathrm{H}, \mathrm{J}=3.2,10.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.72$ (dd, $1 \mathrm{H}, \mathrm{J}=6.6,11.4 \mathrm{~Hz}), 4.62$ (dd, $1 \mathrm{H}, \mathrm{J}=6.4,11.4 \mathrm{~Hz}$ ), 4.66 (d, 1H, J=7.9 Hz), 4.36 (m, 1 H ), $4.08(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=6.5 \mathrm{~Hz}$ ), $3.55(\mathrm{~s}, 3 \mathrm{H}), 2.50(\mathrm{br}, 1 \mathrm{H}, \sim \mathrm{OH})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 166.9,166.3,165.9,133.9,133.7,133.6$, $130.4,130.3,130.2,130.1,130.0,129.9,129.4,129.0$, $128.9,128.8,128.7,102.6,74.6,72.8,69.9,67.7,63.3,57.3$; MS: calcd for $\mathrm{C}_{28} \mathrm{H}_{26} \mathrm{O}_{9} \mathrm{Na} 529.1$, found $\mathrm{m} / \mathrm{z} 529.0(\mathrm{M}+\mathrm{Na})$ .(Garegg, P. J.; Oscarson, S. Carbohydr. Res. 1985, 137, 270-275.)

Methyl 2,3,6-tri-O-benzoyl-4-O-pentafluorophenoxythio-carbonyl- $\beta$-D-galactopyranoside (15).
Methyl 2,3,6-tri-O-benzoyl- $\beta$-D-galactopyranoside (14), $(2.3 \mathrm{~g}, 4.5 \mathrm{mmol})$ gave $2.88 \mathrm{~g}(86 \%)$ purified product 15 as described in Kanie, O.; Crawley, S. C.; Palcic, M. M.; Hindsgaul, O. Carbohydr. Res. 1993, 243, 139-164. $[\alpha]_{D}=9^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.04(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.7$ Hz ), 7.98 (d, 2H, J=7.6 Hz), 7.93 (d, 2H, J=7.7 Hz), $7.58-7.49(\mathrm{~m}, 3 \mathrm{H}), 7.44-7.34(\mathrm{~m}, 6 \mathrm{H}), 6.23(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=3.2$ $\mathrm{Hz}), 5.78(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=7.9 \mathrm{~Hz}), 5.70(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.3,10.4 \mathrm{~Hz})$, $4.75-4.71$ (m, 2H), 4.44 (dd, 1H, J=7.4, 11.0 Hz ), 4.37 ( t , $1 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}), 3.57(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 192.5$, $166.3,166.0,165.5,134.1,133.9,133.7,130.3,130.2$, $130.1,129.6,129.5,128.9,128.8,128.6,102.7,79.9,71.3$, 71.1, 69.9, 61.5, 57.6; MS: calcd for $\mathrm{C}_{35} \mathrm{H}_{25} \mathrm{O}_{9} \mathrm{SF}_{5} \mathrm{Na} 755.1$, found $\mathrm{m} / \mathrm{z} 755.1(\mathrm{M}+\mathrm{Na})$.

Methyl 2,3,6-tri-O-benzoyl-4-deoxy- $\beta$-D-galactopyranoside (16).

Methyl 2,3,6-tri-O-benzoyl-4-O-pentafluorophenox-ythiocarbonyl- $\beta$-D-galactopyranoside (15), ( $2.65 \mathrm{~g}, 3.62$ $\mathrm{mmol})$ gave $1.53 \mathrm{~g}(86 \%)$ of the desired compound 16 as described in Kanie, O; Crawley, S. C.; Palcic, M. M.; Hindsgaul, O. Carbohydr: Res. 1993, 243, 139-164. $[\alpha]_{D}=57.4^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.07(\mathrm{~d}, 2 \mathrm{H}$, $\mathrm{J}=7.3 \mathrm{~Hz}), 8.00(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}), 7.95(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz})$, 7.58 (t, 1H, J=7.4 Hz), 7.53-7.40 (m, 4H), 7.39-7.34 (m,
$4 \mathrm{H}), 5.41(\mathrm{~m}, 2 \mathrm{H}), 4.60(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}), 4.51(\mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}=5.8,11.6 \mathrm{~Hz}), 4.46(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=4.4,11.6 \mathrm{~Hz}), 4.06(\mathrm{~m}, 1 \mathrm{H})$, $2.47(\mathrm{~m}, 1 \mathrm{H}), 1.88(\mathrm{~m}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 166.7,166.3$, 165.9, 133.7, 133.6, 133.5, 130.2, 130.1, 130.0, 129.7, 128.9, 128.8, 128.7, 102.6, 72.9, 71.9, 70.0, 66.2, 57.4, 33.4; MS: calcd for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{O}_{8} \mathrm{Na} 513.1$, found $\mathrm{m} / \mathrm{z} 513.0(\mathrm{M}+\mathrm{Na})$. (Lin, T.-H.; Kovac, P.; Glaudemans, C. P. J. Carbohydr. Res. 1989, 141, 228-238.)

1,2:5,6-Di-O-isopropylidene-3-O-(methylthio)thiocarbo-nyl- $\alpha$-D-glucofuranose (22).

Compound 22 was prepared as previously described in 93\% yield (see Zhiyuan, Z.; Magnusson, G. Carbohydr. Res. 1994, 262, 79-101). $[\alpha]_{D}=-34^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 5.91(\mathrm{~m}, 2 \mathrm{H}), 4.68(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=3.77 \mathrm{~Hz}), 4.31(\mathrm{~m}$, $1 \mathrm{H}), 4.10(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=5.6,8.7 \mathrm{~Hz}), 4.05(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=4.6,8.7$ $\mathrm{Hz}), 2.59(\mathrm{~s}, 3 \mathrm{H}), 1.57(\mathrm{~s}, 3 \mathrm{H}), 1.41(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H})$, $1.31(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 112.8,109.7,105.4,84.6$, 83.1, 80.1, 72.7, 67.3, 27.2, 27.0, 26.6, 25.6, 19.7; MS: calcd for $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{O}_{6} \mathrm{~S}_{2} \mathrm{Na} 373.0$, found $\mathrm{m} / \mathrm{z} 372.8(\mathrm{M}+\mathrm{Na})$.
3-Deoxy-1,2:5,6-di-O-isopropylidene- $\alpha$-D-glucofuranose (23).

To a solution containing $22(2.6 \mathrm{~g}, 7.4 \mathrm{mmol})$ and 120 mg of AIBN ( 0.73 mmol ) in 50 mL dry toluene, $5 \mathrm{~mL}(\mathrm{n}-\mathrm{Bu})$ ${ }_{3} \mathrm{SnH}(18.6 \mathrm{mmol})$ was added and the mixture refluxed for 5 hrs under argon. The reaction was then concentrated and the residue was applied to a silica gel column (10:1-8:1 hexane/EtOAc) to give 1.58 g substantially pure product 23 $(87 \%) \cdot[\alpha]_{D}=-9.2^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 5.82(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=3.7 \mathrm{~Hz}$ ), $4.76(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=4.2 \mathrm{~Hz}), 4.19-4.07(\mathrm{~m}, 3 \mathrm{H})$, $3.84(\mathrm{~m}, 1 \mathrm{H}), 2.18(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.9,13.2 \mathrm{~Hz}), 1.77(\mathrm{~m}, 1 \mathrm{H})$, $1.51(\mathrm{~s}, 3 \mathrm{H}), 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 109.6, 107.9, 104.0, 79.4, 77.0, 75.7, 65.6, 33.7, 25.7, 24.9, 24.5, 23.6; MS: calcd for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{5} \mathrm{Na}$ 267.1, found m/z $266.8(\mathrm{M}+\mathrm{Na}$ ). (Barton, D. H. R.; McCombie, S. W. J. Chem. Soc. Perkin Trans. 1 1975, 1574.) ${ }^{22}$

## 1,2,4,6-tetra-O-benzoyl-3-deoxy- $\alpha$-D-glucofuranose (24).

Compound $23(0.59 \mathrm{~g}, 2.4 \mathrm{mmol})$ was treated with a mixture of $9 \mathrm{mLCF} \mathrm{CO}_{2} \mathrm{H}$ and 1 mL of water for 2 hours at $25^{\circ} \mathrm{C}$. The reaction was concentrated under reduced pressure, coevaporated with water $(2 \times 5 \mathrm{~mL})$ and further dried under vacuum. This material was dissolved in 20 mL of anhydrous pyridine, to which $2.2 \mathrm{~mL}(19.3 \mathrm{mmol})$ of benzoyl chloride was added. The mixture was stirred for 10 hr , pyridine removed in vacuo and the remaining oil diluted with 200 mL EtOAc. The organics washed with saturated $\mathrm{NaHCO}_{3}(50 \mathrm{~mL})$, water ( 40 mL ), brine ( 40 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}{ }^{3}$, and purified with silica gel chromatography ( $3: 1$ hexanes/EtOAc) to give 0.89 g product which was used directly without further characterization.

General Strategy for Formation of Protected Ethyl 1-thio-$\beta$-D-hexopyranosides.

Protected ethyl 1-thio- $\beta$-D-hexopyranosides may be formed in accordance with the present invention by the following reaction. A mixture of protected monosaccharide, (ethylthio)-trimethylsilane, and zinc iodide are refluxed for $11 / 2$ to $2^{1 / 2}$ hrs. The reaction is then cooled, diluted, washed, preferably with saturated $\mathrm{NaHCO}_{3}$ solution, water, and then brine. The organics are then dried, and preferably concentrated and resolved to give the desired product. Other conditions, reagents, method steps, solutions and the like of the present method, may be used in accordance with the present invention.

In a typical reaction, a mixture of 3 mmol protected monosaccharide, 1.5 mL (ethylthio)trimethylsilane ( 9.2 mmol ) and 1.95 g zinc iodide ( 6.1 mmol ) in 30 mL dry
dichloromethane was refluxed for 2 hrs under argon atmosphere. The reaction was then cooled and diluted with 200 $\mathrm{mL} \mathrm{CH} \mathrm{Cl}_{2}$, washed successively with saturated $\mathrm{NaHCO}_{3}$ solution ( $2 \times 30 \mathrm{~mL}$ ), water $(30 \mathrm{~mL})$ and brine ( 30 mL ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated and resolved by silica gel chromatography ( $8: 1$ hexanes/EtOAc) to give the desired product.

Ethyl 2,3,4-tri-O-benzoyl-6deoxy-1-thio- $\beta$-D-glucopyranoside (9).
Compound $8(1 \mathrm{~g}, 1.72 \mathrm{mmol})$ gave $731 \mathrm{mg}(81.5 \%)$ of the desired product. $\mathrm{R}_{f}=0.56$ ( $2: 1$ hexane $/$ EtOAc $) ;[\alpha]_{D}=7^{\circ}$ $\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 8.00-7.94(\mathrm{~m}, 4 \mathrm{H}), 7.82$ (dd, 1H, J=1.4, 7.1 Hz), 7.52 (m, 2H), 7.42-7.37 (m, 5H), $7.23(\mathrm{~m}, 2 \mathrm{H}), 5.85(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.6 \mathrm{~Hz}), 5.54(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.7 \mathrm{~Hz})$, $5.35(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.6 \mathrm{~Hz}), 4.80(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.9 \mathrm{~Hz}), 4.92(\mathrm{~m}, 1 \mathrm{H})$, $2.82(\mathrm{~m}, 2 \mathrm{H}), 1.40(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}), 1.26(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 164.8, 164.4, 164.2, 132.3, 132.2, 132.1, $128.8,128.7,128.6,128.2,128.0,127.9,127.4,127.3$, $127.2,82.3,73.9,73.1,72.7,69.8,22.9,16.8,13.7$; MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{O}_{7} \mathrm{SNa} 543.1$, found $\mathrm{m} / \mathrm{z} 542.9(\mathrm{M}+\mathrm{Na})$.
Ethyl 2,3,6-tri-O-benzoyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (17).

Compound $16(1.5 \mathrm{~g}, 3.06 \mathrm{mmol})$ gave 1.24 g desired product $(77.8 \%) .[\alpha]_{D}=56.9^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.08(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.0 \mathrm{~Hz}), 8.00(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.2 \mathrm{~Hz}), 7.96$ $(\mathrm{d}, 2 \mathrm{H}, \mathrm{J}=8.0 \mathrm{~Hz}), 7.60(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=6.9 \mathrm{~Hz}), 7.55-7.48(\mathrm{~m}, 4 \mathrm{H})$, $7.42-7.36(\mathrm{~m}, 4 \mathrm{H}), 5.50-5.44(\mathrm{~m}, 2 \mathrm{H}), 4.76(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.0$ Hz ), 4.51 (dd, 1H, J=5.7, 1.9 Hz), 4.46 (dd, 1H, J=4.4, 11.9 $\mathrm{Hz}), 4.12(\mathrm{~m}, 1 \mathrm{H}), 2.84-2.69(\mathrm{~m}, 2 \mathrm{H}), 2.53(\mathrm{~m}, 1 \mathrm{H}), 1.91$ $(\mathrm{m}, 1 \mathrm{H}), 1.27(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 166.6$, $166.2,165.9,133.7,133.6,130.2,130.1,129.8$, 129.7, 128.8. 184.2, 74.0, 73.0, 71.5, 66.3, 33.6, 24.7, 15.4; MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{O}_{7} \mathrm{SNa} 543.1$, found $\mathrm{m} / \mathrm{z} 543.1(\mathrm{M}+\mathrm{Na})$.
Ethyl 2,4,6-tri-O-benzoyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (25).
Compound $24(0.89 \mathrm{~g}, 1.5 \mathrm{mmol})$ gave 0.79 substantially pure product $(90 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.14-7.96(\mathrm{~m}, 6 \mathrm{H})$, $7.63-7.40(\mathrm{~m}, 9 \mathrm{H}), 5.32-5.21(\mathrm{~m}, 2 \mathrm{H}), 4.79(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.7$ $\mathrm{Hz}), 4.67(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=2.9,12.0 \mathrm{~Hz}), 4.46(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=6.0,12.0$ $\mathrm{Hz}), 4.09(\mathrm{~m}, 1 \mathrm{H}), 2.96(\mathrm{~m}, 1 \mathrm{H}), 2.78(\mathrm{~m}, 2 \mathrm{H}), 2.00(\mathrm{~m}, 1 \mathrm{H})$, 1.27 (t, 3H, J=7.4 Hz); MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{O}_{7} \mathrm{SNa} 543.1$, found $\mathrm{m} / \mathrm{z} 543.1(\mathrm{M}+\mathrm{Na})$.
Ethyl 3,4,6-tri-O-benzoyl-2-deoxy-1-thio- $\beta$-D-glucopyranoside (40).

Compound 39 ( $1.72 \mathrm{~g}, 2.96 \mathrm{mmol}$ ) gave two products, 0.74 g the desired $\beta$ isomer ( $48 \%$ yield) and 0.5 g the $\alpha$ isomer ( $32 \%$ yield). $\beta$ isomer: $[\alpha]_{D}=120^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right.$ ); IR:_2962, 2871, 1723, 1601, 1450, 1314, 1270, 1107, 708, $686 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.06-7.93(\mathrm{~m}, 6 \mathrm{H}), 7.51-7.36$ $(\mathrm{m}, 9 \mathrm{H}), 5.69-5.64(\mathrm{~m}, 1 \mathrm{H}), 5.60-5.56(\mathrm{~m}, 2 \mathrm{H}), 4.80(\mathrm{~m}$, $1 \mathrm{H}), 4.57(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=2.7,12.0 \mathrm{~Hz}), 4.52(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=12.0,5.5$ $\mathrm{Hz}), 2.72-2.54(\mathrm{~m}, 3 \mathrm{H}), 2.41-2.35(\mathrm{~m}, 1 \mathrm{H}), 1.30(\mathrm{t}, 3 \mathrm{H}$, $\mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 165.2,164.6,164.5,132.3$, $132.1,132.0,128.8,128.7,128.6,128.4,128.1,127.4$, $127.3,78.5,69.4,67.3,62.4,34.4,23.8$. MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{O}_{7} \mathrm{SNa} 543.1$, found $\mathrm{m} / \mathrm{z} 543.1(\mathrm{M}+\mathrm{Na})$. $\alpha$ isomer: $[\alpha]_{D}=-46^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right)$ IR: 2961, 2923, 1732, 1717, 1269, $1108,1099,708,685 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.12-7.93(\mathrm{~m}$, $6 \mathrm{H}), 7.54-7.37(\mathrm{~m}, 9 \mathrm{H}), 5.56(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.7 \mathrm{~Hz}), 5.46(\mathrm{~m}$, $1 \mathrm{H}), 4.87(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=11.8,1.7 \mathrm{~Hz}), 4.60(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.1,12.0$ $\mathrm{Hz}), 4.48(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=5.9,12.0 \mathrm{~Hz}), 4.06(\mathrm{~m}, 1 \mathrm{H}), 2.83-2.68$ $(\mathrm{m}, 2 \mathrm{H}), 2.65-2.64(\mathrm{~m}, 1 \mathrm{H}), 2.08(\mathrm{~m}, 1 \mathrm{H}), 1.32(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 166.6,166.3,165.9,134.1,133.8$, $133.7,133.4,130.6,130.1,130.0,129.7,129.5,128.9$,
128.7, 80.3, 77.1, 73.0, 70.5, 64.3, 37.1, 25.5, 15.5. MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{O}_{7} \mathrm{SNa} 543.1$, found: $\mathrm{m} / \mathrm{z} 543.0(\mathrm{M}+\mathrm{Na})$.
Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- $\beta$-D-gulopyranoside (30).

Compound $29(0.75 \mathrm{~g}, 1.07 \mathrm{mmol})$ gave 0.65 g of the desired compound ( $94 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) 8.18-7.87 (m, 8H), 7.54-7.27 (m, 12H), 5.95 (t, J=3.5 Hz, 1H), 5.67 (dd, $\mathrm{J}=3.3,10.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.61(\mathrm{~m}, 1 \mathrm{H}), 5.27(\mathrm{~d}, \mathrm{~J}=10.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.64(\mathrm{~m}, 2 \mathrm{H}), 4.50(\mathrm{dd}, \mathrm{J}=3.8,9.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.84(\mathrm{~m}, 2 \mathrm{H})$, 1.34 (t, J=7.4 Hz, 3H); MS: calcd for $\mathrm{C}_{36} \mathrm{H}_{32} \mathrm{O}_{9} \mathrm{SNa} 663.2$, found $\mathrm{m} / \mathrm{z} 663.1(\mathrm{M}+\mathrm{Na})$.
Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- $\beta$-D-allopyranoside (35).

Compound $34(0.97 \mathrm{~g}, 1.38 \mathrm{mmol})$ gave 0.85 g desired product $(95 \%) .[\alpha]_{D}=12.7^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right)$ $8.07-7.32(\mathrm{~m}, 20 \mathrm{H}), 6.26(\mathrm{t}, \mathrm{J}=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.56(\mathrm{dd}, \mathrm{J}=2.8$, $10.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.51(\mathrm{dd}, \mathrm{J}=2.9,10.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{dd}, \mathrm{J}=2.5$, $12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{~m}, 1 \mathrm{H}), 4.47(\mathrm{dd}, \mathrm{J}=5.3,12.0 \mathrm{~Hz}, 1 \mathrm{H})$, $2.80(\mathrm{~m}, 2 \mathrm{H}), 1.29(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 171.6, 166.6, 165.7, 165.2, 134.0, 133.9, 133.8, 133.6, $130.5,130.3,130.2,130.1,130.0,129.9,129.6,129.3$, $129.2,128.9,128.8,81.3,73.7,69.7,68.9,68.0,63.9,24.3$, 15.5; MS: calcd for $\mathrm{C}_{36} \mathrm{H}_{32} \mathrm{O}_{9} \mathrm{SNa} 663.1$, found $\mathrm{m} / \mathrm{z} 663.0$ (M+Na).
General Strategy for O-Benzoyl to O-Benzyl Conversion.
O-Benzoyl may be converted to O-Benzyl according to the following method or other methods known to those skilled in the art. Protected ethyl 1-thio- $\beta$-D-hexopyranoside are dissolved in dry MeOH and toluene to which a sodium methoxide solution is added. The mixture is then stirred, preferably for about $1^{1 / 2}$ to $2^{1 / 2}$ hrs at room temperature and optionally neutralized. The organics are preferably concentrated and the corresponding unprotected 1 -ethylthio- $\beta$-Dglucopyranoside purified, and then dissolved. NaH is then added and the reaction is stirred for about $1^{1 / 2}$ to $2^{1 / 2}$ hrs at room temperature followed by the addition of benzyl bromide and stirring, preferably overnight. The mixture may then be diluted, washed with $\mathrm{H}_{2} \mathrm{O}$, brine, and organics, dried, concentrated, and purified to give the purified product.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.

In a typical reaction, 1.4 mmol of protected ethyl 1-thio-$\beta$-D-hexopyranoside was dissolved in 10 mL dry MeOH and 3 mL toluene to which 0.25 mL of a sodium methoxide solution ( $25 \% \mathrm{NaOMe}$ in methanol) was added. The mixture was stirred for 2 hr at room temperature and neutralized with 1 N acetic acid. The organics were concentrated and the corresponding unprotected 1 -ethylthio- $\beta$-D-glucopyranoside purified by silica gel chromatography (10:1 hexane/ EtOAc) which was then dissolved in 10 mL dry DMF and $323 \mathrm{mg} 65 \% \mathrm{NaH}(8.0 \mathrm{mmol})$ was added. The reaction was stirred for 2 hr at room temperature followed by the addition of 1 mL benzyl bromide ( 8.3 mmol ) and continued stirring overnight. The mixture was then diluted with 150 mL EtOAc, washed with $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{~mL})$, brine ( 30 mL ) and the organics dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated and purified by silica gel chromatography ( $10: 1$ hexane/EtOAc) to give the purified product.

Ethyl 2,3,4-tri-O-benzyl-6-deoxy-1-thio- $\beta$-D-glucopyranoside (10).

Compound $9(0.7 \mathrm{~g}, 1.35 \mathrm{mmol})$ gave $480 \mathrm{mg}(75 \%)$ of purified product. $[\alpha]_{D}=5.8^{\circ}\left(\mathrm{c}=1.0, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR 6 $\left(\mathrm{CDCl}_{3}\right) 7.40-7.29(\mathrm{~m}, 15 \mathrm{H}), 4.95-4.85(\mathrm{~m}, 4 \mathrm{H}), 4.77(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}), 4.65(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.5 \mathrm{~Hz}), 4.48(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.8$
$\mathrm{Hz}), 3.66(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=8.9 \mathrm{~Hz}), 3.46-3.38(\mathrm{~m}, 2 \mathrm{H}), 3.23(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{J}=9.2 \mathrm{~Hz}), 2.84-2.70(\mathrm{~m}, 2 \mathrm{H}), 1.35-1.27(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 138.9,138.5,138.4,128.9,128.8,128.7,128.4$, $128.3,128.2,128.1,86.8,85.2,83.8,82.5,76.2,75.9,75.8$, $25.4,18.5,15.5$. MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{34} \mathrm{O}_{4} \mathrm{SNa} 501.2$, found $\mathrm{m} / \mathrm{z} 501.1(\mathrm{M}+\mathrm{Na})$.

Ethyl 2,3,6-tri-O-benzyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (18).
Compound 17 ( $0.85 \mathrm{~g}, 1.63 \mathrm{mmol}$ ) gave $675 \mathrm{mg}(86 \%)$ purified product. $[\alpha]_{D}=40^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right),{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 7.45-7.31 (m, 1SH), 4.92 (d, 1H, J=10.3 Hz), 4.86 (d, 1H, $\mathrm{J}=10.3 \mathrm{~Hz}), 4.74(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7 \mathrm{~Hz}), 4.69(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7$ $\mathrm{Hz}), 4.62(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.0 \mathrm{~Hz}), 4.58(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.0 \mathrm{~Hz}), 4.49$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=9.7 \mathrm{~Hz}), 3.69-3.62(\mathrm{~m}, 3 \mathrm{H}), 3.50(\mathrm{~m}, 1 \mathrm{H}), 3.36$ (dd, 1H, J=8.7, 9.4 Hz ), 2.82-2.75 (m, 2H), $2.23(\mathrm{~m}, 1 \mathrm{H})$, $1.54(\mathrm{~m}, 1 \mathrm{H}), 1.35(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $138.8,138.7,138.5,128.8,128.7,128.6,128.2,128.1$, $128.0,85.5,82.3,80.6,76.0,75.4,73.9,72.9,72.3,34.4$, 25.2, 15.6 MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{34} \mathrm{O}_{4} \mathrm{SNa} 501.2$, found $\mathrm{m} / \mathrm{z}$ $501.0(\mathrm{M}+\mathrm{Na})$.

Ethyl 2,4,6-tri-O-benzyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (26).

Compound 25 ( $608 \mathrm{mg}, 1.17 \mathrm{mmol}$ ) gave 364 mg substantially pure product ( $65 \%$ ). $[\alpha]_{D}=-11.8^{\circ}$ ( $\mathrm{c}=1, \mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.44-7.14(\mathrm{~m}, 15 \mathrm{H}), 4.74(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.6$ $\mathrm{Hz}), 4.66-4.56(\mathrm{~m}, 4 \mathrm{H}), 4.50(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.4 \mathrm{~Hz}), 4.45(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=11.4 \mathrm{~Hz}$ ), 3.83 (d, 1H, J=10.7 Hz), 3.69 (dd, 1H, J=4.4, $10.7 \mathrm{~Hz}), 3.49(\mathrm{~m}, 2 \mathrm{H}), 3.35(\mathrm{~m}, 1 \mathrm{H}), 2.79(\mathrm{~m}, 2 \mathrm{H}), 2.69(\mathrm{~m}$, $\mathrm{H}), 1.54(\mathrm{~m}, 1 \mathrm{H}), 1.35(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ : $138.8,138.4,135.8,128.9,128.7,128.4,128.2,128.1$, $127.4,86.9,81.3,75.6,73.8,73.3,72.5,71.6,69.8,45.3$, 36.7, 25.1, 20.3, 15.5; MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{34} \mathrm{O}_{4} \mathrm{SNa} 501.2$, found $\mathrm{m} / \mathrm{z} 501.0(\mathrm{M}+\mathrm{Na})$.

## Ethyl 2,3,4,6-tetra-O-benzyl-1-thio- $\beta$-D-gulopyranoside

 (31).Compound $30(0.6 \mathrm{~g}, 0.94 \mathrm{mmol})$ gave 330 mg substantially pure product ( $60 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.32-7.17$ ( m , $20 \mathrm{H}), 4.94(\mathrm{~d}, \mathrm{~J}=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.55(\mathrm{~m}, 2 \mathrm{H}), 4.40(\mathrm{~m}, 4 \mathrm{H})$, $4.22(\mathrm{~m}, 2 \mathrm{H}), 4.00(\mathrm{t}, \mathrm{J}=6.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.60-3.42(\mathrm{~m}, 5 \mathrm{H})$, $2.66(\mathrm{~m}, 2 \mathrm{H}), 1.22(\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 3 \mathrm{H})$; MS: calcd for $\mathrm{C}_{36} \mathrm{H}_{40} \mathrm{O}_{5} \mathrm{SNa} 607.2$, found $\mathrm{m} / \mathrm{z} 607.0(\mathrm{M}+\mathrm{Na})$.
Ethy1 2,3,4,6-tetra-O-benzyl-1-thio- $\beta$-D-allopyranoside (36).

Compound 35 ( $0.85 \mathrm{~g}, 1.33 \mathrm{mmol}$ ) gave 496 mg substantially pure product $(64 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.4-7.22$ ( m , 20 H ), 5.05 (d, J=9.7 Hz, 1H), 4.86 (d, J=11.8 Hz, 1H), 4.80 (d, J=11.8 Hz, 1H), 4.69-4.40 (m, 6H), $4.13(\mathrm{~m}, 1 \mathrm{H}), 4.03$ (dd, J=3.1, $9.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.47 (dd, J=2.3, $9.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.29 (dd, J=2.3, $9.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.75(\mathrm{~m}, 2 \mathrm{H}), 1.32(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 139.4,138.9,138.3,138.2,128.9,128.8$, 128.7, 128.6, 128.5, 128.3, 128.2, 128.1, 128.0, 127.9, $127.8,82.0,79.2,76.1,75.4,74.7,73.9,73.8,72.8,71.9$, 69.9, 25.1, 15.6; MS: calcd for $\mathrm{C}_{36} \mathrm{H}_{40} \mathrm{O}_{5} \mathrm{SNa} 607.2$, found $\mathrm{m} / \mathrm{z} 607.0(\mathrm{M}+\mathrm{Na})$.
General Phosphorylation Strategy (Method A: via Ethyl 1 -thio- $\beta$-D-hexopyranoside).

Phosphorylation may take place in accordance with the present invention by the following reaction, which involves ethyl 1-thio- $\beta$-D-hexopyranoside. The ethyl 1-thio- $\beta$-Dhexopyranoside may be ethyl 1-thio- $\beta$-D-hexopyranoside prepared according to the methods described herein or ethyl 1 -thio- $\beta$-D-hexopyranoside prepared by other methods.
According to this method, protected ethyl 1-thio- $\beta$-Dhexopyranoside and dibenzyl phosphate are co-evaporated,
preferably two times from dry toluene and further dried under high vacuum overnight to which N -iodosuccinamide and dry molecular sieves are preferably added. The mixture is then dissolved, preferably in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, cooled to about $-40^{\circ} \mathrm{C}$. to about $-20^{\circ} \mathrm{C}$., preferably about $-30^{\circ} \mathrm{C}$. and trifluoromethane-sulfonic acid is added. The reaction mixture is substantially maintained at the cooled temperature for about 20 to about 40 minutes, preferably about 30 min with stirring. Preferably, the mixture is then diluted, and washed with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ and/or saturated $\mathrm{NaHCO}_{3}, \mathrm{H}_{2} \mathrm{O}$, and brine. The organics are then preferably dried, filtered, concentrated and purified to give the desired product.

Other conditions, steps, reagents, solutions and the like of the present method may be used in accordance with the present invention.

In a typical reaction, 0.84 mmol protected ethyl 1-thio-$\beta$-D-hexopyranoside and 1.44 mmol dibenzyl phosphate were co-evaporated two times from dry toluene and further dried under high vacuum overnight to which 1.24 mmol of N -iodosuccinamide and 300 mg dry molecular sieves were added. The mixture was then dissolved in 10 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, cooled to $-30^{\circ} \mathrm{C}$. and $25 \mu \mathrm{l}$ trifluoromethanesulfonic acid ( 0.28 mmol ) was added. The reaction mixture was maintained at $-30^{\circ} \mathrm{C}$. for 30 min with stirring and then diluted with 100 mL EtOAc, washed with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ $(20 \mathrm{~mL})$ and saturated $\mathrm{NaHCO}_{3}(20 \mathrm{~mL}), \mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$, and brine ( 20 mL ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, concentrated and purified by chromatography on silica gel ( $3: 1$ hexane/EtOAc) to give the desired product.

Dibenzyl-(2,3,4-tri-O-benzyl-6-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (11).

Compound 10 ( $400 \mathrm{mg}, 0.84 \mathrm{mmol}$ ) gave $0.44 \mathrm{mg}(76 \%)$ of the desired product. $[\alpha]_{D}=22.8^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.38-7.28(\mathrm{~m}, 25 \mathrm{H}), 5.93(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.2,6.6 \mathrm{~Hz})$, $5.30(\mathrm{~m}, 4 \mathrm{H}), 5.18(\mathrm{~m}, 3 \mathrm{H}), 5.09(\mathrm{~m}, 2 \mathrm{H}), 4.67(\mathrm{~m}, 2 \mathrm{H}), 3.94$ $(\mathrm{m}, 1 \mathrm{H}), 3.64(\mathrm{~m}, 1 \mathrm{H}), 3.18(\mathrm{~m}, 1 \mathrm{H}), 1.21(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 138.9,138.5,138.0,128.9,128.8,128.7$, 128.4, 128.3, 95.8, 95.7, 94.0, 76.0, 75.7, 73.6, 69.7, 17.3; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 2.58; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{43} \mathrm{O}_{8} \mathrm{PNa} 717.2$, found $\mathrm{m} / \mathrm{z} 717.3(\mathrm{M}+\mathrm{Na})$.

Dibenzyl-(2,3,6-tri-O-benzyl-4-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (19).

Compound 18 ( $512 \mathrm{mg}, 1.07 \mathrm{mmol}$ ) gave $0.565 \mathrm{~g}(76 \%)$ substantially pure product. $[\alpha]_{D}=28.2^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.29-7.10(\mathrm{~m}, 25 \mathrm{H}), 5.91(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.2,6.6$ $\mathrm{Hz}), 4.72-4.57(\mathrm{~m}, 4 \mathrm{H}), 4.41(\mathrm{~m}, 2 \mathrm{H}), 4.02(\mathrm{~m}, 1 \mathrm{H}), 3.81$ $(\mathrm{m}, 1 \mathrm{H}), 3.48(\mathrm{~m}, 1 \mathrm{H}), 3.34(\mathrm{~m}, 2 \mathrm{H}), 2.04-2.00(\mathrm{~m}, 1 \mathrm{H})$, $1.60-1.48(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 138.6,138.4,138.2$, $137.8,137.7,128.3,128.2,128.1,127.9,127.8,127.7$, $127.6,127.5,127.4,97.3,91.9,83.6,80.2,78.1,74.9,73.3$, 73.0, 72.2, 72.0, 71.9, 70.6, 66.7, 52.7, 33.4; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 1.25$; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{43} \mathrm{O}_{8} \mathrm{PNa} 717.2$, found $\mathrm{m} / \mathrm{z} 717.2(\mathrm{M}+\mathrm{Na})$.

Dibenzyl-(2,4,6-tri-O-benzyl-3-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (27).

Compound 26 ( $270 \mathrm{mg}, 0.56 \mathrm{mmol}$ ) gave 0.31 g substantially pure product ( $79 \%, \alpha / \beta=2: 1$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $7.32-7.21$ (m, 25H), 5.96 (dd, 1H, J=2.8, 6.6 Hz), 5.06 (m, $4 \mathrm{H}), 4.66(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7 \mathrm{~Hz}), 4.56(\mathrm{~m}, 3 \mathrm{H}), 4.42(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=12.0 \mathrm{~Hz}), 4.38(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.3 \mathrm{~Hz}), 3.82(\mathrm{~m}, 1 \mathrm{H}), 3.66(\mathrm{~m}$, $3 \mathrm{H}), 3.49(\mathrm{~m}, 1 \mathrm{H}), 2.54(\mathrm{~m}, 0.5 \mathrm{H}), 2.40(\mathrm{~m}, 1 \mathrm{H}), 1.85(\mathrm{~m}$, $1 \mathrm{H}), 1.56(\mathrm{~m}, 0.5 \mathrm{H}) ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.54,0.17$; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{43} \mathrm{O}_{8} \mathrm{PNa} 717.2$, found $\mathrm{m} / \mathrm{z} 717.2(\mathrm{M}+\mathrm{Na})$.

Dibenzyl-(3,4,6-tri-O-benzoyl-2-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (41).

Compound $40(460 \mathrm{mg}, 0.88 \mathrm{mmol})$ gave 0.49 g of substantially pure product ( $75 \%$ ) after silica gel chromatography ( $3: 1-2: 1$ hexane/EtOAc. $[\alpha]_{D}=19^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) 8.05-7.94 (m, 6H), 7.53-7.51 (m, 3H), $7.41-7.34(\mathrm{~m}, 16 \mathrm{H}), 5.96(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=1.6,7.2 \mathrm{~Hz}), 5.68(\mathrm{~m}$, $2 \mathrm{H}), 5.16(\mathrm{~m}, 4 \mathrm{H}), 4.51-4.43(\mathrm{~m}, 2 \mathrm{H}), 4.35(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.1$, $12.0 \mathrm{~Hz}), 2.56(\mathrm{~m}, 1 \mathrm{H}), 2.04(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $166.0,165.6,165.3,135.5,135.4,133.3,133.2,133.0$, $129.8,129.7,129.6,129.3,128.9,128.6,128.4,128.3$, $128.1,127.9,95.9,70.1,69.5,69.2,68.9,62.6 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.32$; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{37} \mathrm{O}_{11} \mathrm{PNa} 759.1$, found: $\mathrm{m} / \mathrm{z} 759.1(\mathrm{M}+\mathrm{Na})$.

## Dibenzyl-(2,3,4,6-tetra-O-benzyl- $\alpha-D$-gulopyranosyl)

Phosphate (32).
Compound $31(120 \mathrm{mg}, 0.21 \mathrm{mmol})$ gave 50 mg of the desired compound ( $30 \%$ ) and 38 mg of the $\beta$ isomer ( $23 \%$ ). $\alpha$ isomer: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.30-6.90(\mathrm{~m}, 30 \mathrm{H}), 5.95(\mathrm{dd}$, $\mathrm{J}=3.7,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{~m}, 5 \mathrm{H}), 4.64(\mathrm{~m}, 2 \mathrm{H}), 4.49-4.30$ $(\mathrm{m}, 8 \mathrm{H}), 3.80(\mathrm{~m}, 2 \mathrm{H}), 3.60(\mathrm{~d}, \mathrm{~J}=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{~m}, 2 \mathrm{H})$; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.8$; MS: calcd for $\mathrm{C}_{48} \mathrm{H}_{49} \mathrm{O}_{9} \mathrm{PNa} 823.3$, found $\mathrm{m} / \mathrm{z} 823.3(\mathrm{M}+\mathrm{Na}) . \beta$ isomer: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $7.25-7.15(\mathrm{~m}, 30 \mathrm{H}), 5.61(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.05-4.99(\mathrm{~m}$, $4 \mathrm{H}), 4.60(\mathrm{~d}, \mathrm{~J}=12 \mathrm{~Hz}, 1 \mathrm{H}), 4.42-4.34(\mathrm{~m}, 4 \mathrm{H}), 4.26$ (d, $\mathrm{J}=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.17(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.65(\mathrm{~m}, 2 \mathrm{H}), 3.48(\mathrm{~m}$, $2 \mathrm{H}), 3.48(\mathrm{~m}, 2 \mathrm{H}), 3.43(\mathrm{dd}, \mathrm{J}=1.3,13.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right)-1.1$; MS: calcd for $\mathrm{C}_{48} \mathrm{H}_{49} \mathrm{O}_{9} \mathrm{PNa} 823.3$, found $\mathrm{m} / \mathrm{z} 823.3$ (M+Na).
Dibenzyl-(2,3,4,6-tetra-O-benzyl- $\alpha$-D-allopyranosyl) Phosphate (37).
Compound 36 ( $169 \mathrm{mg}, 0.29 \mathrm{mmol}$ ) gave 70 mg the desired compound ( $30 \%$ ) and 64 mg of the $\beta$ isomer ( $28 \%$ ). $\alpha$ isomer: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.34-7.13$ (m, 30H), 6.04 (dd, $\mathrm{J}=3.6,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.11-4.92(\mathrm{~m}, 4 \mathrm{H}), 4.89(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.84(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.75(\mathrm{~d}, \mathrm{~J}=11.8 \mathrm{~Hz}, 1 \mathrm{H})$, $4.59-4.37(\mathrm{~m}, 6 \mathrm{H}), 4.23(\mathrm{~m}, 1 \mathrm{H}), 3.73(\mathrm{dd}, \mathrm{J}=3.0,10.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.66(\mathrm{dd}, \mathrm{J}=2.5,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.54(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 139.4,138.4,138.3,137.9,136.5,136.4,128.9$, $128.8,128.7,128.7,128.5,128.4,128.3,128.2,128.1$, $128.0,127.6,95.1,76.2,74.4,73.9,73.3,71.8,69.6,69.2$, 68.7, 68.6; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.27$; MS: calcd for $\mathrm{C}_{48} \mathrm{H}_{49} \mathrm{O}_{9} \mathrm{PNa} 823.3$, found $\mathrm{m} / \mathrm{z} 823.3(\mathrm{M}+\mathrm{Na})$. $\beta$ isomer: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.37-7.07(\mathrm{~m}, 30 \mathrm{H}), 5.63(\mathrm{t}, \mathrm{J}=7.7 \mathrm{~Hz}$, $1 \mathrm{H}), 5.00(\mathrm{~m}, 4 \mathrm{H}), 4.79(\mathrm{~d}, \mathrm{~J}=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, \mathrm{~J}=11.9$ $\mathrm{Hz}, 1 \mathrm{H}), 4.63(\mathrm{~d}, \mathrm{~J}=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.51(\mathrm{~d}, \mathrm{~J}=11.9 \mathrm{~Hz}, 1 \mathrm{H})$, 4.46 (d, J=12.1 Hz, 1H), 4.36 (m, 3H), 4.09 (dd, J=1.4, 9.7 $\mathrm{Hz}, 1 \mathrm{H}), 4.02(\mathrm{~s}, 1 \mathrm{H}), 3.64(\mathrm{dd}, \mathrm{J}=3.7,11.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.58$ (dd, J=1.5, $11.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.51(\mathrm{dd}, \mathrm{J}=2.3,9.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.32$ (dd, J=2.3, $7.9 \mathrm{~Hz}, 1 \mathrm{H} ;{ }^{3}{ }^{3} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.76$; MS: calcd for $\mathrm{C}_{48} \mathrm{H}_{49} \mathrm{O}_{9} \mathrm{PNa} 823.3$, found $\mathrm{m} / \mathrm{z} 823.3(\mathrm{M}+\mathrm{Na})$.

General Phosphorylation Strategy (Method B: via Glycosyl Halide).
According to another embodiment, phosphorylation may take place in accordance with the present invention by the following reaction, which involves glycosyl halide. The glycosyl halide may be glycosyl halide prepared according to the methods described herein or glycosyl halide prepared by other methods.

According to this method, protected D-hexose is dissolved in acetic acid to which HBr in acetic acid was added dropwise at about $0^{\circ} \mathrm{C}$. The reaction is allowed to warm to room temperature and stirred for about $1^{1 / 2}$ to about $21 / 2 \mathrm{hrs}$. The mixture is then diluted with cold $\mathrm{CHCl}_{3}$, washed successively with cold saturated $\mathrm{NaHCO}_{3}$ solution, $\mathrm{H}_{2} \mathrm{O}$ and
brine, and the organics were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The crude protected- $\alpha$-D-pyranosyl bromide may be used directly without further purification. A mixture of dibenzyl phosphate, silver triflate, 2,4,6-collidine and activated $4 \AA$ molecular sieves in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ is stirred at room temperature in the absence of light for about 1 hr . The mixture was then cooled to about $-30^{\circ} \mathrm{C}$. to about $-50^{\circ}$ C., preferably about $-40^{\circ} \mathrm{C}$., to which a solution of the crude protected- $\alpha$-D-pyranosyl bromide in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ is added in dropwise fashion. The reaction mixture is kept at substantially the same cool temperature for about $11 / 2$ to about $2^{1 / 2}$ M hrs, allowed to warm to room temperature and stirred, preferably overnight. The corresponding filtrate is preferably diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, washed with saturated $\mathrm{CuSO}_{4}$, $\mathrm{H}_{2} \mathrm{O}$, and brine, and the organics aree dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. Purification yields substantially pure product.

Other conditions, steps, reagents, solutions and the like of the present method may be used in accordance with the present invention.

Suitably protected D-hexose ( 0.64 mmol ) was dissolved in 5 mL acetic acid to which $5 \mathrm{~mL} 33 \% \mathrm{HBr}$ in acetic acid was added dropwise at $0^{\circ} \mathrm{C}$. The reaction was allowed to warm to room temperature and stirred for 2 hr . The mixture was then diluted with 100 mL cold $\mathrm{CHCl}_{3}$, washed successively with cold saturated $\mathrm{NaHCO}_{3}$ solution ( $3 \times 30 \mathrm{~mL}$ ), $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{~mL})$ and brine ( 20 mL ), and the organics were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The crude pro-tected- $\alpha$-D-pyranosyl bromide was used directly without further purification. A mixture of dibenzyl phosphate (1.80 $\mathrm{mmol})$, silver triflate ( 1.80 mmol ), 2,4,6-collidine ( 3.0 mmol ) and 0.5 g activated $4 \AA$ molecular sieves in 10 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was stirred at room temperature under argon atmosphere in the absence of light for 1 hr . The mixture was then cooled to $-40^{\circ} \mathrm{C}$. to which a solution of the crude protected-$\alpha$-D-pyranosyl bromide in 10 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added in dropwise fashion. The reaction mixture was kept at $-40^{\circ} \mathrm{C}$. for 2 hr , allowed to warm to room temperature and stirred overnight. The corresponding filtrate was diluted with 100 $\mathrm{mLCH}_{2} \mathrm{Cl}_{2}$, washed with saturated $\mathrm{CuSO}_{4}(2 \times 20 \mathrm{~mL}), \mathrm{H}_{2} \mathrm{O}$ $(20 \mathrm{~mL})$ and brine $(20 \mathrm{~mL})$, and the organics were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. Purification by silica gel chromatography ( $1: 1$ hexanes/EtOAc) gave substantially pure product.
Dibenzyl-(2,3,4,6-tetra-O-benzoyl- $\alpha$-D-altropyranosyl)
Phosphate (45).
Perbenzoylated D-altrose (44), ( $0.675 \mathrm{~g}, 0.96 \mathrm{mmol}$ ) gave 0.58 g substantially pure product ( $70 \%$ overall). $[\alpha]_{D}=40^{\circ}$ $\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.05(\mathrm{~m}, 6 \mathrm{H}), 7.82(\mathrm{dd}$, $\mathrm{J}=1.2,7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.53-7.22(\mathrm{~m}, 22 \mathrm{H}), 5.87(\mathrm{~m}, 3 \mathrm{H}), 5.38$ (d, J=3.1 Hz, 1H), $5.04(\mathrm{~m}, 4 \mathrm{H}), 4.90(\mathrm{dd}, \mathrm{J}=3.0,10.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.58$ (dd, J=2.4, $12.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.38 (dd, J=3.9, 12.3 Hz , $1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 166.4,165.4,165.3,164.8,135.7$, 134.3 , 133.9, 133.5, 130.5, 130.4, 130.3, 130.1, 129.6, 129.3, 129.1, 129.0, 128.9, 128.8, 128.4, 128.3, 70.2, 70.1, $70.0,69.5,69.4,67.1,66.9,65.5,63.0 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ -0.02 ; MS: calcd for $\mathrm{C}_{48} \mathrm{H}_{41} \mathrm{O}_{13} \mathrm{NaP} 879.2$, found $\mathrm{m} / \mathrm{z} 879.2$ (M+Na).

Dibenzyl-(2,3,4,6-tetra-O-benzoyl- $\alpha$-D-idopyranosyl) Phosphate (49)

Perbenzoylated D-idose (48), ( $0.32 \mathrm{~g}, 0.46 \mathrm{mmol}$ ) gave 270 mg substantially pure product. ( $69 \%$ overall). $[\alpha]_{D}=11.4^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) 8.11-7.88 (m, $8 \mathrm{H}), 7.40-7.19(\mathrm{~m}, 22 \mathrm{H}), 6.0(\mathrm{~d}, \mathrm{~J}=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.68(\mathrm{~m}$, $1 \mathrm{H}), 5.46(\mathrm{~m}, 1 \mathrm{H}), 5.22(\mathrm{~m}, 1 \mathrm{H}), 5.07-5.01(\mathrm{~m}, 5 \mathrm{H}), 4.60$ (dd, J=7.0, 11.5 Hz, 1H), 4.53 (dd, J=5.8, 11.5 Hz, 1H); ${ }^{13} \mathrm{C}$

NMR $\left(\mathrm{CDCl}_{3}\right) 166.0,165.1,164.7,164.3,135.2,133.6$, $133.5,133.1,130.1,130.0,129.9,129.7,129.6,129.4$, $128.9,128.6,128.5,128.4,128.3,128.2,127.9,127.8,94.9$, 69.7, 69.5, $65.9,65.7,62.6 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.1$; MS: calcd for $\mathrm{C}_{48} \mathrm{H}_{41} \mathrm{O}_{13} \mathrm{NaP}$ 879.2, found $\mathrm{m} / \mathrm{z} 879.1(\mathrm{M}+\mathrm{Na})$.
Dibenzyl-(2,3,4,6-tetra-O-acetyl- $\alpha$-D-talopyranosyl) Phosphate (53).
Peracylated D-talose (52), ( $0.248 \mathrm{~g}, 0.636 \mathrm{mmol}$ ) gave 0.436 g substantially pure product ( $52 \%$ overall) $[\alpha]_{D}=40^{\circ}$ $\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.37(\mathrm{~m}, 10 \mathrm{H}), 5.68(\mathrm{dd}$, $\mathrm{J}=1.3,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.31(\mathrm{~m}, 1 \mathrm{H}), 5.20(\mathrm{t}, \mathrm{J}=3.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.11$ $(\mathrm{m}, 4 \mathrm{H}), 5.04(\mathrm{~d}, \mathrm{~J}=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.30(\mathrm{dd}, \mathrm{J}=1.3,6.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.11(\mathrm{dd}, \mathrm{J}=11.3,6.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.99(\mathrm{dd}, \mathrm{J}=11.3,6.7 \mathrm{~Hz}$, $1 \mathrm{H}), 2.12(\mathrm{~s}, 3 \mathrm{H}), 2.11(\mathrm{~s}, 3 \mathrm{H}), 1.99(\mathrm{~s}, 3 \mathrm{H}), 1.92(\mathrm{~s}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 170.7, 170.4, 169.9, 169.8, 135.7, 135.6, $135.5,129.2,129.1,129.0,128.5,128.4,96.3,70.3,68.9$, $67.0,65.6,64.9,61.7,21.1,21.0,20.9 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ -0.18; HRMS (FAB) caled for $\mathrm{C}_{28} \mathrm{H}_{34} \mathrm{O}_{13} \mathrm{P} 609.1737$, found $\mathrm{m} / \mathrm{z} 609.1747(\mathrm{M}+\mathrm{H})$.
General Strategy for Final Deprotection and Conversion to the Sodium Salt.
Final Deprotection and Conversion to sodium salt may take place in accordance with the present invention by the following reaction.

According to this method, protected $\alpha$-D-pyranosyl phosphate is dissolved in MEOH, $\mathrm{NaHCO}_{3}$ solution and $10 \%$ $\mathrm{Pd} / \mathrm{C}$ are added. The mixture is stirred overnight at room temperature under hydrogen atmosphere after which the catalyst is removed, preferably by filtration and the filtrate concentrated. The aqueous layer is preferably extracted, and then partitioned and submitted to an anion exchange column eluted with water, $0.1 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}, 0.2 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$ and $0.3 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$. The product eluted with $0.2 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$ and these fractions are pooled and co-evaporated with ethanol, preferably several times to remove excess $\mathrm{NH}_{4} \mathrm{HCO}_{3}$. The obtained sugar phosphate ammonium salt is subsequently dissolved in water and applied to a cationexchange column ( $\mathrm{Na}^{+}$type) eluted with mL water. The product containing fractions are collected and lyophilized to give the desired product as the sodium salt.

Other conditions, steps, reagents, solutions and the like of the present method may be used in accordance with the present invention.
In a typical reaction, the protected $\alpha$-D-pyranosyl phosphate ( 0.5 mmol ) was dissolved in $15 \mathrm{~mL} \mathrm{MeOH}, 1.5 \mathrm{~mL} 1 \mathrm{~N}$ $\mathrm{NaHCO}_{3}$ solution and $150 \mathrm{mg} 10 \% \mathrm{Pd} / \mathrm{C}$ were added. The mixture was stirred overnight at room temperature under hydrogen atmosphere after which the catalyst was removed by filtration and the filtrate concentrated to approximately a 10 mL volume. The aqueous layer was extracted with 10 mL of EtOAc, and then partitioned and submitted to an anion exchange column (Dowex $1 \times 8,1.2 \times 12 \mathrm{~cm}$ ) eluted with 100 mL water, $100 \mathrm{~mL} 0.1 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}, 100 \mathrm{~mL} 0.2 \mathrm{M}$ $\mathrm{NH}_{4} \mathrm{HCO}_{3}$ and $100 \mathrm{~mL} 0.3 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$. The product eluted with $0.2 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$ and these fractions were pooled and co-evaporated with ethanol several times to remove excess $\mathrm{NH}_{4} \mathrm{HCO}_{3}$. The obtained sugar phosphate ammonium salt was subsequently dissolved in 5 mL water and applied to an AG-X8 cation-exchange column ( $\mathrm{Na}^{+}$type) eluted with 100 mL water. The product containing fractions were collected and lyophilized to give the desired product as the sodium salt.
Disodium 6-deoxy- $\alpha$-D-glucopyranosyl Phosphate (12).
Compound $11(350 \mathrm{mg}, 0.5 \mathrm{mmol})$ gave $85 \mathrm{mg}(58 \%)$ of the desired sodium salt. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.37(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.4$,
$7.2 \mathrm{~Hz}), 3.98(\mathrm{~m}, 1 \mathrm{H}), 3.70(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz}), 3.45(\mathrm{~m}, 1 \mathrm{H})$, $3.09(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz}), 1.24(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 93.74,75.78,73.3,72.9,68.2,17.2 ;{ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 3.02; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{8} \mathrm{P} 243.0269$, found $\mathrm{m} / \mathrm{z} 243.0277$ (M+H).
Disodium 4-deoxy- $\alpha$-D-glucopyranosyl Phosphate (20).
Compound 19 ( $342 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) gave 78 mg of the title compound ( $55 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.49$ (dd, $1 \mathrm{H}, \mathrm{J}=3.4,7.32$ $\mathrm{Hz}), 4.16(\mathrm{~m}, 1 \mathrm{H}), 3.99(\mathrm{~m}, 1 \mathrm{H}), 3.65(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.2,12.0$ Hz ), 3.55 (dd, $1 \mathrm{H}, \mathrm{J}=6.0,12.0 \mathrm{~Hz}$ ), $3.41(\mathrm{~m}, 1 \mathrm{H}), 1.99-1.95$ $(\mathrm{m}, 1 \mathrm{H}), 1.44(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 95.1,73.8,69.5$, 67.4, 64.0, 34.3; ${ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 1.52; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{8} \mathrm{P} 243.0269$, found $\mathrm{m} / \mathrm{z} 243.0260(\mathrm{M}+\mathrm{H})$.
Disodium 3-deoxy- $\alpha$-D-glucopyranosyl Phosphate (28).
Compound 27 ( $270 \mathrm{mg}, 0.39 \mathrm{mmol}$ ) gave 65 mg title compound ( $58 \%$ ) as a $2: 1 \alpha / \beta$ mixture. ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 5.33 (dd, 1H, J=3.2, 7.3 Hz ), 3.92-3.50 (m, 5H), $3.46(\mathrm{~m}, 1 \mathrm{H})$, $2.33(\mathrm{~m}, 0.43 \mathrm{H}), 2.12(\mathrm{~m}, 1 \mathrm{H}), 1.81(\mathrm{~m}, 1 \mathrm{H}), 1.54(\mathrm{~m}$, $0.43 \mathrm{H}) ;{ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 3.39, 3.12; HRMS ( FAB ): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{8} \mathrm{P} 243.0269$, found $\mathrm{m} / \mathrm{z} 243.0267(\mathrm{M}+\mathrm{H})$.

Disodium 2-deoxy- $\alpha$-D-glucopyranosyl Phosphate (43).
Debenzylation of 41 ( $329 \mathrm{mg}, 0.447 \mathrm{mmol}$ ) was accomplished using the general strategy described above. After the filtrate was concentrated to approximately a 10 mL volume, the solution was cooled to $0^{\circ} \mathrm{C}$. and 1.5 mL 1 N NaOH solution was added in dropwise manner. The mixture was then stirred at room temperature for 4 hr and subsequently neutralized with 1.0 N acetic acid. The final work-up was accomplished as described in the general strategy to give 69 mg title compound ( $53 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.53(\mathrm{~m}, 1 \mathrm{H})$, $4.01(\mathrm{~m}, 1 \mathrm{H}), 3.88-3.84(\mathrm{~m}, 3 \mathrm{H}), 3.72(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=6.2,12.7$ Hz ), $3.31(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.4 \mathrm{~Hz}$ ), $2.19(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=5.0,12.9 \mathrm{~Hz}$ ), $1.66(\mathrm{~m}, \mathrm{H}) ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 2.68$; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{8} \mathrm{P} 243.0269$, found $\mathrm{m} / \mathrm{z} 243.0268(\mathrm{M}+\mathrm{H})$.
Disodium $\alpha$-D-gulopyranosyl Phosphate (33).
Compound $32(35 \mathrm{mg}, 0.044 \mathrm{mmol})$ gave 7.1 mg of the title compound $(55 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.15$ (dd, J=3.0, 7,7 $\mathrm{Hz}, 1 \mathrm{H}), 4.04(\mathrm{~m}, 2 \mathrm{H}), 3.79(\mathrm{~m}, 2 \mathrm{H}), 3.64(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 96.1,75.3,71.6,70.2,70.0,62.2 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 2.9$; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{9} \mathrm{P}$ 259.0218, found $\mathrm{m} / \mathrm{z} 259.0231(\mathrm{M}+\mathrm{H})$.
Disodium $\alpha$-D-allopyranosyl Phosphate (38).
Compound 37 ( $63 \mathrm{mg}, 0.079 \mathrm{mmol}$ ) gave 18 mg substantially pure product $(77 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.44$ (dd, $\mathrm{J}=3.5$, $7.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{~m}, 1 \mathrm{H}), 4.00(\mathrm{~m}, 1 \mathrm{H}), 3.90(\mathrm{dd}, \mathrm{J}=1.9$, $12.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.76 (m, 2H), 3.65 (dd, J=3.0, $10.4 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 95.8,75.1,72.6,71.8,68.1,62.6 . ;{ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.39; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{9} \mathrm{P}$ 259.0218, found $\mathrm{m} / \mathrm{z} 259.0217(\mathrm{M}+\mathrm{H})$.

Disodium $\alpha$-D-altropyranosyl Phosphate (47).
Using the strategy described for 43, compound 45 (260 $\mathrm{mg}, 0.3 \mathrm{mmol}$ ) gave 62 mg of the desired sodium salt ( $67 \%$ overall). ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.29$ (d, J=8.4 Hz, 1H), 4.14 (m, $1 \mathrm{H}), 3.98(\mathrm{~m}, 1 \mathrm{H}), 3.94(\mathrm{t}, \mathrm{J}=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{dd}, \mathrm{J}=2.4$, $12.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.82 (dd, J=3.5, $12.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.77 (dd, J=6.5, $12.3 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 94.9, 70.6, 70.5, 70.0, 64.8, 61.4; ${ }^{31}$ P NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.05; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12}$ $\mathrm{O}_{9} \mathrm{P} 259.0218$, found $\mathrm{m} / \mathrm{z} 259.0211(\mathrm{M}+\mathrm{H})$.
Disodium $\alpha$-D-idopyranosyl Phosphate (51).
Using the strategy described for 43, compound 49 (213 $\mathrm{mg}, 0.25 \mathrm{mmol}$ ) gave 61 mg of the title compound ( $62 \%$ overall). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}\right) 5.14$ (dd, $\left.\mathrm{J}=3.5,7.7 \mathrm{~Hz}, 1 \mathrm{H}\right), 4.24$ (m, 1H), $3.85(\mathrm{dd}, \mathrm{J}=8.9,12.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.75(\mathrm{~m}, 2 \mathrm{H}), 3.60$
$(\mathrm{t}, \mathrm{J}=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.32(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 99.1,75.8$, 75.5, 74.1, 63.9, 53.2; ${ }^{31} \mathrm{P}$ NMR ( $\mathrm{CDCl}_{3}$ ) 2.98; HRMS (FAB): caled for for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{9} \mathrm{P} 259.0218$, found $\mathrm{m} / \mathrm{z}$ $259.0208(\mathrm{M}+\mathrm{H})$.
Disodium $\alpha$-D-talopyranosyl Phosphate (55).
Using the strategy described for 43, compound 53 (436 $\mathrm{mg}, 0.72 \mathrm{mmol}$ ) gave 157 mg of the title compound ( $72 \%$ ).
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.48(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.11(\mathrm{~m}, 1 \mathrm{H}), 3.98$ $(\mathrm{t}, \mathrm{J}=3.2,1 \mathrm{H}), 3.92(\mathrm{~m}, 1 \mathrm{H}), 3.88(\mathrm{~m}, 1 \mathrm{H}), 3.82(\mathrm{dd}, \mathrm{J}=11.1$, $7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.75$ (dd, J=11.7, $4.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.19 (q, J=7.3 $\mathrm{Hz}, 10 \mathrm{H}), 1.28(\mathrm{t}, \mathrm{J}=7.4,15 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right): 94.1,70.2$, 68.8, 67.6, 62.8, 59.6; ${ }^{31}$ P NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 0.52; HRMS (FAB): calcd for $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{9} \mathrm{P} 259.0218$, found $\mathrm{m} / \mathrm{z} 259.0209(\mathrm{M}+\mathrm{H})$.

The following compounds were prepared, preferably according to the methods described herein.
(58) Thymidine 5 '-( $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 563.0705$; found $\mathrm{m} / \mathrm{z} 563.0679(\mathrm{M}+\mathrm{H})$.
(59) Uridine $5^{\prime}$-( $\alpha$-D-glucopyranosyl diphosphate). HRMS ( FAB ): calc for $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{17} \mathrm{~N}_{2} \mathrm{P}_{2} 565.0507$; found $\mathrm{m} / \mathrm{z}$ $565.0472(\mathrm{M}+\mathrm{H})$.
(60) Thymidine 5 '-(2-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2}$ 547.0704; found $\mathrm{m} / \mathrm{z} 547.0714(\mathrm{M}+\mathrm{H})$.
(61) Uridine 5'-(2-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0510(\mathrm{M}+\mathrm{H})$.
(62) Thymidine 5 '-(3-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB) calc for $\mathrm{C}_{15} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2}$ 547.0704; found $\mathrm{m} / \mathrm{z} 547.0720(\mathrm{M}+\mathrm{H})$.
(63) Uridine $5^{\prime}$-(3-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0485(\mathrm{M}+\mathrm{H})$.
(64) Thymidine 5 '-(4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB) calc for $\mathrm{C}_{15} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2}$ 547.0704; found $\mathrm{m} / \mathrm{z} 547.0693(\mathrm{M}+\mathrm{H})$.
(65) Uridine 5'-(4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{16} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0500(\mathrm{M}+\mathrm{H})$.
(66) Thymidine 5'-(6-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB) calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2}$ 547.0704; found $\mathrm{m} / \mathrm{z} 547.0730(\mathrm{M}+\mathrm{H})$.
(67) Uridine $5^{\prime}$-(6-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{O}_{15} \mathrm{~N}_{2} \mathrm{P}_{2} 549.0506$; found $\mathrm{m} / \mathrm{z}$ $549.0492(\mathrm{M}+\mathrm{H})$.
(68) Thymidine $5^{\prime}$-( $\alpha$-D-mannopyranosyl diphosphate). HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z} 563.0701$ $(\mathrm{M}+\mathrm{H})$.
(69) Uridine $5^{\prime}$-( $\alpha$-D-mannopyranosyl diphosphate). HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z} 565.0503(\mathrm{M}+\mathrm{H})$.
(70) Thymidine $5^{\prime}$-( $\alpha$-D-galactopyranosyl diphosphate). HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z} 563.0710$ $(\mathrm{M}+\mathrm{H})$.
(71) Uridine 5'-( $\alpha$-D-galactopyranosyl diphosphate). HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z} 565.0508(\mathrm{M}+\mathrm{H})$.
(72) Thymidine $5^{\prime}$-( $\alpha$-D-allopyranosyl diphosphate). HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z} 563.0715(\mathrm{M}+\mathrm{H})$.
(73) Uridine $5^{\prime}$-( $\alpha$-D-allopyranosyl diphosphate). HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z} 565.0507(\mathrm{M}+\mathrm{H})$.
(74) Thymidine $5^{\prime}$-( $\alpha$-D-altropyranosyl diphosphate). HRMS (FAB) calc 563.0705; found $\mathrm{m} / \mathrm{z} 563.0699$ $(\mathrm{M}+\mathrm{H})$.
(75) Uridine 5'-( $\alpha$-D-altropyranosyl diphosphate). HRMS (FAB): calc 565.0507; found $\mathrm{m} / \mathrm{z} 565.0511(\mathrm{M}+\mathrm{H})$.
(76) Thymidine 5'-( $\alpha$-D-gulopyranosyl diphosphate). HRMS (FAB) calc 563.0705; found $\mathrm{m} / \mathrm{z} 563.00712$ (M+H).
(77) Uridine $5^{\prime}$-( $\alpha$-D-gulopyranosyl diphosphate). HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z} 565.0512(\mathrm{M}+\mathrm{H})$.
(78) Thymidine 5'-( $\alpha$-D-idopyranosyl diphosphate). HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z} 563.0708(\mathrm{M}+\mathrm{H})$
(79) Uridine $5^{\prime}$-( $\alpha$-D-idopyranosyl diphosphate). HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z} 565.0507(\mathrm{M}+\mathrm{H})$.
(80) Thymidine $5^{\prime}$-( $\alpha$-D-talopyranosyl diphosphate). HRMS (FAB) calc 563.0705 ; found $\mathrm{m} / \mathrm{z} 563.0710(\mathrm{M}+\mathrm{H})$.
(81) Uridine $5^{\prime}$-( $\alpha$-D-talopyranosyl diphosphate). HRMS (FAB): calc 565.0507 ; found $\mathrm{m} / \mathrm{z} 565.0499(\mathrm{M}+\mathrm{H})$.

Enzyme Purification.
E. coli-prfbA-C (from Professor Hung-wen Liu (Dept. of Chem., Univ. of MN)) was grown in 2 L superbroth, $100 \mu \mathrm{~g}$ $\mathrm{mL}^{-1}$ ampicillin divided among two 4 L baffled flasks for 18 hours at $37^{\circ} \mathrm{C}$. Cells were harvested by centrifugation ( $5000 \times \mathrm{g}, 20 \mathrm{~min}, 4^{\circ} \mathrm{C}$.), washed twice with buffer A ( 50 mM potassium phosphate buffer, 1 mM EDTA, pH 7.5 ), resuspended in buffer A ( $4 \times$ weight) and split into two equal volumes. Each was sonicated by three 40 second bursts at $0^{\circ}$ C. followed by centrifugation ( $4400 \times \mathrm{g}, 20 \mathrm{~min}, 4^{\circ} \mathrm{C}$.) to remove cellular debris and a further 1.3 -fold dilution of the supernatant with buffer A. To the combined supernatant (167 mL ) was added $31.5 \mathrm{~mL} 5 \%$ streptomycin sulfate in a dropwise fashion followed by gentle stirring ( $1 \mathrm{hr}, 4^{\circ} \mathrm{C}$.) and centrifugation $\left(14,000 \times \mathrm{g}, 30 \mathrm{~min}, 4^{\circ} \mathrm{C}\right.$.) to remove precipitate. The supernatant was diluted ( 0.1 -fold 1 M potassium phosphate buffer, pH 7.5 ) followed by the slow addition of ammonium sulfate crystals to $65 \%$ saturation, gentle stirring ( $7.5 \mathrm{hr}, 4^{\circ} \mathrm{C}$.) and centrifugation ( $4200 \times \mathrm{g}, 30 \mathrm{~min}, 4^{\circ} \mathrm{C}$.). The precipitated protein was dissolved in a minimum amount of buffer A and dialyzed against buffer B ( 20 mM Tris. $\mathrm{HCl}, 1 \mathrm{mM}$ EDTA, pH 7.5 ). The dialysate was applied to a column of DE52 $(3 \mathrm{~cm} \times 15 \mathrm{~cm})$ which was washed with 50 mL buffer B and then eluted with a linear gradient (buffer B, $0-500 \mathrm{mM} \mathrm{NaCl}, 1.0 \mathrm{~mL} \mathrm{~min}^{-1}$ ). The $\mathrm{E}_{p}$ fractions (which eluted in the range of $35-75 \mathrm{mM} \mathrm{NaCl}$ ) were combined ( 24 mL ) and concentrated to 1 mL . Aliquots ( $300 \mu \mathrm{~L}$ ) were further resolved by FPLC (S-200, $20 \times 70 \mathrm{~cm}, 50 \mathrm{mM}$ Tris. $\mathrm{HCl}, 200 \mathrm{mM} \mathrm{NaCl}, \mathrm{pH} 7.5$ ). The $\mathrm{E}_{p}$ fractions were combined ( 7 mL ), concentrated ( $64 \mathrm{mg} \mathrm{min}{ }^{-1}$ ) and stored in aliquots ( 5,20 , and $200 \mu \mathrm{~L}$ ) at $-80^{\circ} \mathrm{C}$. until their use.
General Methods.
Infrared spectra were recorded on a Perkin Elmer 1600 series FTIR spectrophotometer. ${ }^{1} \mathrm{H}$ NMR spectra were obtained on a Bruker AMX $400(400 \mathrm{MHz})$ and are reported in parts per million ( $\delta$ ) relative to either tetramethylsilane ( 0.00 ppm ) or $\mathrm{CDCl}_{3}\left(7.25 \mathrm{ppm}\right.$ ) for spectra run in $\mathrm{CDCl}_{3}$ or relative to $\mathrm{D}_{2} \mathrm{O}(4.82 \mathrm{ppm})$ or $\mathrm{CD}_{3} \mathrm{OD}(3.35 \mathrm{ppm})$ for spectra run in $\mathrm{D}_{2} \mathrm{O}$. Coupling constants (J) are reported in hertz. ${ }^{13} \mathrm{C}$ NMR are reported in $\delta$ relative to $\mathrm{CDCl}_{3}(77.00$ $\mathrm{ppm})$ or $\mathrm{CD}_{3} \mathrm{OD}(49.05 \mathrm{ppm})$ as an internal reference and ${ }^{31} \mathrm{P}$ NMR spectra are reported in $\delta$ relative to $\mathrm{H}_{3} \mathrm{PO}_{4}(0.00$ ppm in $\mathrm{D}_{2} \mathrm{O}$ ). Routine mass spectra were recorded on a PE SCIEX API 100 LC/MS mass spectrometer and HRMS was accomplished by the University of California, Riverside Mass Spectrometry Facility. Optical rotations were recorded on a Jasco DIP-370 polarimeter using a 1.0 dm cell at the room temperature ( $25^{\circ} \mathrm{C}$.) and the reported concentrations. Melting points were measured with Electrothermal 1A-9100 digital melting bpoint instrument. Chemicals used were reagent grade and used as supplied except where noted. Analytical TLC was performed on Whatman AL Sil G/UV silica gel 60 plates. Compounds were visualized by spraying
$\mathrm{I}_{2} / \mathrm{KI} / \mathrm{H}_{2} \mathrm{SO}_{4}$ or by dipping the plates in a cerium sulfateammonium molybdate solution followed by heating. Liquid column chromatography was performed using forced flow of the indicated solvent on E. Merck silica gel $60(40-63 \mu \mathrm{~m})$ and high pressure liquid chromatography was performed on a RAININ Dynamax SD-200 controlled with Dynamax HPLC software.
Although the above methods of IR, NMR, Mass Spec, and HRMS were those used in these examples of the present invention, as indicated above, it would be known to those skilled in the art that other instruments, conditions, types of techniques, and the like may be used for visualization of compounds, to identify compounds and determine their concentrations and purity.

## General Strategy for Azide Formation.

Azides in accordance with the present invention may be formed according to the following method. Protected glycoside is dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The mixture is cooled to about $0^{\circ} \mathrm{C}$. and pyridine and $\left(\mathrm{CF}_{3} \mathrm{SO}_{2}\right)_{2} \mathrm{O}$ are added. The reaction was stirred for approximately 30 min at about $0^{\circ} \mathrm{C}$. and then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organics were washed with water, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The resulting crude residue was dissolved, preferably in anhydrous DMF, to which was added $\mathrm{NaN}_{3}$. The reaction was subsequently stirred, preferably overnight, at room temperature and then diluted with EtOAc. The organics were washed with water, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. Preferably, product purification was accomplished by flash chromatography.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.
In a typical reaction, the appropriately protected glycoside ( 2.1 mmol ) was dissolved in 10 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The mixture was cooled to $0^{\circ} \mathrm{C}$. to which was added pyridine ( 6.3 mmol ) and $\left(\mathrm{CF}_{3} \mathrm{SO}_{2}\right)_{2} \mathrm{O}(3.2 \mathrm{mmol})$. The reaction was stirred 30 $\min$ at $0^{\circ} \mathrm{C}$. and then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(150 \mathrm{~mL})$. The organics were washed with water ( 30 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The resulting crude residue was dissolved in 10 mL anhydrous DMF, to which was added $\mathrm{NaN}_{3}(407 \mathrm{mg}, 6.3 \mathrm{mmol})$. The reaction was subsequently stirred overnight at room temperature and then diluted with EtOAc ( 250 mL ). The organics were washed with water $(2 \times 30 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. Product purification was accomplished by flash chromatography (4:1 hexane/EtOAc).

Ethyl 4-azide-2,3,6-tri-O-benzyl-4-deoxy-1-thio- $\beta$-D-glucopyranoside (FIG. 3 (b) (94)).
Compound (AGCH) $93(310 \mathrm{mg}, 0.63 \mathrm{mmol})^{8}$ gave 285 $\mathrm{mg}(88 \%)$ desired product. $[\alpha]_{D}=62.3^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.40-7.31(\mathrm{~m}, 15 \mathrm{H}), 4.94(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz})$, $4.92(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.3 \mathrm{~Hz}), 4.84(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.6 \mathrm{~Hz}), 4.73(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=10.3 \mathrm{~Hz}), 4.64(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.0 \mathrm{~Hz}), 4.56(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.0$ Hz ), 4.44 (d, 1H, J=9.6 Hz), 3.77 (dd, $1 \mathrm{H}, \mathrm{J}=1.8,10.9 \mathrm{~Hz}$ ), $3.71-3.62(\mathrm{~m}, 2 \mathrm{H}), 3.54(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.4 \mathrm{~Hz}), 3.45(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.8$ $\mathrm{Hz}), 3.3(\mathrm{~m}, 1 \mathrm{H}), 2.84-2.69(\mathrm{~m}, 2 \mathrm{H}), 1.33(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 137.8,137.6,128.3,128.2,128.1,128.0$, $127.8,127.7,127.5,85.0,84.6,81.3,77.8,75.5,75.3,73.3$, 69.1, 61.9, 24.8, 15.0; MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{SNa}$ 542.2 , found $\mathrm{m} / \mathrm{z} 542.0(\mathrm{M}+\mathrm{Na})$.

Ethyl 4-azide-3-O-benzoyl-2-O-benzyl-4,6-dideoxy-1-thio-$\beta$-D-glucopyranoside (FIG. 3 (b) (106)).
Compound (FIG. 3(b) (105)) gave 0.78 g ( $87.4 \%$ ) substantially pure product. $[\alpha]_{D}=38^{\circ}\left(\mathrm{c}=1, \mathrm{CHC}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.03(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=8.2 \mathrm{~Hz}), 7.60(\mathrm{~m}, 1 \mathrm{H}), 7.47(\mathrm{t}, 2 \mathrm{H}$,
$\mathrm{J}=7.5 \mathrm{~Hz}), 7.30(\mathrm{~s}, 2 \mathrm{H}), 7.12(\mathrm{~m}, 3 \mathrm{H}), 5.43(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz})$, $4.80(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.7 \mathrm{~Hz}), 4.34(\mathrm{~m}, 2 \mathrm{H}), 3.54(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.5$ $\mathrm{Hz}), 3.44(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.9 \mathrm{~Hz}), 2.79(\mathrm{~m}, 2 \mathrm{H}), 1.42$ $(\mathrm{d}, 3 \mathrm{H}, \mathrm{J}=6.0 \mathrm{~Hz}), 1.34(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $165.9,137.5,133.8,130.2,129.8,128.9,128.7,128.6$, $128.2,85.4,79.6,77.7,76.7,75.4,75.1,66.7,25.7,19.0$, 15.4; MS: calcd for $\mathrm{C}_{22} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{SNa} 450.1$, found $\mathrm{m} / \mathrm{z}$ $450.0(\mathrm{M}+\mathrm{Na})$.

Ethyl 3-O-benzoyl-2-O-benzyl-6-deoxy-1-thio- $\beta$-D-galactopyranoside (FIG. 3 (b) (105)).

Ethyl 2,3,4-tri-O-acetyl-6-deoxy-1-thio- $\beta$-D-galactopyranoside (FIG. $3(b)$ (104)), $2.72 \mathrm{~g}, 8.14 \mathrm{mmol}$ ) was dissolved in 30 mL MeOH to which $1.2 \mathrm{~mL} 25 \%$ sodium methoxide was added. From this reaction, 1.58 g ( $93.3 \%$ ) ethyl 6 -deoxy-1-thio- $\beta$-D-glactopyranoside was obtained after purification which was combined with TsOH ( $140 \mathrm{mg}, 0.73$ $\mathrm{mmol})$ and 2,2-dimethoxypropane ( $1.9 \mathrm{~mL}, 15.4 \mathrm{mmol}$ ) in 15 mL anhydrous DMF. The reaction was stirred overnight at room temperature, diluted with 200 mL EtOAc and washed successively with saturated $\mathrm{NaHCO}_{3}$ solution ( 50 mL ) and water ( 30 mL ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and purified via silica gel chromatography ( $3: 1$ hexane/EtOAc) to afford 1.73 g ( $86 \%$ ) of purified ethyl 6-deoxy-3,4-O-isopropylidene-1-thio- $\alpha$-D-galactopyranoside. $[\alpha]_{D}=11.9^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 4.19(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}), 4.01(\mathrm{~m}, 2 \mathrm{H}), 3.84(\mathrm{dq}, 1 \mathrm{H}, \mathrm{J}=1.7,13.1 \mathrm{~Hz})$, $3.50(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=6.2,10.2 \mathrm{~Hz}), 2.71(\mathrm{~m}, 2 \mathrm{H}), 1.59(\mathrm{~s}, 3 \mathrm{H})$, 1.37 (d, 3H, J=6.6 Hz), 1.33 (s, 3H), 1.28 (t, 3H, J=7.5 Hz); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 110.2,85.5,79.5,76.8 .73 .2,72.3,28.6$, 26.7, 24.6, 17.2, 15.6; MS: calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{O}_{4} \mathrm{SNa}$ 271.1, found $\mathrm{m} / \mathrm{z} 270.9(\mathrm{M}+\mathrm{Na})$.

The obtained ethyl 6-deoxy-3,4-O-isopropylidene-1-thio-$\alpha$-D-galactopyranoside ( $1.50 \mathrm{~g}, 6.0 \mathrm{mmol}$ ) was combined with of $60 \%$ sodium hydride $(0.36 \mathrm{~g}, 9 \mathrm{mmol})$ and benzyl bromide ( $1.44 \mathrm{~mL}, 12.1 \mathrm{mmol}$ ) in 20 mL dry DMF. The reaction was stirred overnight and $1.7 \mathrm{~g}(83 \%)$ ethyl 2-O-benzyl-6-deoxy-3,4-O-isopropylidene-1-thio- $\alpha$-D-galactopyranoside was obtained after the typical work up and purification. $[\alpha]_{D}=-2.8^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $7.35(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 7.25(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}), 7.09(\mathrm{~m}, 1 \mathrm{H})$, 4.77 (d, 1H, J=11.4 Hz), 4.69 (d, 1H, J=11.4 Hz), 4.31 (d, $1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}$ ), $4.11(\mathrm{~m}, 1 \mathrm{H}), 3.96$ (dd, 1H, J=2.0, 15.6 Hz ), 3.73 (m, 1H), 3.36 (dd, 1H, J=6.7, 19.8 Hz ), 2.64 (m, 2H), $1.42(\mathrm{~s}, 3 \mathrm{H}), 1.29(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.6 \mathrm{~Hz}), 1.27(\mathrm{~s}, 3 \mathrm{H}), 1.22(\mathrm{t}, 3 \mathrm{H}$, $\mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 137.9,128.3,128.2,127.6$, $109.5,83.3,79.7,78.0,76.5,73.4,72.4,28.0,26.4,24.4$, $16.8,14.8$; MS: calcd for $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{SNa} 361.1$, found $\mathrm{m} / \mathrm{z}$ $361.0(\mathrm{M}+\mathrm{Na})$.

The obtained ethyl 2-O-benzyl-6-deoxy-3,4-O-isopropy-lidene-1-thio- $\beta$-D-galactopyranoside ( $1.82 \mathrm{~g}, 5.38 \mathrm{mmol}$ ) was dissolved in a mixture solution including 15 mL 0.5 M HCl and 45 mL MeOH and the mixture was subsequently refluxed for 30 min . The reaction was cooled to room temperature, neutralized with solid $\mathrm{NaHCO}_{3}$, and the resulting mixture concentrated. The concentrate was diluted with EtOAc $(250 \mathrm{~mL})$, washed with water $(2 \times 20 \mathrm{~mL})$ and brine $(20 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and purified by flash chromatography ( $1: 1$ hexane/EtOAc) to give $1.51 \mathrm{~g}(94 \%)$ substantially pure ethyl 2-O-benzyl-6-deoxy-1-thio- $\beta$-D-galactopyranoside. $[\alpha]_{D}=8.4^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 7.42-7.29 (m, 5H), $4.97(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}), 4.67(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=11.0 \mathrm{~Hz}), 4.40(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.6 \mathrm{~Hz}), 3.75(\mathrm{~m}, 1 \mathrm{H}), 3.61(\mathrm{~m}$, $2 \mathrm{H}), 3.45(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.3 \mathrm{~Hz}), 2.78$ (m, 2H), 2.48 (d, 1H, J=5.0 $\mathrm{Hz}), 2.14(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=5.0 \mathrm{~Hz}), 1.32(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 138.5,129.0,128.7,128.5,85.1,79.3,77.6,75.7$,
75.6, 74.8,72.2, 25.4, 16.9, 15.4; MS: calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{4} \mathrm{SNa} 321.1$, found $\mathrm{m} / \mathrm{z} 321.0(\mathrm{M}+\mathrm{Na})$.

To a solution of ethyl 2-O-benzyl-6-deoxy-1-thio- $\beta$-Dgalactopyranoside ( $1.03 \mathrm{~g}, 3.45 \mathrm{mmol}$ ) and DMAP ( 126 mg , 1.0 mmol ) in 10 mL of dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $-30^{\circ} \mathrm{C}$. was added $\mathrm{Et}_{3} \mathrm{~N}(1.92 \mathrm{~mL}, 13.8 \mathrm{mmol})$. Benzoyl chloride ( $0.4 \mathrm{~mL}, 3.45$ mmol ) was added to this mixture in a dropwise fashion, and the stirred at $-30^{\circ} \mathrm{C}$. for 3 hr . The reaction was then quenched by the addition of $\mathrm{MeOH}(2 \mathrm{~mL})$ and the mixture was gradually warmed to room temperature after which the resulting mixture was diluted with EtOAc ( 250 mL ). The solution was washed with saturated $\mathrm{NaHCO}_{3}$ solution ( $2 \times 20$ mL ), water ( 30 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated and purified by flash chromatography ( $3: 1$ to $1: 1$ hexane/EtOAc) to give $1.12 \mathrm{~g}(80 \%)$ of the title product. $[\alpha]_{D}=96.9^{\circ}(\mathrm{c}=1$, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 8.09-8.03(\mathrm{~m}, 2 \mathrm{H}), 7.56(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{J}=7.4 \mathrm{~Hz}), 7.49(\mathrm{~m}, 2 \mathrm{H}), 7.22(\mathrm{~m}, 2 \mathrm{H}), 7.18(\mathrm{~m}, 3 \mathrm{H}), 5.28$ (dd, 1H, J=3.0, 9.6 Hz ), 4.87 (d, 1H, J=10.6 Hz), $4.67(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=10.6 \mathrm{~Hz}), 4.56(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.7 \mathrm{~Hz}), 4.12(\mathrm{~m}, 1 \mathrm{H}), 3.85(\mathrm{t}$, $1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}), 3.80(\mathrm{~m}, 1 \mathrm{H}), 2.81(\mathrm{~m}, 2 \mathrm{H}), 1.93(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=6.7 \mathrm{~Hz}), 1.36(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 166.2,138.0$, $133.7,130.2,130.1,128.9,128.7,128.6,128.2,85.7,78.1$, $77.6,76.5,76.0,74.7,70.9,25.6,16.9,15.4$; MS: calcd for $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{5} \mathrm{SNa} 425.1$, found $\mathrm{m} / \mathrm{z} 425.2(\mathrm{M}+\mathrm{Na})$.
Strategy for Formation of Protected Ethyl 1-thio- $\beta$-D-hexopyranosides.

Ethyl 1-thio- $\beta$-D-hexopyranosides may be generally formed as set forth above. The following method is another exemplary embodiment of such method used in accordance with the present invention. In a typical reaction, a mixture of 4.0 mmol protected monosaccharide, 1.5 mL (ethylthio) trimethylsilane ( 8.0 mmol ) and 1.95 g zinc iodide ( 7.8 mmol ) in 30 mL dry dichloromethane was refluxed for 30 min under argon atmosphere. The reaction was then cooled, 50 mL water was added after which the mixture was extracted with chloroform ( $3 \times 50 \mathrm{~mL}$ ). The combined organic extracts were washed successively with water ( 30 mL ), saturated $\mathrm{NaHCO}_{3}$ solution ( 30 mL ) and brine ( 30 mL ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated and resolved by silica gel chromatography ( $2: 1$ hexanes/ $\mathrm{EtOAc})$ to give the desired product.

Again, variations of this method may be made in accordance with the present invention, as would be apparent to one skilled in the art.
Ethyl 2,4,6-tri-O-acetyl-3-azide-3-deoxy-1-thio- $\beta$-D-glucopyranoside (FIG. $\mathbf{3}(b)$ (99)).

Compound (FIG. 3 (b) (99)) ( $1.5 \mathrm{~g}, 4.0 \mathrm{mmol}$ ) gave 1.26 $\mathrm{g}(83.5 \%)$ title compound. $[\alpha]_{D}=-49.4^{\circ}\left(\mathrm{c}=0.5, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 4.95(\mathrm{~m}, 2 \mathrm{H}), 4.43(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.9 \mathrm{~Hz}), 4.19$ (dd, 1H, J=4.1, 12.4 Hz ), $4.09(\mathrm{~m}, 1 \mathrm{H}), 3.65(\mathrm{~m}, 2 \mathrm{H}), 2.68$ $(\mathrm{m}, 2 \mathrm{H}), 2.12(\mathrm{~s}, 3 \mathrm{H}), 2.10(\mathrm{~s}, 3 \mathrm{H}), 1.24(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz})$, $2.06(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 171.0, 169.6, 169.6, 84.2, $76.8,70.3,68.7,66.1,62.6,24.4,21.2,21.1,21.0,15.1 ; \mathrm{MS}$ : calcd for $\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{7} \mathrm{SNa} 398.1$, found $\mathrm{m} / \mathrm{z} 397.9(\mathrm{M}+\mathrm{Na})$.

Ethy1 2,3,4-tri-O-acetyl-6-azide-6-deoxy-1-thio- $\beta$-D-glucopyranoside (FIG. $\mathbf{3}(b)$ (88)).

Compound (FIG. $3(b)(87))(680 \mathrm{mg}, 1.8 \mathrm{mmol})^{6}$ gave $590 \mathrm{mg}(86 \%)$ of the desired title compound. $[\alpha]_{D}=-17.5^{\circ}$ $\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 5.23(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.4 \mathrm{~Hz}), 5.02$ $(\mathrm{m}, 2 \mathrm{H}), 4.54(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.0 \mathrm{~Hz}), 3.62(\mathrm{~m}, 1 \mathrm{H}), 3.37(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{J}=6.5,13.5 \mathrm{~Hz}$ ), $3.30(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=2.8,13.5 \mathrm{~Hz}), 2.73(\mathrm{~m}$, $2 \mathrm{H}), 2.07(\mathrm{~s}, 3 \mathrm{H}), 2.04(\mathrm{~s}, 3 \mathrm{H}), 2.02(\mathrm{~s}, 3 \mathrm{H}), 1.28(\mathrm{t}, 3 \mathrm{H}$, $\mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 170.0,169.4,169.2,83.0$, $77.2,73.6,69.7,69.3,51.0,23.6,20.6,20.5,14.6$. MS: calcd for $\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{7} \mathrm{SNa} 398.1$, found $\mathrm{m} / \mathrm{z} 397.5(\mathrm{M}+\mathrm{Na})$.

Ethyl 2,3,4-tri-O-acetyl-6-deoxy-1-thio- $\beta$-D-galactopyranoside (FIG. 3 (b) (104)).

Compound (FIG. 3(b) (103)) ( 6.1 mmol ) gave 1.73 g ( $83 \%$ ) of the substantially pure product. $[\alpha]_{D}=-17.5^{\circ}(\mathrm{c}=1$, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 5.28(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=3.3 \mathrm{~Hz}), 5.22(\mathrm{t}$, $1 \mathrm{H}, \mathrm{J}=9.9 \mathrm{~Hz}$ ), $5.05(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.4,9.9 \mathrm{~Hz}), 4.46(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=9.9 \mathrm{~Hz}), 3.82(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=6.4,12.8 \mathrm{~Hz}), 2.74(\mathrm{~m}, 2 \mathrm{H}), 2.17$ (s, 3H), 2.06 (s, 3H), 1.98 (s, 3H), 1.28 (t, 3H, J=7.4 Hz), $1.22(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.4 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 171.0, 170.6, 170.1, 83.9, 77.6, 73.6, 72.7, 70.8, 67.7, 24.5, 21.3, 21.1, 21.0, 16.8, 15.1; MS: calcd for $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{O}_{7} \mathrm{SNa} 357.1$, found $\mathrm{m} / \mathrm{z} 356.6(\mathrm{M}+\mathrm{Na})$.

General Strategy for O-Acetyl to O-Benzyl Conversion.
O-Acetyl may be converted to O-Benzyl according to the following method or other methods known to those skilled in the art. Protected ethyl 1-thio- $\beta$-D-hexopyranoside was dissolved in dry MeOH and toluene to which a sodium methoxide solution is added. The mixture was stirred for about $2^{1 / 2} \mathrm{M}$ to about $3^{1 / 2} \mathrm{hrs}$. at room temperature and neutralized. The organics are then concentrated and the corresponding crude unprotected 1 -ethylthio- $\beta$-D-glucopyranoside directly dissolved in dry DMF. To this mixture NaH and benzyl bromide is added. The reaction is stirred at room temperature, preferably overnight. The mixture was then diluted with EtOAc, washed with $\mathrm{H}_{2} \mathrm{O}$, brine and organics, dried, concentrated, and purified to give the purified product.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.

In a typical reaction, 2.8 mmol of protected ethyl 1-thio-$\beta$-D-hexopyranoside was dissolved in 20 mL dry MeOH and 5 mL toluene to which 0.5 mL of a sodium methoxide solution ( $25 \% \mathrm{NaOMe}$ in methanol) was added. The mixture was stirred for 3 hr at room temperature and neutralized with DOWEX 50W X8-100 resin. The organics were concentrated and the corresponding crude unprotected 1-ethylthio-$\beta$-D-glucopyranoside directly dissolved in 15 mL dry DMF. To this mixture $330 \mathrm{mg} 60 \% \mathrm{NaH}(8.25 \mathrm{mmol})$ and 1.6 mL benzyl bromide was added. The reaction was stirred at room temperature overnight. The mixture was then diluted with 200 mL EtOAc, washed with $\mathrm{H}_{2} \mathrm{O}(2 \times 30 \mathrm{~mL})$, brine ( 30 mL ) and the organics dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated and purified by silica gel chromatography ( $8: 1$ hexane/EtOAc) to give the purified product.

Ethyl 3-azide-2,4,6-tri-O-benzyl-3-deoxy-1-thio- $\beta$-D-glucopyranoside (FIG. 3 (b) (100)).

Compound (FIG. 3 (b) (99)) ( $1.05 \mathrm{~g}, 2.8 \mathrm{mmol}$ ) gave 1.03 $\mathrm{g}(71 \%)$ of the desired title compound. $[\alpha]_{D}=-13.6^{\circ}$ (c=1, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.36-7.28(\mathrm{~m}, 15 \mathrm{H}), 4.90(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=10.1 \mathrm{~Hz}$ ), $4.79(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.6 \mathrm{~Hz}), 4.74(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.1$ $\mathrm{Hz}), 4.60(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.1 \mathrm{~Hz}), 4.54-4.47(\mathrm{~m}, 2 \mathrm{H}), 4.43(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=9.6 \mathrm{~Hz}), 3.70(\mathrm{~m}, 1 \mathrm{H}), 3.57(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.1 \mathrm{~Hz}), 3.45(\mathrm{~m}$, $2 \mathrm{H}), 3.26(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz}), 2.75(\mathrm{~m}, 2 \mathrm{H}), 1.32(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.3$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 138.4, 137.9, 137.8, 129.5, 129.2, 129.1, 128.9, 128.8, 128.7, 128.6, 128.5, 128.4, 128.3, $128.2,128.1,128.0,127.7,85.7,80.4,79.7,76.7,75.8,75.3$, $73.9,72.5,71.0,69.1,25.6,15.6$; MS: caled for $\mathrm{C}_{29} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{SNa} 542.2$, found $\mathrm{m} / \mathrm{z} 542.0(\mathrm{M}+\mathrm{Na})$.

Ethyl 6-azide-2,3,4-tri-O-benzyl-6-deoxy-1-thio- $\beta$-D-glucopyranoside (FIG. 3(b) (89)).

Compound (FIG. 3 (b) ( 88 )) ( $560 \mathrm{mg}, 1.5 \mathrm{mmol}$ ) gave 645 $\mathrm{mg}(85 \%)$ of the desired product. $[\alpha]_{D}=7.9^{\circ}\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.32-7.16(\mathrm{~m}, 15 \mathrm{H}), 4.87(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.9$ Hz ), $4.86(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}), 4.79(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.2 \mathrm{~Hz}), 4.76$ (d, 1H, J=11.0 Hz), $4.66(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}), 4.50(\mathrm{~d}, 1 \mathrm{H}$,
$\mathrm{J}=11.0 \mathrm{~Hz}), 4.43(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}), 3.62(\mathrm{~m}, 1 \mathrm{H}), 3.43-3.35$ (m, 4H), $3.24(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=6.0,13.1 \mathrm{~Hz}$ ), $2.72(\mathrm{~m}, 2 \mathrm{H}), 1.26$ (t, 3H, J=1.5 Hz); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 138.2,137.7,137.5$, $128.5,128.4,128.3,128.2,128.0,127.8,127.7,86.3,84.6$, 81.5, 78.4, 78.2, 75.7, 75.4, 75.1, 51.3, 24.4, 14.9; MS: calcd for $\mathrm{C}_{29} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{SNa} 542.2$, found $\mathrm{m} / \mathrm{z} 541.9(\mathrm{M}+\mathrm{Na})$.
General Strategy for Conversion of Azides to Acetamides.
Azides may be converted to Acetamides according to the following method or other methods known to those skilled in the art. Benzyl-protected ethyl 1-thio- $\beta$-D-azidodeoxyhexopyranoside and $\mathrm{SnCl}_{2}$ are combined in acetonitrile. To this mixture thiophenol and $\mathrm{Et}_{3} \mathrm{~N}$ are added and the reaction is stirred for about $1 / 2$ to about $11 / 2 \mathrm{hr}$ at room temperature. The mixture is then diluted with EtOAc and washed, preferably with 2 N NaOH , water, and brine. The organics are dried, preferably over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated to dryness and the crude residue dissolved in dry pyridine. To this mixture 2 mL acetic anhydride is added and the reaction stirred, preferably overnight, at room temperature. The reaction is concentrated and purified directly by silica gel chromatography ( $3: 2$ to $1: 1$ hexane/EtOAc) to give the purified product.
Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.
In a typical reaction, benzyl-protected ethyl 1 -thio- $\beta$-Dazidodeoxyhexopyranoside ( 2.8 mmol ) and $\mathrm{SnCl}_{2}(1.73$ mmol ) were combined in 10 mL of acetonitrile. To this mixture thiophenol ( 6.9 mmol ) and $\mathrm{Et}_{3} \mathrm{~N}(5.2 \mathrm{mmol})$ were added and the reaction was stirred for 1 hr at room temperature under argon atmosphere. The mixture was then diluted with EtOAc ( 150 mL ) and washed with 2 N NaOH $(2 \times 2 \mathrm{~mL})$, water $(20 \mathrm{~mL})$ and brine ( 30 mL ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, concentrated to dryness and the crude residue dissolved in 10 mL dry pyridine. To this mixture 2 mL acetic anhydride was added and the reaction stirred overnight at room temperature. The reaction was concentrated and purified directly by silica gel chromatography ( $3: 2$ to $1: 1$ hexane/EtOAc) to give the purified product.
Ethy1 3-acetamido-2,4,6-tri-O-benzyl-3-deoxy-1-thio- $\beta$-Dglucopyranoside (FIG. 3(b) (101)).
Compound (FIG. 3(b) (100)) ( $600 \mathrm{mg}, 1.15 \mathrm{mmol}$ ) gave $523 \mathrm{mg}(85 \%)$ of the desired product. $[\alpha]_{D}=-5.4^{\circ}(\mathrm{c}=1$, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 7.27-7.17(\mathrm{~m}, 15 \mathrm{H}), 5.57(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=8.6 \mathrm{~Hz}), 4.73(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.2 \mathrm{~Hz}), 4.55-4.39(\mathrm{~m}, 5 \mathrm{H}), 3.97$ (dd, 1H, J=8.3, 16.5 Hz ), 3.66 (m, 2H), 3.58 (dd, 1H, J=4.1, $10.8 \mathrm{~Hz}), 3.50(\mathrm{~m}, 1 \mathrm{H}), 3.44(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=8.4 \mathrm{~Hz}), 2.71(\mathrm{~m}, 2 \mathrm{H})$, $1.60(\mathrm{~s}, 3 \mathrm{H}), 1.24(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $170.0,137.9,137.8,137.7,128.7,128.3,128.2,128.0$, $127.9,127.7,127.6,85.3,80.1,78.8,75.4,73.7,73.5,73.1$, $69.4,55.7,25.2,23.4,15.0 ;$ MS: calcd for $\mathrm{C}_{31} \mathrm{H}_{37} \mathrm{NO}_{5} \mathrm{SNa}$ 558.2 , found $\mathrm{m} / \mathrm{z} 558.0(\mathrm{M}+\mathrm{Na})$.

Ethyl 4-acetamido-2,3,6-tri-O-benzyl-4-deoxy-1-thio- $\beta$-Dglucopyranoside (FIG. $\mathbf{3}(b)(95)$ ).
Compound (FIG. 3 (b) (94)) ( $640 \mathrm{mg}, 1.23 \mathrm{mmol}$ ) gave 530 mg desired product $(80 \%)[\alpha]_{D}=-36.6^{\circ}\left(\mathrm{c}=0.5, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.32-7.10(\mathrm{~m}, 15 \mathrm{H}), 5.13(\mathrm{br}, 1 \mathrm{H}), 4.85$ (d, 1H, J=10.2 Hz), 4.75 (d, 1H, J=11.7 Hz), $4.70(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=10.7 \mathrm{~Hz}), 4.64(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}), 4.43(\mathrm{~m}, 3 \mathrm{H})$, $3.64-3.47(\mathrm{~m}, 5 \mathrm{H}), 3.37(\mathrm{~m}, 1 \mathrm{H}), 2.67(\mathrm{~m}, 2 \mathrm{H}), 1.61(\mathrm{~s}, 3 \mathrm{H})$, $1.25(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 170.3,138.3$, $138.0,137.8,128.5,128.3,128.2,128.1,127.8,127.7$, $127.6,84.9,82.1,81.7,78.1,75.3,74.7,73.4,70.0,52.6$, 24.9, 23.3, 15.1; MS: calcd for $\mathrm{C}_{31} \mathrm{H}_{37} \mathrm{NO}_{5} \mathrm{SNa} 558.2$, found $\mathrm{m} / \mathrm{z} 557.9(\mathrm{M}+\mathrm{Na})$.

Ethyl 6-acetamido-2,3,4-tri-O-benzyl-6-deoxy-1-thio- $\beta$-Dglucopyranoside (FIG. 3 (b) (90)).

Compound (FIG. 3(b) (89)) ( $502 \mathrm{mg}, 0.97 \mathrm{mmol}$ ) gave 450 mg desired product ( $87 \%$ ). $[\alpha]_{D}=-20.4^{\circ} \quad(\mathrm{c}=1.0$, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.51-7.23(\mathrm{~m}, 15 \mathrm{H}), 5.87(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=4.6 \mathrm{~Hz}), 4.99-4.78(\mathrm{~m}, 4 \mathrm{H}), 4.73(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}), 4.63$ ( $\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=10.4 \mathrm{~Hz}$ ), $4.45(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}), 3.70-3.60(\mathrm{~m}$, $2 \mathrm{H}), 3.52(\mathrm{~m}, 1 \mathrm{H}), 3.41-3.34(\mathrm{~m}, 3 \mathrm{H}), 2.74(\mathrm{~m}, 2 \mathrm{H}), 1.95(\mathrm{~s}$, $3 \mathrm{H}), 1.32(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 169.8, 138.1, 137.6, 137.5, 128.4, 128.3, 128.2, 128.1, 127.9, 127.8, $127.6,86.2,85.1,81.5,78.5,77.1,75.6,75.4,75.1,39.9$, 25.2, 23.1, 15.1; MS: calcd for $\mathrm{C}_{31} \mathrm{H}_{37} \mathrm{NO}_{5} \mathrm{SNa} 558.2$, found $\mathrm{m} / \mathrm{z} 558.2$ (M+Na).

## Phosphorylation Procedure.

As set forth in the methods above, phosphorylation according to the present invention may occur via a protected ethyl 1-thio- $\beta$-D-hexopyranoside. The following method is another exemplary embodiment of such method used in accordance with the present invention.

In a typical reaction, 1.13 mmol protected ethyl 1-thio-$\beta$-D-hexopyranoside and 1.7 mmol dibenzyl phosphate were co-evaporated two times from dry toluene and further dried under high vacuum for 4 hr to which 1.36 mmol of N -iodosuccinamide and 500 mg of dry molecular sieves were added. The mixture was then dissolved in 10 mL dry dichloromethane, cooled to $-30^{\circ} \mathrm{C}$. and $30 \mu \mathrm{~L}$ of trifluoromethanesulfonic acid ( 0.34 mmol ) was added. The reaction was maintained at $-30^{\circ} \mathrm{C}$. for 30 min with stirring and then diluted with EtOAc ( 150 mL ), washed with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(20 \mathrm{~mL})$, saturated $\mathrm{NaHCO}_{3}(20 \mathrm{~mL}), \mathrm{H}_{2} \mathrm{O}(20$ mL ), and brine ( 30 mL ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, concentrated and purified by chromatography on silica gel ( $3: 1$ hexane/EtOAc) to give the desired product.

Again, variations of this method may be made in accordance with the present invention, as would be apparent to one skilled in the art. Dibenzyl-(3-azide-2,4,6-tri-O-benzyl-3-deoxy- $\alpha$-D-glucopyranosyl) phosphate (FIG. 3(b) (100a)).

Compound (FIG. $3(b)(100))(590 \mathrm{mg}, 1.55 \mathrm{mmol})$ gave $700 \mathrm{mg}(84 \%)$ of the title compound. $[\alpha]_{D}=57.8^{\circ}(\mathrm{c}=1$, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.45-7.27$ (m, 25H), 5.99 (dd, $1 \mathrm{H}, \mathrm{J}=3.2,6.8 \mathrm{~Hz}$ ), $5.11-5.05(\mathrm{~m}, 4 \mathrm{H}), 4.84(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.6$ $\mathrm{Hz}), 4.82(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.4 \mathrm{~Hz}), 4.72(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.5 \mathrm{~Hz}), 4.60$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=12.0 \mathrm{~Hz}), 4.49(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.7 \mathrm{~Hz}), 4.46(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=12.1 \mathrm{~Hz}), 3.84(\mathrm{~m}, 2 \mathrm{H}), 3.68(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.0,10.9 \mathrm{~Hz})$, $3.57(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}), 3.48(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 137.9, 137.6, 137.4, 136.2, 136.1, 136.0, 129.0, 128.9, 128.7, 128.6, 128.5, 128.4, 128.3, 128.1, 128.0, 94.9, 77.4, $76.1,75.7,75.3,74.0,73.2,72.4,69.9,69.8,69.7,69.6,67.9$, 65.2; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) \quad 0.82$; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{42} \mathrm{~N}_{3} \mathrm{O}_{8} \mathrm{PNa} 758.2$, found $\mathrm{m} / \mathrm{z} 758.2(\mathrm{M}+\mathrm{Na})$.
Dibenzyl-(3-acetamido-2,4,6-tri-O-benzyl-3-deoxy- $\alpha$-Dglucopyranosyl) Phosphate (FIG. 3(b) (101a)).

Compound (FIG. $3(b)(101))(490 \mathrm{mg}, 0.91 \mathrm{mmol})$ gave $480 \mathrm{mg}(70 \%)$ of the desired product. $[\alpha]_{D}=52^{\circ}(\mathrm{c}=1$, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.40-7.19(\mathrm{~m}, 25 \mathrm{H}), 5.94(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{J}=3.2,6.7 \mathrm{~Hz}), 5.07(\mathrm{br}, 1 \mathrm{H}), 4.97(\mathrm{~m}, 4 \mathrm{H}), 4.63(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=11.7 \mathrm{~Hz}), 4.56(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.0 \mathrm{~Hz}), 4.39(\mathrm{~m}, 4 \mathrm{H}), 3.90(\mathrm{~m}$, $2 \mathrm{H}), 3.85(\mathrm{~m}, 2 \mathrm{H}), 3.56(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.3,11.0 \mathrm{~Hz}), 3.39(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{J}=1.6,11.0 \mathrm{~Hz}), 1.76(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 170.7$, $137.8,137.5,137.4,135.6,135.5,128.4,128.3,128.2$, $127.9,127.8,127.7,127.6,127.5,94.9,77.2,75.1,74.0$, 73.3, 72.8, 72.3, 69.3, 69.4, 69.0, 67.9, 53.4, 23.4; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.62$; MS: calcd for $\mathrm{C}_{43} \mathrm{H}_{45} \mathrm{NO}_{9} \mathrm{PNa} 774.3$, found $\mathrm{m} / \mathrm{z} 774.3(\mathrm{M}+\mathrm{Na})$.

Dibenzyl-(4-azide-2,3,6-tri-O-benzyl-4-deoxy- $\alpha$-D-glucopyranosyl) Phosphate. (FIG. 3 (b) (94a))

Compound (FIG. 3(b) (94)) ( $280 \mathrm{mg}, 054 \mathrm{mmol}$ ) gave $316 \mathrm{mg}(80 \%)$ of the desired product. $[\alpha]_{D}=105.8^{\circ}(\mathrm{c}=1$, $\mathrm{CHCl}_{3}$ ); 7.28-7.14 (m, 25H), $5.85(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.2,6.8 \mathrm{~Hz})$, 5.12-4.96 (m, 5H), $4.82(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.6 \mathrm{~Hz}), 4.71-4.66(\mathrm{~m}$, $2 \mathrm{H}), 4.57(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.3 \mathrm{~Hz}), 4.50(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.1 \mathrm{~Hz}), 4.37$ (d, 1H, J=12.1 Hz), 3.68-3.47 (m, 5H), 3.37 (dd, 1H, J=1.5, $11.0 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 137.7,137.6,137.5,135.7$, $135.6,128.5,128.4,128.3,128.2,128.2,128.0,127.9$, $127.8,127.7,127.6,127.5,95.4,78.9,78.8,75.6,75.5,73.4$, $72.9,71.5,69.3,69.2,69.2,67.9,60.8 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ 0.82 ; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{42} \mathrm{~N}_{3} \mathrm{O}_{8} \mathrm{PNa} 758.2$, found $\mathrm{m} / \mathrm{z}$ 758.0 (M+Na).

5 Dibenzyl-(4-acetamido-2,3,6-tri-O-benzyl-4-deoxy- $\alpha$-Dglucopyranosyl) Phosphate. (FIG. 3 (b) (95a))

Compound (FIG. 3(b) (95)) ( $430 \mathrm{mg}, 0.80 \mathrm{mmol}$ ) gave $389 \mathrm{mg}(65 \%)$ of the desired product. $[\alpha]_{D}=35^{\circ}$ (c=1, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) 7.23-7.09(\mathrm{~m}, 25 \mathrm{H}), 5.82(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{J}=3.2,6.8 \mathrm{~Hz}$ ), 5.47 (d, 1H, J=8.5 Hz), $5.01-4.93$ (m, $4 \mathrm{H}), 4.69(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7 \mathrm{~Hz}), 4.59(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.1 \mathrm{~Hz}), 4.54$ $(\mathrm{m}, 2 \mathrm{H}), 4.35(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.9 \mathrm{~Hz}), 4.30(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.9 \mathrm{~Hz})$, $3.92(\mathrm{~m}, 1 \mathrm{H}), 3.86(\mathrm{~m}, 1 \mathrm{H}), 3.77(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.7 \mathrm{~Hz}), 3.55(\mathrm{~m}$, $1 \mathrm{H}), 3.39(\mathrm{~m}, 2 \mathrm{H}), 1.65(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 170.0$, $138.2,137.7,137.3135 .7,135.6,135.5128 .3,128.2,128.1$, $128.0,127.9,127.8,127.7,127.6,127.4,95.6,79.4,79.3$, 77.2, 74.5. 73.3, 72.8, 72.2, 69.2, 50.8, 23.2; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.71$; MS: calcd for $\mathrm{C}_{43} \mathrm{H}_{46} \mathrm{NO}_{9} \mathrm{PNa} 774.3$, found $\mathrm{m} / \mathrm{z} 774.3$ (M+Na)
Dibenzyl-(6-azide-2,3,4-tri-O-benzyl-6-deoxy- $\alpha$-D-glucopyranosyl) Phosphate. (FIG. 3 (b) (89a))

Compound (FIG. $3(b)(89)$ ) ( $430 \mathrm{mg}, 0.76 \mathrm{mmol}$ ) gave $285 \mathrm{mg}(51 \%)$ of the desired product and 160 mg the $\beta$ isomer. $[\alpha]_{D}=41.5^{\circ} \quad\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \quad \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right)$ $7.28-7.16(\mathrm{~m}, 25 \mathrm{H}), 5.87(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.2,6.7 \mathrm{~Hz}), 5.05-4.92$ (m, 4H), $4.85(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.9 \mathrm{~Hz}), 4.81(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz})$, $4.71(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.3 \mathrm{~Hz}), 4.70(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.9 \mathrm{~Hz}), 4.60(\mathrm{~d}$, $1 \mathrm{H}, \mathrm{J}=11.3 \mathrm{~Hz}), 4.50(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.0 \mathrm{~Hz}), 3.81(\mathrm{~m}, 2 \mathrm{H}), 3.54$ (dt, $1 \mathrm{H}, \mathrm{J}=9.5,3.1 \mathrm{~Hz}$ ), $3.47(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz}$ ), $3.18(\mathrm{~m}, 2 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 138.2,137.7,137.3,135.7,135.6,128.5$, $128.4,128.3,128.0,127.9,127.8,127.7,127.6,95.1,80.7$, $79.3,79.2,77.1,75.5,75.1,73.0,71.9,69.3,69.2,69.2,69.1$, $50.7 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) \quad 0.75$; MS: calcd for $\mathrm{C}_{41} \mathrm{H}_{42} \mathrm{~N}_{3} \mathrm{O}_{8} \mathrm{PNa} 758.2$, found $\mathrm{m} / \mathrm{z} 758.1(\mathrm{M}+\mathrm{Na})$.
Dibenzyl-(6-acetamido-2,3,4-tri-O-benzyl-6-deoxy- $\alpha$-Dglucopyranosyl) Phosphate. (FIG. 3 (b) (90a))
Compound (FIG. 3(b) (90)) ( $430 \mathrm{mg}, 0.80 \mathrm{mmol}$ ) gave $389 \mathrm{mg}(65.0 \%)$ of the desired product. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $7.27-7.19(\mathrm{~m}, 25 \mathrm{H}), 6.00(\mathrm{br}, 1 \mathrm{H}), 5.68(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.4,5.5$ $\mathrm{Hz})$, 4.99-4.93 (m, 4H), 4.85(d, 1H, J=11.9 Hz), $4.76(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=10.6 \mathrm{~Hz}), 4.72(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.5 \mathrm{~Hz}), 4.65(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.5$ $\mathrm{Hz}), 4.60(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.5 \mathrm{~Hz}), 4.57(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=10.5 \mathrm{~Hz}), 3.81$ $(\mathrm{m}, 2 \mathrm{H}), 3.49(\mathrm{dt}, 1 \mathrm{H}, \mathrm{J}=3.5,9.4 \mathrm{~Hz}), 3.44(\mathrm{~m}, 2 \mathrm{H}), 3.24(\mathrm{t}$, $1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 178.5,138.3,138.0$, $137.9,136.2,136.1,129.0,128.9,128.8,128.5,128.4$, $128.3,128.1,95.5,75.7,75.6,74.6,73.9,73.4,72.9,70.0$, 69.9, 69.6, 69.5, 68.5, 54.0, 29.9; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.53$; MS: calcd for $\mathrm{C}_{43} \mathrm{H}_{46} \mathrm{NO}_{9} \mathrm{PNa} 774.6$, found $\mathrm{m} / \mathrm{z} 774.3$ ( $\mathrm{M}+\mathrm{Na}$ ).
Dibenzyl-(4-azide-3-O-benzoyl-2-O-benzyl-4,6-dideoxy- $\alpha$ -D-glucopyranosyl) Phosphate. (FIG. 3(b) (106a))

Compound (FIG. $3(b)(106))(323 \mathrm{mg}, 0.76 \mathrm{mmol})$ gave $350 \mathrm{mg}(72 \%)$ substantially pure product. $[\alpha]_{D}=100.1^{\circ}$ $\left(\mathrm{c}=1, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 8.16(\mathrm{~m}, 2 \mathrm{H}), 7.62(\mathrm{~m}$, $1 \mathrm{H}), 7.49(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.9 \mathrm{~Hz}), 7.34-714(\mathrm{~m}, 11 \mathrm{H}), 5.98(\mathrm{dd}$,
$1 \mathrm{H}, \mathrm{J}=3.2,7.1 \mathrm{~Hz}), 5.68(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.8 \mathrm{~Hz}), 5.13-5.05(\mathrm{~m}$, $4 \mathrm{H}), 4.68(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.1 \mathrm{~Hz}), 4.50(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.1 \mathrm{~Hz})$, $3.81(\mathrm{~m}, 1 \mathrm{H}), 3.68(\mathrm{dt}, 1 \mathrm{H}, \mathrm{J}=3.0,9.8 \mathrm{~Hz}), 3.28(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=10.0$ Hz ), $1.24(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 185.8$, $137.2,136.1,136.0,133.8,130.3,129.9,129.0,128.9$, $128.8,128.6,128.4,128.1,95.1,77.7,76.5,72.7,72.1,70.1$, 70.1, 69.7, 69.7, 68.7, 66.3, 18.6; ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}\right) 0.52$; MS: calcd for $\mathrm{C}_{34} \mathrm{H}_{34} \mathrm{~N}_{3} \mathrm{O}_{8} \mathrm{PNa} 666.2$ found $\mathrm{m} / \mathrm{z} 666.2$ ( $\mathrm{M}+\mathrm{Na}$ ).
Strategy for Final Deprotection and Conversion to the Sodium Salt.

Set forth above is a general strategy for final deprotection and conversion to the sodium salt according to the present invention. The following method is another exemplary embodiment of such method used in accordance with the present invention.

In a typical reaction, the protected $\alpha$-D-pyranosyl phosphate ( 0.5 mmol ) was dissolved in 15 mL MeOH, 1.5 mL 1 N $\mathrm{NaHCO}_{3}$ solution and $150 \mathrm{mg} 10 \% \mathrm{Pd} / \mathrm{C}$ were added. The mixture was stirred overnight at room temperature under hydrogen atmosphere after which the catalyst was removed by filtration and the filtrate concentrated and redissolved 10 mL water. The aqueous layer was extracted with EtOAc ( 10 mL ), and then submitted to an anion exchange column (Dowex $1 \times 8,1.2 \times 12 \mathrm{~cm}$ ) eluted with 100 mL water, 100 mL $0.1 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}, 100 \mathrm{~mL} 0.2 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$ and 100 mL 0.3 $\mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$. The product eluted with $0.2 \mathrm{M} \mathrm{NH}_{4} \mathrm{HCO}_{3}$ and these fractions were pooled and co-evaporated with ethanol several times to remove excess $\mathrm{NH}_{4} \mathrm{HCO}_{3}$. The obtained sugar phosphate ammonium salt was subsequently dissolved in 5 mL water and applied to an AG-X8 cation-exchange column ( $\mathrm{Na}^{+}$type) eluted with 100 mL water. The product containing fractions were collected and lyophilized to give the desired product as the sodium salt.

Again, variations of this method may be made in accordance with the present invention, as would be apparent to one skilled in the art.

Disodium (3-amino-3-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3(b) (96)).

Dibenzyl-(3-azide-2,4,6-tri-O-benzyl-3-deoxy- $\alpha-D-g l u-$ copyranosyl) phosphate ( $250 \mathrm{mg}, 0.34 \mathrm{mmol}$ ) gave 68 mg ( $66 \%$ ) of the title compound. $[\alpha]_{D}=68.1^{\circ}\left(\mathrm{c}=1, \mathrm{H}_{2} \mathrm{O}\right) ;{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.46(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.0,7.0 \mathrm{~Hz}), 3.93$ (m, 1H), 3.85 (m, 1H), 3.74 (dd, 1H, J=4.5, 12.5 Hz ), 3.69 (m, 1H), 3.58 $(\mathrm{m}, 1 \mathrm{H}), 3.45(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 91.9$, $71.2,68.4,65.3,59.3,54.8 ;{ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.85; HRMS: calcd for $\mathrm{C}_{6} \mathrm{H}_{13} \mathrm{NO}_{8} \mathrm{P} 258.0379$, found $\mathrm{m} / \mathrm{z} 258.0372$ ( $\mathrm{M}+\mathrm{H}$ ).
Disodium-(3-acetamido-3-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3 (b) (97)).

Dibenzyl-(3-acetamido-2,4,6-tri-O-benzyl-3-deoxy- $\alpha$-Dglucopyranosyl) phosphate ( $280 \mathrm{mg}, 0.37 \mathrm{mmol}$ ) gave 59 $\mathrm{mg}(53 \%)$ of the desired product. $[\alpha]_{D}=93.6^{\circ}\left(\mathrm{c}=1, \mathrm{H}_{2} \mathrm{O}\right)$; ${ }^{1}$ HNMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.43(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=2.9,6.5 \mathrm{~Hz}), 4.07(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{J}=10.3 \mathrm{~Hz}$ ), 3.88 (dd, $1 \mathrm{H}, \mathrm{J}=2.5,9.7 \mathrm{~Hz}$ ), $3.80(\mathrm{~m}, 1 \mathrm{H})$, $3.71(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=4.7,12.3 \mathrm{~Hz}), 3.57(\mathrm{~m}, 1 \mathrm{H}), 3.40(\mathrm{t}, 1 \mathrm{H}, 10.1$ Hz ), 2.01(s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 174.4, 92.8, 71.6, 69.5, 67.1, 59.8, 53.3, 21.5; ${ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.07; HRMS: calcd for $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{9} \mathrm{P} 300.0484$, found $\mathrm{m} / \mathrm{z} 300.0478(\mathrm{M}+\mathrm{H})$.

Disodium-(4-amino-4-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3(b) (91)).

Dibenzyl-(4-azide-2,3,6-tri-O-benzyl-4-deoxy- $\alpha-D-g l u-$ copyranosyl) phosphate ( $350 \mathrm{mg}, 0.476 \mathrm{mmol}$ ) gave 77 mg 6 ( $54 \%$ ) the desired product. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.46$ (dd, 1 H , $\mathrm{J}=3.2,7.1 \mathrm{~Hz}), 4.11(\mathrm{~m}, 1 \mathrm{H}), 3.90-3.75(\mathrm{~m}, 3 \mathrm{H}), 3.59(\mathrm{~m}$,
$1 \mathrm{H}), 3.13(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=10.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 92.9, 71.3, 68.8, 68.2, 59.8, 51.7; ${ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.80; HRMS: calcd for $\mathrm{C}_{6} \mathrm{H}_{13} \mathrm{NO}_{8} \mathrm{P} 258.0379$, found $\mathrm{m} / \mathrm{z} 258.0372(\mathrm{M}+\mathrm{H})$.

Disodium-(4-acetamido-4-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3 (b) (92)).
Dibenzyl-(4-acetamido-2,3,6-tri-O-benzyl-4-deoxy- $\alpha$-Dglucopyranosyl) phosphate ( $370 \mathrm{mg}, 0.50 \mathrm{mmol}$ ) gave 120 $\mathrm{mg}(71 \%)$ of the desired product. $[\alpha]_{D}=109.2^{\circ}\left(\mathrm{c}=1, \mathrm{H}_{2} \mathrm{O}\right)$; ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.44$ (dd, $1 \mathrm{H}, \mathrm{J}=3.3,7.2 \mathrm{~Hz}$ ), $3.89(\mathrm{~m}, 1 \mathrm{H})$, $3.76(\mathrm{~m}, 2 \mathrm{H}), 3.64(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J} 12.4,1.2 \mathrm{~Hz}), 3.53(\mathrm{~m}, 2 \mathrm{H})$, $1.99(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 173.8, 93.2, 71.4, 70.4, 69.8, $60.0,50.6,21.3 ;{ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 1.93; HRMS: calcd for $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{9} \mathrm{P} 300.0484$, found $\mathrm{m} / \mathrm{z} 300.0499(\mathrm{M}+\mathrm{H})$.

Disodium-(6-amino-6-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3 (b) (85)):
Dibenzyl-(6-azide-2,3,4-tri-O-benzyl-6-deoxy- $\alpha$-D-glucopyranosyl) phosphate ( $360 \mathrm{mg}, 0.49 \mathrm{mmol}$ ) gave 85 mg ( $57 \%$ ) of the title compound. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.47(\mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}=3.5,6.8 \mathrm{~Hz}$ ), $4.14(\mathrm{dt}, 1 \mathrm{H}, \mathrm{J}=2.5,12.6 \mathrm{~Hz}$ ), $3.78(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{J}=9.5 \mathrm{~Hz}), 3.55(\mathrm{~m}, 2 \mathrm{H}), 3.33(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.3 \mathrm{~Hz}), 3.07(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{J}=10.3,12.9 \mathrm{~Hz}$ ) ${ }^{13}{ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 94.1, 73.4, 72.5, 72.4, 72.3, 68.6, 41.0; ${ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.80; HRMS: calcd for $\mathrm{C}_{6} \mathrm{H}_{13} \mathrm{NO}_{8} \mathrm{P} 258.0379$, found $\mathrm{m} / \mathrm{z} 258.0388(\mathrm{M}+\mathrm{H})$.
Disodium-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3 (b) (86)).
Dibenzyl-(6-acetamido-2,3,4-tri-O-benzyl-6-deoxy- $\alpha$-Dglucopyranosyl) phosphate ( $340 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) gave 124 $\mathrm{mg}(79.4 \%)$ of the desired product. $[\alpha]_{D}=60.5^{\circ}\left(\mathrm{c}=1, \mathrm{H}_{2} \mathrm{O}\right)$; ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) 5.39(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=3.2,6.5 \mathrm{~Hz}), 3.95(\mathrm{t}, 1 \mathrm{H}$, $\mathrm{J}=7.1 \mathrm{~Hz}), 3.73(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.4 \mathrm{~Hz}), 3.54(\mathrm{~m}, 1 \mathrm{H}), 3.45(\mathrm{~m}$, $1 \mathrm{H}), 3.34(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=6.7,14.1 \mathrm{~Hz}), 3.25(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.5 \mathrm{~Hz})$, $1.99(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 178.4, 97.4, 76.7, 75.9, 74.8, 73.8, 43.9, 25.6; ${ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$ ) 2.98; HRMS: caled for $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{9} \mathrm{P} 300.0484$, found $\mathrm{m} / \mathrm{z} 300.0482(\mathrm{M}+\mathrm{H})$
Disodium-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl) Phosphate (FIG. 3(b) (102))
Dibenzyl-(4-azide-3-O-benzoy1-2-O-benzyl-4,6-
dideoxy- $\alpha$-D-glucopyranosyl) phosphate ( $300 \mathrm{mg}, 0.466$ mmol ) was dissolved in a mixture of 10 mL of MeOH and 2 mL of toluene. To this solution was added 1.4 mL 1 N NaOH and 100 mg of $10 \% \mathrm{Pd} / \mathrm{C}$ and the reaction stirred overnight under hydrogen atmosphere. The catalyst was removed by filtration, the filtrate concentrated to a volume of 4 mL , cooled to $0^{\circ} \mathrm{C}$., and 0.7 mL 1 N NaOH solution was added in a dropwise fashion. The mixture was stirred for 3 hr at $0^{\circ} \mathrm{C}$., neutralized with 1 N HOAc and the product purified via anion exchange as described in the general procedure above to give $86 \mathrm{mg}(67 \%)$ of the substantially pure product. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right) 5.44$ (dd, $1 \mathrm{H}, \mathrm{J}=3.2,6.7 \mathrm{~Hz}$ ), $4.24(\mathrm{~m}, 1 \mathrm{H}), 3.88(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=9.7 \mathrm{~Hz}), 3.56(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=1.3,9.4$ $\mathrm{Hz}), 2.94(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=10.3 \mathrm{~Hz}), 1.32(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}\right) ~ 92.9,71.5,68.3,64.2,56.6,16.2 ;{ }^{31} \mathrm{P}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right)$ 2.16. HRMS: calcd for $\mathrm{C}_{6} \mathrm{H}_{13} \mathrm{NO}_{7} \mathrm{P} 242.0429$, found $\mathrm{m} / \mathrm{z} 242.0441(\mathrm{M}+\mathrm{H})$

## $\mathrm{E}_{p}$-Catalyzed Conversion.

A reaction containing 2.5 mM NTP, 5.0 mM sugar phosphate, 5.5 mM MgCl 2 and 10 U inorganic pyrophosphatase in a total volume of $50 \mu \mathrm{~L} 50 \mathrm{mM}$ potassium phosphate buffer, pH 7.5 at $37^{\circ} \mathrm{C}$. was initiated by the addition of 3.52 $\mathrm{UE}_{p}(1 \mathrm{U}=$ the amount of protein needed to produce $1 \mu \mathrm{~mol}$ TDP- $\alpha$-D-glucose $\mathrm{min}^{-1}$ ) The reaction was incubated with slow agitation for 30 min at $37^{\circ} \mathrm{C}$., quenched with MeOH ( $50 \mu \mathrm{~L}$ ), centrifuged ( $5 \mathrm{~min}, 14,000 \times \mathrm{g}$ ) and the supernatant was stored at $-20^{\circ} \mathrm{C}$. until analysis by HPLC. Samples (30
$\mu \mathrm{L}$ ) were resolved on a Sphereclone 5u SAX column (250× 4.6 mm ) fitted with a guard column ( $30 \times 4.6 \mathrm{~mm}$ ) using a linear gradient ( $50-200 \mathrm{mM}$ potassium phosphate buffer, pH $\left.5.0,1.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}, \mathrm{~A}_{275} \mathrm{~nm}\right)$.

The following compounds were prepared, preferably 5 according to the methods described herein:
(109) Thymidine $5^{5}$-(6-amino-6-deoxy- $\alpha-\mathrm{D}$-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 562.0839 ; found $\mathrm{m} / \mathrm{z} 562.0837(\mathrm{M}+\mathrm{H})$.
(110) Uridine 5 '-( 6 -amino-6-deoxy- $\alpha-\mathrm{D}$-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 564.0632; found $\mathrm{m} / \mathrm{z} 564.0640(\mathrm{M}+\mathrm{H})$.
(111) Thymidine 5'-(4-amino-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 562.0839; found $\mathrm{m} / \mathrm{z} 562.0848(\mathrm{M}+\mathrm{H})$.
(112) Uridine 5 '-(4-amino-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 564.0632; found $\mathrm{m} / \mathrm{z} 564.0638(\mathrm{M}+\mathrm{H})$.
(113) Thymidine 5'-(3-amino-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 562.0839 ; found $\mathrm{m} / \mathrm{z} 562.0835(\mathrm{M}+\mathrm{H})$.
(114) Uridine $5^{\prime}$-(3-amino-3-deoxy- $\alpha-\mathrm{D}$-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 564.0632; found $\mathrm{m} / \mathrm{z} 564.0622(\mathrm{M}+\mathrm{H})$.
(115) Thymidine 5'-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl 2 diphosphate). HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 562.0839; found $\mathrm{m} / \mathrm{z} 562.0842(\mathrm{M}+\mathrm{H})$.
(116) Uridine 5 '-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 564.0632; found $\mathrm{m} / \mathrm{z} 564.0630(\mathrm{M}+\mathrm{H})$.
(117) Thymidine $5^{\prime}$-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 604.0945$; found $\mathrm{m} / \mathrm{z} 604.0953(\mathrm{M}+\mathrm{H})$.
(118) Uridine 5'-(6-acetamido-6-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2}$ 606.0737; found $\mathrm{m} / \mathrm{z} 606.0732(\mathrm{M}+\mathrm{H})$.
(119) Thymidine $5^{\prime}$-(4-acetamido-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2}$ 604.0945; found $\mathrm{m} / \mathrm{z} 604.0940(\mathrm{M}+\mathrm{H})$.
(120) Uridine 5 '-(4-acetamido-4-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2}$ 606.0737; found $\mathrm{m} / \mathrm{z} 606.0730(\mathrm{M}+\mathrm{H})$.
(121) Thymidine 5 '-(3-acetamido-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 604.0945$; found $\mathrm{m} / \mathrm{z} 604.0947(\mathrm{M}+\mathrm{H})$.
(122) Uridine 5 '-(3-acetamido-3-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2}$ 606.0737; found $\mathrm{m} / \mathrm{z} 606.0735(\mathrm{M}+\mathrm{H})$.
(123) Thymidine $5^{\prime}$-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{16} \mathrm{~N}_{3} \mathrm{P}_{2} 604.0945$; found $\mathrm{m} / \mathrm{z} 604.0951(\mathrm{M}+\mathrm{H})$.
(124) Uridine $5^{\prime}$-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{17} \mathrm{~N}_{3} \mathrm{P}_{2}$ 606.0737; found $\mathrm{m} / \mathrm{z} 606.0738(\mathrm{M}+\mathrm{H})$.
(125) Thymidine $5^{\prime}$-(4-amino-4,6-dideoxy-a-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{14} \mathrm{~N}_{3} \mathrm{P}_{2} 546.0889$; found $\mathrm{m} / \mathrm{z} 546.0895(\mathrm{M}+\mathrm{H})$.
(126) Uridine $5^{\prime}$-(4-amino-4,6-dideoxy- $\alpha$-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{15} \mathrm{~N}_{3} \mathrm{P}_{2}$ 548.0682; found $\mathrm{m} / \mathrm{z} 548.0673(\mathrm{M}+\mathrm{H})$.

Structure-Based Engineering of $\mathrm{E}_{P}$ Expression, Purification and Mutagenesis of $\mathrm{E}_{p}$. $\mathrm{E}_{p}$ may be modified in accordance with the present invention according to the following method: $\mathrm{E}_{p}$ and $\mathrm{E}_{p}$ mutants are expressed and purified by methods known in the art. For seleno-methionine-labeled protein, the expresion
vector was transformed into the methionine auxotroph $E$. coli B834 and grown, preferably overnight at a temperature of about $25^{\circ} \mathrm{C}$. to about $35^{\circ} \mathrm{C}$., preferably about $30^{\circ} \mathrm{C}$. in the presence of seleno-methionine. Seleno-methionine-labeled $\mathrm{E}_{p}$ is purified using the standard protocol but in the presence of DTT. All $\mathrm{E}_{p}$ mutant gene cassettes are generated by a two-step PCR approach. Mutant genes are subsequently characterized by dsDNA sequencing of both strands
According to a preferred method, expression and purification and $\mathrm{E}_{p}$ and $\mathrm{E}_{p}$ mutants were accomplished as described in Jiang, J., Biggins, J. B. \& Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP- and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000). For seleno-methioninelabeled protein, the expresion vector was transformed into the methionine auxotroph $E$. coli B834 and grown overnight at $30^{\circ} \mathrm{C}$. in the presence of $50 \mathrm{mg} \mathrm{L}^{-1}$ seleno-methionine. Seleno-methionine-labeled $\mathrm{E}_{p}$ was purified using the standard protocol but in the presence of 5 mM DTT. No additional proteolysis or modifications during this process were observed by mass spectrometry. All $\mathrm{E}_{p}$ mutant gene cassettes were generated by a two-step PCR approach. Mutant genes were subsequently characterized by dsDNA sequencing of both strands.
Crystallization. A general crystallization technique that may be used in accordance with the present invention, is as follows: Purified $\mathrm{E}_{p}$ is concentrated in a buffer, and crystallized in a hanging drop by vapor diffusion at approximately room temperature ( $20^{\circ} \mathrm{C}$.). $\mathrm{E}_{p}$-dTTP crystals are obtained against reservoir containing TTP, 2.0 M ammonium phosphate, 0.1 M Tris. $\mathrm{HCl}, \mathrm{pH} 8.5$, and 20 mM MgCl 2 . Crystals grow with two monomers (half of the Ep tetramer) in the asymmetric unit. The $\mathrm{E}_{p}$-UDP-Glc crystals were obtained against a reservoir containing 2 mM UDP-Glc, 1.9 M ammonium sulfate, isopropanol.

According to an exemplary method, the purified $\mathrm{E}_{p}$ was concentrated to $20 \mathrm{mg} \mathrm{mL}^{-1}$ in a buffer containing 10 mM $\mathrm{KCl}, 2 \mathrm{mM} \mathrm{MgCl} \mathrm{I}_{2}$ and 10 mM HEPES, pH 7.2 , and crystallized in a hanging drop by vapor diffusion at room temperature ( $20^{\circ} \mathrm{C}$.). The $\mathrm{E}_{p}$-dTTP crystals were obtained against reservoir containing $2 \mathrm{mM} \mathrm{TTP}, 2.0 \mathrm{M}$ ammonium phosphate, 0.1 M Tris. $\mathrm{HCl}, \mathrm{pH} 8.5$, and $20 \mathrm{mM} \mathrm{MgCl}{ }_{2}$. Crystals grow in the tetragonal space group $\mathrm{P}_{3} 2_{1}{ }_{2}$ ( $\mathrm{a}=\mathrm{b}=120 \AA, \mathrm{c}=94 \AA$ ) with two monomers (half of the $\mathrm{E}_{p}$ tetramer) in the asymmetric unit. The $\mathrm{E}_{P}$-UDP-Glc crystals were obtained against a reservoir containing 2 mM UDPGle, 1.9 M ammonium sulfate, and $7.5 \%$ isopropanol. These crystals grow in the orthorhombic space group $\mathrm{P}_{1} 2_{1} 2_{1}$ ( $a=93 \AA, b=112 \AA, c=132 \AA$ ) with four monomers (one tetramer) in the asymmetric unit.

Data Collection and Structure Determination.
Data may be collected and structure determination made according to methods that would be known to those skilled in the art, including, for example, x-ray crystallography.

According to an exemplary embodiment, crystals were harvested and flash frozen in the cold stream of an X-Stream cooling system (Rigaku) in the mother liquor with added $20-25 \%$ glycerol as a cryoprotectant. Data was collected either in house using a Rigaku RAXIS-IV imaging plate area detector, or at the NSLS Brookhaven beamline X9B. Oscillation photographs were integrated, scaled and merged using DENZO and SCALEPACK. (Otwinowski, Z. \& Minor, W. Data Collection and Processing., Sawyer, L., Isaacs, N. \& Bailey, S. Ed. SERC Daresbury Laboratory: Warrington, UK. 556-562 (1993).) Subsequent calculations were performed with the CCP4 program suite. (CCP4, The CCP4 suite: programs for X-ray crystallography. Acta Crystallogr.

D, 50, 760-763 (1994).) The $\mathrm{E}_{p}$-UDP-Glc structure was determined using the single wavelength anomalous diffraction phasing method. (Hendrickson, W. A., Determination of Macromolecular Structures from Anomalous Diffraction of Synchrotron Radiation. Science, 254, 51-58 (1991).) Only the dataset collected at the wavelength of the selenium absorption peak was processed. Peak wavelength anomalous data were input to the program SnB to identify the location of the Se atoms. Twenty peaks from the best solution were refined using MLPHARE (CCP4) employing only the peak wavelength anomalous differences in the resolution range 35 to 2 A . Additional Se sites were located using anomalousdifference fourier maps. The final round of MLPHARE consisted of 47 Se sites. Seven of these sites correspond to Se-methionines with dual sidechain conformation. The phases calculated from MLPHARE had a figure of merit of 0.34 which was improved to 0.72 by density modification with the program DM (CCP4). The resulting electron density map was clearly interpretable, indicating also the correct handedness of the Se substructure. The map was further improved using free atom refinement and the automatic chain tracing procedure of the wARP program. Out of the 1156 residues, the main chain of 1003 were automatically traced and very clear density could be seen for the rest of the structure. The unambiguous tracing and sequence assignment of the $\mathrm{E}_{p}$ tetramer was completed using the 0 program. (Jones, T. A., et al., Improved Methods for Building Protein Models in Electron Density Maps and the Location of Errors in these Models. Acta Crystallogr., A47, 110-119 (1991).) Refinement of the model by conventional least-squares algorithm was done with XPLOR. (Brünger, A. T., X-PLOR v. 3.1 Manual. New Haven: Yale University (1993).) The final refined $\mathrm{E}_{p}$ tetramer model at 2.0 A resolution had a free R-value (Brünger, A. T., Free R Value: A Novel Statistical Quantity for Assessing the Accuracy of Crystal Structures. Nature, 355, 472-475 (1992)) of 22.3\% and included 9938 non-hydrogen atoms in 1156 well-ordered residues (1-289 in each monomer) and 762 water molecules. In our determination, electron density was lacking for only the 3 C-terminal residues of each monomer. Restrained refinement of 4 temperature factors was monitored throughout by the free

R-factor criterion as set forth in Liu, H.-w. \& Thorson, J. S. Pathways and Mechanisms in the Biogenesis of Novel Deoxysugars by Bacteria. Ann. Rev. Microbiol. 48, 223-256 (1994) and Johnson, D. A. \& Liu, H.-w. Mechanisms and Pathways from Recent Deoxysugar Biosynthesis Research. Curr. Opin. Chem. Biol. 2, 642-649 (1998)). Stereochemical analysis of the refined model using PROCHECK (CCP4 suite) revealed main-chain and side-chain parameters better than, or within, the typical range of values for protein structures determined at $2.0 \AA$ resolution (overall G-factor, 2.2). None of the $E_{p}$ residues fell in the disallowed region of the Ramachandran plot. (Ramachandran, G. N., Ramakrishnan, C. \& Sasisekharan, V. Stereochemistry of Polypeptide Chain Configuration. J. Molec. Biol., 7, 95-99 (1963).) The $\mathrm{E}_{p}$-dTTP structure was determined using the Molecular Replacement (MR) method, with our $\mathrm{E}_{p}$-UDP-Glc structure as a search model and the program XPLOR. The final refined model (half of the $\mathrm{E}_{p}$ tetramer) at $2.1 \AA$ resolution had a free R value ${ }^{35}$ of $23.5 \%$ and included 5017 non-hydrogen atoms in 578 well defined in the electron density map amino acids (1-289 for each monomer), and 387 water molecules. The PROCHECK overall G-factor is 2.5 , and none of the $\mathrm{E}_{p}$ residues fell in the disallowed region of the Ramachandran plot.

Enzyme Assays and Determination of Steady State Kinetic Parameters.

Assays for product formation and steady state kinetics were accomplished using conditions similar to those described. in Jiang, J., Biggins, J. B. \& Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP- and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000). For the mutant pool assays, an aliquot which contained an eqimolar ratio of each mutant ( $60 \mu \mathrm{~g}$ ) was utilized.
While the present invention is described with respect to particular examples and preferred embodiments, it is understood that the present invention is not limited to these examples and embodiments. The present invention as claimed therefore, includes variations from the particular examples and preferred embodiments described herein, as will be apparent to one of skill in the art.

SEQUENCE LISTING


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Ile Val Leu Ala Val Asn Tyr Arg Pro Glu Ile Met Glu Lys Phe Leu
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Leu Gly Lys Asp Asp Ser Pro Phe Phe Val Leu Asn Ser Asp Val Ile
100
105
Cys Asp Tyr Pro Phe Lys Glu Leu Leu Glu Phe His Lys Ala His Gly

Asp Glu Gly Thr Ile Val Val Thr Lys Val Glu Glu Pro Ser Lys Tyr | 135 |
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| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Leu | Thr | Leu | $\begin{aligned} & \text { Thr } \\ & 20 \end{aligned}$ | Leu | Pro | Lys | ro | $\begin{aligned} & \text { Leu } \\ & 25 \end{aligned}$ | al | Glu | he | Gly | $\begin{aligned} & \text { Asn } \\ & 30 \end{aligned}$ | Arg Pro |
| Met | Ile | $\begin{aligned} & \text { Leu } \\ & 35 \end{aligned}$ | His | Gln | Ile | Glu | $\begin{aligned} & \text { Ala } \\ & 40 \end{aligned}$ | Leu | Ala | Ala | Ala | $\begin{aligned} & \text { Gly } \\ & 45 \end{aligned}$ | Val | Thr Asp |
| Ile | $\begin{aligned} & \text { Val } \\ & 50 \end{aligned}$ |  | La |  |  | $\begin{aligned} & \text { Tyr } \\ & 55 \end{aligned}$ | Arg |  | lu | Val | Met $60$ | Val | Ser | Thr Leu |
| $\begin{aligned} & \text { Lys } \\ & 65 \end{aligned}$ | Lys | Tyr | lu | Glu | $\begin{aligned} & \text { Glu } \\ & 70 \end{aligned}$ | Tyr | $1 y$ | al | er | $\begin{aligned} & \text { Ile } \\ & 75 \end{aligned}$ | Thr | Phe | Ser | $\begin{gathered} \text { Val } \begin{array}{c} \mathrm{Glu} \\ 80 \end{array} \end{gathered}$ |
| Glu | Glu | ro | eu | $\begin{aligned} & \text { Gly } \\ & 85 \end{aligned}$ | r | a | $1 y$ | Pro | $\begin{aligned} & \text { Leu } \\ & 90 \end{aligned}$ | ys | eu | la | $1 \mathrm{u}$ | $\begin{aligned} & \text { Glu Val } \\ & 95 \end{aligned}$ |
| Leu | Lys | $y s$ | Asp <br> 100 | A | Ser | ro | he | Phe <br> 105 | Val | Leu | An | er | Asp <br> 110 | Val Ile |
| Cys | Asp | $\begin{aligned} & \text { Tyr } \\ & 115 \end{aligned}$ | P | Phe | Lys | Glu | $\begin{aligned} & \text { Leu } \\ & 120 \end{aligned}$ | Ala | sp | ne | His | $\begin{aligned} & \text { Lys } \\ & 125 \end{aligned}$ | Ala | His Gly |
| Ala | Ala <br> 130 | Gly |  | Ile | $1$ | $\begin{gathered} \text { Ala } \\ 135 \end{gathered}$ | Thr | Lys | al | sp | $\begin{aligned} & \text { Glu } \\ & 140 \end{aligned}$ | ro | Ser | ys Tyr |
| $\begin{gathered} \text { Gly } \\ 145 \end{gathered}$ | Val | Ile | 1 | S | $\begin{aligned} & \text { Asp } \\ & 150 \end{aligned}$ | Arg | $s p$ | Thr | Pro | $\begin{aligned} & \text { Asn } \\ & 155 \end{aligned}$ | Leu | Ile | Asp | $\begin{array}{r} \text { Arg Phe } \\ 160 \end{array}$ |
| Val | Glu | Lys | $0$ | $\begin{aligned} & \mathrm{Val} \\ & 165 \end{aligned}$ | Glu | Phe | al | Gly | Asn <br> 170 | rg | Ile | sn | $1 \mathrm{a}$ | $\begin{aligned} & \text { Gly Leu } \\ & 175 \end{aligned}$ |
| Tyr | e | Leu | $180$ |  | Ser V | Val | Le | $\begin{aligned} & \text { Asp } \\ & 185 \end{aligned}$ | u | e | lu | et | $\begin{aligned} & \text { Arg } \\ & 190 \end{aligned}$ | Pro Thr |
| Ser | Ile | $\begin{aligned} & \text { Glu } \\ & 195 \end{aligned}$ | Lys | Glu | r | he | $\begin{aligned} & \text { Pro } \\ & 200 \end{aligned}$ | Ile | eu | al | Glu | $\begin{aligned} & \mathrm{Gln} \\ & 205 \end{aligned}$ | Lys | Gln Leu |
| Tyr | $\begin{aligned} & \text { Ser } \\ & 210 \end{aligned}$ | Phe | Asp | eu | lu | $\begin{aligned} & \text { Gly } \\ & 215 \end{aligned}$ | Tyr | $\operatorname{Trp}$ | t | sp | $\begin{aligned} & \text { Val } \\ & 220 \end{aligned}$ | Gly | $\ln$ | ro Lys |
| $\begin{aligned} & \text { Asp } \\ & 225 \end{aligned}$ | Phe | Leu |  | ly | $\begin{aligned} & \text { Thr } \\ & 230 \end{aligned}$ | Cys | $u$ | Tyr | u | $\begin{aligned} & \text { Thr } \\ & 235 \end{aligned}$ | Ser | Leu | Ser | $\begin{aligned} & \text { ys } \begin{array}{l} \text { Lys } \\ 240 \end{array} \end{aligned}$ |
| His | Pro | Glu | Lys | $\begin{aligned} & \text { Leu } \\ & 245 \end{aligned}$ | ys | Lys | $l u$ | Lys | $\begin{aligned} & \text { Tyr } \\ & 250 \end{aligned}$ | Val | His | Gly | Gly | $\begin{aligned} & \text { Asn Val } \\ & 255 \end{aligned}$ |
| Leu | Ile | Asp | $\begin{aligned} & \text { Pro } \\ & 260 \end{aligned}$ | Thr | la | Lys | le | $\begin{aligned} & \text { His } \\ & 265 \end{aligned}$ | Pro | er | la | eu | $\begin{aligned} & \text { Ile } \\ & 270 \end{aligned}$ | Gly Pro |
| Asn | al | $\begin{aligned} & \text { Thr } \\ & 275 \end{aligned}$ | Ile | Gly |  | $s n$ | $\begin{aligned} & \mathrm{Val} \\ & 280 \end{aligned}$ | Val | al | Gly | glu | $\begin{aligned} & \text { Gly } \\ & 285 \end{aligned}$ | Ala | Arg Ile |
| Gln | $\begin{aligned} & \text { Arg } \\ & 290 \end{aligned}$ | Ser | al | u | u | $\begin{gathered} \text { Ala } \\ 295 \end{gathered}$ | Asn | Ser | $\mathrm{Gln}$ | Jal | $\begin{aligned} & \text { Lys } \\ & 300 \end{aligned}$ | Asp | His | Ala Trp |
| $\begin{aligned} & \text { Val } \\ & 305 \end{aligned}$ | Lys | Ser | r | Ile | $\begin{aligned} & \text { Val } \\ & 310 \end{aligned}$ | Gly | $r p$ | sn | Ser | $\begin{aligned} & \text { Arg } \\ & 315 \end{aligned}$ | Ile | Gly | Lys | $\begin{array}{r} \text { Trp Ala } \\ 320 \end{array}$ |
| Arg | Thr | Glu | Gly | $\begin{aligned} & \text { Val } \\ & 325 \end{aligned}$ | Thr | Val | Leu | Gly | $\begin{aligned} & \text { Asp } \\ & 330 \end{aligned}$ | Asp | Val | Glu | Val | $\begin{aligned} & \text { Lys Asn } \\ & 335 \end{aligned}$ |


<210> SEQ ID NO 9
$<211>$ LENGTH: 445
<212> TYPE: PRT
$<213>$ ORGANISM: Thermotoga maritima
$<400>$ SEQUENCE: 9

Asp Pro Ser Gly Tyr Gly Arg Val Ile Gln Asp Gly Asp Lys Tyr Arg

Asp Ala Val Asn Phe Ala Glu Lys Val Arg Val Val Arg Thr Asp Asp

| Leu Leu Glu Ile Thr Gly Val Asn Thr Arg Lys Thr Leu Val Trp Leu |  |
| ---: | :--- |
| 210 | 215 |

Glu Glu Gln Leu Arg Met Arg Lys Ile Glu Glu Leu Leu Glu Asn Gly
225
Val Thr Ile Leu Asp Pro Ala Thr Thr Tyr Ile His Tyr Ser Val Glu
245
250
Asp Cys Glu Ile Gly Asn Asn Val Lys Ile Thr Arg Ser Glu Cys Phe290295300
Lys Ser Val Ile Glu Asp Asp Val Ser Val Gly Pro Phe Ala Arg Leu
305

Arg Glu Gly Thr Ile Leu Lys Lys Ser Ser Lys Ile Gly Asn Phe Val Glu Ile Lys Lys Ser Thr Ile Gly Glu Gly Thr Lys Ala Gln His Leu

## -continued


$<210>$ SEQ ID NO 10
$<211>$ LENGTH: 456
$<212>$ TYPE: PRT
$<213>$ ORGANISM: Bacillus subtilis
$<400>$ SEQUENCE $: 10$

-continued

$<210>$ SEQ ID NO 11
$<211>$ LENGTH: 257
$<212>$ TYPE : PRT
$<213>$ ORGANISM: Yersinia pseudotuberculosis
$<400>$ SEQUENCE : 11


$<210>$ SEQ ID NO 12
$<211>$ LENGTH: 257
$<212>$ TYPE: PRT
$<213>$ ORGANISM: Salmonella typhimurium
$<400>$ SEQUENCE : 12

Glu

What is claimed is:

1. An isolated mutant $\mathrm{E}_{p}$ nucleotidylyltransferase, wherein the mutant is SEQ ID NO:1 mutated from leucine to threonine at position 89 (L89T), said isolated mutant possessing a different substrate specifity for sugar phos- 5 phates than the corresponding non-mutated nucleotidylyltransferase.
2. A method of altering nucleotidylyltransferase substrate specificity comprising mutating an isolated nucleic acid
sequence encoding the nucleotidylyltransferase as set forth in SEQ ID NO:1 to replace leucine with threonine at residue 89 (L89T), the encoded mutant nucleotidylyltransferase possessing a different substrate specificity for sugar phosphates than the corresponding non-mutated nucleotidylyltransferase.
