

(12) United States Patent

Thorson et al.

US 7,906,460 B2 (10) **Patent No.:** (45) **Date of Patent:** *Mar. 15, 2011

(54) ACTIVE-SITE ENGINEERING OF **NUCLEOTIDYLYLTRANSFERASES AND** GENERAL ENZYMATIC METHODS FOR THE SYNTHESIS OF NATURAL AND "UNNATURAL" UDP-AND TDP-NUCLEOTIDE SUGARS

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Notice: Subject to any disclaimer, the term of this

patent is extended or adjusted under 35

U.S.C. 154(b) by 1072 days.

This patent is subject to a terminal dis-

claimer.

(21) Appl. No.: 11/536,208

Filed: (22)Sep. 28, 2006

(65)**Prior Publication Data**

US 2007/0178487 A1 Aug. 2, 2007

Related U.S. Application Data

- Division of application No. 10/013,542, filed on Dec. 13, 2001, now Pat. No. 7,122,359.
- (60) Provisional application No. 60/254,927, filed on Dec. 13, 2000.
- (51) Int. Cl. C40B 40/06 (2006.01)C12N 9/10 (2006.01)
- (52) **U.S. Cl.** **506/16**; 435/193
- (58) Field of Classification Search None See application file for complete search history.

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ABSTRACT

The present invention provides mutant nucleotidylyl-transferases, such as E_n, having altered substrate specificity; methods for their production; and methods of producing nucleotide sugars, which utilize these nucleotidylyl-transferases. The present invention also provides methods of synthesizing desired nucleotide sugars using natural and/or modified E_p or other nucleotidyltransferases; and nucleotide sugars synthesized by the present methods. The present invention further provides new glycosyl phosphates, and methods for making them.

4 Claims, 32 Drawing Sheets

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FIG. 1

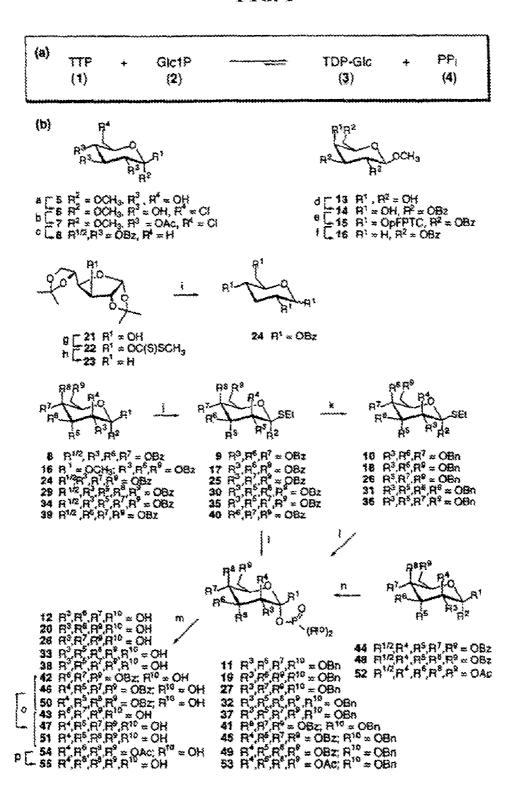


FIG. 2

Substrate	TTP Conv. (%)*	Reten. (mi	n) ^b UTP Conv.	(%)* Reten. (min)*
2° 40 10 40	99.3 ±	0.1 4	i,1 99.5	± 0.7 3.7 ^d
43 490 24	25.5 ±	0.4 4	i.2 22.3 :	± 0.4
28 HO HO	o+, °o } oe:s≠ _`}o	0.9 4	l.3 6.5 ±	: 0.3 ^f 3.7
20 HO 10	98.3 ±	1.6 4	.4 99.3 :	± 0.4 3.9
12 HRO 140	98.2 ±	1.7 4	.3 99.1 :	± 0.8 3.9
56° HO THO	99.5 ±	0.1 4	.1 17.9 :	£ 1.7 3.7
57° HO HO	56.8 ±	0.4 4	.2 32.7 :	£ 2.7 3.7 d
38 HO HO HO	14.8 ±	0.1 4.	.0 -	, h
47 HO 7	5.4±0),4 4.	.0 -	h h
31, 51 and 55	_ #	٠.	, h	, h

FIG. 3

FIG. 4

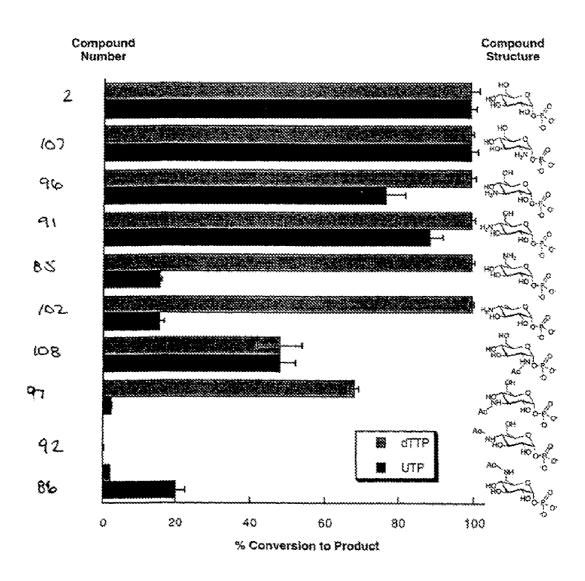


FIG. 5

FIG. 6

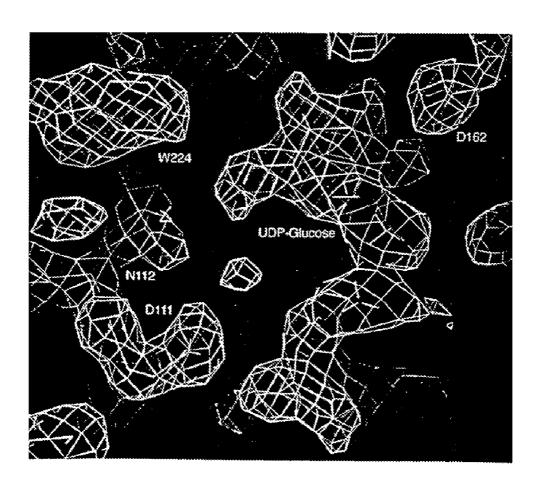
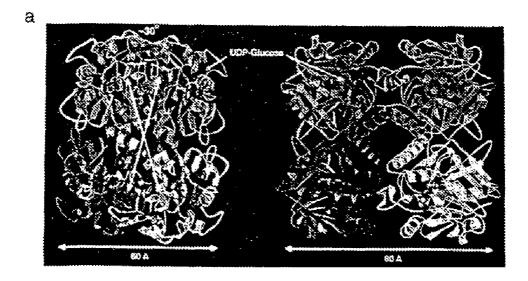


FIG. 7



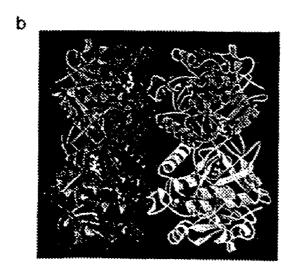


FIG. 8(a)

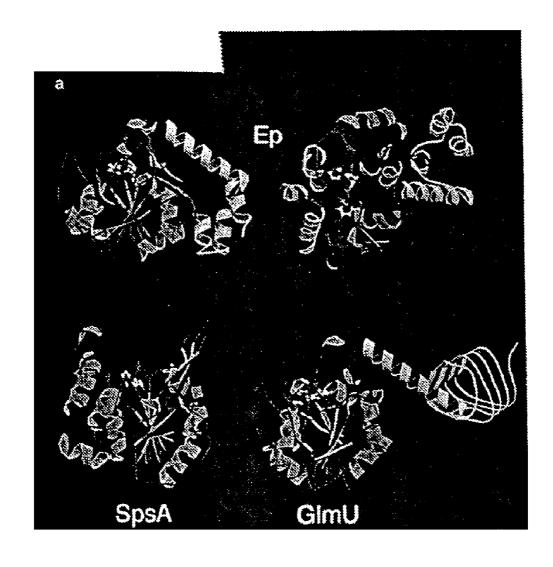
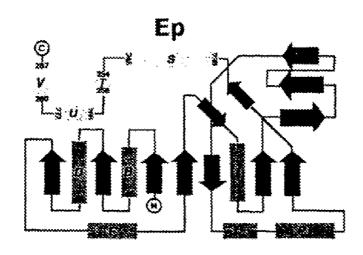
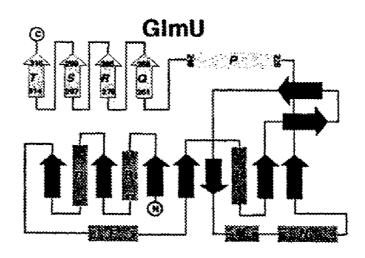


FIG. 8(b)





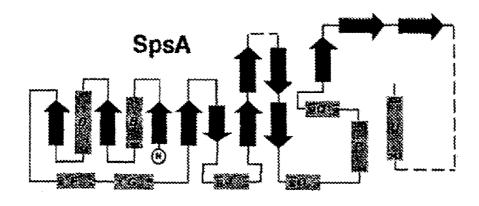


FIG. 9

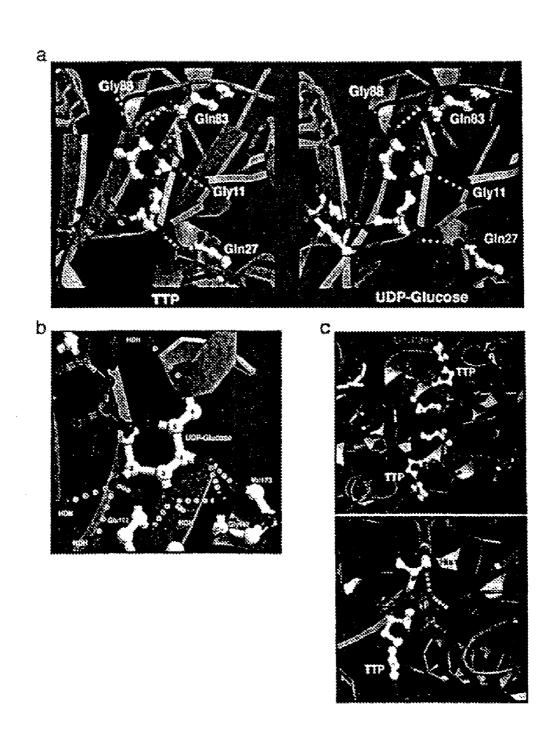


FIG. 1θ

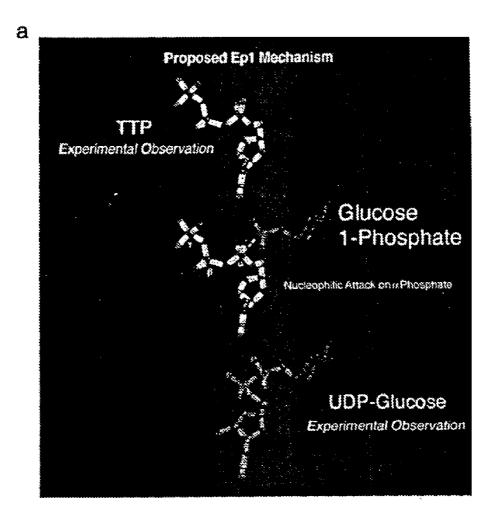


FIG. 10 cont.

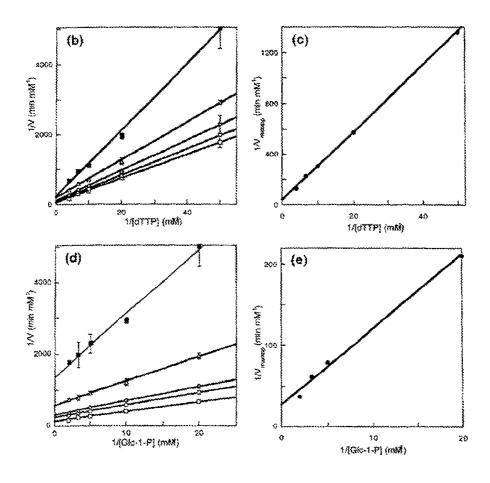


FIG. 11

	Substrate	Wild-Type E _p % Conv.	Mutant Pool % Conv.	Trp224His % Conv.	Thr201Ala % Conv.
1	190 Car Tolk	99.3 ± 0.8	90,7 ≵ 2,8	98.6 ± 1.1	99.3 ± 0.1
2	Part of the state	4\$.5 ± 9.4	18.3 ± 3.4	16.9 ± 3.1	89.5 ± 1.6
\$	S. Carton	. \$	18.7 ± 3.8	36.5 ± 0.5	. \$
ł	of contract	97.2 ± 2.9	48.0 ± 0.9	72.3 ± 0.5	23.5 ± 2.6
	the state of the s	74.8 ± 0.1	\$. §	<u>.</u> %
	Total Total	è	97.9 ± 2.1	68.5 ± 1.1	, §



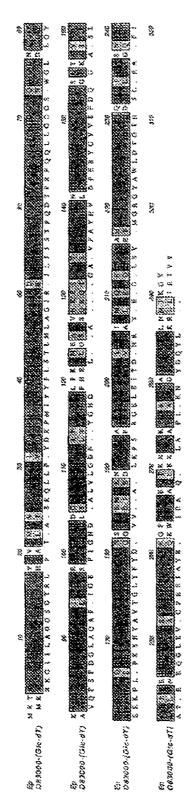


FIG. 1

ခွာလက**း**က

with Gle-I-P Uridylyttransferases SP DRIGOGRAPH N CANGETZ (CSC-L) N CANGETZ (CSC-L) S CALL MYCAE (OSC-L)

Ep with Glc-1-P Adenylyltransferases

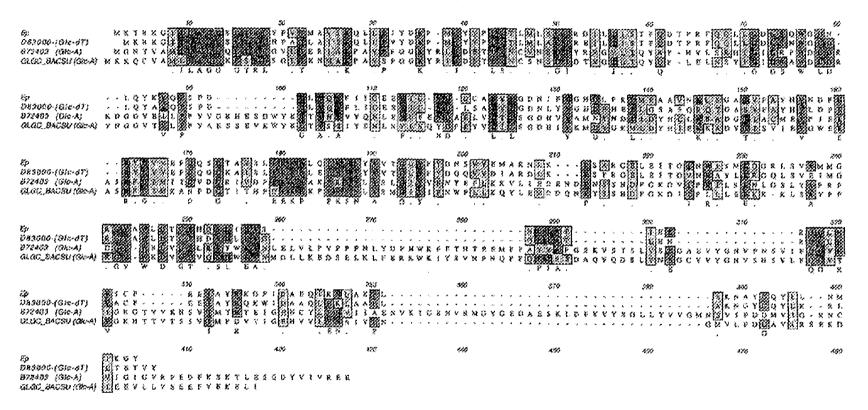
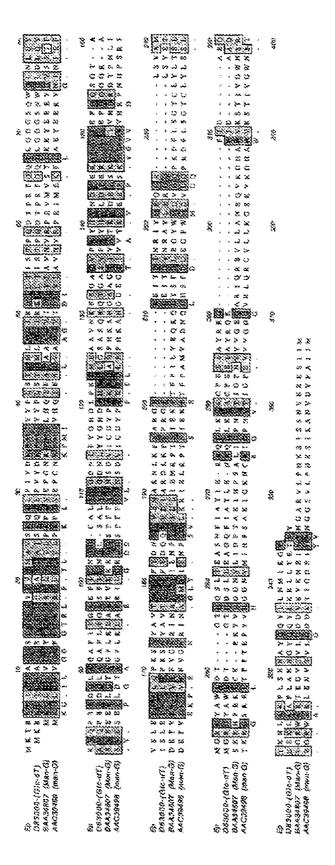
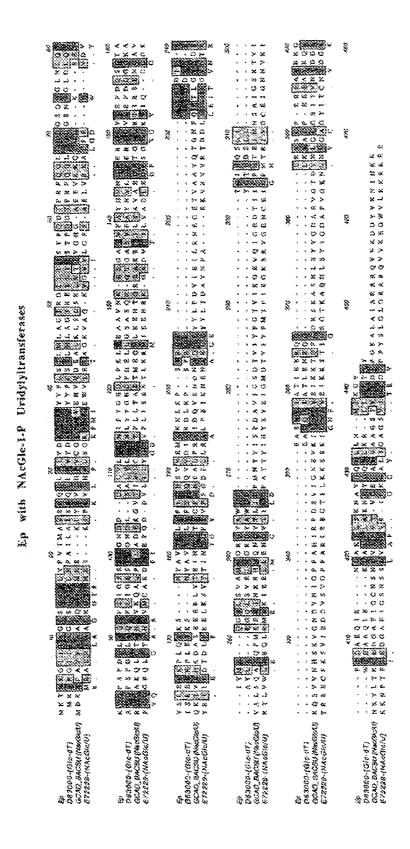


FIG. 14

with Man-1-P Guanylystrunsferases





*** Ep with Glc-1-P Cytidylyttransferases 69
082000-(010-07)
1866. 84.17 (00-07)
1866. 84.17 (00-07)
1869. 786. 786. 00-07)

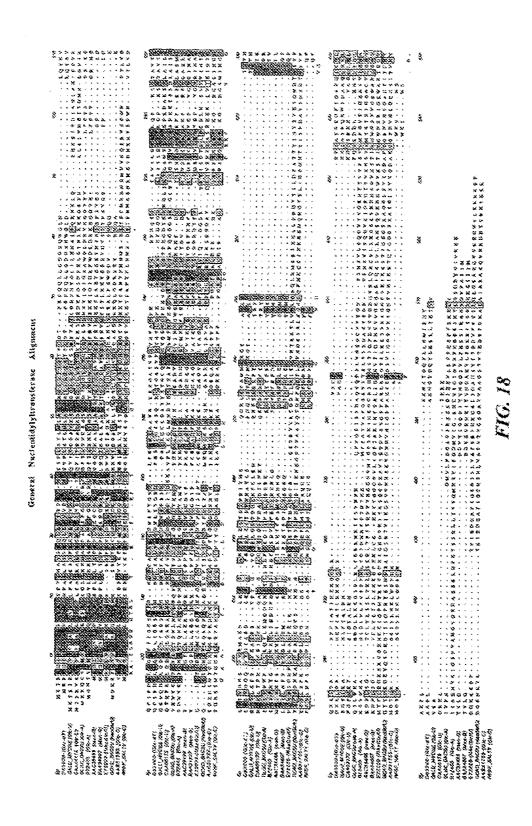


FIG. 19(A)

			\$	mallest Sum	<u> </u>
			High	Probability	
	Sequences pi	roducing high-scoring segment pairs:	Score	P(M)	23
_			2000	^	
	gi 1710100	CLUCOSE-1-PHOSPHATE THYMIDYLYLTRANSFERAS	151.0	р р	1
	gi 9957817	Glucose-1-phosphate thymidylyltransferas.	1507	0	1
	gi 9957822	Glucese-1-phosphate thymidylyltrensferas.	1499 1497		
	gi 9957847	Glucose-1-phosphate thymidylyitransferas.		Q B	ì
	gi 9957866	Glucose-i-phosphate thymidylyltransferas	1496		1
	gi 9957852	Glucose-1-phosphate thymidylyltrensferes.	1488	0	1.
	gi 9957857	Glucose-1-phosphate thymidylyitransferas	1450	õ	1.
	gi 9957836	Glucose-1-phosphate thymidylyltransferas	1444	0	1
	g1 1073702	RfbA protein - Shigella flexmeri (strain	1440	C C	1
	gi 141362	GLUCOSE-1-FROSPHATE THYMIDYLYLTRAMSFERAS	1437	0	3 .
	gi 9957831	Glucose-1-phosphate thymidylyitransferas	1429	<u>o</u>	1
	gi 9957841	Glucose-1-phosphate thymidylyltransferas	1424	e .	ī
	gi 2507297	GLOCOSE-1-PHOSPHATE THYMIDYLYLARAMSFERAS_	1431	0	1
	gi 2121141	Glucose-1-phosphate thymidylyltransferas	1408	0	1
	gi 9957862	Glucose-1-phosphate thymidylyltransferss	1389	2.70~178	1
	gi 9957627	Glucose-1-phosphate thymidylyltransferas	1356	7.1e-178	1.
	gi 585826	GLUCOSE-I-PHOSPHATE THYMIDYLYLTRAMSPERAS_	1067	2.26-171	3
13.	gi 11348597	Glucose-1-phosphate thymidylyltransferas	1185	2.8e-154	1
	gi 3135675	Putative glucose-l-phosphata thymidyltra	1139	6.3 a -148	1
	gi 3608394	Putative glucose-1-phosphate thymidyl tr	1112	3.40~144	3
21.	gi 1666508	RfhA (Leptospira interrogans)	1103	5.9e-143	3.
22.	g1 4234804	RmlA [Leptospira borgpetersenii]	2092	1.9e-141	3.
23.	gi 1881544	Glucose-l-phosphate thymidyl transferase.	1073	8.16-139	1
24.	gi 2500162	GLICOSE-1-PROSPHATE THIMIDVINIATRANSFERAS	1070	2.1e-138	1
25.	gi 7471939	Glucose-1-phosphate thymidylyltransferas.	1069	2.9e-138	1
26.	gi 7434861	@lucose-1-phosphate thymidylyltransferas	1064	1.4e~137	1
27.	gd 4200433	Cps2L [Streptococcus pneumomiae]	1056	1.8e~136	3
28.	gi 3320399	Glucose-1-phosphate thymidyl transferase.	1055	2.5e-136	1
29.	gi 7592815	D-glucose-1-phosphate thymidylyltransfer	1051	8.8e-136	1
	gi 5545318	Glucose-1-phosphate thymidylyltransferas	1045	5,9e~135	1
31,	gi 1944160	Glucose-1-phosphate-thymidylyltransferas	1045	5.9e-335	3.
	gi 4406249	Glucose-1-phosphate thymidylyl transfera	1039	4e-134	3.
	gi 3832506	Glucose-1-phosphate thymidylyl transfera.	1039	4e-134	1
	gi 1710101	GLIXOSE-1-PROSPHATE THYMIDYLYLTRANSFERAS	1036	1e-133	1
	gi 3907610	Glucose-1-phosphate thimidylyl transfers	1033	2.7e~133	3
	gi 9977936	GIAKOSE-1-PHOSPHATE THYMIDYLYLTRANSFERAS	1031	5.1e~133	1
	gi 1098479	Glucose-1-phosphate thymidyl transferase	1629	9.6e-133	1
	gi 7434867	Probable glucose-i-phosphate thymidylylt	1023	6.56-132	1
	gi 9978667	GLUCOSE-1-PHOSPHATE THYMIDYLYLTRANSFERAS	1019		1
	gi 585825	GLUCOSE-1-PHOSPHATE THYMIDYIXLTRANSFERAS.	1010		1
	qi 2500161	PROBABLE CLUCOSE-1-PHOSPHATE THYMICXLYLA	1007		1
	gi 2507298	GLOCOSE-1-PHOSPHATE THYMIDYLYL/TEAMSYERAS	998	1.8e~128	1
	2-1				

FIG. 19(B)

4.3.	gi[1710102	GLUCOSE-1-Phospeate ThymilylyLyltpinisperas	598	1.80-128	1
44.	gi 5931969	Glucose-1-phosphate thymidyltransferase	549	6.2e-128	2
45.	gi[11261716	Glucose-1-phosphate thymidylyltransferas	984	1.5e~126	2.
46.	gi 5199111	Glucose-1-phosphate thymidyl transferase	540	6.5e~125	2
	gi 7434866	Glucose-1-phosphate thymidylyltransferas.	366	4.8e~124	1
	gi 1710029	GLUCOSE-1-PHOSPHATE THYMIDYLYLTRANSPERAS.	948	1.50-121	3
			466		3
	gi 1314579	Glucose-1-phosphate thymidylyltransferas.		3.5e-120	
	gi 1890601	ExpA7 [Sinorhizobium meliloti]	551	5.99~120	2
	gi;6677502	Putative glucose-1-phosphate thymidyl tr.	933	1.7e-119	3
52.	gi 6688595	RmlA protein [Legionalla pneumophila]	933	1.7e-119	1.
53.	gi 148192	Similar to Streptomyces griseus StrU pro	543	1e~118	2
54.	gi 421098	Hypothetical protein 6292 - Escherichia	536	9.3e-118	2
	gi 7434863	Glucose-1-phosphate thymidylyltransferas.	518	6.2∞-1.17	2
	gi 9133016	Potative dTDP-1-glucose synthase; NotY [907	6.6e-116	1
	gi 9714086	Glucose-1-phosphate thymidyltransferase	511	6.68-114	2
		Putative CTDP-glucose synthase (Strepton.	883	1.4e-112	1
	gi 6018314				
	gi 3789899	Alpha-D-glucose-1-phosphate thymidylyltr	883	1.9e-112	1
	gi 7688728	NovV (Streptomyces apheroides)	504	1.46-111	3
	gi 11095238	DTDF-glucose synthese; glucose-1-phosphs	499	9.5e-111	2
62.	gi 10800782	Glucose-l-phosphate thymidyltransferase	865	4.le-IIO	3.
63.	gi 4884772	TDP-glucose synthase homolog (Streptomyc	863	7.80-110	1
64.	gi 5921158	Glucose-1-phosphate thymidyltransferase	859	2.5e~109	3.
	gi 4884768	TDP-glucose synthase (Streptomyces spect	683	2.6e-107	2
	gi 5579435	SpcK [Streptomyces flavopersicus]	670	3e-105	2
	gi 4033331	DTDP-glucose synthese Actinoplanes sp	452	4.4e-103	2
	gi 580705	CAC3 (Azerhizobium caulinodans)	798	7.3e-101	ĩ
	. 2	Probable glucose-1-phosphate thymidylylt	798	7.3e~101	ĩ
	gi 1072851				
	g1 2804683	Glucose-l-phosphate thymidyl transferase	758	2.4e~95	1
	gi 2804721	Glucose-1-phosphate thymidyl transferase	737	1.9e-92	1
	gi 2127533	Glucose-i-phosphate thymidylyltransferas	466	1.7e-89	3
	gi 1944620	Glucose-1-phosphate thymidylyltransferas	652	1e-80	3
74.	gi 4574161	Glucose-l-phosphats thymidyl transferase	651	1.46-80	1
75.	gi 730818	SPORE COAT POLYGACCHARIDE BIOSYNTHESIS P.	268	1.3e-50	4
76.	gi 10175985	Spore coat polysaccharide synthesis (glu.	261	6.15-44	3
?7.	gi 11279395	Glucose-1-phosphate thymidylyltransferas.	279	4.86-42	3:
	gi 7329194	DTDP-D-glucose synthase (Streptomyces an.	279	4.86~42	3
	gi 4731596	BlmD (Streptomyces bluensis)	3.75	le-41	4
	gi 4406265	Glucose-1-phosphate thymidylyl transfera.	365	4.20-41	2
	gi 7448174	Glucose-1-phosphate thymidylyltransferas	250	5.98-39	3
	gi 7448157	Hypothetical protein - Symechocystis sp	220	8.8e-39	3
			181	1.7e-38	ě.
	gi 11279397	Probable dTDF-1-glucose synthese (import			
	gi 280334	StrD protein - Streptomyces griseus	156	1.1e-37	4
	gi 134991	GLICOSE-1-PHOSPHATE THEMIDYLYLTRANSFERAS	156	1.le-37	4.
86.	gi 11279398	Glucose-l-phosphate thymidylyltransferas	153	3e-37	4
87.	gi 3256958	StrD (Streptonyces glaudescens)	153	6.1e~36	4
88.	gi 11497938	Gluccse-1-phosphate thymidylyltransferas	309	7.le-35	4
89.	gi 975621	Glucose-1-phosphate thymidylyltransferes	160	7.1e-35	4
99,	gi 7481914	DNDP-glucose synthase - Streptomyces vir.	158	3.3e-34	4
	gi 7448156	Gluncae-1-phosphate thymidylyltransferas	181	4.5e-34	4
	gi 6002933	MMDP-glucose synthetase [Streptomyces fr	163	8.36~34	4
	gi 7448164	Glucose-l-phosphate thymidylyltransferas.	131	1.le-33	5
			234	1.2e-32	Š
	gi 7448155	Probable glucose-1-phosphate thymidylylt			
	gi 4240414	NDP-homose synthetase homolog [Streptomy	156	1.86-32	4
	gi 2209217	Glucose-1-phosphete thymidyl transferase	267	1.4e-27	1
	gi 6015546	Glucose-1-phosphate thymidylyltransferas	156	5.4e-27	4
	gi 6933890	Putative TDP-glucose synthase (Streptomy	228	1.8e-26	3
99	gi 4884958	Glucose-1-phosphate thymidylyltranaferas	263	2,€⊕-25	2

FIG. 19(C)

			40 f f 2 2 2 4 2 2 4 4 1 4 1 4 1 4 1 4 1 4 1 4	4 2 23	0.00	
		7473500	Probable glucose-i-phosphote thymidylylt	217	9.66-23	ij
191.	gi	1346094	UTPGLIXOSE-1-SHOSPHATE URIDYLYLTRANSFZ	\$ 54	3.3e-19	3
102.	gi	10956341	PXO1-94 (Bacillus amphracis)	138	3.4e~19	3
103.	ai.	10176376	UTP-glucose-1-phosphate uridylyltransfer	126	6.5e-19	3
		7521440	Probable sugar-phosphate nucleotydyl tra.	139	le-18	3
		7821719	Sugar-phosphate nucleotydyi transferase	140	2.5e~18	3
		6138856	UTP-glucose-1-phosphate wridylylcransfer.	121	4.99~18	3
				130	9.2e-18	3
		2501471	PURATIVE DIFCLICOSE-1-PHOSPHRIK UNIDAL.			
		10580603	Glucose-1-phosphate thymidylyltransferas	119	1.98-17	4
		6380274	96% identity with amino acids 1-24 of E	330	3.12-17	2
126.	33	7434850	UTPglucose-1-phosphate uridylyltramsfs	130	3.5e-17	3
111.	gi	2501469	UTPGLUCCSE-1~FHOSPHATE URIDYLYLARANSFE	132	3.88~17	€,
112.	gi.	7434856	UTFglucose-1-phosphate uridylyltransfe.	233	4.6e-17	3
113.	gi	2601467	UTFGLUCOSE-1-PHOSPHATE URIDYLYLTRANSFE.	131	4.60~17	3
		10802706	Glucose-1-F thymidylyltransfersse (Carbo.	154	6.4e-17	4
		10992777	Glucose-1-phosphate thymidylyluransferas	150	8.7e-17	ń.
		7739964	Putative UDP-gluccee pyrophosphorylass (130	1.76+16	3
		555004	Glucose-1-phosphate unidylyltransferase	132	1.7e~16	3
		7434852	UTPglucose-1-phosphate uridylyitransfe.	109	1.8e-16	3
			Ciucose-1-phosphate unidylyltransferase	133	2.4e-16	3
		33.92048				
		3550619	UTP-qlucoss-i-phosphate uridylyltransfer.	133	3.12-16	3
		1177038	PUTATIVE UTPGLUCOSE-1-PECSPHATE UNIDYL	123	3.36-16	3
		10174923	OTP-glucose-1-phosphate unidylyltransfer	111.	3.4e-16	3
		3777501	Putative GDP-mannose pyrophosphorylase [150	6.5e-16	3
124.	gi.	3970893	GDF-saurose pyrophosphorylase (Candida a	150	6.5e-16	3
125.	31	3777593	Putative GDF-mannose pyrophosphorylase {	150	€.5e~16	3
126.	31	7295813	001129 gene product [alt 1] [Drosophila	115	81-86.8	2
		4340429	MDF-hexase synthetase hamaleg (Streptomy	112	8.3&-16	2
		7448155	Probable glucose-1-phosphate thymidylylt	116	16-15	5
		2127932	Glucose-1-phosphate thymidylyltraneferas.	143	1.3e~15	4
		3323397	Mannose-1-phosphate quanyiyitransferase	152	1.7e~15	4
		585225	DIPGLIXOSE-1-PHOSPHATE URIDYLYLTRANSFEL.	137	1.8e-15	3
		7434851	FTFgiucose-1-phosphate uridylyltransfe.	109	2.1e-15	3
			UTD-glucose-1-phosphate unidylyltransfer.	133	3.40-15	3
	:	10176341				3
		2501470	PTPGLUCOSE-1-PHOSPHATE CEIDYLYLTRANSFE.	130	3.70-15	
		7497318	Hypothetical protein C43Cl.5 - Caenorhab.	171	5.4e-15	3
		10174033	Mannose-1-phosphate guamyltransferase (2	129	1.10-14	4
		7331158	GDF-mannose pyrophosphorylase (Pichia su.	135	1.3e-14	3
138.	QĹ	6320148	Mannose-1-phosphate guanylingnaferase, G.	159	4.8a-14	3
		10579718	Glucose-l-phosphate thymidylyltransfersa	105	5e-14	3
140.	gx	7448165	Mannose-1-phosphate guanyltransferase PA	113	6.3 e -14	爻
141.	Çi i.	2121149	Glucose-1-phosphate thymicyltransferass	168	6.50-16	3.
142.	gi	1.0640626	Marmose-i-phosphate gramyltransferase ra.	133	76-14	3
143.	gi	894304	Mannose-1-phosphare gnanyltransferage (3.	148	8.8e-14	3
		9055355	DTDF-glucose synthese (Streptomyces rimo.	1.67	90-14	3.
		10579656	Glucose-1-phosphate thymidylyitransferes.	148	1,1e-13	2
125	200	4760690	GDF-mannose pyrophosphorylase [Candida g	144		3
2.50	25.5	7649503	Putative nucleotide phosphorylese (Strep.	86	5.58-13	3
		7446158	Glucose-1-phosphate thymidylyltransferas	196	6.Se-13	5
		7448179	Probable rmlA2 protein - Mycobacterium t	102	8.66~13	2
		10880965	Putative VIP-gluonse-1-phosphate unidyly	133	3.48-13	3
	-	6015731	Glucose-1-phosphace thymidylyltnensferas	1.1.5	9.494.13	3
		7434855	VTPglucose-1-phosphate unidylyltransfe	2.07	1.24-12	3
		11352828	VTFglucose-1-phosphate uridylyltransfa	266	1.48-13	3
3.54.	gi.	4884955	Glucose-1-phosphate thymidylyltransfaras	3.57	2.26~32	λ
165.	gá l	4378170	UTF-glucose-1-phosphate unidyltransferas	2.2.5	2.8e~22	3
156.	gi	7451544	Mannosc-1-phosphate quanyitransferase	125	3.36-13	3
		•				

FIG. 19(D)

~ ****	110000100	THE CONTRACTOR OF THE PROPERTY AND ASSESSMENT OF THE ASSESSMENT	200	6.2e-13	3
	11/2503468	UTPCLUCOSE-1-FROSPERTE UNIDILITATIONEL.	107		
	21 7492163	Manacae-1-phosphate quanyltransferase	136	1.69-11	3
	gi 11361675	OTF-glucose-1-thosphate unidylyitransfe.	191	1.7e-11	3
160. 9	gi 7448173	Probable glucose-i-phosphate thymidylylt	83	1.50-11	4
161. 9	31. 7434849	UTPglacese-1-phosphate uridylyltransfa	128	20-11	2
162.	gi 11261677	UTD-glucose-1-phosphate uridylyltransfer	106	4. \$%~Il	3
	zi 11201687	Probable UTPglucose-1-phosphate unidyl	107	5.70-11	3
	gi 7381245	UDSG-pyrophosphorylase (Acetobacter Myli-	1.03	3.76-11	3
	gi 3373537	WP-glucose-1-phosphata unidylyltransfer	115	89-11	3
		ODP-glucose pyrophosphorylase [Sulfoldbu	322	9.48-11	4
	gi 6015664		126	9.90-11	3
	gi 10579698	Glucose-1-phosphate thymidylyltransferas			
	gi 7448161	Probable mannose-1-phosphate quenylyltma.	135	1.3e~10	3
	gi 11261691	WFglucose-1-phosphare unidylyltransia.	107	1.40~10	3
	gi}7448163	Slucese-1-phosphate thymidylyltransferas	79	1.5e-10	4.
171. 3	gi[3559991 -	UDP-glucces pyrophosphoryless (Pseudomon.,	106	2.4e~10	3
172.	gi 1169833	(TEPGLECOSE-1-MESSMATE BEIGGENLERANSFE.	117	2.7e-10	3
173.	gi 2117938	WFglucose-l-phosphare uridylyltransfe.	308	3.4e~10	3
	gi 130929	UTPCENCOSE-1-FHOSPHATE URIDYLYLTFAMEYE	129	4.79-10	2
	gi 541005	ExoN procein - Bhischium meliloti	112	7.16-10	3:
	gi 7649599	Purative mannose-1-phosphete quanyicrans	105	7.Ze-10	3
	gi 6066425	Mennose-1-phosphate quamyitransferase (L.	3.30	8e-10	3
		UTSGLUCOSE-1-EKOSPHATE URINYLYLTDABUPE.	112	8.1 4- 10	3
	gi (463035		138	9.1e-10	l.
	gi 7434854	Probable OFFglucose-1-phosphate unidyl.			
	gi 10579655	Glucoss-i-phosphate thymidylyltmansferas	90	34-05	3
	9114103334	SDF-manuese pyropheaphorylase (Solanum t	113	20~09	3
	gi 4234784	Undonom (Leptospira borgreteramii)	9.3	2e-09	3
193. 9	g: 5814381	Unimown [Leptospira interrogans]	95	26-09	.3
164.	g1 7448169	Probable membes-l-phosphate quanyltrans	115	3.70-09	2
	gi [3185089	utpGlicobe-1-frosphate uriditaltrapspel.	125	5.7e-09	3
	gi 7648154	Manacas-l-physphate quenylyltransferace	2.04	6.59 - 99	3
	gi 111261685	PTPglucose-1-phosphate uridylyltransfe.	93	7.3e-39	3
	gá 3319929	GalS protein [Pecushacterium carotovorum.	131	8,4e~09	3.
	gi 116099	UTPGLUCOSE-1-PHOSPHATE DELUYLYLTRANSFE.	3.03	9.48-09	3
	gi 7521439	Probable sugar phosphate transferase APE.	325	1.86-08	\$
	gi 17269958	SEE-mannese pyrophosphorylase like prote.	107	3.10.08	\tilde{z}
			89	\$.7e-08	2
	gi 7448158	Mannose-1-phosphate quantitransferase	84	7.68-03	3
	g\$ 7447292	Probable glucose-1-phosphate thymidylylt			
	gi 1360733	WTFgiucose-1-phosphate uridylyltcansfe.	122	8.78-08	2
	gi 130926	DIPCLUDOSE-1-PROSPENTE UNILYLYLANDER.	755	8.88~08	2
	gi [96762	OLDdynose-y-byosphane mighlypranege"	755	8.84-99	2
187.	gi. 2501466 -	UTPGLUCCEE-1-PHOSPHRIE URIDYLYLTHMMSTE	122	9,86-08	2
198.	g4:10803043	UDP-glucose pyrophosphorylase (Kaenophil	3.3.3	1.6 e ~07	3
199.	gi 7448172	Probable glucose-i phosphata transferase.	80	3.9a-07	:3
200.	gi 7434875	Slucose-l-phosphate adenyiyitransferase	93	3.16-07	2
	gi. 585168	UTPGLECOSE-I-FROSURATE URIDYLYLYBANSFE.	119	3.2e-07	Ž
	gi. 2133477	Pyrophosphorylase ppp-1 homolog - Caenon.	120	3.8e-07	1
	91 3041673	EUTATIVE TRANSLATION INTTIATION FACTOR R.	120	2.86-97	2
	gi 7206597	C. elegans (PPP-1) putative translation	120	2 8e-07	ì
		Probable super-phosphate nucleaticyl tra.	95	3.8e-07	2
	g\$ {11354079		118	5.26-07	3.
	gi 10436247	Unnamed protein product (Notes Septems)			
	gi 249433.3	THANSLATION INITIATION FACTOR RIF-28 GAN.	822	5.3a-07 5.3a-07	3.
308.	gi 3366779	Bukaryotic translation initiation factor.	118	5.20-07	2
	gi. 7434857	WWglucose-l-phosphate uridylyltrausie	88	?.35007	3
210.	g4 (11281689 -	UTPglucose-1-phosphats unidylyluransis	<u>44</u>	9.46-97	2
	gi{1978827	Pyrophosphorylase 1 - Caentrhalditis bri	318	9.50-07	1
212.	gi }11353788	Mannose-l-phosphate guanyltransferase-re.	31	1.2e-08	.3
	gi 639245	ImbO protein - Streptomyces lincolnemsis	84	1.26-66	3
		·			

FIG. 19(E)

	gi 11347184	Probable sugar-phosphote moleculdyltran	32	1.3e-08	2
	gi 7451542	Mypothatical protein - Symechocystis sp	85	1.4e-96	3
	gi 11261683	UTPglucose-l-phosphare uridylyltransfe	88	2.84~06	2
	gi 111498742	Glucose-1-phosphace cytidylyltxansierase.	3.3.4	1.9e-08	1
218,	gi. 11279398	Marmose-1-phosphate guanylyitransierase	3.95	2.4e~08	2
239.	gi{10436672	Unnamed protein product (Home sapiens)	208	4.10-08	.2
220.	gi 7019397	GDF-mannose pyrophosphorylase B (Romo sa.,	108	4.le~06	3
223.	gi 11431494	GDF-mannose pyrophosphorylase & (Homo sall	108	4.10-06	3
332.	gi[265795	Glucose-1-phosphate thymidylyl-transfera	111	4.8e-56	3
223.	gi 348416	Glucose-1-phosphats thymidylyltransferss	111	4.89-95	3.
	gi 7448171	Hypothetical protein - Symechocystis sp	99	1.2s-05	2
225.	gi 586920	RABOTHERICAL EROTEIN IN 200 2. FERRICA	92	1.66-05	2
226.	gi 3320397	Putative glycerol-2-phosphate (Streptoco	3,06	1.7e-05	2
	gi 3818694	Cps23fN [Streptococcus gnaumoniae]	106	1.70-05	3
238.	gi(743488%	Gluccse-1-phosphate ademylyltransferase	88	1.80-05	3
229.	gi 729582	GUXCOSE-1-PHOSPHATE ADENVLYLYRAMSFERASE	82	2.76~05	Ą
230.	gi 232171	CLUCOSE-1- Skosphats albentlytabansperase	78	3.18-65	3
331.	gi 7434848	DTPglucose-1-phosphate uridylyltmansfe.	64	3.2e-05	4
232.	gi 10280002	Glucose-1-phosphare thymidylyltransferes.	79	3.4e-05	3
233.	gi\11352831	Virulence factor XV0591 (imported) - Xyi	76	5.le-05	3
334.	gi 11351550	Probable mucleotidyl transferase PACS97	22	5.6e-95	3
235.	gd 4545244	Unknown (Pseudomenas asmiginosa)	77	5.86-95	3
236.	gi 11261781	Glucose-l-phosphate ademylyltransferese	32	7,99~05	2
237.	gi 2811060	GLUCOSE-1-REOSPHATE ADENYLYLARAMSFEEASE	77	8.3.8-05	3
238.	gi]7473559	Probable mannose-1-phosphate quanyitrans	77	0.00018	3
239.	gi 7448152	Glucose-1-phosphate thymidylyltransferas	7.5	0.00019	3
243.	gi[2731779	ADP-glucose pyrophosphorylase [Thermus c	74	0.00041	3
24%.	gi[74298 4 8	Glucose-1-phosphate thymidylylsransfaras	9.4	0.00082	2
242.	gi 10638133	OTP-glucose-1-phosphata unidylyltxansfer	79	0.00084	3
263.	gi.{1063 62 06	WTP-glucose-i-phosphate unidylyltransfer	70	0.00054	3
244.	gi:11559596	Mukaryotic initiation factor cIFIS gamma.	98	0.66657	3.
245.	gi 73030 5 7	CO8190 gess product (Erosophila melanoga.	38	0.00075	3,
	gi [7488395	Translation regulator CCDS homolog F1913.	94	0.0011	1
	gi [8051798	Putative transferase (Amycolatopsis orie	87	0.0018	2
348.	gi}10638687	UTP-glucose-1-phosphate uridylyltransisr	70	5.00%	3
249.	gi 7469529	Hypothetical protein - Symechocystis sp	70	0.002	3
250.	gi}31497858	Glucuse-1-phosphace thymidylylcransfersa	81	0.002	3
	gi[132501	GLUCOSE-1-PHOSPHATE CYTHALATARAFERASE	8.5	0.003	1
	gi 10638183	UTF-glucose-1-phosphate unidylyltransfer	73,	0,0038	3
	g: 10638186	WTF-glucose-1-phosphate unidelyletransfer	71	9,0025	3
	gi 10638309	MP-glucose-i-phosphate uridylyltransfer	70	0.0025	3
	gi 10638346	WW-glucose-1-phosphata uridylyltransfer	70	0.0025	2
	gi 6041791	Autative translation instinction featur E.	91	0.0028	1
	gi 10173763	Glucose-1-phosphate adenylyltranaferase	86	0.0089	3
	g4 (3834671	ADP-glucose pyrophosphozylase (Shodospir.	57	0.0931	đ.
	gi 10638185	WF-glucose-1-phosphats unidylyitransfer.	76	0.0034	2
	gi 5315114	ADF-glucose pyrophosphorylase small subc.	83	0.0083	3
	g\$1232170	CLUCUSE-1-FROSPHATE ADENILVIPRANSFERASE	83	0.006	2
	g1 10640388	Gluctse-1-phosphace thymidylyltransferss.	57	5.00%	3
	g1 20638173	WP-glucose-i-phosphate unidylyltransfer.	76	0.0062	3
	g1 10638177	VIP-glocose-l-phosphare unidylyltransier	79	0.0062	2
	gi 20638180	UTP-glucose-1-phosphate uridyiyitranofer	35	S.9952	3
	gi 10638195	WP-glucoss-1-phosphata uridylyitransfer	79	0.0062	3
	gi 19638165	WTF-glucose-1-phosphate uridylyltransfer	70	0.0062	3
	gi 10638203	WF-qlucose-1-phosphace unidylylaransies.	804	0.0062	3
	gi.]381.1033	GLUCOSE-1-PROSPERTE ADENTIATARASPERASE	73	5.8878	2
279.	GT 8852256	Nausosa-l-phosphate-guanyltransferase-li	86	8.0086	3

FIG. 19(F)

27%.	gi 406922	Homology to ENP pyrophosphorylass M76548	87	0.0099	1
	ga 111261806	Probable glucose-1-phosphate adenylyitra	73	9.511	2
	94 7671532	Glocose-1-phosphate adenylyltransferase	79	0.611	2
	91 633874	Alpha-D-glucose cytidylyltransferase/ Ep.	36	0.014	3.
	gi 485384	Alpha-D-glucose-1-phosphate cyticylyltra.	88	0.014	3.
	gi \$21276	Glucose-1-phosphate cytidylyltransferase	88	0.014	3.
	91 10638193	UTP-glucose-1-phosphate unidylyltransfer.	70	0.015	2
	gi 10638156	WEP-glucose-1-phosphate uridylyltransfer	84	0.035	2
	gi 10638153	VTP-glucose-1-phosphate unitylyltransfer	54	0.035	3
	gi 10638168	UTP-glucose-1-phosphane unidylyltransfer.	54	0.015	ğ
	gi 2146023	Into protein - Streptomyces lincolnessis	73	9.016	2
	gi 2558972	Data (Vibrio anguillamum)	83	9.017	2
	gi 1237080	ADP-glucose pyrophosphorylase (Pisum sat	80	0.018	3
		Glucose-1-phosphete cyticylylcznsferse.	85	0.019	ã
	gi [7447201	Probable licC protein (licC) - syphilis	83	9.619	ì
	gi 7531163	GINCOSE-1-PROSPRATE ADENYANATRANSFERASE	80	9.023	2
	gi 121263		80	0.023	
	gi 100675	Glucose-I-phosphate adenylyltransferase			2
	gi 5917789	ADP-glucose pyrophosphorylase small subt	80	0.024	
	gi 8633019	T22C5.13 [Arabidopsis thalisms]	73	0.024	2
	gi 1575754	ADF glucose pyrophosphorylase small subu.	60	0.024	2
	gi 13347147	Probable sugar nucleotidyltransferase Cj	84	0.028	1
	gi. 21403	ADF-glucose gyrophosphorylass: glucose-l-	86	0.028	3
	gi 100426	Glucosa-1-phosphate ademylyltransferase	80	0.028	2
	gi 633678	ADF-glucose pyrophosphonylase (Spinecia	80	0.038	3:
	gi 10638150	UNP-glucose-1-phosphate unidylyltransfer.	76	0.038	3
	år 3330032	Glucosa~l~phosphate adenylyltransferasa	89	0.03	35
	GT 532723	GLACOSE-1-PHOSPHATE ADSTYLYLTRADSFERASE	80	0.03	33
	gi 7340287	Small subunit ADF glucose pyrophosphoryl	80	9.03	2
	gi 1707939	GLUCUSE-I-FROSPEATE ADMINITIANSFERASE	80	0.031	2
	gi 1237082	MDP-glucose pyrophosphorylase (Pisum sat	80	0.032	2
301.	gi 1707943	GLINCOSE-1-PROSPHATE RUSHYLYL/TRANSFERASS	80	0.032	2
	gi 1707340	GLUCOSE-1-PHOSPHATS ADSTYLYLTRANSFERASE	90	0.032	3
303.	gi. 3015514	ADFG pyrophosphorylase small subunit [Ar.,	80	0.033	2
304.	gi (1071859	Glucose-l-phosphate adenylyitransferass	80	0.033	3
305.	gi 1325984	ADP-glucose pyrophosphorylase small subu.	80	0.033	3
306.	gi 232164	GLICOSE-1-PROSPHATE ADEXILYLITHANSPERASE	ដង	0-033	5
307.	gi 7434881	Glucose-1-phosphate adenylyitransfersse	80	0.033	2
308.	g1 1707930	GLUCOSE-1-PHOSPHATE ADENYLYLYRANSFERASE	82	9.933	22
309.	gi 7434879	Glucose-1-phosphate adenylyltransferase	80	9.933	2
310.	G7 \$652084	ADP-glucose pyrophosphorylase small subu.	80	0.033	2
344.	çi 7434871	Glucope-l-phosphate adenylyltransferase	80	0.933	2
	gi 7434891	Glucose-i-phosphate adenylyltransferase	62	9.944	2
	gi 1707938	CLUCOSE-1-PHOSPHATE ADENVLYLATRANSPERASE	52	9.044	2
	gi 2149021	ADPG pyrophosphorylase large subunit (Ar	75	0.044	2
	g1:4586350	Glucose-i-phosphate adenylyitransferase	75	0.044	2
	gi 1707923	GLIKTOSE-1-PHOSPHRITE ADEINILVINRANSYERASE	52	0.044	2
	GT 100280	Glucope-1-phosphate adenylyltransferase	62	0.045	2
	gi. 5091.608	Identical to gb/D50317 ADF glucose pyrop	62	0.045	5
	g1 7434885	Glucose-i-phosphace ademylylcransferase	73	0.057	3
	gs. 7688095	ACF-glucose pyrophosphorylase small subu.	35	0.059	2
	g1 7434886	Glucose-1-phosphace ademylylcransfersse	54	0.064	3
	gi 7488396	Translation regulator GCD6 homolog TSAZL.	81	0.065	1
	gi 5320417	Translation initiation factor eIF-28 eps.	81	0.065	1
	gi 7521184	Probable manuse-1-phosphate quanyltrans	64	0.065	3
	gi 1197640	Dina (Yersinia enterocolitica (type 0:8))	86	5.588	3
	gi 154448	ADF-qlucose pyrophosphorylase (Synechocy.	86	8.088	1
	gi 7447199	Glucose-1-phosphate cytidylyltransferasa.	80	0.000	ì
366	Set inceres	CARCOLOGY & PASSASTANCE COSCIONAS ALLEGANOS	W-16	See A See Asked	

FIG. 19(G)

328.	94 (2707944	GLIXOSE-1-PROSENATE ADMINISTRANSFERASE	86	0.088	.1
339.	gi 3023677	fromable translation initiation factor ru	77	0.08	3
335.	gi 121293	Charare-1-proferte algeriatiraneferase	62	0.2	3
331.	gi}7434883	Glucose-i-phosphate adenylyitransferase	73	0.1	3
332.	gi 97183	Lic-1 protein C - Heemophilus influenzae.	75	0.13	2
	gd 7404390	FICE BECEFFE	33	9.12	3.
	gi 9797361	Probable mannose-i-phosphate guanyltrans	55	0.13	\mathfrak{A}
	93 7434873	Glucose-1-phosphane adenyiylinansfarese	7.2	0.14	2
	GE 2707933	CLIETOSE-1-PROSPRATE ADENTLATRANSFERASE	72	0.38	2
	gi 3023576	PROGRABLE TRANSLATION INITIATION FACTOR EL.	78	0.16	3.
	gi 1707932	GIXCOSE-1-ETOSPHATE ADMYLYLTFAMSFYTRASE	71	0.17	2
	gi 1707929	CLINCES-1-PRESIDENTA ADENVIATARANSPERASE	69	5.18	3
	gi 7671230	ADP-glucose pyrophosphorylase caralytic	69	0.28	3
	Si 7008160	Glucuse-i-phosphate adenylyltransferses	63	9,31	2
	94 11923507	Futative glucose-1-F-cytidylyltransferss.	77	0.21	ã
	gi 1840114	ADP-gluccee pyrophosphorylase large suint.	69	0.24	2
	gi 5739461	Galf (Excherichie coli)	SĄ	0.27	3
	gi 2506458	GLICOSE-I-PHUGFHATE ADENCATARANSFERASS	76	0,28	3.
	gi 7434976	Gloccae-1-phosphate adamyivitransferase	72	0.29	Ž.
	G1 1778436	ADF-plucese pyrophosphorylase large sulm.	89	0.31	2
	gi 7448187	Probable glucese-1-phosphate thymidylylt	75	0.36	ĩ
	gi 7434893	Glucose 1-phosphate adenylyltransferase	80	0.39	3
	gi 5882732	Similar to objAF135433 CDF-manuse pyrop	74	0.46	ž
	G1 6646773	Printive CCP-manucse pyrophosphorylass:	74	0.46	1
	gi 7434858	Glucese-1-phosphete adenylyltransferase	57	0.43	<u>.</u>
	G1 7636883	Gincose-1-phosphate ademylyltransfermer	57	0.49	3
	gi 5917791	ADP-glucose pyrophosphorylase large suba	59	0.5	ã
	91 7471938	Giucose-1-phosphate adenylyltransferase	72	0.53	3
	gi 7471937	Glucose-l-phosphate adenvivitransfersee	53	0.59	Š
	gi 31386853	PROBABINE GLIDOSE-1-PROSPERTE ADENVLYETHA.	68	0.8	2
	gi 7434869	Glucese-1-phosphete adenylyltranefersse	62	0,€	2
	gi 7492700	Probable mannose-1-phosphate gausyl tran.	\$7	0.52	2
	gi 2139037	Glucose-1-phosphete ademylyltransferase	72	0.69	à
	gi 2146810	Glucose-1-phosphate adenylylcransferace	72	0.69	3
	gi 7448100	COP-B-acetylgiucosamine pyrophosphorylas	48	0.78	4
	gi 135927	ODE-M-ACETYLOGECOSABINE PYROPHORPLAS.	43	8.79	
	gi 7522214	Olucese-1-phosphete admylyltransferse	86	0.88	5 2
	91 4544432	Putative GDP-manucae pyrophosphurylass	7ô	0.89	î
	gill##39507	Mannose-1-phosphote guanyitronsferace re.	70	0.89	7
	gi 7447200	Glucose-l-phosphate cyticylyltransceraes.	70	0.89	3
	gilkanass	GLUCUSE-1-PROSPIRATE ADENYLVLTRANSFERASE	63	0.93	2
	93 2383290	Ribescral protein S4 homolog (Prypaneaum.	54	0.94	
	gi 5923897	All-glucose pyrophosphorylase lerge sukt.	63	0.95	3 2
	gi. 7434864	Glucose-l-phosphate adamylylcransferase	53	0.95	ž
	gi 5626364	Galf-like (Bradyrhizobium japonicum)	63	9.99	1
	gi 111359799	Type 20 Protein Phosphatase related prot	69	0.95	*
	gi 7434874	Glucosa : phosphers adenyiyitzanaisa proces	59	9.98	\$
	gi 7543739	Pypothetical protein; 66083-64412 (Arabia	68	0.98	ž
		ADP-glucose pyrophosphorylese (Thomass b.	83	9.99	ŝ
	gá (5783,881 gá (58832075	Ann-glacose pyraphosphorylass (lposses b	53 53	4	2
	gi 7331959	Contains similarity to Piem femilies FFG.	54	ì	3
		Fibronecin-binding protein - Streptococc.	56 56	3.	.3
	gi (479426 ~ (476920	Sannasa-1-phosphane guarrylyltransforase -	90 66	3. }	3.
	94 (476970 (2211686	ANN-glucose pyrophosphonylase large sulu.	63.	3.	ž Ž
	gi 3211959	ADF-glucose pyrophosphorylase large subu.	38 58	3. 1	3
200.	g£ [7671238	cause grandente propriétaire plante y comes destrites dell'étée	.00		۵

FIG. 20(a)

FIG. 20(b)

FIG. 21

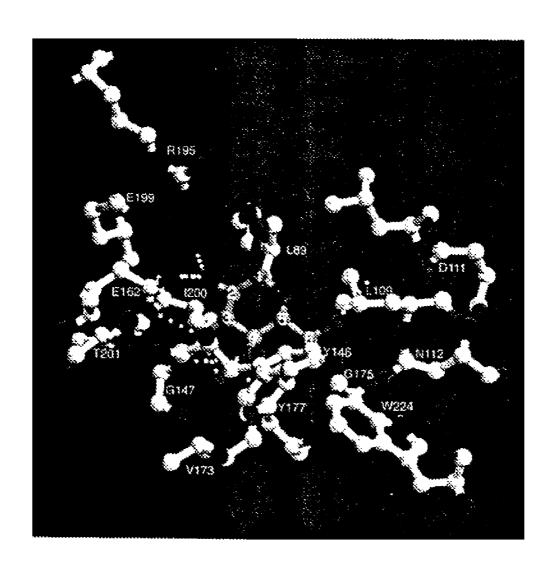


FIG. 22

FIG. 23

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ACTIVE-SITE ENGINEERING OF NUCLEOTIDYLYLTRANSFERASES AND GENERAL ENZYMATIC METHODS FOR THE SYNTHESIS OF NATURAL AND "UNNATURAL" UDP-AND TDP-NUCLEOTIDE SUGARS

This application is a divisional application of U.S. patent application Ser. No. 10/013,542, filed Dec. 13, 2001, now U.S. Pat. No. 7,122,359, which claims the benefit of U.S. Application Ser. No. 60/254,927, filed Dec. 13, 2000 each of which are incorporated herein in there entirety.

FIELD OF THE INVENTION

The present invention is directed to nucleotidylyl-transferases and mutant nucleotidylyltransferases having altered substrate specificity and methods for their production.

The present invention is also directed to methods of synthesizing desired nucleotide sugars using natural and/or 20 mutant $\rm E_p$ or other nucleotidyltransferases, preferably $\rm E_p$ or other nucleotidylytransferases modified by the present methods. Additionally, the present invention is directed to nucleotide sugars synthesized by the present methods.

The present invention is further directed to new glycosyl ²⁵ phosphates, and methods for making them.

BACKGROUND OF THE INVENTION

Many bioactive metabolites possess unusual carbohy- 30 drates required for molecular recognition. (See for example, Liu, H.-w.; Thorson, J. S. Ann. Rev. Microbiol., 1994, 48, 223-256; Weymouth-Wilson, A. C. Nat. Prod. Rep. 1997, 14, 99-110; In Macrolide Antibiotics, Chemistry, Biology and Practice; Omura, S. Ed., Academic Press: New York; 1984; 35 Johnson, D. A.; Liu, H.-w. Curr. Opin. Chem. Biol. 1998, 2, 642-649; and Trefzer, A.; Salas, J. A.; Bechthold, A. Nat. Prod. Rep. 1999, 16, 283-299.) In fact, roughly 70% of current lead compounds in modern drug discovery derive directly from natural products, many of which are glycosy- 40 lated metabolites. (See Thorson, J. S. et al. Nature's Carbohydrate Chemists: The Enzymatic Glycosylation of Bioactive Bacterial Metabolites. Curr. Org. Chem. manuscript in press, (2000); and references therein and Weymouth-Wilson, A. C. The Role of Carbohydrates in Biologically Active Natural 45 Products. Nat. Prod. Rep. 14, 99-110 (1997)). Examples of pharmaceutically important glycosylated metabolites include, for example, amphotericin, megalomicin/erythromycin, mithramycin, doxorubicin, vancomycin and calicheamicin, as shown in FIG. 5. While it is known that the 50 sugar moieties of these pharmaceutically important metabolites often define their corresponding biological activity, (see Weymouth-Wilson, A. C., The Role of Carbohydrates in Biologically Active Natural Products, Nat. Prod. Rep. 14, 99-110 (1997)), efficient methods to systematically alter these essen- 55 tial carbohydrate ligands are still lacking.

In metabolite biosynthesis, glycosylation begins with the nucleotidylyltransferase-catalyzed activation of a sugar phosphate as a nucleotide diphosphosugar (NDP-sugar) donor. After activation, a number of enzymatic processing for reactions often occur (e.g., deoxygenation, transamination, oxidation/reduction, epimerization, alkylation, and decarboxylation) prior to the culminating glycosyltransferase-catalyzed attachment to the aglycon. (Liu, H.-w. & Thorson, J. S. Pathways and Mechanisms in the Biogenesis of Novel 65 Deoxysugars by Bacteria. Ann. Rev. Microbiol. 48, 223-256 (1994); Kirschning, A., Bechtold, A. F-W. & Rohr, J. Chemi-

2

cal and Biochemical Aspects of Deoxysugars and Deoxysugar Oligosaccharides. Top. Curr. Chem. 188, 1-84 (1997); Johnson, D. A. & Liu, H.-w. Mechanisms and Pathways from Recent Deoxysugar Biosynthesis Research. Curr. Opin. Chem. Biol. 2, 642-649 (1998); Hallis, T. M. & Liu, H.-w. Learning Nature's Strategies for Making Deoxy Sugars: Pathways, Mechanisms, and Combinatorial Applications. Acc. Chem. Res. 32, 579-588 (1999); Johnson, D. A. & Liu, H.-w. In Comprehensive Chemistry of Natural Product Chemistry (Barton, D.; Nakanishi; K.; Meth-Cohn, O. eds), Elsevier Science, Oxford, 311, (1999); Trefzer, A., Salas, J. & Bechthold, A. Genes and Enzymes Involved in Deoxysugar Biosynthesis in Bacteria. Nat. Prod. Rep. 16, 283-299 (1999); and Bechthold, A. & Rohr, J. In New Aspects of Bioorganic 15 Chemistry (Diederichsen, U.; Lindhorst, T. K.; Wessjohann, L.; Westerman, B., eds.) Wiley-VCH, Weinheim, 313,

The glycosyltransferases that incorporate these essential ligands are thought to rely almost exclusively upon UDP- and TDP-nucleotide sugars: however some have demonstrated promiscuity towards the sugar donor, (e.g., Gal, D-galactose; Glc, D-glucose; Man, D-mannose; NTP, nucleotide triphosphate; pFPTC, pentafluorophenoxythiocarbonyl; TDP, thymidine diphosphate; TMP, thymidine monophosphate; TTP, thymidine triphosphate; UDP, uridine diphosphate.) Genetic experiments suggest that downstream glycosyltransferases in secondary metabolism are promiscuous with respect to their NDP-sugar donor, setting the stage for the expansion of "combinatorial biosynthesis" approaches to change metabolite glycosylation. (See Madduri, K. et al., Production of the antitumor drug epirubicin (4'-epidoxorubicin) and its precursor by a genetically engineered strain of Streptomyces peucetius Nat. Biotech. 16, 69-74 (1998); and Hutchinson, C. R. Combinatorial Biosynthesis for New Drug Discovery. Curr. Opin. Microbiol. 1, 319-329 (1998).) This information has led to the exploitation of the carbohydrate biosynthetic machinery to manipulate metabolite glycosylation, (Madduri, K.; Kennedy, J.; Rivola, G.; Inventi-Solari, A.; Filppini, S.; Sanuso, G.; Colombo, A. L.; Gewain, K. M.; Occi, J. L.; MacNeil, D. J.; Hutchinson, C. R. Nature Biotech. 1998, 16, 69-74; and Zhao, L.; Ahlert, J.; Xue, Y.; Thorson, J. S.; Sherman, D. H.; Liu, H.-w. J. Am. Chem. Soc., 1999, 121, 9881-9882 and references therein), revitalizing interest in methods to expand the repertoire of available UDP- and TDP-sugar nucleotides. (See Zhao, Y.; Thorson, J. S. J. Org. Chem. 1998, 63, 7568-7572; and Elhalabi, J. M.; Rice, K. G. Cur. Med. Chem. 1999, 6, 93-116.)

These in vivo methods are limited by both a particular host's biosynthetic machinery and the specific host's tolerance to each newly constructed metabolite. Further, in vitro progress in this area is limited by the availability of the required NDP-sugar substrates. (Solenberg, P. J. et al., Production of Hybrid Glycopeptide Antibiotics in vitro and in *Streptomyces toyocaensis*. Chem. & Biol. 4, 195-202 (1997).)

Thus, there is a need for a greater variety of available NDP-sugar substrates.

Salmonella enterica LT2 α -D-glucopyranosyl phosphate thymidylyltransferase (E $_p$) is a member of the prevalent nucleotidylyltransferase family responsible for the reversible conversion of α -D-hexopyranosyl phosphate and NTP to the corresponding NDP-sugar nucleotide and pyrophosphate. Of the many nucleotidylyl-transferases studied, the NDP-sugar nucleotide-forming thymidylyltransferases have received the least attention in prior work. (See Lindquist, L.; Kaiser, R.; Reeves, P. R.; Lindberg, A. A. Eur. J. Biochem. 1993, 211, 763-770, and Gallo, M. A.; Ward J.; Hutchinson, C. R. Microbiol. 1996, 142, 269-275.) Even in E $_p$, substrate specificity

studies prior to the work of the present inventors were limited to only a few available hexopyranosyl phosphates. (See Lindquist, L.; Kaiser, R.; Reeves, P. R.; Lindberg, A. A. Eur. J. Biochem. 1993, 211, 763-770.)

3

SUMMARY OF THE INVENTION

The present invention is directed to methods of engineering or mutating nucleotidylyltransferases, such as \mathbf{E}_p , to vary their specificity in a directed manner. The invention is also 10 directed to nucleotidylyl-transferases and mutated nucleotidyltransferases, preferably \mathbf{E}_p or other nucleotidyltransferases modified by the present methods. The present invention is further directed to mutant \mathbf{E}_p and other nucleotidyltransferases with altered substrate specificity, 15 methods for their production, and methods of producing nucleotide sugars, which utilize these nucleotidyl-transferases

The present invention is also directed to methods of synthesizing desired nucleotide sugars using natural and/or 20 mutated ${\rm E}_p$ or other nucleotidylyltransferases, preferably ${\rm E}_p$ or other nucleotidylyltransferases mutated by the present methods. Additionally, the present invention is directed to nucleotide sugars synthesized by the present methods.

Examples of nucleotide sugars produced the present meth- 25 ods (that is, via the exploitation of the promiscuity of E_n) include, but are not limited to Thymidine 5'-(α-D-glucopyranosyl diphosphate) (58); Uridine 5'-(α-D-glucopyranosyl diphosphate) (59); Thymidine 5'-(2-deoxy-α-D-glucopyranosyl diphosphate) (60); Uridine 5'-(2-deoxy-α-D-glucopyranosyl diphosphate) (61); Thymidine 5'-(3-deoxy-α-D-glucopyranosyl diphosphate) (62); Uridine 5'-(3-deoxy-α-Dglucopyranosyl diphosphate) (63); Thymidine 5'-(4-deoxyα-D-glucopyranosyl diphosphate) (64); Uridine 5'-(4-deoxyα-D-glucopyranosyl diphosphate) (65); Thymidine 5'-(6- 35 deoxy-α-D-glucopyranosyl diphosphate) (66); Uridine 5'-(6deoxy-α-D-glucopyranosyl diphosphate) (67); Thymidine 5'-(α-D-mannopyranosyl diphosphate) (68); Uridine 5'-(α-D-mannopyranosyl diphosphate) (69); Thymidine 5'-(α-Dgalactopyranosyl diphosphate) (70); Uridine 5'-(α-D-galac- 40 diphosphate) (71); Thymidine 5'-(α-Dtopyranosyl allopyranosyl diphosphate) (72); Uridine 5'-(α-Ddiphosphate) (73); Thymidine 5'-(α-Dallopyranosyl altropyranosyl diphosphate) (74); Uridine $5'-(\alpha-D$ diphosphate) (75); Thymidine 5'-(α-D- 45 altropyranosyl gulopyranosyl diphosphate) (76); Uridine $5'-(\alpha-D$ gulopyranosyl diphosphate) (77); Thymidine 5'-(α-Didopyranosyl diphosphate) (78); Uridine $5'-(\alpha-D$ diphosphate) (79); Thymidine 5'-(α -Didopyranosyl talopyranosyl diphosphate) (80);Uridine 5'-(α-D- 50 talopyranosyl diphosphate) (81); Thymidine 5'-(6-amino-6deoxy-α-D-glucopyranosyl diphosphate) (109); Uridine 5'-(6-amino-6-deoxy-α-D-glucopyranosyl diphosphate) (110); Thymidine 5'-(4-amino-4-deoxy-α-D-lucopyranosyl diphosphate) (111); Uridine 5'(4-amino-4-deoxy-α-D-glucopyra- 55 nosyl diphosphate) (112); Thymidine 5'-(3-amino-3-deoxyα-D-glucopyranosyl diphosphate) (113); Uridine 5'-(3amino-3-deoxy-α-D-glucopyranosyl diphosphate) (114); 5'-(2-amino-2-deoxy-α-D-glucopyranosyl diphosphate) (115); Uridine 5'-(2-amino-2-deoxy-α-D-glu- 60 copyranosyl diphosphate) (116); Thymidine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate) (117); Uri-5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate) (118); Thymidine 5'-(4-acetamido-4-deoxy-α-D-glucopyranosyl diphosphate) (119); Uridine 5'-(4-aceta- 65 mido-4-deoxy-α-D-glucopyranosyl diphosphate) (120); Thymidine 5'-(3-acetamido-3-deoxy-α-D-glucopyranosyl

4

diphosphate) (121); Uridine 5'-(3-acetamido-3-deoxy- α -D-glucopyranosyl diphosphate) (122); Thymidine 5'-(2-acetamido-2-deoxy- α -D-glucopyranosyl diphosphate) (123); Uridine 5'-(2-acetamido-2-deoxy- α -D-glucopyranosyl diphosphate) (124); Thymidine 5'-(4-amino-4,6-dideoxy- α -D-glucopyranosyl diphosphate) (125); and Uridine 5'-(4-amino-4,6-dideoxy- α -D-glucopyranosyl diphosphate) (126). Nucleotide sugars such as these, and methods for making them, are provided by the present invention.

Examples of nucleotide sugars according to the present invention, which may be produced by designed mutants of E_n include, but are not limited to, Thymidine 5'-(6-acetamido-6deoxy-α-D-glucopyranosyl diphosphate) (117); Uridine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate) (118); Thymidine 5'-(α-D-glucopyran-6-uronic acid diphosphate) (130); Uridine 5'-(α-D-glucopyran-6-uronic acid diphosphate) (131); Thymidine 5'-(2-acetamido-2-deoxy-α-D-glucopyranosyl diphosphate) (123); Uridine 5'-(2-acetamido-2-deoxy-α-D-glucopyranosyl diphosphate) (124); Thymidine 5'-(4-amino-4,6-dideoxy-α-D-glucopyranosyl diphosphate) (125); Uridine 5'-(4-amino-4,6-dideoxy-α-Dglucopyranosyl diphosphate) (126); Thymidine 5'-(α-D-arabinopyranosyl diphosphate) (128); and Uridine 5'-(α-D-arabinopyranosyl diphosphate) (129). These nucleotide sugars, and methods for making them, are provided by the present invention.

The present invention is also directed to new glycosyl phosphates, and methods for making them. Examples of these new glycosyl phosphates and methods for synthesizing them are represented for example in FIG. 1(b).

The present inventors have discovered that \mathbf{E}_p is pliable in terms of its substrate specificity. The present inventors have also discovered the three dimensional structure of \mathbf{E}_p and the molecular details of \mathbf{E}_p substrate recognition.

In general, the present invention provides a very rapid method of converting sugar phosphates to nucleotide diphosphosugars.

The present invention will broadly impact efforts to understand and exploit the biosynthesis of glycosylated bioactive natural products, many of which are pharmacologically useful. (See Thorson, J. S.; Shen, B.; Whitwam, R. E.; Liu, W.; Li, Y.; Ahlert, J. Bioorg. Chem., 1999, 27, 172-188; Whitwam, R. E.; Ahlert, J.; Holman, T. R.; Ruppen, M.; Thorson, J. S. J. Am. Chem. Soc., 2000, 122, 1556-1557; Thorson, J. S.; Sievers, E. L.; Ahlert, J.; Shepard, E.; Whitwam, R. E.; Onwueme, K. C.; Ruppen, M. Cur. Pharm. Des., 2000, manuscript in press; and J. S. Thorson, T. J. Hosted Jr., J. Jiang, J. B. Biggins, J. Ahlert, M. Ruppen, Curr. Org. Chem. 2000.)

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1. FIG. 1(a) depicts a reaction according to the present invention, catalyzed by E_p . In this reaction, the enzyme catalyzes the reversible conversion of an α -D-hexopyranosyl phosphate (such as an α -D-glucopyranosyl phosphate)(2) and NTP, such as TTP (1) to the corresponding NDP-sugar nucleotide (for example a TDP-sugar nucleotide, such as TDP-Glc)(3) and pyrophosphate (4). Glc1P (2) depicted in the reaction of FIG. 1(a) is a commercially available α -D-hexopyranosyl phosphate (although other α -D-hexopyranosyl phosphates that may be used in accordance with the present invention may include those synthesized from free sugars.)

FIG. $\mathbf{1}(b)$ depicts the synthesis of α -D-hexopyranosyl phosphates.

FIG. 2. \mathbb{E}_p -Catalyzed Conversion of Substrates (a) Percent conversion= $[A_p/(A_p+A_T)]\times 100$, where A_P is the NDP-sugar

product peak integration and A_T represents the NTP peak integration. HRMS for all observed products reported in the supporting information. (b) Standard retention times: TDP, 4.5 min; TTP, 7.2 min; UDP, 4.0 min; UTP, 6.1 min. (c) Commercially available. (d) Coelutes with commercially available standard. (e) Product hydrolysis observed (43, 7.6% TDP and 10.2% UDP). (f) Adjusted for the 2:1 α/β -28. (g) In contrast to previously published studies (See Lindquist, L; Kaiser, R.; Reeves, P. R.; Lindberg, A. A., Eur J. Biochem, 1993, 211, 763-770). (h) No products observed.

FIG. 3. FIG. 3(a) sets forth a reaction according to the present invention, catalyzed by E_p . FIG. 3(b) shows an overview of the key steps in the described syntheses of E_p substrates analogs. The box highlights the point from which the aminodeoxy-α-D-glucose phosphate series and N-acetyl- 15 aminodeoxy-α-D-glucose phosphate series diverge. The reaction conditions of the steps are as follows: (a) TMSSEt, ZnI₂ (84.2% overall yield); (b) i) MeONa, ii) NaH, BnBr (77.3% average overall yield, two steps) (c) i) SnCl₂, PhSH, Et₃N, ii) Ac₂O, pyr (84.0% average overall yield, two steps); 20 (d) i) Tf₂O, pyr, ii) NaN₃ (87.7% average overall yield, two steps); (e) i) NaOMe, ii) CH₃CH(OCH₃)₂CH₃, TsOH, iii) NaH, BnBr, iv) HCl/MeOH, v) BzCl, DMAP, Et₃N (87.3% average overall yield, five steps); final steps (not shown): i) phosphorylation, ii) reductive deprotection, iii) cation 25 exchange to give the Na⁺ salt (44.4% average overall yield).

FIG. 4. Percent conversion to product using substrates according to the present invention.

FIG. **5**. Examples of pharmacologically important glycosylated metabolites. The general nucleotidylyl-transferase-catalyzed formation of NDP-sugars is highlighted in the box while the carbohydrate ligands of each metabolite are accentuated in red. Note the difference between erythromycin from *S. erythrea* and megalomicin from *M. megalomicea* is the addition of a third sugar megosamine (highlighted by the arrow).

sequence.

FIG. **13**. Shows alignme Adenylyltransferases.

FIG. **15**. Shows alignme 1-P Guanylyltransferases.

FIG. **16**. Shows alignme 1-P Guanylyltransferases.

FIG. 6. Representative region of the density-modified experimental electron density map showing the substrate binding pocket in the \mathbf{E}_p UDP-Glc structure configured at 1.2.sigma.

FIG. 7 Quaternary structure of E_p bound to UDP-Glc or dTTP. (a) Two 90 degree views of the E_p tetramer bound to four molecules of UDP-Glc. (b) The E_p tetramer bound to eight molecules of dTTP.

FIG. **8**. Structures of the E_p monomer and structural 45 otides the homologs SpsA and GlmU. The β strands and α helices of the α/β open sheet Rossmann fold are shown in red and green respectively, while variable regions are shown in yellow. (a) Two 90 degree views of the E_p monomer (upper) and the corresponding structures of SpsA (lower left) and GlmU 50 mutants (lower right). (b) The folding topology of E_p , SpsA, and GlmU. trating strating structures of the sugar FIG. 2

FIG. 9. Close up views of the E_p active site. Hydrogen bonds are depicted by green dashed lines. (a) Interactions between E_p and the dTTP substrate (left) and the UDP-Glc 55 product (right). (b) Interactions between E_p and the glucose moiety in the sugar binding pocket. (c) Two different views of dTTP bound in the 'accessory' site at the monomer interface. The different chains of the tetramer are labeled either in blue (chain-A) or red (chain-B). The β -phosphate of dTTP hydrogen bonds with both His117 of chain-A and Gly221 of chain-B

FIG. 10. (a) The proposed enzymatic mechanism based on the structures of substrate- and product-bound E_p . (b) The determination of E_p steady state kinetic parameters. The conditions for the E_p assay conditions and HPLC resolution of reactants and products were similar to those described in

6

Jiang, J., Biggins, J. B. & Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP-and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000). The Lineweaver-Burke plots of from assays (done in triplicate) varying dTTP concentration as a function of α -D-glucose-1-phosphate concentration (mM): 0.5 (\square), 0.3 nM (\circ), 0.2 (\diamond), 0.1 (Δ) and 0.05 (\blacksquare). (c) Secondary plot from FIG. 6b (dTTP K_m=0.7±0.2; V_{max}=0.03±0.01 mM min⁻¹). (d) The Lineweaver-Burke plots of assays (done in triplicate) varying α -D-glucose-1-phosphate concentration as a function of dTTP concentration (mM): 0.25 (\square), 0.15 nM (\circ), 0.1 (\diamond), 0.05 (Δ) and 0.02 (\blacksquare). (e) Secondary plot from FIG. 6d (α -D-glucose-1-phosphate K_m=0.3±0.1; V_{max}=0.03±0.02 mM min⁻¹).

FIG. 11. Percent conversion of sugar phosphates according to the present invention by wild-type and mutant enzymes. The alterations from native substrate (Glc-1-P, 1) are highlighted in red. For the mutant pool, mutants Asp41Asn, Glu62Asp, Thr201A and Trp224His were pooled, concentrated and an aliquote constituting 60 μ g of each mutant (corresponding to 3.5 U E_p) was utilized for the assay.

Percent conversion was determined as described in Jiang, J., Biggins, J. B. & Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP-and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000).

§Represents less than 5% conversion to product.

FIG. 12. Shows alignment of the Thymidylyltransferase sequence.

FIG. 13. Shows alignment of the \mathbf{E}_p sequence with Glc-1-P Uridylyltransferases.

FIG. 14. Shows alignment of the ${\rm E}_p$ sequence with Glc-1-P Adenylyltransferases.

FIG. 15. Shows alignment of the E_p sequence with Man-1-P Guanylyltransferases.

FIG. 16. Shows alignment of the E_p sequence with NAcGlc-1-P Uridylyltransferases.

FIG. 17. Shows alignment of the ${\rm E}_p$ sequence with Glc-1-P Cytidylyltransferases.

FIG. 18. Shows general Nucleotidylyltransferase Alignment.

FIG. 19. FIG. 19 is a BLASR analysis for E_p sequences, showing sequences producing high-scoring segment pairs.

FIG. 20. FIGS. 20(a) and 20(b) depict NDP-sugar nucleotides that may be prepared using nucleotydylyl-transferases as enzymes in accordance with the present invention.

FIG. 21. Interaction between \mathbf{E}_p and the glucose moiety in the sugar binding pocket.

FIG. 22. Summary of sugar phosphate accepted by E_p and mutants

FIG. 23. One dimensional representation of FIG. 21 illustrating some of the important contacts and potential sites for engineering promiscuity of nucleotidyly-transferases.

DETAILED DESCRIPTION OF THE EXEMPLARY EMBODIMENTS

The present inventors discovered that the *Salmonella enterica* LT2 rmlA-encoded α -D-glucopyranosyl phosphate thymidylyltransferase (E $_p$), (also referred to as dTDP-glucose synthase, dTDP-glucose pyrophosphorylase, thymidine diphosphoglucose pyrophosphorylase and thymidine diphosphate glucose pyrophosphorylase), which catalyzes the conversion of α -D-glucopyranosyl phosphate (Glc-1-P) and dTTP to dTDP- α -D-glucose (TDP-Glc) and pyrophosphate (PP $_i$), displays unexpected promiscuity toward both its nucleotide triphosphate (NTP) and its sugar phosphate substrates.

Through a substrate specificity reevaluation of *Salmonella enterica* LT2 α -D-glucopyranosyl phosphate thymidylyltransferase ($\rm E_p$), the present inventors made the surprising discovery that this enzyme can convert a wide variety of phosphates, including for example, α -D-hexopyranosyl phosphates, including, but not limited to, deoxy- α -D-glucopyranosyl, aminodeoxy- α -D-hexopyranosyl and acetamidodeoxy- α -D-hexopyranosyl phosphates to their corresponding dTDP- and UDP-nucleotide sugars.

This discovery led to the invention by the present inventors of general chemo-enzymatic methods of rapidly generating nucleotide diphosphosugar reagents. These methods allow for the provision of a substrate set for developing in vitro glycosylation systems, which are useful for, inter alia, in vitro production of known bioactive metabolites and of new bioactive metabolites.

 α -D-Hexopyranosyl Phosphates and Methods of Making the Same

An embodiment of the invention includes α -D-hexopyranosyl phosphates, methods including combining these phosphates with NTP in the presence of nucleotidylyl-transferase, which may be wild type or mutated, and nucleotide sugars produced by converting such hexopyranosyl phosphates using nucleotidylyl-transferases, such as E_p .

 E_p is encoded by rmlA, which was previously known as rfbA (Reeves et al. Trends Microbiol. 1996, 4, 495-502). The rmlA-encoded E_p was overexpressed in *E. coli* to provide the desired E_p as >5% of the total soluble protein. The corresponding E_n was purified to near homogeniety with a specific 30 activity of 110 U mg⁻¹, a 2-fold improvement over the previously reported values. (See Lindquist, L.; Kaiser, R.; Reeves, P. R.; Lindberg, A. A. Eur. J. Biochem., 1993, 211, 763-770.) An (NH₄)2SO₄ precipitate of E. coli-prfbA-C crude extracts was dialyzed against buffer B (20 mM 35 Tris.HCl, 1 mM EDTA, pH 7.5) The dialysate was resolved by anion exchange (DE52, 3×15 cm, 50 mL buffer B wash followed by a linear gradient of 0-500 mM NaCl, 1.0 mL min⁻¹) and the E_n fractions combined, concentrated and further resolved by FPLC gel filtration (S-200, 2×70 cm, 50 mM 40 Tris.HCl, 200 mM NaCl, pH 7.5). The purified E_p was stored in aliquots (-80° C.) until used.

Although α -D-glucopyranosyl phosphate (2) (FIG. 2), α -D-mannopyranosyl phosphate (compound 56) (FIG. 2) and α -D-galactopyranosyl phosphate (57) (FIG. 2) were commercially available for examination as potential substrates for E_p , most of the α -D-hexopyranosyl phosphates examined were synthesized from free sugars.

For synthetically derived α -D-hexopyranosyl phosphates, particularly glycosyl phosphates, a general phosphorylation 50 strategy from the appropriately protected precursor relied upon

i) anomeric activation via the ethy 1-thio-β-D-pyranoside [to form e.g., Ethyl 2,3,4-tri-O-benzoyl-6-deoxy-1-thio-β-D-glucopyranoside (9), Ethyl 2,3,6-tri-O-benzoyl-4-deoxy-1-55 thio-β-D-glucopyranoside (17), Ethyl 2,4,6-tri-O-benzoyl-3-deoxy-1-thio-β-D-glucopyranoside (25), Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio-β-D-glucopyranoside (30), Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio-β-D-allopyranoside (35) and Ethyl 3,4,6-tri-O-benzoyl-2-deoxy-1-thio-β-D-glucopyranoside (40) (The α/β-40 mixture (1:1.5) was chromatographically resolved.) (FIG. 1(b))],

ii) deprotection/reprotection [to form e.g., thyl 2,3,4-tri-O-benzyl-6-deoxy-1-thio-β-D-glucopyranoside (10), Ethyl 2,3, 6-tri-O-benzyl-4-deoxy-1-thio-β-D-glucopyranoside (18), 65 Ethyl 2,4,6-tri-O-benzyl-3-deoxy-1-thio-β-D-glucopyranoside (26), Ethyl 2,3,4,6-tetra-O-benzyl-1-thio-β-D-gulopyra-

noside (31), and Ethyl 2,3,4,6-tetra-O-benzyl-1-thio- β -D-allopyranoside (36) (FIG. $\mathbf{1}(b)$)],

iii) phosphorylation [to form e.g., Dibenzyl-(2,3,4-tri-Obenzyl-6-deoxy-α-D-glucopyranosyl) phosphate (11), Dibenzyl-(2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranosyl) phosphate (19), Dibenzyl-(2,4,6-tri-O-benzyl-3-deoxy-α-D-glucopyranosyl) phosphate (27), Dibenzyl-(2,3,4,6-tetra-O-benzyl-α-D-gulopyranosyl) phosphate (32), Dibenzyl-(2,3,4,6-tetra-O-benzyl-α-D-allopyranosyl) phosphate (37), and Dibenzyl-(3,4,6-tri-O-benzoyl-2-deoxy-α-D-glucopyranosyl) phosphate (41) (FIG. 1(b))], and

iv) complete deprotection [to form e.g., Disodium 6-deoxy- α -D-glucopyranosyl phosphate (12), Disodium 4-deoxy- α -D-glucopyranosyl phosphate (20), Disodium 3-deoxy- α -D-glucopyranosyl phosphate (28), Disodium α -D-gulopyranosyl phosphate (33), Disodium α -D-allopyranosyl phosphate (38) and Disodium 2-deoxy- α -D-glucopyranosyl phosphate (43) (FIG. 1(b))].

In FIG. 1(b): (a) Ph₃P, CCl₄; (b) Ac₂O, pyr; (c) (i) LiAlH₄, (ii) AcOH/HCl, (iii) BzCl, pyr; (d) BzCl, pyr; (e) pFPTC-Cl, DMAP; (f) (n-Bu)3SnH; (g)(i) NaH, imidazole; (ii) CS₂; (iii) CH₃I; (h) AIBN, (n-Bu)₃SnH; (i) (i) CF₃CO₂H, (ii) BzCl, pyr; (j) EtS-TMS, ZnI₂; (k) (i) NaOMe; (ii) NaH; (iii) BnBr; (l) (i) (BnO)2P(O)OH, NIS; (m) H₂, Pd/C; (n) (i) HBr; (ii) (BnO)2P(O)OH, silver triflate, 2,4,6-collidine; (O) NaOH; (p) AcOH/HCl. In each case, cation exchange provided the Na+ salt.

The overall yield of this four-step phosphorylation strategy ranged from 19%-28% including the final ion exchange. FIG. 1(b) shows these glycosyl phosphates and methods for synthesizing them. These glycosyl phosphates, and methods for making them, are provided by the present invention.

The present method includes anomerically activating an ethyl 1-thio-β-D-pyranoside to form a compound having the formula 1

$$R^{7}$$
 R^{6}
 R^{9}
 R^{4}
 R^{4}
 R^{3}
 R^{2}
 R^{2}
 R^{2}

wherein R² is OCH₃, OBz, or OH, R³ is OH, OAc, or OBz, R⁴ is H, OH, or a halogen atom, and R⁵, R⁶, R⁷, R⁸, and R⁹ are each OBz,

and three or more of R³, R⁵, R⁶, R⁷, R⁸, and R⁹ are OBz substituents; deprotecting the OBz substituents to convert at least one such substituent to a OBn substituent; phosphorylating to form a compound of the formula 2,

$$R^{7}$$
 R^{6} R^{5} R^{3} R^{1} R^{1

wherein R¹ is OCH₃, OBz, OAc or OH, R² is OCH₃, OBz, OAc or OH,

9

R³ is OH, OAc, or OBz,

R⁴ is H, OH, OBz, OAc or a halogen atom,

R⁵, R⁶, R⁷, R⁸, and R⁹ are each OBz or OAc, and R¹⁰ is OH, or OBn.

wherein at least four of R³, R⁴, R⁵, R⁶, R⁷, R⁸, R⁹, and R¹⁰ are independently OBn or OBz substituents; and

deprotecting to convert any OBn substituents to OH substituents

Preferably, the α -D-hexopyranosyl phosphate is a glycosyl phosphate. Also included are α -D-hexopyranosyl phosphates, preferably glycosyl phosphates synthesized by these methods. Preferably these α -D-hexopyranosyl phosphates are selected from the group consisting of deoxy- α -D-glucopyranosyl, aminodeoxy- α -D-hexopyranosyl and acetamidodeoxy- α -D-hexopyranosyl phosphates.

The present invention also includes a method that includes providing isolated E_p having the formula 3

$$R^7$$
 R^6 R^4 R^4 R^5 R^3 R^1 R^1 R^{10}

wherein R¹ is OCH₃, OBz, OAc or OH, R² is OCH₃, OBz, OAc or OH, R³ is OH, OAc, or OBz, R⁴ is H, OH, OBz, OAc or a halogen atom, R⁵, R, R⁷, R⁸, and R⁹ are each OBz or OAc, and R¹⁰ is OH, or OBn,

wherein at least four of R³, R⁴, R⁵, R⁶, R⁷, R⁸, R⁹, and R¹⁰ are independently OH or OBz substituents.

Alternatively, phosphorylation of Dibenzyl-(2,3,4,6-tetra-O-benzoyl-α-D-altropyranosyl) phosphate (45), Dibenzyl-(2,3,4,6-tetra-O-benzoyl-α-D-idopyranosyl) phosphate (49) 40 Dibenzyl-(2,3,4,6-tetra-O-acetyl- α -D-talopyranosyl) phosphate (53) (FIG. $\mathbf{1}(b)$) via the glycosyl halide followed by complete deprotection gave the glycosyl phosphates Disodium α-D-altropyranosyl phosphate (47), Disodium α-Didopyranosyl phosphate (51) and Disodium α -D-talopyrano- 45 syl phosphate (55) as depicted in FIG. $\mathbf{1}(b)$ in an overall yield ranging from 37%-47%. The 6-deoxy precursor 1,2,3,4-tetra-O-benzoyl-6-deoxy-α,β-D-glucopyranose (8) may be synthesized by LiAlH₄ reduction and subsequent benzoylation of the halide Methyl 2,3,4-tri-O-acetyl-6-chloro-6-deoxy-α-D- 50 glucopyranoside (7). (See Anisuzzaman, A. K. M.; Whistler, R. L. Carbohydr. Res. 1978, 61, 511-518.). For the 4-deoxy progenitor, deoxygenation at C-4 may be accomplished by selective benzoylation of methyl β-D-galactopyranoside Methyl β -D-galactopyranoside (13) (as depicted in FIG. **1**(*b*)) 55 to provide the desired tribenzolated Methyl 2,3,6-tri-O-benzoyl-β-D-galactopyranoside (14) (54%) as well as the tetrabenzolated derivative (19%). Subsequent C-4 activation Methyl 2,3,6-tri-O-benzoyl-4-O-pentafluorophenoxythiocarbonyl- β -D-galactopyranoside (15) and $(n-Bu)_3SnH$ 60 reductive 4-deoxygenation were accomplished as described in Kanie, O.; Crawley, S. C.; Palcic, M. M.; Hindsgaul, O. Carbohydr. Res. 1993, 243, 139-164 to give the desired 4-deoxy precursor Methyl 2,3,6-tri-O-benzoyl-4-deoxy-β-D-galactopyranoside (16). The 3-deoxy predecessor 1,2,4,6-65 tetra-O-benzoyl-3-deoxy-α-D-glucofuranose (24) (FIG. 1(b)) was synthesized from 1,2:5,6-di-O-isopropylidene- α - 10

D-glucofuranose (21) by reduction of the previously reported furanose 1,2:5,6-Di-O-isopropylidene-3-O-(methylthio) thiocarbonyl- α -D-glucofuranose (22) (See Zhiyuan, Z.; Magnusson, G. Carbohydr. Res. 1994, 262, 79-101), while the 2-deoxy precursor (39) derived from a commercial source.

Thus, another embodiment of the present invention includes methods of making $\alpha\text{-D-hexopyranosyl}$ phosphates, which include, but are not limited to, phosphorylating a phosphate selected from the group consisting of Dibenzyl-(2,3,4,6-tetra-O-benzyol- $\alpha\text{-D-altropyranosyl})$ phosphate, Dibenzyl-(2,3,4,6-tetra-O-benzyl- $\alpha\text{-D-idopyranoxyl})$ phosphate, and Dibenzyl-(2,3,4,6-tetra-O-acetyl- $\alpha\text{-D-talopyranosyl})$ phosphate via a glycosyl halide; and deprotecting to form a glycosyl phosphate selected from the group consisting of Disodium $\alpha\text{-D-altropyranosyl}$ phosphate, Disodium $\alpha\text{-Didopyranosyl}$ phosphate. The present invention also includes $\alpha\text{-D-hexopyranosyl}$ phosphates prepared according to this method.

Nucleotide Sugars and Methods of Synthesizing the Same The present invention includes methods of making nucleotide sugars, which include combining $\alpha\text{-D-hexopyranosyl}$ phosphate and NTP in the presence of at least one mutated nucleotidylyltransferase. Other methods according to the present invention include combining $\alpha\text{-D-hexopyranosyl}$ phosphate and NTP other than TTP in the presence of at least one nucleotidylyltransferase, and combining NTP and $\alpha\text{-D-hexopyranosyl}$ phosphate other than Glc1P in the presence of at least one nucleotidylyltransferase.

The present invention includes a method of synthesizing nucleotide sugars that includes combining a nucleotidylyltransferase, α -D-glucopyranosyl phosphate, Mg⁺2, NTP and inorganic pyrophosphatase, and incubating. Preferably, the incubating is at a temperature of from about 30° C. to about 45° C., preferably about 33° C. to about 42° C., even more preferably about 37° C. for about 20 to about 40 minutes, preferably about 25 to about 35 minutes, and even more preferably about 30 minutes. The nucleotydylyltransferase according to these methods may include one or more natural and/or mutated nucleotidylyltransferases, such as natural and/or mutated E_p.

The present invention further includes nucleotide sugars made by the methods described herein.

Nucleotide sugars of the present invention may be selected from the group consisting of TDP-sugar, GDP-sugar, CDPsugar, UDP-sugar, and ADP-sugar and combinations thereof.

In embodiments utilizing or including mutated nucleotidy-lyltransferases, a preferred mutated nucleotidylyltransferase is $\rm E_p$ mutated at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177.

In embodiments utilizing or including mutated nucleotidylyl transferases, a preferred mutated nucleotidylyl transferases is \mathbf{E}_p mutated at one or more amino acids in its active site, its divalent cation binding site, and/or its auxiliary site.

Likewise, other preferred mutated nucleotidylyl transferases include nucleotidylyl transferases mutated at one or more amino acids in their active sites, divalent cation binding sites, and/or their auxiliary sites.

To evaluate the synthetic utility of purified thymidylyl-transferase, E_p , α -D-hexopyranosyl phosphate, Mg⁺2 and NTP were incubated at about 37° C. for about 30 min and the extent of product formation determined by HPLC. The results of these assays are illustrated in FIG. 2. Confirmation of product formation was based upon HPLC co-elution with commercially available standards and/or HPLC isolation and

high resolution mass spectroscopy of the product. (For select compounds, product peaks were lyopholized and submitted directly for HRMS (FAB) analysis.) As controls, little or no product formation was observed in the absence of E_n , glycosyl phosphate, Mg⁺2, or NTP. A reaction containing 5 mM⁻⁵ NTP, 10 mM sugar phosphate and 5.5 mM MgCl₂ in a total volume of 50 uL 50 mM potassium phosphate buffer, pH 7.5 at 37° C. was initiated by the addition of 3.52 U E_n (1 U=the amount of protein needed to produce 1 µmol TDP-D-glucose min⁻¹). The reaction was incubated with slow agitation for 30 min at 37° C., quenched with MeOH (50 μL), centrifuged (5 min, 14,000×g) and the supernatant was stored at -20° C. until analysis by HPLC. Samples (20 µL) were resolved on a Sphereclone 5u SAX column (250×4.6 mm) fitted with a guard column (30×4.6 mm) using a linear gradient (20-60 mM potassium phosphate buffer, pH 5.0, 1.5 mL min⁻¹, A₂75

The following nucleotide sugars are non-limiting examples of nucleotide sugars according to the present inven- $_{20}$ tion, which may preferably be produced in accordance with one or more of the methods described herein, and in particular the reactions of FIG. 2: (58) Thymidine 5'-(α-D-glucopyranosyl diphosphate) (HRMS (FAB) calc for C₁6H₂5O₁6N₂P₂ 563.0705. found m/z 563.0679 (M+H)). (59) Uridine 5'-(α - 25 D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁4H₂3O₁7N₂P₂ 565.0507. found m/z 565.0472 (M+H)). (60) Thymidine 5'-(2-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0714 (M+H)). (61) Uridine 5'-(2-deoxy- α -Dglucopyranosyl diphosphate) (HRMS (FAB): calc C₁4H₂3O₁6N₂P₂ 549.0506. found m/z 549.0510 (M+H)). (62) Thymidine 5'-(3-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB) calc for $C_16H_25O_15N_2P_2$ 547.0704. 35 found m/z 547.0720 (M+H)). (63) Uridine 5'-(3-deoxy- α -Dglucopyranosyl diphosphate) (HRMS (FAB): calc $C_14H_23O_16N_2P_2$ 549.0506. found m/z 549.0485 (M+H)). (64) Thymidine 5'-(4-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0693 (M+H)). (65) Uridine 5'-(4-deoxy-α-Dglucopyranosyl diphosphate) (HRMS (FAB): calc $C_14H_23O_16N_2P_2$ 549.0506. found m/z 549.0500 (M+H)). (66) Thymidine 5'-(6-deoxy-α-D-glucopyranosyl diphos- 45 phate) (HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0730 (M+H)). (67) Uridine 5'-(6-deoxy- α -Dglucopyranosyl diphosphate) (HRMS (FAB): calc $C_14H_23O_16N_2P_2$ 549.0506. found m/z 549.0492 (M+H)). (68) Thymidine 5'- $(\alpha$ -D-mannopyranosyl diphosphate) (HRMS (FAB) calc 563.0705. found m/z 563.0701 (M+H)). (69) Uridine 5'-(α-D-mannopyranosyl diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0503 (M+H)). (70) Thymidine 5'-(α-D-galactopyranosyl diphosphate) (HRMS 55 (FAB) calc 563.0705. found m/z 563.0710 (M+H)). (71) Uridine 5'-(α-D-galactopyranosyl diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0508 (M+H)). (72) Thymidine 5'-(α-D-allopyranosyl diphosphate) (HRMS (FAB) calc 563.0705. found m/z 563.0715 (M+H)). (73) Uridine 5'-(α -D-allopyranosyl diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0507 (M+H)). (74) Thymidine 5'-(α-D-altropyranosyl diphosphate) (HRMS (FAB) calc 563.0705. found m/z 563.0699 (M+H)). (75) Uridine 5'-(α -D-altropyranosyl ₆₅ diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0511 (M+H)). (76) Thymidine 5'- $(\alpha$ -D-gulopyranosyl

12

diphosphate) (HRMS (FAB) calc 563.0705. found m/z 563.00712 (M+H)). (77) Uridine 5'-(α-D-gulopyranosyl diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0512 (M+H)). (78) Thymidine 5'-(α-D-idopyranosyl diphosphate) (HRMS (FAB) calc 563.0705. found m/z 563.0708 (M+H)). (79) Uridine 5'-(α-D-idopyranosyl diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0507 (M+H)). (80) Thymidine 5'-(α-D-talopyranosyl diphosphate) (HRMS (FAB) calc 563.0705. found m/z 563.0710 (M+H)). and (81) Uridine 5'-(α-D-talopyranosyl diphosphate) (HRMS (FAB): calc 565.0507. found m/z 565.0499 (M+H)). although data is not depicted for all products

Other nucleotide sugars in accordance with the present invention include, but are not limited to, the following: (109) Thymidine 5'-(6-amino-6-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0837 (M+H)). (110) Uridine 5'-(6amino-6-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0640 (M+H)). (111) Thymidine 5'-(4-amino-4-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0848 (M+H)). (112) Uridine 5'-(4-amino-4-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0638 (M+H)). (113) Thymidine 5'-(3-amino-3-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0835 (M+H)). (114) Uridine 5'-(3-amino-3-deoxyα-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $C_15H_24O_16N_2P_2$ 564.0632. found m/z 564.0622 (M+H)). (115) Thymidine 5'-(2-amino-2-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0842 (M+H)). (116) Uridine 5'-(2amino-2-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0630 (M+H)). (117) Thymidine 5'-(6-acetamido-6deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁8H₂8O₁6N₃P₂ 604.0945. found m/z 604.0953 (M+H)). (118) Uridine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁7H₂6O₁₇N₃P₂ 606.0737. found m/z 606.0732 (M+H)). (119) Thymidine 5'-(4-acetamido-4-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁8H₂8O₁6N₂P₂ 604.0945. found m/z 604.0940 (M+H)). (120) Uridine 5'-(4acetamido-4-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 606.0737. found m/z 606.0730 (M+H)). (121) Thymidine 5'-(3-acetamido-3deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁8H₂8O₁6N₃P₂ 604.0945. found m/z 604.0947 (M+H)). (122) Uridine 5'-(3-acetamido-3-deoxy- α -D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 606.0737. found m/z 606.0735 (M+H)). (123) Thymidine 5'-(2-acetamido-2-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁8H₂8O₁6N₃P₂ 604.0945. found m/z 604.0951 (M+H)). (124) Uridine 5'-(2acetamido-2-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 606.0737. found m/z 606.0738 (M+H)). (125) Thymidine 5'-(4-amino-4,6dideoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁6H₂6O₁4N₃P₂ 546.0889. found m/z 546.0895 (M+H)). and (126) Uridine 5'-(4-amino-4,6-dideoxy- α -Dglucopyranosyl diphosphate) (HRMS (FAB): calc for $C_15H_24O_{15}N_3P_2$ 548.0682. found m/z 548.0673 (M+H)).

Further nucleotide sugars in accordance with the present invention include, but are not limited to, the following:

FIG. 2 illustrates the utility of E_p as a catalyst/reagent to simplify the synthesis of useful nuleotide sugars—of the twelve glycosyl phosphate tested (which include all possible 15 α -D-hexoses and monodeoxy α -D-glucoses), all produce product with both TTP and UTP under the conditions described. These yield might be further improved by using pryophosphatase to drive the equilibrium of the reaction. An examination of accepted α -D-hexopyranosyl phosphates to 20 with TTP suggests that E_p prefers pyranosyl phosphates, which are predicted to exist predominately as ⁴C₁ conformers [e.g., (12), (20), (28), (43), α-D-glucopyranosyl phosphate (2), α -D-mannopyranosyl phosphate (56), and α -D-galactopyranosyl phosphate (57) (FIGS. 1 and 2)], while those predicted to not adopt the ⁴C₁ conformation [e.g., Ethyl 2,3,4,6tetra-O-benzyl-1-thio-β-D-gulopyranoside (31),allopyranosyl phosphate (38), α -D-altropyranosyl phosphate (47), α -D-idopyranosyl phosphate (51) and α -D-talopyranosyl phosphate (55) (FIGS. 1 and 2)] show less activity.

Regarding specific interactions required for conversion, analysis of the corresponding deoxy series [(12), (20), (28) and (43) (FIGS. 1 and 2)] implicates only a single required hydroxyl (C-2), the removal of which impairs the yield by 35 >70%. A similar trend is observed in the UTP series with two exceptions, glycosides (28) and α -D-mannopyranosyl phosphate (56) (FIGS. 1 and 2). Cumulatively, these results suggest that, while the C-2 hydroxyl is important for turnover, alterations at C-3 in the context of UTP result in adverse 40 cooperativity.

Aminodeoxy- α -D-hexapyranosyl phosphates and acetamidodeoxy- α -D-hexapyranosyl phosphates are each examples of α -D-hexapyranosyl phosphates that may be used in accordance with the present invention. A direct comparison 45 of the aminodeoxy- α -D-glucopyranosyl phosphate series to their corresponding acetamidodeoxy analogs provides insight pertaining to the ability of the E_p active site to accommodate additional steric bulk.

Of the aminodeoxy-α-D-glucopyranosyl phosphates 50 examined, only two, 2-amino-2-deoxy-α-D-glucopyranosyl phosphate (107) (FIG. 4) and 2-acetamido-2-deoxy-α-D-glucopyranosyl phosphate (108) (FIG. 4), were commercially available. The syntheses of the remaining analogs diverged from the key intermediates Ethyl 6-azide-2,3',4-tri-O-benzyl-55 6-deoxy-1-thio-β-D-glucopyranoside (89), Ethyl 4-azide-2, 3,6-tri-O-benzyl-4-deoxy-1-thio-β-D-glucopyranoside (94) and Ethyl 3-azide-2,4,6-tri-O-benzyl-3-deoxy-1-thio-β-D-glucopyranoside (100) (FIG. 3(b)).

Thus, the present invention includes a method of making 60 aminodeoxy- α -D-glucopyranosyl phosphates comprising converting an intermediate selected from the group consisting of ethyl 6-azide-2,3,4-tri-O-benzyl-6-deoxy-1-thio- β -D-glucopyranoside (89), ethyl 4-azide-2,3,6 tri-O-benzyl-4-deoxy-1-thio- β -D-glucopyranoside (94), and ethyl 3-azide-2,4,6-65 tri-O-benzyl-3-deoxy-1-thio- β -D-glucopyranoside (100) to a corresponding amide.

Ethyl 1-thio-β-D-pyranosides (89) and (100) derived from previously reported glycosides (FIG. 3(b)(87)) (see V. Maunier, P. Boullanger, D. Lafont, Y. Chevalier, Carbohydr. Res. 1997, 299, 49-57) and FIG. **3**(*b*)(98) (W. A. Greenberg, E. S. Priestley, P. S. Sears, P. B. Alper, C. Rosenbohm, M. Hendrix, S.-C. Hung, C.-H. Wong, J. Am. Chem. Soc. 1999, 121, 6527-6541), respectively), while (94) was synthesized, from the previously reported compound (93) (FIG. 3(b)) (P. J. Garegg, I. Kvarnstrom, A. Niklasson, G. Niklasson, S. C. T. Svensson, J. Carbohydr. Chem. 1993, 12, 933-953) in a manner similar to that of the deoxy-α-D-glucopyranosyl phosphate syntheses described herein. Specifically, this strategy invoked a protection scheme to selectively expose the position of substitution followed by activation (via TsCl or Tf₂O) and SN² displacement by sodium azide. From the divergent point (FIG. 3 (89), (94) and (100)), an efficient azide selective SnCl₂ reduction followed by acetylation gave the desired ethyl 1-thio-β-D-pyranoside precursors (90), (95), and (101). Finally, the subsequent phosphorylation of FIG. 3(b) (89), (90), (94), (95), (100), and (101) was accomplished by reaction with dibenzyl phosphate as previously described where the culminating reductive deprotection also led to the conversion of the FIG. 3(b) (89), (94), and (100) azides to the desired amines. As an aminodideoxy sugar representative, 4-amino-4,6-dideoxy- α -D-glucopyranosyl phosphate (FIG. 3(b) (102)) was also synthesized from peracetylated D-fucose (FIG. 3(b) (103)) as illustrated in FIG. 3 using a similar strategy.

To evaluate the synthetic utility of thymidylyl-transferase, purified E_n , α -D-glucopyranosyl phosphate, Mg⁺2, NTP and inorganic pyrophosphatase were incubated at 37° C. for 30 min and the extent of product formation determined by HPLC. The inorganic pyrophosphatase was included to drive the reaction forward. A reaction containing 2.5 mM NTP, 5.0 mM sugar phosphate, 5.5 mM MgCl₂ and 10 U inorganic pyrophosphatase in a total volume of 50 µL 50 mM potassium phosphate buffer, pH 7.5 at 37° C. was initiated by the addition of 3.52 U E_n (1 U=the amount of protein needed to produce 1 mol TDP-D-glucose min⁻¹). The reaction was incubated with slow agitation for 30 min at 37° C., quenched with MeOH (50 μ L), centrifuged (5 min, 14,000×g) and the supernatant was stored at -20° C. until analysis by HPLC. Samples (30 μL) were resolved on a Sphereclone 5 u SAX column (150×4.6 mm) fitted with a SecurityGuard™ cartridge (Phenomenex; Torrance, Calif.) using a linear gradient (50-200 mM potassium phosphate buffer, pH 5.0, 1.5 mL min^{-1} , A₂75 nm).

The results of these assays are illustrated in FIG. **4**. For each assay, confirmation of product formation was based upon high resolution mass spectroscopy of HPLC-isolated products and, also in some cases, HPLC co-elution with commercially available standards. (Allosteric activation is common for the nucleotidylyltransferase family (for examples see: M. X. Wu, J. Preiss, Arch. Biochem. Biophys. 1998, 358, 182-188; and D. A. Bulik, P. van Ophem, J. M. Manning, Z. Shen, D. S. Newburg, E. L. Jarroll, J. Biol. Chem. 2000, 275, 14722-14728) although data is not yet available pertaining to the allosteric effectors of E_p .) As controls, no product formation was observed in the absence of E_p , glucopyranosyl phosphate, Mg^+2 , or NTP.

The nucleotide sugars (109)-(126) set forth above are examples of nucleotide sugars of the present invention, which may be produced in accordance with the methods described herein, and in particular the reactions diagramed in FIG. 4. A comparison of the aminodeoxy- α -D-glucopyranosyl phosphate/dTTP assay results (FIG. 4 (85), (91), (96), and (107)) to the E_p native reaction (FIG. 4, (2)/dTTP) reveals that amino

substitution has little or no effect on product formation, and, with the exception of compound (85) (FIG. 4), a similar phenomenon is observed in presence of UTP.

The divergence of compound (85) from this trend is consistent with UTP-dependent E_p "adverse cooperativity" in the presence of certain hexopyranosyl phosphates, as described herein. This phenomenon is perhaps attributable to allosteric activation by dTTP. Evaluation of the acetamidodeoxy- α -D-glucopyranosyl phosphate/dTTP assays (FIG. 4 compounds (86), (92), (97) and (108)), in comparison to their non-acetylated counterparts (FIG. 4 (85), (91), (96) and (107)), reveal that a bulky N-acetyl group at C-2 or C-3 (FIG. 4 (97) and (108)) is well-tolerated while the identical C-4 or C-6 substitution (FIG. 4 (92)) and (86)) results in less activity. Given that these effects most likely derive from unfavored steric interactions, it follows that the E_p active site is able to accommodate additional C-2/C-3 bulk while sterics limit the allowed C-4/C-6 substitutions.

Surprisingly, product formation from FIG. 4 (86)/UTP was 20 markedly increased (8-fold) in comparison to (86)/dTTP. This is the first example to contradict the typical adverse UTP-dependent effect upon yields observed, as illustrated by FIG. 4 compounds (85) and (97). Finally, a comparison of aminodideoxy-α-D-glucopyranosyl phosphate (FIG. 4 (102) 25 (The product of this reaction, thymidine 5'-(4-amino-4,6dideoxy-α-D-glucopyranosyl diphosphate), is an important critical intermediate in the formation of the calicheamicin aryltetrasaccharide to that of (FIG. 4 (96)) reveals C-6 deoxygenation does not effect dTTP-dependent E_p catalysis but 30 greatly diminishes UTP-dependent conversion. However, given independent deoxygenation at C-6 or amino substitution at C-4 (FIG. 4 (91)) each has no effect on product yield, data from independent substitutions may not be reliable in predicting the effects of multiple substitutions on product 35

FIG. 4 illustrates the utility of E_p as a catalyst/reagent to simplify the synthesis of useful nucleotide sugar pools—of the nine substrate analogs tested, all provide product with dTTP and with dUTP under the conditions described. Further, seven with dTTP and four with UTP provide appreciable product (>50% conversion) under the conditions described.

Nucleotide sugars produced via the exploitation of the promiscuity of E_p include, but are not limited to, compounds (58)-(81), (109)-(126) and those set forth in FIGS. **20**(*a*) and 45 **20**(*b*).

Nucleotidylyltransferases

Structure-Based Engineering of E_n

The present inventors have determined the first three dimensional structures of this unique enzyme in complex 50 with the product UDP- α -D-glucose (UDP-Glc) and with the substrate dTTP at 1.9 Å and 2.0 Å resolution, respectively. A three dimensional structure of E_p is depicted in FIG. 6. This discovery has facilitated the elucidation of the molecular details of E_p substrate recognition. The structures reveal the catalytic mechanism of thymidylyltransferases, which is further supported by new kinetic data. The present inventors have also used structure-based engineering or mutations of E_p to produce modified enzymes. These inventive enzymes are capable of utilizing "unnatural" sugars previously not 60 accepted by wild-type E_p .

Structure Determination

The E_p -UDP-Glc structure was determined using selenomethionine-labeled protein crystals and a data-set collected at a wavelength corresponding to the selenium absorption peak. 65 A representative portion of the experimental electron density is shown in FIG. 6. The E_p -dTTP structure was subsequently

20

determined by molecular replacement using the ${\rm E}_p$ -UDP-Glc monomer structure as a search model.

Overview of the E_p Structure

The structure of the biologically active E_n tetramer is illustrated in FIG. 7. The present model is refined at 2.0 Å resolution to an R factor of 18.3% with restrained temperature factors and good stereochemistry. FIG. 7a shows \mathbf{E}_p in complex with UDP-Glc and FIG. 7b displays the E_p -dTTP complex. The two tetrameric structures are very similar with r.m.s.d. for equivalent C_{α} positions=1.0 Å. The enzyme has overall dimensions of about 80 Å×80 Å×60 Å and a compact tertiary structure generated by four monomers packing tightly against each other along two two-fold axes of symmetry drawn on the leftmost panel of FIG. 7. The overall surface area buried during tetramer formation is approximately 10,000 Å, equivalent to the surface of one monomer. The monomer interactions are dominated by helix-helix packing of the four large helices in the center of the E_p tetramer and surrounding extensive loop-loop interactions involving multiple van der Waals contacts, hydrogen bonds, and salt bridges. The active site pockets of the monomers are located close to, but not overlapping with, the monomer interface.

The E_p monomer (FIG. 8) is a two-domain molecule with overall size of approximately 50 Å×50 Å×50 Å. The domain containing the active site is dominated by a large seven-stranded mixed central β -sheet, with an unusual left-handed twist, packed against three α -helices on one side and another three α -helices on the other. Its extensive hydrophobic core contains no cavities and is dominated by aromatic side chains.

This domain has overall resemblance, including the location of the active site in a large pocket on the top of the β -sheet, to other nucleotide binding proteins (see Vrielink, A., Ruger, W., Dreissen, H. P. C. & Freemont, P. S. Crystal Structure of the DNA Modifying Enzyme β-Glucosyltransferase in the Presence and Absence of the Substrate Uridine Diphosphoglucose, EMBO J. 13, 3413-3422 (1994); Charnock, S. J. & Davies, G. J. Structure of the Nucleotide-Diphospho-Sugar Transferase, SpsA from Bacillus subtilis, in Native and Nucleotide-Complexed Forms. Biochem. 38, 6380-6385 (1999); Gastinel, L. N. Cambillau, C. & Bourne, Y. Crystal Structures of the Bovine 4Galatosyltransferase Catalytic Domain and Its Complex with Uridine Diphosphogalactose, EMBO J. 18, 3546-3557 (1999); Ha, S., Walker, D., Shi, Y. & Walker, S. The 1.9 Å Crystal Structure of Escherichia coli MurG, a Membrane-Associated Glycosyltransferase Involved in Peptidoglycan Biosynthesis. Prot. Sci. 9, 1045-1052 (2000); and Brown, K., Pompeo, F., Dixon, S., Mengin-Lecreulx, D., Cambillau, C. & Bourne, Y. Crystal Structure of the Bifunctional N-Acetylglucosamine 1-phosphate uridylyltransferase from Escherichia coli: A Paradigm for the Related Pyrophosphorylase Superfamily. EMBO J. 18, 4096-4107 (1999)), containing the α/β open-sheet Rossmann fold. (Rossmann, M. G., et al., Evolutionary and structural relationship among dehydrogenases, in The Enzymes, I. P. D. Boyyer, Editor. Academic Press: New York. p. 61-102 (1975); and Branden, C. & Tooze, J. Introduction to Protein Structure. New York: Garlan Publishing, Inc. (1991).) The second E_n domain (represented by yellow in (FIG. 8), packing tightly to the side of the active-site domain, contains four α -helices and a two-stranded β -sheet and is involved in the inter-monomer packing interactions forming the \mathbf{E}_{p} tetramer.

Structural Homology to Glycosyltransferases and Uridylyltransferases

The present inventors' elucidation of the structure of E_p represents the first such elucidation of a structure of a thymidylyltransferase. Comparison of the structure with the contents of the FSSP database, (Holm, L. & Sander, C. Tour-

ing Protein Fold Space with Dali/FSSP. Nucleic Acids Res., 26, 316-319 (1998)) revealed that the overall E_p fold is different from other previously determined structures. The closest structural homologs of E_p are the SpsA glycosyltransferase from *Bacillus subtilis* and the functionally related *E*. coli enzyme GlmU. GlmU is a bifunctional enzyme containing acetyltransferase and uridylyltransferase domains, respectively. FIG. 8 illustrates these three proteins, highlighting the structurally similar regions. As expected, the structural homology lies within the nucleotide-sugar binding domains. The active sites of the enzymes are located in pockets on top of the large β -sheet, although the precise positioning differs between glycosyltransferases and nucleotidyltransferases and involves secondary structure elements, which are not structurally equivalent. The three-dimensional structures of two other sugar-phosphate transferring enzymes, α-D-galactopyranosyl phosphate (Gal-1-P) uridylyltransferase and kanamycin nucleotidyltransferase are

known, but do not activate sugars and both differ structurally 20

and functionally from E_p .

Active Site Interactions: Substrate and Product Binding FIG. 8 shows two 90° views of the E_p active site pocket. In both of the E_p -dTTP and E_p -UDP-Glc structures, the experimental electron density for the dTTP and UDP-Glc is excel- 25 lent. E_n utilizes both dTTP and UTP, but not CTP, and FIG. 9a illustrates the structural basis for this substrate specificity. Specifically, the exocyclic N3 and O4 ring atoms of both dT and U are hydrogen bonded to Gln83. In addition, O4 hydrogen bonds to the main chain N of Gly88 while O2 is bound to 30 the main chain N of Gly11. Finally, the 3'-hydroxyl group of the pentose forms a hydrogen bond with Gln27. The substrate dTTP also makes extensive van der Waals contacts with Leu9, Leu89 and Leu109, which form a hydrophobic bed for the nucleoside, and position 5 of the pyrimidine base is far 35 enough from any protein atom to allow an easy fit for the methyl group of dT in the pocket. The phosphate groups of dTTP lie in an extended position firmly held in place by multiple interactions with the main chain nitrogen atoms of Ser13, Gly14, and Thr15, and with the catalytically important 40 Mg⁺2 (see below). The γ-phosphate also makes a hydrogen bond with Thr15 and both the α - and γ -phosphates bind Arg16. The nearby Arg145, Lys163, and Arg195 create a favorable electrostatic environment, but do not interact directly with dTTP.

The E_n product, UDP-Glc, is bound along the diameter of the surface pocket. The nucleoside sits in the active site in virtually the same conformation as the substrate dTTP, with the addition of a hydrogen bond between the 2'-hydroxyl of the ribose and the main chain O of Gly11. In the glucose- 50 binding pocket, as illustrated on (NAT) FIG. 5b, the hydroxyl groups O2, O3 and O4 of the glucose moiety are directly hydrogen-bonded to protein residues, while O6 is bound to E_n via a water molecule. Gln162 binds both O2 and O3, the main chain N of Gly147 binds both O3 and O4, and the main chain 55 O of Val173 binds O4. The side chain of ThR201 is also close to both O2 and O3. In addition, four well-ordered water molecules, shown on FIG. 5b, bridge E_n and the glucose moiety. Leu109, Leu89, and Ile200 make van der Waals contacts with the underside of the hexose ring and Trp224 and 60 Tyr146 close the glucose binding pocket which would prevent bulkier sugars, for example disaccharides, from binding. In the E_{ν} -UDP-Glc structure, the phosphate groups are now twisted away from their straight conformation in dTTP so that they can connect the nucleoside with the hexose—see also 65 FIG. 10. The phosphates are also much more solvent exposed and do not interact with any main chain atoms, but instead,

22

with the side chains of the positively charged Arg16, Lys163, and Arg195, as well as with water molecules.

Divalent Cation Binding Site

The activity of nucleotidyltransferases is strictly dependent on a divalent cation involved in catalysis via stabilizing the leaving PP_i (See Kornfeld, S. & Glaser, L. J. Biol. Chem. 236, 1791-1794 (1961)). Crystallographic data generated by the present inventors allow for the identification of the location of this cofactor and, in this region, a Mg⁺2 electron density feature, larger than a water molecule and chemically ideal for a metal location, was modeled. Indeed the Mg⁺2 is 2.6 Å away from the β-phosphate oxygen and is also coordinated by the side chain of Gln26, main chain O of Gly11, main chain nitrogens of Ser13 and Gly14, and a water molecule. This region (particularly Gly10 to Gly15) is mostly disordered in the E_p -UDP-Glc structure, indicating that the Mg⁺2, in addition to electrostatically stabilizing the leaving group, also plays a structural role in folding the substratebinding region of E_{ν} around itself to fix the NTP at an optimal position for the catalytic event.

A Secondary dTTP-Binding Site and Possible Allosteric

The structure of E_p -dTTP, disclosed herein (FIG. 7), indicates that the E_p tetramer binds eight molecules of dTTP four in the active site pockets on top of the β -sheet, and four in an auxiliary sites at the interface between the subunits. FIG. 9c shows a close-up of a dTTP molecule in the auxiliary site. There are fewer contacts between E_p and dTTP here than in the active site. As a result, CTP, which is not accepted by E_n , could easily fit in the auxiliary site. The dTTP base and the ribose in the secondary site interact with one E_n monomer, including hydrogen bonds to the main chain N of Gly116 and Ser152, and van der Waals contacts with Leu46, Tyr115 and Ile249. The dTTP phosphates, on the other hand, interact primarily with residues of an adjacent E_p monomer, including Arg220 and Gly221.

Several other nucleotidylyltransferases are under allosteric control by metabolites distinct from their products or substrates. The presence of an auxiliary site strongly suggests that E_n is also under allosteric control. Indeed, binding of an effector in this hydrophobic pocket at the monomer interface could alter the relative orientation of the E_p monomers, thus altering the conformation or the access to the active site. Given the non-specific nature of the observed interactions, and the fact that nucleotidylyl-transferase effectors are generally not substrates, the putative E_n allosteric effector is most likely not dTTP.

The E_p Catalytic Mechanism

Before the present experiments, two conflicting hypotheses for nucleotidylyltransferase catalysis were suggested. Lindquist and co-workers proposed a ping-pong bi-bi mechanism for E_p , the necessary prerequisite for which is the formation of an enzyme-substrate covalent intermediate. (See Lindquist, L., Kaiser, R., Reeves, P. R. and Lindberg, A. A., Purification, Characterization and HPLC Assay of Salmonella Glucose-1-Phosphate Thymidylylphospherase from the cloned rfbA Gene, Eur. J. Biochem, 211, 763-770 (1993). Alternatively, in a related enzyme, Frey and co-workers had previously presented evidence for inverted geometry about the α-phosphate upon attack by Glc-1-P which led the authors to propose a single displacement mechanism for the entire nucleotidylytransferase family. (Sheu, K.-F. R., Richard, J. P. & Frey, P. A. Stereochemical Courses of Nucleotidyl-transferase and Phosphotransferase Action. Uridine Diphosphate Glucose Pyrophosphorylase, Galactose-1-Phosphate Uridylyltransferase, Adenylate Kinase, and Nucleoside Diphosphate Kinase Biochem. 18, 5548-5556 (1979)).

In the present invention, a comparison of the topology of the ${\rm E}_p$ -bound substrate (dTTP) to the ${\rm E}_p$ -bound product (UDP-Glc) (FIG. 10a) suggests that the Glc-1-P oxygen nucleophile must directly attack the α -phosphate of dTTP. In this reaction, the formation of a phosphodiester bond on one side of the α -phosphate atom is simultaneous with the breaking of a phosphodiester bond on the opposing face (to give PP_i as the leaving group). Consistent with an ${\rm S}_{N2}$ type mechanism, the bond undergoing formation in the structure disclosed herein is "in-line" or 180 degrees away from the leaving group and thus, the two oxygen atoms bonded to the phosphate invert their geometry upon bond formation. Although the α -phosphate here is not chiral, both the reactant (substrate) and product topologies, as well as the architecture of the active site, clearly suggest that an inversion has occurred

The present inventors evaluated the E_p steady state kinetics in order to further probe the enzymatic mechanism. The intersecting patterns observed in FIG. 10b and FIG. 10d are consistent with the structural data in support of a single displacement mechanism rather than the previously postulated pingpong bi bi (double displacement) mechanism. Finally, the E_p -dTTP crystals were soaked in a solution containing 2 mM of either Glc-1-P or D-Glc, in addition to the 2 mM dTTP and 25 Mg⁺2 already present. The glucose soaks did not significantly alter the electron density in the active site. On the other hand, Glc-1-P soaks quickly caused deterioration of the crystal diffraction quality. Data collected with crystals soaked for 30 min revealed electron density in the active site that was an 30 average of the density in our EP-dTTP and E_p-UDP-Glc crystals. Therefore, the phosphate of Glc-1-P is necessary for binding by E_p , and the lack of any observable E_p -UMP covalent intermediate in these experiments further supports the single displacement mechanism.

Active-Site Engineering

Two sugar phosphates not utilized by wild-type E_n and two additional sugar phosphates poorly utilized by the enzyme were selected to test rational engineering of E_p substrate promiscuity. Specifically, 6-acetamido-6-deoxy-α-D-glucopy- 40 ranosyl phosphate (FIG. 11 (86)) is not well-accepted and α-D-glucopyranuronic acid 1-(dihydrogen phosphate) (FIG. 11 (127)) is not accepted by E_p , and 2-acetamido-2-deoxy- α -D-glucopyranosyl phosphate (FIG. 11 (108)) and α -Dallopyranosyl phosphate (FIG. 11 (38)) lead to poor conver- 45 sion. Because a representative "unnatural" sugar phosphate was believed to be efficiently converted only by wild-type E_{n} , 4-amino-4,6-dideoxy-α-D-glucopyranosyl phosphate (FIĜ. 11 (102)) was also tested with all mutants. Sugar phosphates not utilized by a wild-type enzyme, e.g., E_p , and sugar phosphates poorly utilized by the enzyme may be referred to herein as "unnatural" with respect to that enzyme.

Structure-based modeling reveals steric and/or electrostatic infringements may be the limiting factor in the conversion of "unnatural" sugar phosphates. In an attempt to relieve 55 these constraints, three mutants were constructed. In particular, a Thr201Ala mutant and Glu162Asp were believed to decrease the steric interference at the sugar positions C-2 and/or C-3 for compounds (108) and (38), while a Trp224His substitution was designed to decrease steric constraints at C-6 of the substrate (e.g. compound (86)). Furthermore, the glucuronic acid derivative (127) offers the unique challenge of engineering electrostatic balance and the Trp224His variant was predicted to provide a partial positive charge to assist in (127)-binding in addition to steric relief. Alternatively, Asp-111 (6 Å from the substrate C-6-OH) was predicted to result in the electrostatic repulsion of substrates containing a nega-

tive charge at the C-6 of the sugar phosphate. Thus, an additional mutant (Asp111Asn) was constructed to eliminate this effect.

As a rapid means to assay the entire pool of the four newly constructed mutants, the mutants were combined and the mixture directly tested for ability to convert compounds (2), (108), (89), (102), (38), and (127). FIG. 11 shows that the mutant pool was able to turn over all but one (5) of the sugar phosphates tested. Those substrates turned over include (86) and (127), the two sugar phosphates not accepted or poorly accepted by wild-type E_p .

The following nucleotide sugars were produced by the reactions of FIG. 11: (117) Thymidine 5'-(6-acetamido-6deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $C_18H_28O_16N_3P_2$ 604.0945. found m/z 604.0953 (M+H)). (118) Uridine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for $C_17H_26O_17N_3P_2$ 606.0737. found m/z 606.0732 (M+H)). (130) Thymidine 5'-(α-D-glucopyran-6-uronic acid diphosphate) (HRMS (FAB): calc for C₁6H₂3O₁7N₂P₂ 577.0472. found m/z 577.0465 (M+H)). (131) Uridine 5'-(α-D-glucopyran-6-uronic acid diphosphate) (HRMS (FAB): calc for C₁5H₂1O₁8N₂P₂ 579.2774. found m/z 579.2775 (M+H)). (123) Thymidine 5'-(2-acetamido-2-deoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁8H₂8O₁6N₃P₂ 604.0945. found m/z 604.0951 (M+H)). (124) Uridine 5'-(2acetamido-2-deoxy- α -D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 606.0737. found m/z 606.0738 (M+H)). (125) Thymidine 5'-(4-amino-4,6dideoxy-α-D-glucopyranosyl diphosphate) (HRMS (FAB): calc for C₁6H₂6O₁4N₃P₂ 546.0889. found m/z 546.0895 (M+H)). (126) Uridine 5'-(4-amino-4,6-dideoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $C_15H_24O_15N_3P_2$ 548.0682. found m/z 548.0673 (M+H)) 35 although data is not depicted for all products.

A deconvolution of the mutant pool, by individual mutant analysis, revealed the Trp224His mutation as responsible for converting both (86) and (127). Thr102Ala, on the other hand, was responsible for the 2-fold increase in the conversion of (108). The remaining two mutants (Asp41Asn and Glu162Asp, not shown in FIG. 11) failed to enhance conversion, over wild-type E_p , of any of the tested putative substrates. Yet, cumulatively, this small set of directed mutants was able to successfully turn over three of four targeted "unnatural" substrates. Of particular interest is the Trp224His mutant, which displays enhanced promiscuity without affecting wild-type traits. This E_p variant will serve as an excellent foundation for second generation double mutants. Finally, the demonstrated ability to test mutant sets via pooling will rapidly expedite the development of this methodology.

In E_p , amino acids that make contacts or near contacts to the sugar in the active site include V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177. These amino acids may be mutated in order to alter the specificity of E_p , as demonstrated herein. Any mutation that alters the specificity may be made and tested, as taught herein, to determine its effect on the specificity of E_p for its substrate and the efficiency of conversion of substrate to product.

Thus, the present invention includes a nucleotidylyl-transferase mutated at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177. Preferably the nucleotidylyltransferase is E_p .

An embodiment of the present invention is directed to a nucleotidylyltransferase mutated such that it is capable of having a different substrate specificity than a non-mutated

nucleotidylyltransferase. Examples include nucleotidylyltransferases having a substrate specificity for GTP, ATP, CTP, TTP or UTP. Further provided are methods of altering nucleotidylyltransferase substrate specificity comprising mutating the nucleotidylyltransferase at one or more amino acids selected from the group consisting of V173, G147, W224, N112, G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177. Preferably the nucleotidylyltransferase is E_p. Also provided are nucleotidylyl-transferases, so modified.

The present invention also includes purine or pyrimidine triphosphate type nucleotidylyltransferases set forth in FIG. 19, and purine or pyrimidine triphosphate type nucleotidylyltransferases mutated at one or more amino acids selected from the group consisting of V173, G147, W224, N112, 15 G175, D111, E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177.

Further, sequence comparison reveals that many nucleotidyltransferases bear high degrees of sequence identity to E_n The substrate specificity of such enzymes may be altered, 20 using methods described herein for E_n , at amino acids that make contacts or near contacts to the sugar in the active site. These amino acids may be located via sequence comparison with E_p —the contact sites will often be those at the same relative position as V173, G147, W224, N112, G175, D111, 25 E162, T201, I200, E199, R195, L89, L89T, L109, Y146 and Y177 in E_p . FIG. 19 provides a list of nucleotidyltransferases that bear high degrees of sequence identity to E_p . FIGS. 12 to 18 show the alignment of the E_p sequence and those of other representative nucleotidyltransferases. Other nucleotidylyl- 30 transferases may also be mutated at one or more amino acids in their active sites, divalent cation binding sites and/or auxiliary sites.

Methods for mutating proteins are well-known in the art. For the present invention, it is preferable to perform site- 35 directed mutagenesis on the nucleotide encoding the enzyme of interest. In this manner, and using the guidance provided herein, one of skill in the art can make mutations to the codons encoding the amino acids at the sites of the enzyme desired to be changed. Likewise, the use of site directed mutagenesis 40 allows the worker to ensure that each codon desired to be changed is changed to encode a different amino acid from the wild-type molecule. In contrast, the use of random mutagenesis might result in mutated codons encoding the same amino acids as the wild-type codons, due to the degeneracy of the 45 genetic code. Methods for manipulation and mutation of nucleotides, as well as for the expression of recombinant peptides are well known in the art, as exemplified by Sambrook et al., Molecular Cloning: A Laboratory Manual, Second Edition, Cold Spring Harbor Laboratory Press, 1989

References for nucleotidyltransferases in these Figures, include: AAB31755-Glc-1-P Cytitdylyltransferase from Yersinia pseudotuberculosis. See Thorson J S, Lo S F, Ploux O, He X & Liu H W J. Bacteriol. __176:__5483-5493 (1994) [94350832]; AAC39498-Man-1-P Guanylyltransferase from 55 Hypocrea jecorina. Kruszewska, J. S., Saloheimo, M., Penttila, M. & Palamarczyk, G. Direct Submission; B72403-Glc-1-P adenylyltransferase from *Thermotoga maritima* (strain MSB8). Nelson K E, Clayton R A, Gill S R, Gwinn M L, Dodson R J, Haft D H, Hickey E K, Peterson J D, Nelson W 60 C, Ketchum K A, McDonald L, Utterback T R, Malek J A, Linher K D, Garrett M M, Stewart A M, Cotton M D, Pratt M S, Phillips C A, Richardson D, Heidelberg J, Sutton G G, Fleischmann R D, Eisen J A, Fraser C M & et al Nature _399:_323-329 (1999) [99287316]; BAA34807-Man-1-P Guanylyltransferase from Candida albicans. Ohta, A. & Sudoh, M. Direct Submission; CAA06172-Glc-1-P

26

Uridiylyltransferase from Streptococcus pneumoniae. Mollerach M, Lopez R & Garcia E J. Exp. Med. __188:__2047-2056 (1998) [99059828]; D83000-Glc-1-P thymidylyltransferase from Pseudomonas aeruginosa (strain PAO1); Stover C K, Pham X Q, Erwin A L, Mizoguchi S D, Warrener P, Hickey M J, Brinkman F S, Hufnagle W O, Kowalik D J, Lagrou M, Garber R L, Goltry L, Tolentino E, Westbrock-Wadman S, Yuan Y, Brody L L, Coulter S N, Folger K R, Kas A, Larbig K, Lim R, Smith K, Spencer D, Wong G K, Wu Z & Paulsen I T Nature __406:__959-964 (19100) [20437337]; E72229-N-acetylglucosamine-1-phosphate (NacGlc-1-P) Uridylyl-transferase from Thermotoga maritima (strain MSB8). Nelson K E, Clayton R A, Gill S R, Gwinn M L, Dodson R J, Haft D H, Hickey E K, Peterson J D, Nelson W C, Ketchum K A, McDonald L, Utterback T R, Malek J A, Linher K D, Garrett M M, Stewart A M, Cotton M D, Pratt M S, Phillips C A, Richardson D, Heidelberg J, Sutton G G, Fleischmann R D, Eisen J A, Fraser C M & et al Nature _399:_323-329 (1999) [99287316]; GalU_MY-CGE-Glc-1-P Uridylyltransferase from Mycoplasma genitalium. Fraser C M, Gocayne J D, White O, Adams M D, Clayton R A, Fleischmann R D, Bult C J, Kerlavage A R, Sutton G, Kelley M _270:__397-403 (1995) [96026346]; GCAD_ Science BACSU-NacGlc-1-P Uridylyltransferase from Bacillus subtilis. Nilsson D, Hove-Jensen B & Arnvig K Mol. Gen. _218:__565-571 (1989) [90066361]; GLGC_ BACSU-Glc-1-P Adenylyltransferase from Bacillus subtilis. Kiel J A, Boels J M, Beldman G & Venema G Mol. Microbiol. 11: 203-218 (1994) [94195107]; RFB SALTY-Glc-1-P Cytidylyltransferase from Salmonella serovar typhimurium (strain LT2). Jiang X M, Neal B, Santiago F, Lee S J, Romana L K & Reeves P R Mol. Microbiol. __5:_ 695-713 (1991) [91260454].

According to one embodiment of the present invention mutations at amino acid L89T were tested. Such mutations increased the yield of allo-, altro-, talo-, gulo- and ido-derivatives. Wild-type and/or this mutant also led to the production of the new nucleotide sugar compounds set forth in FIGS. 20(a) and (b). Methods of production of such compounds and of the mutant nucleotidylyltransferase are as set forth herein with regard to other compounds and mutant nucleotidylyltransferase. In particular, the compounds may be produced by synthesizing the corresponding sugar phosphate followed by E_p catalyzed conversion of the sugar phosphate to the new products.

The present invention includes the nucleotide sugars of FIGS. 20(a) and 20(b), their corresponding sugar phosphates and nucleotidylyltransferases mutated at L89T, which may convert such sugar phosphates to a nucleotide sugar.

Glycorandomization of Natural Product-Based Metabolites

The wild-type glycosyltransferases in secondary metabolism show significant flexibility with respect to their NDP-sugar donors. Coupled with the presented \mathbf{E}_p -catalyzed production of NDP-sugar donor libraries and the appropriate aglycon, a diverse library of "glycorandomized" structures based upon a particular natural product scaffold can be rapidly generated.

Accordingly, the present invention is also directed to nucleotide sugar libraries including two or more of the nucleotide sugars described herein. More preferably the nucleotide sugars are nucleotide sugars made by the methods described herein, preferably using a natural or mutated nucleotidylyltransferase as a catalyst. The present invention also includes in vitro glycorandomization using such sugar libraries.

Exploiting the promiscuity of wild type \mathbf{E}_p and utilizing the ability conferred by the methods of the present invention to

28 EXAMPLES

rationally engineer variants able to utilize sugar phosphates not previously usable, libraries of previously unavailable nucleotide sugars may be generated. The ability to generate a set of \mathbf{E}_p variants provides the subsequent ability to generate, in a simple one pot reaction, diverse libraries of NDP-sugars. Both sugars that were unknown prior to the present invention and those that could not be synthesized in vitro prior to the present invention may be synthesized using the methods of the present invention. Such libraries of NDP-sugars, in conjunction with downstream glycosyltransferases, form the basis for the in vitro manipulation of metabolite sugar ligands

in a combinatorial fashion (or "glycorandomization").

For example, a diverse library of "glycorandomized" structures based upon the known antitumor agent mithramycin (FIG. 5) may be constructed. Beginning with a small pool of sugar phosphates, e.g., 25 different sugar phosphates, the anticipated library size would be the result of combining 25 different sugars at 5 different positions on mithramycin to give 25⁵, or >9.7 million, distinct mithramycin-based vari- 20 ants. Furthermore, as alterations of the carbohydrate ligands of biologically active metabolites can lead to drastically different pharmacological and/or biological properties, this approach has significant potential for drug discovery. As an example of alterations of the carbohydrate ligands of biologi- 25 cally active metabolites producing dramatically different pharmacological properties, the structure of 4-epidoxorubicin differs from that of doxorubicin, which is more toxic, only in carbohydrate ligands. As an example of alterations of the carbohydrate ligands of biologically active metabolites producing dramatically different biological properties, the structure of erythromycin, an antibiotic, differs from that of megalomicin, a compound with antiviral and antiparasitic activity, only in carbohydrate ligands.

An embodiment of the invention includes incubating a glycotransferase with one or more of the sugars of a nucleotide sugar library according to the present invention, and a molecule capable of being glycosylated.

The present inventors have discovered that E_p is pliable in $_{40}$ terms of its substrate specificity. The present inventors have also discovered the three dimensional structure of E_n and the molecular details of E_p substrate recognition. The present inventors have invented methods of engineering or modifying E_n to vary its specificity in a directed manner, conferring the 45 ability to rationally engineer variants able to utilize sugar phosphates not previously usable. The present inventors have also invented a method for the synthesis of desired nucleotide sugars using both natural and engineered E_p . Thus, the present invention will likely broadly impact efforts to understand and exploit the biosynthesis of glycosylated bioactive natural products, many of which are pharmacologically useful. The ability conferred by the methods of the present invention to alter nucleotidylyltransferase specificity by design allows the creation of promiscuous in vitro systems, which could provide large and diverse libraries of potentially new bioactive metabolites.

The present invention will now be illustrated by the following examples, which show how certain specific representative embodiments of the compounds and methods of the present invention, the compounds, intermediates, process steps, and the like being understood as examples that are intended to be illustrative only. In particular, the invention is not intended to be limited to the conditions, order of the steps and the like specifically recited herein. Rather, the Examples are intended to be illustrative only.

General Methods

Infrared spectra were recorded on a Perkin Elmer 1600 series FTIR spectrophotometer. ¹H NMR spectra were obtained on a Bruker AMX 400 (400 MHz) and are reported in parts per million (δ) relative to either tetramethylsilane (0.00 ppm) or CDCl₃ (7.25 ppm) for spectra run in CDCl₃ and D₂O (4.82 ppm) or CD₃OD (3.35 ppm) for spectra run in D₂O. Coupling constants (J) are reported in hertz. ¹³C NMR are reported in δ relative to CDCl₃ (77.00 ppm) or CD₃OD (49.05 ppm) as an internal reference and ³1P NMR spectra are reported in δ relative to H₃PO₄ (0.00 ppm in D₂O). Routine mass spectra were recorded on a PE SCIEX API 100 LC/MS mass spectrometer and HRMS was accomplished by the University of California, Riverside Mass Spectrometry Facility. Optical rotations were recorded on a Jasco DIP-370 polarimeter using a 1.0 or 0.5 dm cell at the room temperature (25° C.) and the reported concentrations. Melting points were measured with Electrothermal 1A-9100 digital melting point instrument. Chemicals used were reagent grade and used as supplied except where noted. Analytical TLC was performed on either E. Merck silica gel 60 F₂54 plates (0.25 mm) or Whatman AL Sil G/UV silica gel 60 plates. Compounds were visualized by spraying I₂/KI/H₂SO₄ or by dipping the plates in a cerium sulfate-ammonium molybdate solution followed by heating. Liquid column chromatography was performed using forced flow of the indicated solvent on E. Merck silica gel 60 (40-63 μm) and high pressure liquid chromatography was performed on a RAININ Dynamax SD-200 controlled with Dynamax HPLC software.

Although the above methods of IR, NMR, Mass Spec, and HRMS were those used in the examples of the present invention, it would be known to those skilled in the art that other instruments, conditions, types of techniques, and the like may be used to visualize compounds, identify compounds and determine their concentrations and purity.

Methyl 2,3,4-tri-O-acetyl-6-chloro-6-deoxy-α-D-glucopyranoside (7)

Compound 7 was prepared as previously described from methyl α -D-glucopyranoside (5), (7.26 g, 27.7 mmol) in 82% yield (Anisuzzaman, A. K. M.; Whistler, R. L. Carbohydr. Res. 1978, 61, 511-518). R_j=0.34 (2:1 hexane/EtOAc); $[\alpha]_D$ =147° (c=1, CHCl₃); ¹H NMR (CDCl₃) 5.46 (t, 1H, J=9.5 Hz), 4.98 (m, 2H), 4.02 (m, 1H), 3.82 (dd, 1H, J=12.0, 2.5 Hz), 3.73 (dd, 1H, J=12.0, 6.5 Hz), 3.43 (s, 3H), 2.06 (s, 3H), 2.04 (s, 3H), 1.99 (s, 3H); ¹³C NMR (CDCl₃) 170.45, 170.42, 169.96, 96.98, 77.67, 71.08, 70.46, 70.32, 69.13, 55.85, 43.81, 21.07, 21.05. MS: calcd for C₁3H₁9O₈ClNa 360.9. found m/z 360.9 (M+Na).

1,2,3,4-tetra-O-benzoyl-6-deoxy-α,β-D-glucopyranose (8)

Compound 7 (2.9 g, 8.57 mmol) was dissolved in 100 mL dry THF and 1.0 g LiAlH₄ slowly added. The corresponding mixture was refluxed for 10 hr under argon and the reaction quenched with 10 mL MeOH and concentrated. The concentrate was then dissolved in a mixture of 40 mL acetic acid and 10 mL 1N HCl and the reaction stirred at 95° C. for 10 hrs. The reaction was neutralized with 1N NaOH and the organics concentrated, dried over MgSO₄ and purified by silica gel chromatography (4:1 CHCl₃/MeOH). The resulting product was dissolved in 50 mL dry pyridine, 8.0 mL benzoyl chloride

(68.9 mmol) was added and the reaction stirred overnight at room temperature. To the reaction mixture was added to 100 mL saturated NaHCO3 solution and the mixture extracted with CHCl₃ (3×100 mL). The combined organics were washed with H₂O (50 mL), brine (50 mL), dried over Na₂SO₄, concentrated and purified by silica gel chromatography (2:1 hexane/EtOAc) to give 2.8 g (56.2%) of the desired product 8 ($\alpha/\beta=3:2$). This mixture was utilized directly for the next step without further resolution. MS: calcd for C₃4H₂8O₉Na 630.2. found m/z 630.2 (M+Na).

Methyl 2,3,6-tri-O-benzoyl-β-D-galactopyranoside (14)

Methyl β-D-galactopyranoside (13), 3.7 g, 19 mmol) gave the desired product 14 (5.2 g, 54%) and 2.3 g (19%) of the corresponding tetra benzoylated derivative as described in Reist, E. J.; Spencer, R. R.; Calkins, D. F.; Baker, B. R.; Goodman, L. J. Org. Chem. 1965, 2312-2317. $[\alpha]_D$ =7.3° (c=1, CHCl₃); ¹H NMR (CDCl₃) 8.18-7.92 (m, 6H), 7.52-7.38 (m, 8H), 5.77 (dd, 1H, J=8.0, 10.4 Hz), 5.37 (dd, 1H, J=3.2, 10.3 Hz, 1H), 4.72 (dd, 1H, J=6.6, 11.4 Hz), 4.62 (dd, 1H, J=6.4, 11.4 Hz), 4.66 (d, 1H, J=7.9 Hz), 4.36 (m, 1H), ¹³C NMR (CDCl₃): 166.9, 166.3, 165.9, 133.9, 133.7, 133.6, 130.4, 130.3, 130.2, 130.1, 130.0, 129.9, 129.4, 129.0, 128.9, 128.8, 128.7, 102.6, 74.6, 72.8, 69.9, 67.7, 63.3, 57.3; MS: calcd for C₂8H₂6O₉Na 529.1. found m/z 529.0 (M+Na). (Garegg, P. J.; Oscarson, S. Carbohydr. Res. 1985, 137, 270- 30 Perkin Trans. 1 1975, 1574.)²²

Methyl 2,3,6-tri-O-benzoyl-4-O-pentafluorophenoxythiocarbonyl-β-D-galactopyranoside (15)

Methyl 2,3,6-tri-O-benzoyl-β-D-galactopyranoside (14), (2.3 g, 4.5 mmol) gave 2.88 g (86%) purified product 15 as described in Kanie, O.; Crawley, S. C.; Palcic, M. M.; Hindsgaul, O. Carbohydr. Res. 1993, 243, 139-164. $[\alpha]_D = 9^\circ$ (c=1, CHCl₃); ¹H NMR (CDCl₃) 8.04 (d, 2H, J=7.7 Hz), 7.98 (d, 40 2H, J=7.6 Hz), 7.93 (d, 2H, J=7.7 Hz), 7.58-7.49 (m, 3H), 7.44-7.34 (m, 6H), 6.23 (d, 1H, J=3.2 Hz), 5.78 (dd, 1H, J=7.9 Hz), 5.70 (dd, 1H, J=3.3, 10.4 Hz), 4.75-4.71 (m, 2H), 4.44 (dd, 1H, J=7.4, 11.0 Hz), 4.37 (t, 1H, J=7.0 Hz), 3.57 (s, 3H); ¹³C NMR (CDCl₃) 192.5, 166.3, 166.0, 165.5, 134.1, 133.9, 45 133.7, 130.3, 130.2, 130.1, 129.6, 129.5, 128.9, 128.8, 128.6, 102.7, 79.9, 71.3, 71.1, 69.9, 61.5, 57.6; MS: calcd for $C_35H_25O_9SF_5Na$ 755.1. found m/z 755.1 (M+Na).

Methyl 2,3,6-tri-O-benzoyl-4-deoxy-β-D-galactopyranoside (16)

Methyl 2,3,6-tri-O-benzoyl-4-O-pentafluorophenoxythiocarbonyl-β-D-galactopyranoside (15), (2.65 g, 3.62 mmol) gave 1.53 g (86%) of the desired compound 16 as described in 55 Kanie, O; Crawley, S. C.; Palcic, M. M.; Hindsgaul, O. Carbohydr. Res. 1993, 243, 139-164. $[\alpha]_D = 57.4^{\circ} (c=1, CHCl_3);$ ¹H NMR (CDCl₃) 8.07 (d, 2H, J=7.3 Hz), 8.00 (d, 2H, J=7.4 Hz), 7.95 (d, 2H, J=7.3 Hz), 7.58 (t, 1H, J=7.4 Hz), 7.53-7.40 (m, 4H), 7.39-7.34 (m, 4H), 5.41 (m, 2H), 4.60 (d, 1H, J=7.5 60 Hz), 4.51 (dd, 1H, J=5.8, 11.6 Hz), 4.46 (dd, 1H, J=4.4, 11.6 Hz), 4.06 (m, 1H), 2.47 (m, 1H), 1.88 (m, 1H); ¹³C NMR (CDCl₃) 166.7, 166.3, 165.9, 133.7, 133.6, 133.5, 130.2, 130.1, 130.0, 129.7, 128.9, 128.8, 128.7, 102.6, 72.9, 71.9, 70.0, 66.2, 57.4, 33.4; MS: calcd for C₂6H₂6O₈Na 513.1. 65 found m/z 513.0 (M+Na). (Lin, T.-H.; Kovac, P.; Glaudemans, C. P. J. Carbohydr. Res. 1989, 141, 228-238.)

1,2:5,6-Di-O-isopropylidene-3-O-(methylthio)thiocarbonyl-α-D-glucofuranose (22)

Compound 22 was prepared as previously described in 93% yield (see Zhiyuan, Z.; Magnusson, G. Carbohydr. Res. 1994, 262, 79-101). $[\alpha]_{D} = -34^{\circ}$ (c=1, CHCl₃); ¹H NMR (CDCl₃) 5.91 (m, 2H), 4.68 (d, 1H, J=3.77 Hz), 4.31 (m, 1H), 4.10 (dd, 1H, J=5.6, 8.7 Hz), 4.05 (dd, 1H, J=4.6, 8.7 Hz), 2.59 (s, 3H), 1.57 (s, 3H), 1.41 (s, 3H), 1.32 (s, 3H), 1.31 (s, 3H); ¹³C NMR (CDCl₃) 112.8, 109.7, 105.4, 84.6, 83.1, 80.1, 72.7, 67.3, 27.2, 27.0, 26.6, 25.6, 19.7; MS: calcd for $C_14H_2O_6S_2Na$ 373.0. found m/z 372.8 (M+Na).

3-Deoxy-1,2:5,6-di-O-isopropylidene-α-D-glucofuranose (23)

To a solution containing 22 (2.6 g, 7.4 mmol) and 120 mg of AIBN (0.73 mmol) in 50 mL dry toluene, 5 mL (n-Bu) ₃SnH (18.6 mmol) was added and the mixture refluxed for 5 hrs under argon. The reaction was then concentrated and the residue was applied to a silica gel colunn (10:1-8:1 hexane/ EtOAc) to give 1.58 g substantially pure product 23 (87%). $[\alpha]_D = -9.2^{\circ}$ (c=1, CHCl₃); ¹H NMR (CDCl₃) 5.82 (d, 1H, J=3.7 Hz), 4.76 (t, 1H, J=4.2 Hz), 4.19-4.07 (m, 3H), 3.84 (m, 4.08 (t, 1H, J=6.5 Hz), 3.55 (s, 3H), 2.50 (br, 1H, .about.OH); 25 1H), 2.18 (dd, 1H, J=3.9, 13.2 Hz), 1.77 (m, 1H), 1.51 (s, 3H), 1.42 (s, 3H), 1.35 (s, 3H), 1.30 (s, 3H); ¹³C NMR (CDCl₃) 109.6, 107.9, 104.0, 79.4, 77.0, 75.7, 65.6, 33.7, 25.7, 24.9, 24.5, 23.6; MS: calcd for $C_{12}H_2O_5$ Na 267.1. found m/z 266.8 (M+Na). (Barton, D. H. R.; McCombie, S. W. J. Chem. Soc.

1,2,4,6-tetra-O-benzoyl-3-deoxy-α-D-glucofuranose (24)

Compound 23 (0.59 g, 2.4 mmol) was treated with a mixture of 9 mL CF₃CO₂H and 1 mL of water for 2 hours at 25° C. The reaction was concentrated under reduced pressure, coevaporated with water (2×5 mL) and further dried under vacuum. This material was dissolved in 20 mL of anhydrous pyridine, to which 2.2 mL (19.3 mmol) of benzoyl chloride was added. The mixture was stirred for 10 hr, pyridine removed in vacuo and the remaining oil diluted with 200 mL EtOAc. The organics washed with saturated NaHCO₃ (50 mL), water (40 mL), brine (40 mL), dried over Na₂SO₄, and purified with silica gel chromatography (3:1 hexanes/EtOAc) to give 0.89 g product which was used directly without further characterization.

General Strategy for Formation of Protected Ethyl 1-thio- β -D-hexopyranosides.

Protected ethyl 1-thio-β-D-hexopyranosides may be formed in accordance with the present invention by the following reaction. A mixture of protected monosaccharide, (ethylthio)-trimethylsilane, and zinc iodide are refluxed for $1\frac{1}{2}$ to $2\frac{1}{2}$ hrs. The reaction is then cooled, diluted, washed, preferably with saturated NaHCO3 solution, water, and then brine. The organics are then dried, and preferably concentrated and resolved to give the desired product. Other conditions, reagents, method steps, solutions and the like of the present method, may be used in accordance with the present invention.

In a typical reaction, a mixture of 3 mmol protected monosaccharide, 1.5 mL (ethylthio)trimethylsilane (9.2 mmol) and 1.95 g zinc iodide (6.1 mmol) in 30 mL dry dichloromethane was refluxed for 2 hrs under argon atmosphere. The reaction was then cooled and diluted with 200 mL CH₂Cl₂, washed successively with saturated NaHCO₃ solution (2×30 mL), water (30 mL) and brine (30 mL). The organics were dried over $\rm Na_2SO_4$, concentrated and resolved by silica gel chromatography (8:1 hexanes/EtOAc) to give the desired product.

Ethyl 2,3,4-tri-O-benzoyl-6deoxy-1-thio-β-D-glucopyranoside (9)

Compound 8 (1 g, 1.72 mmol) gave 731 mg (81.5%) of the desired product. R_{7} =0.56 (2:1 hexane/EtOAc); $[\alpha]_{D}$ =7° (c=1.0, CHCl₃); 1 H NMR (CDCl₃) 8.00-7.94 (m, 4H), 7.82 10 (dd, 1H, J=1.4, 7.1 Hz), 7.52 (m, 2H), 7.42-7.37 (m, 5H), 7.23 (m, 2H), 5.85 (t, 1H, J=9.6 Hz), 5.54 (t, 1H, J=9.7 Hz), 5.35 (t, 1H, J=9.6 Hz), 4.80 (d, 1H, J=9.9 Hz), 4.92 (m, 1H), 2.82 (m, 2H), 1.40 (d, 3H, J=6.2 Hz), 1.26 (t, 3H, J=7.4 Hz); 13 C NMR (CDCl₃) 164.8, 164.4, 164.2, 132.3, 132.2, 132.1, 128.8, 128.7, 128.6, 128.2, 128.0, 127.9, 127.4, 127.3, 127.2, 82.3, 73.9, 73.1, 72.7, 69.8, 22.9, 16.8, 13.7; MS: calcd for C_{2} 9 H_{2} 8 O_{7} SNa 543.1. found m/z 542.9 (M+Na).

Ethyl 2,3,6-tri-O-benzoyl-4-deoxy-1-thio-β-D-glucopyranoside (17)

Compound 16 (1.5 g, 3.06 mmol) gave 1.24 g desired product (77.8%). $[\alpha]_D$ =56.9° (c=1, CHCl₃); 1 H NMR (CDCl₃) 8.08 (d, 2H, J=8.0 Hz), 8.00 (d, 2H, J=8.2 Hz), 7.96 25 (d, 2H, J=8.0 Hz), 7.60 (t, 1H, J=6.9 Hz), 7.55-7.48 (m, 4H), 7.42-7.36 (m, 4H), 5.50-5.44 (m, 2H), 4.76 (d, 1H, J=9.0 Hz), 4.51 (dd, 1H, J=5.7, 1.9 Hz), 4.46 (dd, 1H, J=4.4, 11.9 Hz), 4.12 (m, 1H), 2.84-2.69 (m, 2H), 2.53 (m, 1H), 1.91 (m, 1H), 1.27 (t, 3H, J=7.6 Hz); 13 C NMR (CDCl₃) 166.6, 166.2, 30 165.9, 133.7, 133.6, 130.2, 130.1, 129.8, 129.7, 128.8, 184.2, 74.0, 73.0, 71.5, 66.3, 33.6, 24.7, 15.4; MS: calcd for C₂9H₂8O₇SNa 543.1. found m/z 543.1 (M+Na).

Ethyl 2,4,6-tri-O-benzoyl-3-deoxy-1-thio-β-D-glucopyranoside (25)

Compound 24 (0.89 g, 1.5 mmol) gave 0.79 substantially pure product (90%). $^1{\rm H}$ NMR (CDCl₃) 8.14-7.96 (m, 6H), 7.63-7.40 (m, 9H), 5.32-5.21 (m, 2H), 4.79 (d, 1H, J=9.7 Hz), 40 4.67 (dd, 1H, J=2.9, 12.0 Hz), 4.46 (dd, 1H, J=6.0, 12.0 Hz), 4.09 (m, 1H), 2.96 (m, 1H), 2.78 (m, 2H), 2.00 (m, 1H), 1.27 (t, 3H, J=7.4 Hz); MS: calcd for C₂9H₂8O₇SNa 543.1. found m/z 543.1 (M+Na).

Ethyl 3,4,6-tri-O-benzoyl-2-deoxy-1-thio-β-D-glucopyranoside (40)

Compound 39 (1.72 g, 2.96 mmol) gave two products, 0.74 g the desired β isomer (48% yield) and 0.5 g the β isomer 50 (32% yield). β isomer: $[\alpha]_D$ =120° (c=1, CHCl₃); IR:_2962, 2871, 1723, 1601, 1450, 1314, 1270, 1107, 708, 686 cm⁻¹; ¹H NMR (CDCl₃) 8.06-7.93 (m, 6H), 7.51-7.36 (m, 9H), 5.69-5.64 (m, 1H), 5.60-5.56 (m, 2H), 4.80 (m, 1H), 4.57 (dd, 1H, J=2.7, 12.0 Hz), 4.52 (dd, 1H, J=12.0, 5.5 Hz), 2.72-2.54 (m, 55 3H), 2.41-2.35 (m, 1H), 1.30 (t, 3H, J=7.4 Hz); ¹³C NMR (CDCl₃): 165.2, 164.6, 164.5, 132.3, 132.1, 132.0, 128.8, 128.7, 128.6, 128.4, 128.1, 127.4, 127.3, 78.5, 69.4, 67.3, 62.4, 34.4, 23.8. MS: calcd for C₂9H₂8O₇SNa 543.1. found m/z 543.1 (M+Na). α isomer: $[\alpha]_D$ =-46° (c=1, CHCl₃) IR: 60 2961, 2923, 1732, 1717, 1269, 1108, 1099, 708, 685 cm⁻¹. ¹H NMR (CDCl₃) 8.12-7.93 (m, 6H), 7.54-7.37 (m, 9H), 5.56 (t, 1H, J=9.7 Hz), 5.46 (m, 1H), 4.87 (dd, 1H, J=11.8, 1.7 Hz), 4.60 (dd, 1H, J=3.1, 12.0 Hz), 4.48 (dd, 1H, J=5.9, 12.0 Hz), 4.06 (m, 1H), 2.83-2.68 (m, 2H), 2.65-2.64 (m, 1H), 2.08 (m, 65 1H), 1.32 (t, 3H, J=7.4 Hz); ¹³C NMR (CDCl₃) 166.6, 166.3, 165.9, 134.1, 133.8, 133.7, 133.4, 130.6, 130.1, 130.0, 129.7,

129.5, 128.9, 128.7, 80.3, 77.1, 73.0, 70.5, 64.3, 37.1, 25.5, 15.5. MS: calcd for $\rm C_29H_28O_7SNa$ 543.1. found: m/z 543.0 (M+Na).

Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio-β-D-gulopyranoside (30)

Compound 29 (0.75 g, 1.07 mmol) gave 0.65 g of the desired compound (94%). 1 H NMR (CDCl₃) 8.18-7.87 (m, 8H), 7.54-7.27 (m, 12H), 5.95 (t, J=3.5 Hz, 1H), 5.67 (dd, J=3.3, 10.3 Hz, 1H), 5.61 (m, 1H), 5.27 (d, J=10.3 Hz, 1H), 4.64 (m, 2H), 4.50 (dd, J=3.8, 9.5 Hz, 1H), 2.84 (m, 2H), 1.34 (t, J=7.4 Hz, 3H); MS: calcd for $C_36H_32O_9SNa$ 663.2. found m/z 663.1 (M+Na).

Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio-β-D-allopyranoside (35)

Compound 34 (0.97 g, 1.38 mmol) gave 0.85 g desired product (95%). [α]_D=12.7° (c=1, CHCl₃); 1 H NMR (CDCl₃) 8.07-7.32 (m, 20H), 6.26 (t, J=2.8 Hz, 1H), 5.56 (dd, J=2.8, 10.1 Hz, 1H), 5.51 (dd, J=2.9, 10.1 Hz, 1H), 4.71 (dd, J=2.5, 12.0 Hz, 1H), 4.56 (m, 1H), 4.47 (dd, J=5.3, 12.0 Hz, 1H), 2.80 (m, 2H), 1.29 (t, J=7.5 Hz, 3H); 13 C NMR (CDCl₃) 171.6, 166.6, 165.7, 165.2, 134.0, 133.9, 133.8, 133.6, 130.5, 130.3, 130.2, 130.1, 130.0, 129.9, 129.6, 129.3, 129.2, 128.9, 128.8, 81.3, 73.7, 69.7, 68.9, 68.0, 63.9, 24.3, 15.5; MS: calcd for C_3 6H₃2O₉SNa 663.1. found m/z 663.0 (M+Na).

General Strategy for O-Benzoyl to O-Benzyl Conversion. O-Benzoyl may be converted to O-Benzyl according to the following method or other methods known to those skilled in the art. Protected ethyl 1-thio-β-D-hexopyranoside are dissolved in dry MeOH and toluene to which a sodium methoxide solution is added. The mixture is then stirred, preferably for about 1½ to 2½ hrs at room temperature and optionally neutralized. The organics are preferably concentrated and the corresponding unprotected 1-ethylthio-β-D-glucopyranoside purified, and then dissolved. NaH is then added and the reaction is stirred for about 1½ to 2½ hrs at room temperature followed by the addition of benzyl bromide and stirring, preferably overnight. The mixture may then be diluted, washed with H₂O, brine, and organics, dried, concentrated, and purified to give the purified product.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in 45 the art may be used in accordance with the present invention.

In a typical reaction, 1.4 mmol of protected ethyl 1-thioβ-D-hexopyranoside was dissolved in 10 mL dry MeOH and 3 mL toluene to which 0.25 mL of a sodium methoxide solution (25% NaOMe in methanol) was added. The mixture was stirred for 2 hr at room temperature and neutralized with 1N acetic acid. The organics were concentrated and the corresponding unprotected 1-ethylthio-β-D-glucopyranoside purified by silica gel chromatography (10:1 hexane/EtOAc) which was then dissolved in 10 mL dry DMF and 323 mg 65% NaH (8.0 mmol) was added. The reaction was stirred for 2 hr at room temperature followed by the addition of 1 mL benzyl bromide (8.3 mmol) and continued stirring overnight. The mixture was then diluted with 150 mL EtOAc, washed with H₂O (30 mL), brine (30 mL) and the organics dried over Na₂SO₄, concentrated and purified by silica gel chromatography (10:1 hexane/EtOAc) to give the purified product.

Ethyl 2,3,4-tri-O-benzyl-6-deoxy-1-thio-β-D-glucopyranoside (10)

Compound 9 (0.7 g, 1.35 mmol) gave 480 mg (75%) of purified product. $[\alpha]_D = 5.8^{\circ}$ (c=1.0, CHCl₃); ¹H NMR

33

 $\begin{array}{l} ({\rm CDCl_3})\,7.40\text{-}7.29\ (m,\,15{\rm H}),\,4.95\text{-}4.85\ (m,\,4{\rm H}),\,4.77\ (d,\,1{\rm H},\,\\ {\rm J=}10.2\ {\rm Hz}),\,4.65\ (d,\,1{\rm H},\,{\rm J=}10.5\ {\rm Hz}),\,4.48\ (d,\,1{\rm H},\,{\rm J=}9.8\ {\rm Hz}),\\ 3.66\ (t,\,1{\rm H},\,{\rm J=}8.9\ {\rm Hz}),\,3.46\text{-}3.38\ (m,\,2{\rm H}),\,3.23\ (t,\,1{\rm H},\,{\rm J=}9.2\ {\rm Hz}),\,2.84\text{-}2.70\ (m,\,2{\rm H}),\,1.35\text{-}1.27\ (m,\,2{\rm H});\,^{13}{\rm C}\ {\rm NMR}\\ ({\rm CDCl_3}):\,138.9,\,138.5,\,138.4,\,128.9,\,128.8,\,128.7,\,128.4,\,^{5}\\ 128.3,\,128.2,\,128.1,\,86.8,\,85.2,\,83.8,\,82.5,\,76.2,\,75.9,\,75.8,\\ 25.4,\,18.5,\,15.5.\ {\rm MS}:\ {\rm calcd}\ {\rm for}\ {\rm C_29H_34O_4SNa}\ 501.2.\ {\rm found}\\ {\rm m/z}\ 501.1\ ({\rm M+Na}). \end{array}$

Ethyl 2,3,6-tri-O-benzyl-4-deoxy-1-thio-β-D-glucopyranoside (18)

Compound 17 (0.85 g, 1.63 mmol) gave 675 mg (86%) purified product. $[\alpha]_D$ =40° (c=1, CHCl₃), 1 H NMR (CDCl₃) 7.45-7.31 (m, 1SH), 4.92 (d, 1H, J=10.3 Hz), 4.86 (d, 1H, J=10.3 Hz), 4.74 (d, 1H, J=11.7 Hz), 4.69 (d, 1H, J=11.7 Hz), 4.62 (d, 1H, J=12.0 Hz), 4.58 (d, 1H, J=12.0 Hz), 4.49 (d, 1H, J=9.7 Hz), 3.69-3.62 (m, 3H), 3.50 (m, 1H), 3.36 (dd, 1H, J=8.7, 9.4 Hz), 2.82-2.75 (m, 2H), 2.23 (m, 1H), 1.54 (m, 1H), 1.35 (t, 3H, J=7.5 Hz); 13 C NMR (CDCl₃) 138.8, 138.7, 20 138.5, 128.8, 128.7, 128.6, 128.2, 128.1, 128.0, 85.5, 82.3, 80.6, 76.0, 75.4, 73.9, 72.9, 72.3, 34.4, 25.2, 15.6 MS: calcd for C_2 9H₃4O₄SNa 501.2. found m/z 501.0 (M+Na).

Ethyl 2,4,6-tri-O-benzyl-3-deoxy-1-thio-β-D-glucopyranoside (26)

Compound 25 (608 mg, 1.17 mmol) gave 364 mg substantially pure product (65%). [α]_D=-11.8° (c=1, CHCl₃); 1 H NMR (CDCl₃) 7.44-7.14 (m, 15H), 4.74 (d, 1H, J=11.6 Hz), 4.66-4.56 (m, 4H), 4.50 (d, 1H, J=9.4 Hz), 4.45 (d, 1H, J=11.4 Hz), 3.83 (d, 1H, J=10.7 Hz), 3.69 (dd, 1H, J=4.4, 10.7 Hz), 3.49 (m, 2H), 3.35 (m, 1H), 2.79 (m, 2H), 2.69 (m, H), 1.54 (m, 1H), 1.35 (t, 3H, J=7.3 Hz); 13 C NMR (CDCl₃): 138.8, 138.4, 135.8, 128.9, 128.7, 128.4, 128.2, 128.1, 127.4, 86.9, 35 81.3, 75.6, 73.8, 73.3, 72.5, 71.6, 69.8, 45.3, 36.7, 25.1, 20.3, 15.5; MS: calcd for C₂9H₃4O₄SNa 501.2. found m/z 501.0 (M+Na).

Ethyl 2,3,4,6-tetra-O-benzyl-1-thio-β-D-gulopyranoside (31)

Compound 30 (0.6 g, 0.94 mmol) gave 330 mg substantially pure product (60%). 1 H NMR (CDCl₃) 7.32-7.17 (m, 20H), 4.94 (d, J=9.8 Hz, 1H), 4.55 (m, 2H), 4.40 (m, 4H), 4.22 45 (m, 2H), 4.00 (t, J=6.4 Hz, 1H), 3.60-3.42 (m, 5H), 2.66 (m, 2H), 1.22 (t, J=7.4 Hz, 3H); MS: calcd for $\rm C_{3}6H_{4}OO_{5}SNa$ 607.2. found m/z 607.0 (M+Na).

Ethyl 2,3,4,6-tetra-O-benzyl-1-thio-β-D-allopyranoside (36)

Compound 35 (0.85 g, 1.33 mmol) gave 496 mg substantially pure product (64%). 1 H NMR (CDCl₃) 7.4-7.22 (m, 20H), 5.05 (d, J=9.7 Hz, 1H), 4.86 (d, J=11.8 Hz, 1H), 4.80 (d, 55 J=11.8 Hz, 1H), 4.69-4.40 (m, 6H), 4.13 (m, 1H), 4.03 (dd, J=3.1, 9.7 Hz, 1H), 3.47 (dd, J=2.3, 9.8 Hz, 1H), 3.29 (dd, J=2.3, 9.8 Hz, 1H), 2.75 (m, 2H), 1.32 (t, J=7.5 Hz, 3H); 13 C NMR (CDCl₃) 139.4, 138.9, 138.3, 138.2, 128.9, 128.8, 128.7, 128.6, 128.5, 128.3, 128.2, 128.1, 128.0, 127.9, 127.8, 60 82.0, 79.2, 76.1, 75.4, 74.7, 73.9, 73.8, 72.8, 71.9, 69.9, 25.1, 15.6; MS: calcd for C_3 6H₄0O₅SNa 607.2. found m/z 607.0 (M+Na).

General Phosphorylation Strategy (Method A: Via Ethyl 1-thio- β -D-hexopyranoside).

Phosphorylation may take place in accordance with the present invention by the following reaction, which involves 34

ethyl 1-thio-β-D-hexopyranoside. The ethyl 1-thio-β-D-hexopyranoside may be ethyl 1-thio-β-D-hexopyranoside prepared according to the methods described herein or ethyl 1-thio-β-D-hexopyranoside prepared by other methods.

According to this method, protected ethyl 1-thio- β -D-hexopyranoside and dibenzyl phosphate are co-evaporated, preferably two times from dry toluene and further dried under high vacuum overnight to which N-iodosuccinamide and dry molecular sieves are preferably added. The mixture is then dissolved, preferably in dry CH₂Cl₂, cooled to about -40° C. to about -20° C., preferably about -30° C. and trifluoromethane-sulfonic acid is added. The reaction mixture is substantially maintained at the cooled temperature for about 20 to about 40 minutes, preferably about 30 min with stirring. Preferably, the mixture is then diluted, and washed with saturated Na₂S₂O₃ and/or saturated NaHCO₃, H₂O, and brine. The organics are then preferably dried, filtered, concentrated and purified to give the desired product.

Other conditions, steps, reagents, solutions and the like of the present method may be used in accordance with the present invention.

In a typical reaction, 0.84 mmol protected ethyl 1-thio- β -D-hexopyranoside and 1.44 mmol dibenzyl phosphate were co-evaporated two times from dry toluene and further dried under high vacuum overnight to which 1.24 mmol of N-io-dosuccinamide and 300 mg dry molecular sieves were added. The mixture was then dissolved in 10 mL dry CH₂Cl₂, cooled to -30° C. and 25 μ L trifluoromethanesulfonic acid (0.28 mmol) was added. The reaction mixture was maintained at -30° C. for 30 min with stirring and then diluted with 100 mL EtOAc, washed with saturated Na₂S₂O₃ (20 mL) and saturated NaHCO₃ (20 mL), H₂O (20 mL), and brine (20 mL). The organics were dried over Na₂SO₄, filtered, concentrated and purified by chromatography on silica gel (3:1 hexane/EtOAc) to give the desired product.

Dibenzyl-(2,3,4-tri-O-benzyl-6-deoxy-α-D-glucopy-ranosyl) Phosphate (11)

Compound 10 (400 mg, 0.84 mmol) gave 0.44 mg (76%) of the desired product. [α]_D=22.8° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.38-7.28 (m, 25H), 5.93 (dd, 1H, J=3.2, 6.6 Hz), 5.30 (m, 4H), 5.18 (m, 3H), 5.09 (m, 2H), 4.67 (m, 2H), 3.94 (m, 1H), 3.64 (m, 1H), 3.18 (m, 1H), 1.21 (d, 3H, J=6.2 Hz); ¹³C NMR (CDCl₃) 138.9, 138.5, 138.0, 128.9, 128.8, 128.7, 45 128.4, 128.3, 95.8, 95.7, 94.0, 76.0, 75.7, 73.6, 69.7, 17.3; ³1P NMR (CDCl₃) 2.58; MS: calcd for C₄1H₄3O₈PNa 717.2. found m/z 717.3 (M+Na).

Dibenzyl-(2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopy-ranosyl) Phosphate (19)

Compound 18 (512 mg, 1.07 mmol) gave 0.565 g (76%) substantially pure product. $[\alpha]_D=28.2^\circ$ (c=1, CHCl₃); $^1\mathrm{H}$ NMR (CDCl₃) 7.29-7.10 (m, 25H), 5.91 (dd, 1H, J=3.2, 6.6 Hz), 4.72-4.57 (m, 4H), 4.41 (m, 2H), 4.02 (m, 1H), 3.81 (m, 1H), 3.48 (m, 1H), 3.34 (m, 2H), 2.04-2.00 (m, 1H), 1.60-1.48 (m, 1H); $^{13}\mathrm{C}$ NMR (CDCl₃) 138.6, 138.4, 138.2, 137.8, 137.7, 128.3, 128.2, 128.1, 127.9, 127.8, 127.7, 127.6, 127.5, 127.4, 97.3, 91.9, 83.6, 80.2, 78.1, 74.9, 73.3, 73.0, 72.2, 72.0, 71.9, 70.6, 66.7, 52.7, 33.4; $^{3}\mathrm{1P}$ NMR (CDCl₃) 1.25; MS: calcd for $\mathrm{C_41H_43O_8PNa}$ 717.2. found m/z 717.2 (M+Na).

Dibenzyl-(2,4,6-tri-O-benzyl-3-deoxy-α-D-glucopy-ranosyl) Phosphate (27)

Compound 26 (270 mg, 0.56 mmol) gave 0.31 g substantially pure product (79%, α/β =2:1). ¹H NMR (CDCl₃) 7.32-

7.21 (m, 25H), 5.96 (dd, 1H, J=2.8, 6.6 Hz), 5.06 (m, 4H), 4.66 (d, 1H, J=11.7 Hz), 4.56 (m, 3H), 4.42 (d, 1H, J=12.0 Hz), 4.38 (d, 1H, J=11.3 Hz), 3.82 (m, 1H), 3.66 (m, 3H), 3.49 (m, 1H), 2.54 (m, 0.5H), 2.40 (m, 1H), 1.85 (m, 1H), 1.56 (m, 0.5H); 3 1P NMR (CDCl₃) 0.54, 0.17; MS: calcd for 5 C₄1H₄3O₈PNa 717.2. found m/z 717.2 (M+Na).

Dibenzyl-(3,4,6-tri-O-benzoyl-2-deoxy-α-D-glucopyranosyl) Phosphate (41)

Compound 40 (460 mg, 0.88 mmol) gave 0.49 g of substantially pure product (75%) after silica gel chromatography (3:1-2:1 hexane/EtOAc. $[\alpha]_D=19^\circ$ (c=1, CHCl $_3$); 1 H NMR (CDCl $_3$) 8.05-7.94 (m, 6H), 7.53-7.51 (m, 3H), 7.41-7.34 (m, 16H), 5.96 (dd, 1H, J=1.6, 7.2 Hz), 5.68 (m, 2H), 5.16 (m, 14H), 4.51-4.43 (m, 2H), 4.35 (dd, 1H, J=3.1, 12.0 Hz), 2.56 (m, 1H), 2.04 (m, 1H); 13 C NMR (CDCl $_3$) 166.0, 165.6, 165.3, 135.5, 135.4, 133.3, 133.2, 133.0, 129.8, 129.7, 129.6, 129.3, 128.9, 128.6, 128.4, 128.3, 128.1, 127.9, 95.9, 70.1, 69.5, 69.2, 68.9, 62.6; 3 1P NMR (CDCl $_3$) 0.32; MS: calcd for 20 C $_4$ 1H $_3$ 7O $_1$ 1PNa 759.1 found: m/z 759.1 (M+Na).

Dibenzyl-(2,3,4,6-tetra-O-benzyl-α-D-gulopyranosyl) Phosphate (32)

Compound 31 (120 mg, 0.21 mmol) gave 50 mg of the desired compound (30%) and 38 mg of the β isomer (23%). α isomer: 1H NMR (CDCl $_3$) 7.30-6.90 (m, 30H), 5.95 (dd, J=3.7, 7.5 Hz, 1H), 4.97 (m, 5H), 4.64 (m, 2H), 4.49-4.30 (m, 8H), 3.80 (m, 2H), 3.60 (d, J=3.4 Hz, 1H), 3.44 (m, 2H); 3 1P NMR (CDCl $_3$) 0.8; MS: calcd for C $_4$ 8H $_4$ 9O $_9$ PNa 823.3. found m/z 823.3 (M+Na). β isomer: 1H NMR (CDCl $_3$) 7.25-7.15 (m, 30H), 5.61 (t, J=7.2 Hz, 1H), 5.05-4.99 (m, 4H), 4.60 (d, J=12 Hz, 1H), 4.42-4.34 (m, 4H), 4.26 (d, J=6.2 Hz, 1H), 4.17 (t, J=6.3 Hz, 1H), 3.65 (m, 2H), 3.48 (m, 2H), 3.48 (m, 2H), 3.43 (dd, J=1.3, 13.5 Hz, 1H); 3 1P NMR (CDCl $_3$) -1.1; MS: calcd for C $_4$ 8H $_4$ 9O $_9$ PNa 823.3. found m/z 823.3 (M+Na).

Dibenzyl-(2,3,4,6-tetra-O-benzyl-α-D-allopyranosyl) Phosphate (37)

Compound 36 (169 mg, 0.29 mmol) gave 70 mg the desired compound (30%) and 64 mg of the β isomer (28%). α isomer: ¹H NMR (CDCl₃) 7.34-7.13 (m, 30H), 6.04 (dd, J=3.6, 7.1 45 Hz, 1H), 5.11-4.92 (m, 4H), 4.89 (d, J=12.0 Hz, 1H), 4.84 (d, J=12.0 Hz, 1H), 4.75 (d, J=11.8 Hz, 1H), 4.59-4.37 (m, 6H), 4.23 (m, 1H), 3.73 (dd, J=3.0, 10.0 Hz, 1H), 3.66 (dd, J=2.5, 10.0 Hz, 1H), 3.54 (m, 2H); ¹³C NMR (CDCl₃) 139.4, 138.4, 138.3, 137.9, 136.5, 136.4, 128.9, 128.8, 128.7, 128.7, 128.5, 50 128.4, 128.3, 128.2, 128.1, 128.0, 127.6, 95.1, 76.2, 74.4, 73.9, 73.3, 71.8, 69.6, 69.2, 68.7, 68.6; ³1P NMR (CDCl₃) 0.27; MS: calcd for C₄8H₄9O₉PNa 823.3. found m/z 823.3 (M+Na). β isomer: ¹H NMR (CDCl₃) 7.37-7.07 (m, 30H), 5.63 (t, J=7.7 Hz, 1H), 5.00 (m, 4H), 4.79 (d, J=11.9 Hz, 1H), 554.72 (d, J=11.9 Hz, 1H), 4.63 (d, J=11.9 Hz, 1H), 4.51 (d, J=11.9 Hz, 1H), 4.46 (d, J=12.1 Hz, 1H), 4.36 (m, 3H), 4.09 (dd, J=1.4, 9.7 Hz, 1H), 4.02 (s, 1H), 3.64 (dd, J=3.7, 11.0 Hz, 1H), 3.58 (dd, J=1.5, 11.0 Hz, 1H), 3.51 (dd, J=2.3, 9.8 Hz, 1H), 3.32 (dd, J=2.3, 7.9 Hz, 1H; ³1P NMR (CDCl₃) 0.76; 60 MS: calcd for C₄8H₄9O₉PNa 823.3. found m/z 823.3 (M+Na).

General Phosphorylation Strategy (Method B: Via Glycosyl Halide).

According to another embodiment, phosphorylation may 65 take place in accordance with the present invention by the following reaction, which involves glycosyl halide. The gly-

36

cosyl halide may be glycosyl halide prepared according to the methods described herein or glycosyl halide prepared by other methods.

According to this method, protected D-hexose is dissolved in acetic acid to which HBr in acetic acid was added dropwise at about 0° C. The reaction is allowed to warm to room temperature and stirred for about $1\frac{1}{2}$ to about $2\frac{1}{2}$ hrs. The mixture is then diluted with cold CHCl₃, washed successively with cold saturated NaHCO₃ solution, H₂O and brine, and the organics were dried over anhydrous Na2SO4 and concentrated. The crude protected-α-D-pyranosyl bromide may be used directly without further purification. A mixture of dibenzyl phosphate, silver triflate, 2,4,6-collidine and activated 4 Å molecular sieves in dry CH₂Cl₂ is stirred at room temperature in the absence of light for about 1 hr. The mixture was then cooled to about -30° C. to about -50° C., preferably about -40° C., to which a solution of the crude protected- α -Dpyranosyl bromide in dry CH₂Cl₂ is added in dropwise fashion. The reaction mixture is kept at substantially the same cool temperature for about 1½ to about 2½ M hrs, allowed to warm to room temperature and stirred, preferably overnight. The corresponding filtrate is preferably diluted with CH₂Cl₂, washed with saturated CuSO₄, H₂O, and brine, and the organics are dried over anhydrous Na2SO4 and concentrated. Purification yields substantially pure product.

Other conditions, steps, reagents, solutions and the like of the present method may be used in accordance with the present invention.

Suitably protected D-hexose (0.64 mmol) was dissolved in 5 mL acetic acid to which 5 mL 33% HBr in acetic acid was added dropwise at $0^{\rm o}\,\rm C.$ The reaction was allowed to warm to room temperature and stirred for 2 hr. The mixture was then diluted with 100 mL cold CHCl₃, washed successively with cold saturated NaHCO₃ solution (×30 mL), H₂O (30 mL) and brine (20 mL), and the organics were dried over anhydrous Na₂SO₄ and concentrated. The crude protected-α-D-pyranosyl bromide was used directly without further purification. A 40 mixture of dibenzyl phosphate (1.80 mmol), silver triflate (1.80 mmol), 2.4,6-collidine (3.0 mmol) and 0.5 g activated 4 Å molecular sieves in 10 mL dry CH₂Cl₂ was stirred at room temperature under argon atmosphere in the absence of light for 1 hr. The mixture was then cooled to -40° C. to which a solution of the crude protected-α-D-pyranosyl bromide in 10 mL dry CH₂Cl₂ was added in dropwise fashion. The reaction mixture was kept at -40° C. for 2 hr, allowed to warm to room temperature and stirred overnight. The corresponding filtrate was diluted with 100 mL CH₂Cl₂, washed with saturated $CuSO_4$ (2×20 mL), H_2O (20 mL) and brine (20 mL), and the organics were dried over anhydrous Na₂SO₄ and concentrated. Purification by silica gel chromatography (1:1 hexanes/EtOAc) gave substantially pure product.

Dibenzyl-(2,3,4,6-tetra-O-benzoyl-α-D-altropyranosyl) Phosphate (45)

Perbenzoylated D-altrose (44), (0.675 g, 0.96 mmol) gave 0.58 g substantially pure product (70% overall). $[\alpha]_D$ =40° (c=1, CHCl₃); 1 H NMR (CDCl₃) 8.05 (m, 6H), 7.82 (dd, J=1.2, 7.2 Hz, 2H), 7.53-7.22 (m, 22H), 5.87 (m, 3H), 5.38 (d, J=3.1 Hz, 1H), 5.04 (m, 4H), 4.90 (dd, J=3.0, 10.0 Hz, 1H), 4.58 (dd, J=2.4, 12.3 Hz, 1H), 4.38 (dd, J=3.9, 12.3 Hz, 1H); 13 C NMR (CDCl₃) 166.4, 165.4, 165.3, 164.8, 135.7, 134.3, 133.9, 133.5, 130.5, 130.4, 130.3, 130.1, 129.6, 129.3, 129.1, 129.0, 128.9, 128.8, 128.4, 128.3, 70.2, 70.1, 70.0, 69.5, 69.4,

67.1, 66.9, 65.5, 63.0; 31P NMR (CDCl $_3$) -0.02; MS: calcd for $C_48H_41O_13NaP$ 879.2. found m/z 879.2 (M+Na).

Dibenzyl-(2,3,4,6-tetra-O-benzoyl-α-D-idopyranosyl) Phosphate (49)

Perbenzoylated D-idose (48), (0.32 g, 0.46 mmol) gave 270 mg substantially pure product (69% overall). $[\alpha]_D=11.4^\circ$ (c=1, CHCl₃); 1 H NMR (CDCl₃) 8.11-7.88 (m, 8H), 7.40-7.19 (m, 22H), 6.0 (d, J=6.5 Hz, 1H), 5.68 (m, 1H), 5.46 (m, 10 H), 5.22 (m, 1H), 5.07-5.01 (m, 5H), 4.60 (dd, J=7.0, 11.5 Hz, 1H), 4.53 (dd, J=5.8, 11.5 Hz, 1H); 13 C NMR (CDCl₃) 166.0, 165.1, 164.7, 164.3, 135.2, 133.6, 133.5, 133.1, 130.1, 130.0, 129.9, 129.7, 129.6, 129.4, 128.9, 128.6, 128.5, 128.4, 128.3, 128.2, 127.9, 127.8, 94.9, 69.7, 69.5, 65.9, 65.7, 62.6; 15 3 P NMR (CDCl₃) 0.1; MS: calcd for C₄8H₄1O₁3NaP 879.2. found m/z 879.1 (M+Na).

Dibenzyl-(2,3,4,6-tetra-O-acetyl- α -D-talopyranosyl) Phosphate (53)

Peracylated D-talose (52), (0.248 g, 0.636 mmol) gave 0.436 g substantially pure product (52% overall) $[\alpha]_D$ =40° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.37 (m, 10H), 5.68 (dd, J=1.3, 6.5 Hz, 1H), 5.31 (m, 1H), 5.20 (t, J=3.7 Hz, 1H), 5.11 25 (m, 4H), 5.04 (d, J=3.0 Hz, 1H), 4.30 (dd, J=1.3, 6.8 Hz, 1H), 4.11 (dd, J=11.3, 6.7 Hz, 1H), 3.99 (dd, J=11.3, 6.7 Hz, 1H), 2.12 (s, 3H), 2.11 (s, 3H), 1.99 (s, 3H), 1.92 (s, 3H); ¹³C NMR (CDCl₃) 170.7, 170.4, 169.9, 169.8, 135.7, 135.6, 135.5, 129.2, 129.1, 129.0, 128.5, 128.4, 96.3, 70.3, 68.9, 67.0, 65.6, 30 64.9, 61.7, 21.1, 21.0, 20.9; ³1P NMR (CDCl₃) -0.18; HRMS (FAB) calcd for C₂8H₃4O₁3P 609.1737. found m/z 609.1747 (M+H).

General Strategy for Final Deprotection and Conversion to the Sodium Salt.

Final Deprotection and Conversion to sodium salt may take place in accordance with the present invention by the following reaction.

According to this method, protected α-D-pyranosyl phosphate is dissolved in MEOH, NaHCO₃ solution and 10% 40 Pd/C are added. The mixture is stirred overnight at room temperature under hydrogen atmosphere after which the catalyst is removed, preferably by filtration and the filtrate concentrated. The aqueous layer is preferably extracted, and then partitioned and submitted to an anion exchange column 45 eluted with water, 0.1M NH₄HCO₃, 0.2M NH₄HCO₃ and 0.3M NH₄HCO₃. The product eluted with 0.2 M NH₄HCO₃ and these fractions are pooled and co-evaporated with ethanol, preferably several times to remove excess NH₄HCO₃. The obtained sugar phosphate ammonium salt is subse- 50 quently dissolved in water and applied to a cation-exchange column (Na type) eluted with mL water. The product containing fractions are collected and lyophilized to give the desired product as the sodium salt.

Other conditions, steps, reagents, solutions and the like of 55 the present method may be used in accordance with the present invention.

In a typical reaction, the protected α -D-pyranosyl phosphate (0.5 mmol) was dissolved in 15 mL MeOH, 1.5 mL 1N NaHCO3 solution and 150 mg 10% Pd/C were added. The 60 mixture was stirred overnight at room temperature under hydrogen atmosphere after which the catalyst was removed by filtration and the filtrate concentrated to approximately a 10 mL volume. The aqueous layer was extracted with 10 mL of EtOAc, and then partitioned and submitted to an anion 65 exchange column (Dowex 1×8, 1.2×12 cm) eluted with 100 mL water, 100 mL 0.1 M NH4HCO3, 100 mL 0.2 M

38

NH₄HCO₃ and 100 mL 0.3 M NH₄HCO₃. The product eluted with 0.2M NH₄HCO₃ and these fractions were pooled and co-evaporated with ethanol several times to remove excess NH₄HCO₃. The obtained sugar phosphate ammonium salt was subsequently dissolved in 5 mL water and applied to an AG-X8 cation-exchange column (Na⁺ type) eluted with 100 mL water. The product containing fractions were collected and lyophilized to give the desired product as the sodium salt.

Disodium 6-deoxy-α-D-glucopyranosyl Phosphate (12)

Compound 11 (350 mg, 0.5 mmol) gave 85 mg (58%) of the desired sodium salt. $^1\mathrm{H}$ NMR (D2O) 5.37 (dd, 1H, J=3.4, 7.2 Hz), 3.98 (m, 1H), 3.70 (t, 1H, J=9.5 Hz), 3.45 (m, 1H), 3.09 (t, 1H, J=9.5 Hz), 1.24 (d, 1H, J=6.2 Hz); $^{13}\mathrm{C}$ NMR (D2O) 93.74, 75.78, 73.3, 72.9, 68.2, 17.2; $^3\mathrm{1P}$ NMR (D2O) 3.02; HRMS (FAB): calcd for $\mathrm{C_6H_{12}O_8P}$ 243.0269. found m/z 243.0277 (M+H).

Disodium 4-deoxy- α -D-glucopyranosyl Phosphate (20)

Compound 19 (342 mg, 0.5 mmol) gave 78 mg of the title compound (55%). 1 H NMR (D₂O) 5.49 (dd, 1H, J=3.4, 7.32 Hz), 4.16 (m, 1H), 3.99 (m, 1H), 3.65 (dd, 1H, J=3.2, 12.0 Hz), 3.55 (dd, 1H, J=6.0, 12.0 Hz), 3.41 (m, 1H), 1.99-1.95 (m, 1H), 1.44 (m, 1H); 13 C NMR (D₂O) 95.1, 73.8, 69.5, 67.4, 64.0, 34.3; 3 1P NMR (D₂O) 1.52; HRMS (FAB): calcd for 6 H₁₂O₈P 243.0269. found m/z 243.0260 (M+H).

Disodium 3-deoxy-α-D-glucopyranosyl Phosphate (28)

Compound 27 (270 mg, 0.39 mmol) gave 65 mg title compound (58%) as a 2:1 α /β mixture. ¹H NMR (D₂O) 5.33 (dd, 1H, J=3.2, 7.3 Hz), 3.92-3.50 (m, 5H), 3.46 (m, 1H), 2.33 (m, 0.43H), 2.12 (m, 1H), 1.81 (m, 1H), 1.54 (m, 0.43H); ³1P NMR (D₂O) 3.39, 3.12; HRMS (FAB): calcd for C₆H₁₂O₈P 243.0269. found m/z 243.0267 (M+H).

Disodium 2-deoxy- α -D-glucopyranosyl Phosphate (43)

Debenzylation of 41 (329 mg, 0.447 mmol) was accomplished using the general strategy described above. After the filtrate was concentrated to approximately a 10 mL volume, the solution was cooled to 0° C. and 1.5 mL 1N NaOH solution was added in dropwise manner. The mixture was then stirred at room temperature for 4 hr and subsequently neutralized with 1.0 N acetic acid. The final work-up was accomplished as described in the general strategy to give 69 mg title compound (53%). ¹H NMR (D₂O) 5.53 (m, 1H), 4.01 (m, 1H), 3.88-3.84 (m, 3H), 3.72 (dd, 1H, J=6.2, 12.7 Hz), 3.31 (t, 1H, J=9.4 Hz), 2.19 (dd, 1H, J=5.0, 12.9 Hz), 1.66 (m, H); ³1P NMR (D₂O) 2.68; HRMS (FAB): calcd for C₆H₁₂O₈P 243.0269. found m/z 243.0268 (M+H).

Disodium α-D-gulopyranosyl Phosphate (33)

Compound 32 (35 mg, 0.044 mmol) gave 7.1 mg of the title compound (55%). 1 H NMR (D₂O) 5.15 (dd, J=3.0, 7.7 Hz, 1H), 4.04 (m, 2H), 3.79 (m, 2H), 3.64 (m, 2H); 13 C NMR (D₂O) 96.1, 75.3, 71.6, 70.2, 70.0, 62.2; 3 1P NMR (D₂O) 2.9; HRMS (FAB): calcd for C₆H₁₂O₉P 259.0218. found m/z 259.0231 (M+H).

Compound 37 (63 mg, 0.079 mmol) gave 18 mg substantially pure product (77%). ¹H NMR (D₂O) 5.44 (dd, J=3.5, 7.5 Hz, 1H), 4.14 (m, 1H), 4.00 (m, 1H), 3.90 (dd, J=1.9, 12.3) Hz, 1H), 3.76 (m, 2H), 3.65 (dd, J=3.0, 10.4 Hz, 1H); 13 C NMR (D₂O) 95.8, 75.1, 72.6, 71.8, 68.1, 62.6; ³1P NMR (D₂O) 2.39; HRMS (FAB): calcd for C₆H₁₂O₉P 259.0218. found m/z 259.0217 (M+H).

Disodium α -D-altropyranosyl Phosphate (47)

Using the strategy described for 43, compound 45 (260 mg, 0.3 mmol) gave 62 mg of the desired sodium salt (67% overall). ¹H NMR (D₂O) 5.29 (d, J=8.4 Hz, 1H), 4.14 (m, 1H), 3.98 (m, 1H), 3.94 (t, J=3.5 Hz, 1H), 3.90 (dd, J=2.4, 12.3 Hz, 1H), 3.82 (dd, J=3.5, 12.4 Hz, 1H), 3.77 (dd, J=6.5, 12.3 Hz, 1H); ¹³C NMR (D₂O) 94.9, 70.6, 70.5, 70.0, 64.8, 61.4; ³1P NMR (D₂O) 2.05; HRMS (FAB): calcd for C₆H₁₂O₉P 259.0218. found m/z 259.0211 (M+H).

Disodium α-D-idopyranosyl Phosphate (51)

Using the strategy described for 43, compound 49 (213 mg, 0.25 mmol) gave 61 mg of the title compound (62% overall). 25 HRMS (FAB) calc 563.0705. found m/z 563.00712 (M+H). ¹H NMR (D₂O) 5.14 (dd, J=3.5, 7.7 Hz, 1H), 4.24 (m, 1H), 3.85 (dd, J=8.9, 12.3 Hz, 1H), 3.75 (m, 2H), 3.60 (t, J=5.0 Hz, 1H), 3.32 (m, 1H); ¹³C NMR (D₂O) 99.1, 75.8, 75.5, 74.1, 63.9, 53.2; ³1P NMR (CDCl₃) 2.98; HRMS (FAB): calcd for $C_6H_{12}O_9P$ 259.0218. found m/z 259.0208 (M+H).

Disodium α -D-talopyranosyl Phosphate (55)

Using the strategy described for 43, compound 53 (436 mg, 0.72 mmol) gave 157 mg of the title compound (72%). ¹H 35 NMR (D₂O) 5.48 (d, J=8.2 Hz, 1H), 4.11 (m, 1H), 3.98 (t, J=3.2, 1H), 3.92 (m, 1H), 3.88 (m, 1H), 3.82 (dd, J=11.1, 7.7 Hz, 1H), 3.75 (dd, J=11.7, 4.4 Hz, 1H), 3.19 (q, J=7.3 Hz, 10H), 1.28 (t, J=7.4, 15H); ¹³C NMR (D₂O): 94.1, 70.2, 68.8, 67.6, 62.8, 59.6; ³1P NMR (D₂O) 0.52; HRMS (FAB): calcd 40 for $C_6H_{12}O_9P$ 259.0218. found m/z 259.0209 (M+H).

The following compounds were prepared, preferably according to the methods described herein.

- (58) Thymidine 5'- $(\alpha$ -D-glucopyranosyl diphosphate). HRMS (FAB) calc for $C_16H_25O_16N_2P_2563.0705$. found m/z 45 563.0679 (M+H).
- (59) Uridine 5'-(α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁4H₂3O₁7N₂P₂ 565.0507. found m/z 565.0472 (M+H).
- (60) Thymidine 5'-(2-deoxy-α-D-glucopyranosyl diphos- 50 phate). HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0714 (M+H).
- (61) Uridine 5'-(2-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc C₁4H₂3O₁6N₂P₂ 549.0506. found m/z 549.0510 (M+H).
- (62) Thymidine 5'-(3-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0720 (M+H).
- (63) Uridine 5'-(3-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc C₁4H₂3O₁6N₂P₂ 549.0506. 60 found m/z 549.0485 (M+H).
- (64) Thymidine 5'-(4-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0693 (M+H).
- (65) Uridine 5'-(4-deoxy-α-D-glucopyranosyl diphos- 65 phate). HRMS (FAB): calc C₁4H₂3O₁6N₂P₂ 549.0506. found m/z 549.0500 (M+H).

40

- (66) Thymidine 5'-(6-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB) calc for C₁6H₂5O₁5N₂P₂ 547.0704. found m/z 547.0730 (M+H).
- (67) Uridine 5'-(6-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc C₁4H₂3O₁6N₂P₂ 549.0506. found m/z 549.0492 (M+H).
- (68) Thymidine 5'-(α -D-mannopyranosyl diphosphate). HRMS (FAB) calc 563.0705. found m/z 563.0701 (M+H).
- (69) Uridine 5'- $(\alpha$ -D-mannopyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0503 (M+H).
- (70) Thymidine 5'-(α-D-galactopyranosyl diphosphate). HRMS (FAB) calc 563.0705. found m/z 563.0710 (M+H).
- (71) Uridine 5'-(α-D-galactopyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0508 (M+H).
- (72) Thymidine 5'- $(\alpha$ -D-allopyranosyl diphosphate). HRMS (FAB) calc 563.0705. found m/z 563.0715 (M+H).
- (73) Uridine 5'-(α -D-allopyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0507 (M+H).
- (74) Thymidine 5'-(α -D-altropyranosyl diphosphate). HRMS (FAB) calc 563.0705. found m/z 563.0699 (M+H).
- (75) Uridine 5'-(α -D-altropyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0511 (M+H).
- (76) Thymidine 5'- $(\alpha$ -D-gulopyranosyl diphosphate).
- (77) Uridine 5'-(α-D-gulopyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0512 (M+H).
- (78) Thymidine 5'-(α -D-idopyranosyl diphosphate). HRMS (FAB) calc 563.0705. found m/z 563.0708 (M+H).
- (79) Uridine 5'-(α-D-idopyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0507 (M+H).
- (80) Thymidine 5'-(α-D-talopyranosyl diphosphate). HRMS (FAB) calc 563.0705. found m/z 563.0710 (M+H).
- (81) Uridine 5'-(α-D-talopyranosyl diphosphate). HRMS (FAB): calc 565.0507. found m/z 565.0499 (M+H).

Enzyme Purification. E. coli-prfbA-C (from Professor Hung-wen Liu (Dept. of Chem., Univ. of Minnesota)) was grown in 2 L superbroth, 100 μg mL⁻¹ ampicillin divided among two 4 L baffled flasks for 18 hours at 37° C. Cells were harvested by centrifugation (5000×g, 20 min, 40° C.), washed twice with buffer A (50 mM potassium phosphate buffer, 1 mM EDTA, pH 7.5), resuspended in buffer A (4x weight) and split into two equal volumes. Each was sonicated by three 40 second bursts at 0° C. followed by centrifugation (4400×g, 20 min, 40° C.) to remove cellular debris and a further 1.3-fold dilution of the supernatant with buffer A. To the combined supernatant (167 mL) was added 31.5 mL 5% streptomycin sulfate in a dropwise fashion followed by gentle stirring (1 hr, 4° C.) and centrifugation (14,000×g, 30 min, 4° C.) to remove precipitate. The supernatant was diluted (0.1-fold 1M potassium phosphate buffer, pH 7.5) followed by the slow addition of ammonium sulfate crystals to 65% saturation, gentle stirring $(7.5 \text{ hr}, 4^{\circ} \text{ C.})$ and centrifugation $(4200 \times \text{g}, 30 \text{ min}, 4^{\circ} \text{ C.})$. The precipitated protein was dissolved in a minimum amount of buffer A and dialyzed against buffer B (20 mM Tris.HCl, 1 mM EDTA, pH 7.5). The dialysate was applied to a column of DE52 (3 cm×15 cm) which was washed with 50 mL buffer B and then eluted with a linear gradient (buffer B, 0-500 mM NaCl, 1.0 mL min⁻¹). The E_p fractions (which eluted in the range of 35-75 mM NaCl) were combined (24 mL) and concentrated to 1 mL. Aliquots (300 µL) were further resolved by FPLC (S-200, 20×70 cm, 50 mM Tris.HCl, 200 mM NaCl, pH 7.5). The E_p fractions were combined (7 mL), concentrated (64 mg min⁻¹) and stored in aliquots (5, 20, and 200 μ L) at -80° C. until their use.

General Methods.

Infrared spectra were recorded on a Perkin Elmer 1600 series FTIR spectrophotometer. ¹H NMR spectra were obtained on a Bruker AMX 400 (400 MHz) and are reported in parts per million (δ) relative to either tetramethylsilane (0.00 ppm) or CDCl₃ (7.25 ppm) for spectra run in CDCl₃ or relative to D₂O (4.82 ppm) or CD₃OD (3.35 ppm) for spectra run in D₂O. Coupling constants (J) are reported in hertz. ¹³C NMR are reported in δ relative to CDCl₃ (77.00 ppm) or CD₃OD (49.05 ppm) as an internal reference and ³1P NMR spectra are reported in δ relative to H_3PO_4 (0.00 ppm in D_2O). Routine mass spectra were recorded on a PE SCIEX API 100 LC/MS mass spectrometer and HRMS was accomplished by the University of California, Riverside Mass Spectrometry Facility. Optical rotations were recorded on a Jasco DIP-370 polarimeter using a 1.0 dm cell at the room temperature (25° C.) and the reported concentrations. Melting points were measured with Electrothermal 1A-9100 digital melting bpoint instrument. Chemicals used were reagent grade and 20 used as supplied except where noted. Analytical TLC was performed on Whatman AL Sil G/UV silica gel 60 plates. Compounds were visualized by spraying I₂/KI/H₂SO₄ or by dipping the plates in a cerium sulfate-ammonium molybdate solution followed by heating. Liquid column chromatogra- ²⁵ phy was performed using forced flow of the indicated solvent on E. Merck silica gel 60 (40-63 µm) and high pressure liquid chromatography was performed on a RAININ Dynamax SD-200 controlled with Dynamax HPLC software.

Although the above methods of IR, NMR, Mass Spec, and $\,^{30}$ HRMS were those used in these examples of the present invention, as indicated above, it would be known to those skilled in the art that other instruments, conditions, types of techniques, and the like may be used for visualization of $\frac{\text{Euryl} \, 2.5.7 \, \text{til} \, 3.600}{35 \, \text{noside} \, (\text{FIG.} \, 3(b) \, (104)), \, 2.72 \, \text{g}, \, 8.14 \, \text{mmol}) \, \text{was dissolved in}}$ compounds, to identify compounds and determine their concentrations and purity.

General Strategy for Azide Formation.

Azides in accordance with the present invention may be formed according to the following method. Protected glyco- 40 side is dissolved in CH_2Cl_2 . The mixture is cooled to about 0° C. and pyridine and (CF₃SO₂)₂O are added. The reaction was stirred for approximately 30 min at about 0° C. and then diluted with CH₂Cl₂. The organics were washed with water, dried over Na₂SO₄ and concentrated. The resulting crude 45 residue was dissolved, preferably in anhydrous DMF, to which was added NaN₃. The reaction was subsequently stirred, preferably overnight, at room temperature and then diluted with EtOAc. The organics were washed with water, dried over Na₂SO₄ and concentrated. Preferably, product 50 purification was accomplished by flash chromatography.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.

(2.1 mmol) was dissolved in 10 mL of CH₂Cl₂. The mixture was cooled to 0° C. to which was added pyridine (6.3 mmol) and (CF₃SO₂)₂O (3.2 mmol). The reaction was stirred 30 min at 0° C. and then diluted with CH₂Cl₂ (150 mL). The organics were washed with water (30 mL), dried over Na₂SO₄ and 60 concentrated. The resulting crude residue was dissolved in 10 mL anhydrous DMF, to which was added NaN₃ (407 mg, 6.3 mmol). The reaction was subsequently stirred overnight at room temperature and then diluted with EtOAc (250 mL). The organics were washed with water (2×30 mL), dried over 65 Na₂SO₄ and concentrated. Product purification was accomplished by flash chromatography (4:1 hexane/EtOAc).

Ethyl 4-azide-2,3,6-tri-O-benzyl-4-deoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (94))

Compound (AGCH) 93 (310 mg, 0.63 mmol)⁸ gave 285 mg (88%) desired product. $[\alpha]_D=62.3^\circ$ (c=1, CHCl₂); ¹H NMR (CDCl₃) 7.40-7.31 (m, 15H), 4.94 (d, 1H, J=9.5 Hz), 4.92 (d, 1H, J=10.3 Hz), 4.84 (d, 1H, J=10.6 Hz), 4.73 (d, 1H, J=10.3 Hz), 4.64 (d, 1H, J=12.0 Hz), 4.56 (d, 1H, J=12.0 Hz), 4.44 (d, 1H, J=9.6 Hz), 3.77 (dd, 1H, J=1.8, 10.9 Hz), 3.71-3.62 (m, 2H), 3.54 (t, 1H, J=9.4 Hz), 3.45 (t, 1H, J=9.8 Hz), 3.3 (m, 1H), 2.84-2.69 (m, 2H), 1.33 (t, 3H, J=7.4 Hz); ¹³C NMR (CDCl₃) 137.8, 137.6, 128.3, 128.2, 128.1, 128.0, 127.8, 127.7, 127.5, 85.0, 84.6, 81.3, 77.8, 75.5, 75.3, 73.3, 69.1, 61.9, 24.8, 15.0; MS: calcd for C₂9H₃3N₃O₄SNa 542.2. 15 found m/z 542.0 (M+Na).

> Ethyl 4-azide-3-O-benzoyl-2-O-benzyl-4,6-dideoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (106))

Compound (FIG. **3**(*b*) (105)) gave 0.78 g (87.4%) substantially pure product. $[\alpha]_D = 38^\circ$ (c=1, CHC₃); ¹H NMR (CDCl₃) 8.03 (d, 2H, J=8.2 Hz), 7.60 (m, 1H), 7.47 (t, 2H, J=7.5 Hz), 7.30 (s, 2H), 7.12 (m, 3H), 5.43 (t, 1H, J=9.8 Hz), 4.80 (d, 1H, J=10.7 Hz), 4.34 (m, 2H), 3.54 (t, 1H, J=9.5 Hz), 3.44 (m, 1H), 3.32 (t, 1H, J=9.9 Hz), 2.79 (m, 2H), 1.42 (d, 3H, J=6.0 Hz), 1.34 (t, 3H, J=7.4 Hz); ¹³C NMR (CDCl₃) 165.9, 137.5, 133.8, 130.2, 129.8, 128.9, 128.7, 128.6, 128.2, 85.4, 79.6, 77.7, 76.7, 75.4, 75.1, 66.7, 25.7, 19.0, 15.4; MS: calcd for C₂2H₂5N₃O₄SNa 450.1. found m/z 450.0 (M+Na).

Ethyl 3-O-benzoyl-2-O-benzyl-6-deoxy-1-thio-β-Dgalactopyranoside (FIG. 3(b) (105))

Ethyl 2,3,4-tri-O-acetyl-6-deoxy-1-thio-β-D-galactopyra-30 mL MeOH to which 1.2 mL 25% sodium methoxide was added. From this reaction, 1.58 g (93.3%) ethyl 6-deoxy-1thio-β-D-glactopyranoside was obtained after purification which was combined with TsOH (140 mg, 0.73 mmol) and 2,2-dimethoxypropane (1.9 mL, 15.4 mmol) in 15 mL anhydrous DMF. The reaction was stirred overnight at room temperature, diluted with 200 mL EtOAc and washed successively with saturated NaHCO₃ solution (50 mL) and water (30 mL). The organics were dried over Na₂SO₄ and purified via silica gel chromatography (3:1 hexane/EtOAc) to afford 1.73 g (86%) of purified ethyl 6-deoxy-3,4-O-isopropylidene-1thio- α -D-galactopyranoside. [α]_D=11.9° (c=1, CHCl₃); ¹H NMR (CDCl₃) 4.19 (d, 1H, J=10.2 Hz), 4.01 (m, 2H), 3.84 (dq, 1H, J=1.7, 13.1 Hz), 3.50 (dd, 1H, J=6.2, 10.2 Hz), 2.71 (m, 2H), 1.59 (s, 3H), 1.37 (d, 3H, J=6.6 Hz), 1.33 (s, 3H), 1.28 (t, 3H, J=7.5 Hz); ¹³C NMR (CDCl₃) 110.2, 85.5, 79.5, 76.8, 73.2, 72.3, 28.6, 26.7, 24.6, 17.2, 15.6; MS: calcd for $C_{11}H_2O_4SNa\ 271.1$. found m/z 270.9 (M+Na).

The obtained ethyl 6-deoxy-3,4-O-isopropylidene-1-thio-In a typical reaction, the appropriately protected glycoside 55 α-D-galactopyranoside (1.50 g, 6.0 mmol) was combined with of 60% sodium hydride (0.36 g, 9 mmol) and benzyl bromide (1.44 mL, 12.1 mmol) in 20 mL dry DMF. The reaction was stirred overnight and 1.7 g (83%) ethyl 2-Obenzyl-6-deoxy-3,4-O-isopropylidene-1-thio-α-D-galactopyranoside was obtained after the typical work up and purification. $[\alpha]_D$ =-2.8° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.35 (d, 2H, J=7.1 Hz), 7.25 (t, 2H, J=7.1 Hz), 7.09 (m, 1H), 4.77 (d, 1H, J=11.4 Hz), 4.69 (d, 1H, J=11.4 Hz), 4.31 (d, 1H, J=9.8 Hz), 4.11 (m, 1H), 3.96 (dd, 1H, J=2.0, 15.6 Hz), 3.73 (m, 1H), 3.36 (dd, 1H, J=6.7, 19.8 Hz), 2.64 (m, 2H), 1.42 (s, 3H), 1.29 (d, 3H, J=6.6 Hz), 1.27 (s, 3H), 1.22 (t, 3H, J=7.4 Hz); ¹³C NMR (CDCl₃) 137.9, 128.3, 128.2, 127.6, 109.5,

 $83.3, 79.7, 78.0, 76.5, 73.4, 72.4, 28.0, 26.4, 24.4, 16.8, 14.8; MS: calcd for <math display="inline">\rm C_18H_26O_4SNa$ 361.1. found m/z 361.0 (M+Na).

The obtained ethyl 2-O-benzyl-6-deoxy-3,4-O-isopropylidene-1-thio-β-D-galactopyranoside (1.82 g, 5.38 mmol) was dissolved in a mixture solution including 15 mL 0.5M HCl and 45 mL MeOH and the mixture was subsequently refluxed for 30 min. The reaction was cooled to room temperature, neutralized with solid NaHCO3, and the resulting mixture concentrated. The concentrate was diluted with EtOAc (250 mL), washed with water (2×20 mL) and brine (20 mL), dried over Na₂SO₄ and purified by flash chromatography (1:1 hexane/EtOAc) to give 1.51 g (94%) substantially pure ethyl 2-O-benzyl-6-deoxy-1-thio-β-D-galactopyranoside. $[\alpha]_D = 8.4^{\circ} (c=1, CHCl_3)$; ¹H NMR (CDCl₃) 7.42-7.29 (m, 5H), 4.97 (d, 1H, J=11.0 Hz), 4.67 (d, 1H, J=11.0 Hz), 4.40 (d, 1H, J=9.6 Hz), 3.75 (m, 1H), 3.61 (m, 2H), 3.45 (t, 1H, J=9.3 Hz), 2.78 (m, 2H), 2.48 (d, 1H, J=5.0 Hz), 2.14 (d, 1H, J=5.0 Hz), 1.32 (m, 6H); ¹³C NMR (CDCl₃) 138.5, 129.0, 128.7, 128.5, 85.1, 79.3, 77.6, 75.7, 75.6, 74.8, 72.2, 25.4, 16.9, 15.4; MS: calcd for C₁5H₂O₄SNa 321.1. found m/z 20 321.0 (M+Na).

To a solution of ethyl 2-O-benzyl-6-deoxy-1-thio-β-D-galactopyranoside (1.03 g, 3.45 mmol) and DMAP (126 mg, 1.0 mmol) in 10 mL of dry CH₂Cl₂ at -30° C. was added Et₃N (1.92 mL, 13.8 mmol). Benzoyl chloride (0.4 mL, 3.45 mmol) was added to this mixture in a dropwise fashion, and the stirred at -30° C. for 3 hr. The reaction was then quenched by the addition of MeOH (2 mL) and the mixture was gradually warmed to room temperature after which the resulting mixture was diluted with EtOAc (250 mL). The solution was washed with saturated NaHCO3 solution (2×20 mL), water (30 mL), dried over Na₂SO₄, concentrated and purified by flash chromatography (3:1 to 1:1 hexane/EtOAc) to give 1.12 g (80%) of the title product. $[\alpha]_D$ =96.9° (c=1, CHCl₃); 1 H NMR (CDCl₃) 8.09-8.03 (m, 2H), 7.56 (t, 1H, J=7.4 Hz), 7.49 (m, 2H), 7.22 (m, 2H), 7.18 (m, 3H), 5.28 (dd, 1H, J=3.0, 9.6 Hz), 4.87 (d, 1H, J=10.6 Hz), 4.67 (d, 1H, J=10.6 Hz), 4.56 (d, 1H, J=9.7 Hz), 4.12 (m, 1H), 3.85 (t, 1H, J=9.8 Hz), 3.80 (m, 1H), 2.81 (m, 2H), 1.93 (d, 1H, J=6.7 Hz), 1.36 (m, 6H); ¹³C NMR (CDCl₃) 166.2, 138.0, 133.7, 130.2, 130.1, 128.9, 128.7, 128.6, 128.2, 85.7, 78.1, 77.6, 76.5, 76.0, 74.7, 70.9, 40 25.6, 16.9, 15.4; MS: calcd for C₂2H₂6O₅SNa 425.1. found m/z 425.2 (M+Na).

Strategy for Formation of Protected Ethyl 1-thio- $\beta\text{-}D\text{-}hex\text{-}opyranosides.}$

Ethyl 1-thio-β-D-hexopyranosides may be generally 45 formed as set forth above. The following method is another exemplary embodiment of such method used in accordance with the present invention. In a typical reaction, a mixture of 4.0 mmol protected monosaccharide, 1.5 mL (ethylthio)trimethylsilane (8.0 mmol) and 1.95 g zinc iodide (7.8 mmol) in 50 mL dry dichloromethane was refluxed for 30 min under argon atmosphere. The reaction was then cooled, 50 mL water was added after which the mixture was extracted with chloroform (3×50 mL). The combined organic extracts were washed successively with water (30 mL), saturated NaHCO₃ 55 solution (30 mL) and brine (30 mL). The organics were dried over Na₂SO₄, concentrated and resolved by silica gel chromatography (2:1 hexanes/EtOAc) to give the desired product.

Again, variations of this method may be made in accordance with the present invention, as would be apparent to one 60 skilled in the art.

Ethyl 2,4,6-tri-O-acetyl-3-azide-3-deoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (99))

Compound (FIG. **3**(*b*) (99)) (1.5 g, 4.0 mmol) gave 1.26 g (83.5%) title compound. $[\alpha]_D$ =-49.4° (c=0.5, CHCl₃); ¹H

44

NMR (CDCl₃) 4.95 (m, 2H), 4.43 (d, 1H, J=9.9 Hz), 4.19 (dd, 1H, J=4.1, 12.4 Hz), 4.09 (m, 1H), 3.65 (m, 2H), 2.68 (m, 2H), 2.12 (s, 3H), 2.10 (s, 3H), 1.24 (t, 3H, J=7.5 Hz), 2.06 (s, 3H); $^{13}\mathrm{C}$ NMR (CDCl₃) 171.0, 169.6, 169.6, 84.2, 76.8, 70.3, 68.7, 66.1, 62.6, 24.4, 21.2, 21.1, 21.0, 15.1; MS: calcd for C₁4H₂1N₃O₇SNa 398.1. found m/z 397.9 (M+Na).

Ethyl 2,3,4-tri-O-acetyl-6-azide-6-deoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (88))

Compound (FIG. **3**(*b*) (87)) (680 mg, 1.8 mmol)⁶ gave 590 mg (86%) of the desired title compound. $[\alpha]_D$ =-17.5° (c=1, CHCl₃); ¹H NMR (CDCl₃) 5.23 (t, 1H, J=9.4 Hz), 5.02 (m, 2H), 4.54 (d, 1H, J=10.0 Hz), 3.62 (m, 1H), 3.37 (dd, 1H, J=6.5, 13.5 Hz), 3.30 (dd, 1H, J=2.8, 13.5 Hz), 2.73 (m, 2H), 2.07 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 1.28 (t, 3H, J=7.4 Hz); ¹³C NMR (CDCl₃) 170.0, 169.4, 169.2, 83.0, 77.2, 73.6, 69.7, 69.3, 51.0, 23.6, 20.6, 20.5, 14.6. MS: calcd for C₁4H₂1N₃O₇SNa 398.1. found m/z 397.5 (M+Na).

Ethyl 2,3,4-tri-O-acetyl-6-deoxy-1-thio-β-D-galactopyranoside (FIG. **3**(*b*) (104))

Compound (FIG. 3(*b*) (103)) (6.1 mmol) gave 1.73 g (83%) of the substantially pure product. [α]_D=-17.5° (c=1, CHCl₃); ¹H NMR (CDCl₃) 5.28 (d, 1H, J=3.3 Hz), 5.22 (t, 1H, J=9.9 Hz), 5.05 (dd, 1H, J=3.4, 9.9 Hz), 4.46 (d, 1H, J=9.9 Hz), 3.82 (dd, 1H, J=6.4, 12.8 Hz), 2.74 (m, 2H), 2.17 (s, 3H), 2.06 (s, 3H), 1.98 (s, 3H), 1.28 (t, 3H, J=7.4 Hz), 1.22 (d, 3H, J=6.4 Hz); ¹³C NMR (CDCl₃) 171.0, 170.6, 170.1, 83.9, 77.6, 73.6, 72.7, 70.8, 67.7, 24.5, 21.3, 21.1, 21.0, 16.8, 15.1; MS: calcd for C₁4H₂O₇SNa 357.1. found m/z 356.6 (M+Na).

General Strategy for O-Acetyl to O-Benzyl Conversion.

O-Acetyl may be converted to O-Benzyl according to the following method or other methods known to those skilled in the art. Protected ethyl 1-thio-β-D-hexopyranoside was dissolved in dry MeOH and toluene to which a sodium methoxide solution is added. The mixture was stirred for about 2½ M to about 3½ hrs. at room temperature and neutralized. The organics are then concentrated and the corresponding crude unprotected 1-ethylthio-β-D-glucopyranoside directly dissolved in dry DMF. To this mixture NaH and benzyl bromide is added. The reaction is stirred at room temperature, preferably overnight. The mixture was then diluted with EtOAc, washed with H₂O, brine and organics, dried, concentrated, and purified to give the purified product.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.

In a typical reaction, 2.8 mmol of protected ethyl 1-thio- β -D-hexopyranoside was dissolved in 20 mL dry MeOH and 5 mL toluene to which 0.5 mL of a sodium methoxide solution (25% NaOMe in methanol) was added. The mixture was stirred for 3 hr at room temperature and neutralized with DOWEX 50W X8-100 resin. The organics were concentrated and the corresponding crude unprotected 1-ethylthio- β -D-glucopyranoside directly dissolved in 15 mL dry DMF. To this mixture 330 mg 60% NaH (8.25 mmol) and 1.6 mL benzyl bromide was added. The reaction was stirred at room temperature overnight. The mixture was then diluted with 200 mL EtOAc, washed with H₂O (2×30 mL), brine (30 mL) and the organics dried over Na₂SO₄, concentrated and purified by silica gel chromatography (8:1 hexane/EtOAc) to give the purified product.

Ethyl 3-azide-2,4,6-tri-O-benzyl-3-deoxy-1-thio- β -D-glucopyranoside (FIG. 3(b) (100))

Compound (FIG. **3**(*b*) (99)) (1.05 g, 2.8 mmol) gave 1.03 g (71%) of the desired title compound. $[\alpha]_D = -13.6^{\circ}$ (c=1,

CHCl₃); ¹H NMR (CDCl₃) 7.36-7.28 (m, 15H), 4.90 (d, 1H, J=10.1 Hz), 4.79 (d, 1H, J=10.6 Hz), 4.74 (d, 1H, J=10.1 Hz), 4.60 (d, 1H, J=12.1 Hz), 4.54-4.47 (m, 2H), 4.43 (d, 1H, J=9.6 Hz), 3.70 (m, 1H), 3.57 (t, 1H, J=9.1 Hz), 3.45 (m, 2H), 3.26 (t, 1H, J=9.5 Hz), 2.75 (m, 2H), 1.32 (t, 1H, J=7.3 Hz); ¹³C ⁵ NMR (CDCl₃) 138.4, 137.9, 137.8, 129.5, 129.2, 129.1, 128.9, 128.8, 128.7, 128.6, 128.5, 128.4, 128.3, 128.2, 128.1, 128.0, 127.7, 85.7, 80.4, 79.7, 76.7, 75.8, 75.3, 73.9, 72.5, 71.0, 69.1, 25.6, 15.6; MS: calcd for C₂9H₃3N₃O₄SNa 542.2. found m/z 542.0 (M+Na).

Ethyl 6-azide-2,3,4-tri-O-benzyl-6-deoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (89))

Compound (FIG. **3**(*b*) (88)) (560 mg, 1.5 mmol) gave 645 mg (85%) of the desired product. [α]_D=7.9° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.32-7.16 (m, 15H), 4.87 (d, 1H, J=10.9 Hz), 4.86 (d, 1H, J=10.2 Hz), 4.79 (d, 1H, J=11.2 Hz), 4.76 (d, 1H, J=11.0 Hz), 4.66 (d, 1H, J=10.2 Hz), 4.50 (d, 1H, J=11.0 Hz), 4.43 (d, 1H, J=9.8 Hz), 3.62 (m, 1H), 3.43-3.35 (m, 4H), 3.24 co (dd, 1H, J=6.0, 13.1 Hz), 2.72 (m, 2H), 1.26 (t, 3H, J=1.5 Hz); ¹³C NMR (CDCl₃) 138.2, 137.7, 137.5, 128.5, 128.4, 128.3, 128.2, 128.0, 127.8, 127.7, 86.3, 84.6, 81.5, 78.4, 78.2, 75.7, 75.4, 75.1, 51.3, 24.4, 14.9; MS: calcd for C₂9H₃3N₃O₄SNa 542.2. found m/z 541.9 (M+Na).

General Strategy for Conversion of Azides to Acetamides. Azides may be converted to Acetamides according to the following method or other methods known to those skilled in the art. Benzyl-protected ethyl 1-thio-β-D-azidodeoxy-hexopyranoside and SnCl₂ are combined in acetonitrile. To this mixture thiophenol and Et₃N are added and the reaction is stirred for about ½ to about ½ to about 1½ hr at room temperature. The mixture is then diluted with EtOAc and washed, preferably with 2N NaOH, water, and brine. The organics are dried, preferably over Na₂SO₄, concentrated to dryness and the crude residue dissolved in dry pyridine. To this mixture 2 mL acetic anhydride is added and the reaction stirred, preferably overnight, at room temperature. The reaction is concentrated and purified directly by silica gel chromatography (3:2 to 1:1 hexane/EtOAc) to give the purified product.

Other conditions, reagents, solutions and the like of the present method, or other methods known to those skilled in the art may be used in accordance with the present invention.

In a typical reaction, benzyl-protected ethyl 1-thio- β -D-azidodeoxyhexopyranoside (2.8 mmol) and SnCl $_2$ (1.73 45 mmol) were combined in 10 mL of acetonitrile. To this mixture thiophenol (6.9 mmol) and Et $_3$ N (5.2 mmol) were added and the reaction was stirred for 1 hr at room temperature under argon atmosphere. The mixture was then diluted with EtOAc (150 mL) and washed with 2N NaOH (2×2 mL), water 50 (20 mL) and brine (30 mL). The organics were dried over Na $_2$ SO $_4$, concentrated to dryness and the crude residue dissolved in 10 mL dry pyridine. To this mixture 2 mL acetic anhydride was added and the reaction stirred overnight at room temperature. The reaction was concentrated and purified directly by silica gel chromatography (3:2 to 1:1 hexane/ EtOAc) to give the purified product.

Ethyl 3-acetamido-2,4,6-tri-O-benzyl-3-deoxy-1-thio-β-D-glucopyranoside (FIG. 3(*b*) (101))

Compound (FIG. 3(*b*) (100)) (600 mg, 1.15 mmol) gave 523 mg (85%) of the desired product. [α]_D=-5.4° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.27-7.17 (m, 15H), 5.57 (d, 1H, J=8.6 Hz), 4.73 (d, 1H, J=11.2 Hz), 4.55-4.39 (m, 5H), 3.97 (dd, 1H, J=8.3, 16.5 Hz), 3.66 (m, 2H), 3.58 (dd, 1H, J=4.1, 10.8 Hz), 3.50 (m, 1H), 3.44 (t, 1H, J=8.4 Hz), 2.71 (m, 2H),

 $1.60\,(s,3H),1.24\,(t,3H,J=7.4\,Hz);^{13}C\,NMR\,(CDCl_3)\,170.0,\,137.9,\,137.8,\,137.7,\,128.7,\,128.3,\,128.2,\,128.0,\,127.9,\,127.7,\,127.6,\,85.3,\,80.1,\,78.8,\,75.4,\,73.7,\,73.5,\,73.1,\,69.4,\,55.7,\,25.2,\,23.4,\,15.0;\,MS:\,calcd\,\,for\,\,C_31H_37NO_5SNa\,\,558.2.\,\,found\,\,m/z\,\,558.0\,\,(M+Na).$

Ethyl 4-acetamido-2,3,6-tri-O-benzyl-4-deoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (95))

Compound (FIG. 3(*b*) (94)) (640 mg, 1.23 mmol) gave 530 mg desired product (80%) [α]_D=-36.6° (c=0.5, CHCl₃); 1 H NMR (CDCl₃) 7.32-7.10 (m, 15H), 5.13 (br, 1H), 4.85 (d, 1H, J=10.2 Hz), 4.75 (d, 1H, J=11.7 Hz), 4.70 (d, 1H, J=10.7 Hz), 4.64 (d, 1H, J=10.2 Hz), 4.43 (m, 3H), 3.64-3.47 (m, 5H), 3.37 (m, 1H), 2.67 (m, 2H), 1.61 (s, 3H), 1.25 (t, 3H, J=7.4 Hz); 13 C NMR (CDCl₃) 170.3, 138.3, 138.0, 137.8, 128.5, 128.3, 128.2, 128.1, 127.8, 127.7, 127.6, 84.9, 82.1, 81.7, 78.1, 75.3, 74.7, 73.4, 70.0, 52.6, 24.9, 23.3, 15.1; MS: calcd for C₃1H₃7NO₅SNa 558.2. found m/z 557.9 (M+Na).

Ethyl 6-acetamido-2,3,4-tri-O-benzyl-6-deoxy-1-thio-β-D-glucopyranoside (FIG. **3**(*b*) (90))

Compound (FIG. 3(b) (89)) (502 mg, 0.97 mmol) gave 450 mg desired product (87%). [α]_D=-20.4° (c=1.0, CHCl₃); 1 H NMR (CDCl₃) 7.51-7.23 (m, 15H), 5.87 (d, 1H, J=4.6 Hz), 4.99-4.78 (m, 4H), 4.73 (d, 1H, J=10.2 Hz), 4.63 (d, 1H, J=10.4 Hz), 4.45 (d, 1H, J=9.8 Hz), 3.70-3.60 (m, 2H), 3.52 (m, 1H), 3.41-3.34 (m, 3H), 2.74 (m, 2H), 1.95 (s, 3H), 1.32 (t, 3H, J=7.4 Hz); 13 C NMR (CDCl₃) 169.8, 138.1, 137.6, 137.5, 128.4, 128.3, 128.2, 128.1, 127.9, 127.8, 127.6, 86.2, 85.1, 81.5, 78.5, 77.1, 75.6, 75.4, 75.1, 39.9, 25.2, 23.1, 15.1; MS: calcd for C₃1H₃7NO₅SNa 558.2. found m/z 558.2 (M+Na).

Phosphorylation Procedure.

As set forth in the methods above, phosphorylation according to the present invention may occur via a protected ethyl 1-thio- β -D-hexopyranoside. The following method is another exemplary embodiment of such method used in accordance with the present invention.

In a typical reaction, 1.13 mmol protected ethyl 1-thio-β-D-hexopyranoside and 1.7 mmol dibenzyl phosphate were co-evaporated two times from dry toluene and further dried under high vacuum for 4 hr to which 1.36 mmol of N-iodosuccinamide and 500 mg of dry molecular sieves were added. The mixture was then dissolved in 10 mL dry dichloromethane, cooled to -30° C. and 30 μL of trifluoromethanesulfonic acid (0.34 mmol) was added. The reaction was maintained at -30° C. for 30 min with stirring and then diluted with EtOAc (150 mL), washed with saturated Na₂S₂O₃ (20 mL), saturated NaHCO₃ (20 mL), H₂O (20 mL), and brine (30 mL). The organics were dried over Na₂SO₄, filtered, concentrated and purified by chromatography on silica gel (3:1 hexane/EtOAc) to give the desired product.

Again, variations of this method may be made in accordance with the present invention, as would be apparent to one skilled in the art. Dibenzyl-(3-azide-2,4,6-tri-O-benzyl-3-deoxy-α-D-glucopyranosyl) phosphate (FIG. 3(b) (100a)).

Compound (FIG. 3(b) (100)) (590 mg, 1.55 mmol) gave 700 mg (84%) of the title compound. [α]_D=57.8° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.45-7.27 (m, 25H), 5.99 (dd, 1H, J=3.2, 6.8 Hz), 5.11-5.05 (m, 4H), 4.84 (d, 1H, J=10.6 Hz), 4.82 (d, 1H, J=11.4 Hz), 4.72 (d, 1H, J=11.5 Hz), 4.60 (d, 1H, J=12.0 Hz), 4.49 (d, 1H, J=10.7 Hz), 4.46 (d, 1H, J=12.1 Hz), 3.84 (m, 2H), 3.68 (dd, 1H, J=3.0, 10.9 Hz), 3.57 (t, 1H, J=9.8 Hz), 3.48 (m, 2H); ¹³C NMR (CDCl₃) 137.9, 137.6, 137.4, 136.2, 136.1, 136.0, 129.0, 128.9, 128.7, 128.6, 128.5, 128.4,

128.3, 128.1, 128.0, 94.9, 77.4, 76.1, 75.7, 75.3, 74.0, 73.2, 72.4, 69.9, 69.8, 69.7, 69.6, 67.9, 65.2; 3 1P NMR (CDCl $_3$) 0.82; MS: calcd for C $_4$ 1H $_4$ 2N $_3$ O $_8$ PNa 758.2. found m/z 758.2 (M+Na).

Dibenzyl-(3-acetamido-2,4,6-tri-O-benzyl-3-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (101a))

Compound (FIG. 3(b) (101)) (490 mg, 0.91 mmol) gave 480 mg (70%) of the desired product. $[\alpha]_D=52^\circ$ (c=1, CHCl₃); $^1\mathrm{H}$ NMR (CDCl₃) 7.40-7.19 (m, 25H), 5.94 (dd, 1H, J=3.2, 6.7 Hz), 5.07 (br, 1H), 4.97 (m, 4H), 4.63 (d, 1H, J=11.7 Hz), 4.56 (d, 1H, J=12.0 Hz), 4.39 (m, 4H), 3.90 (m, 2H), 3.85 (m, 2H), 3.56 (dd, 1H, J=3.3, 11.0 Hz), 3.39 (dd, 1H, J=1.6, 11.0 Hz), 1.76 (s, 3H); $^{13}\mathrm{C}$ NMR (CDCl₃) 170.7, 137.8, 137.5, 137.4, 135.6, 135.5, 128.4, 128.3, 128.2, 127.9, 127.8, 127.7, 127.6, 127.5, 94.9, 77.2, 75.1, 74.0, 73.3, 72.8, 72.3, 69.3, 69.4, 69.0, 67.9, 53.4, 23.4; $^{3}\mathrm{IP}$ NMR (CDCl₃) 0.62; MS: calcd for $\mathrm{C_43H_46NO_9PNa}$ 774.3. found m/z 774.3 (M+Na).

Dibenzyl-(4-azide-2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (94a))

Compound (FIG. 3(b) (94)) (280 mg, 054 mmol) gave 316 mg (80%) of the desired product. $[\alpha]_D$ =105.8° (c=1, CHCl₃); 7.28-7.14 (m, 25H), 5.85 (dd, 1H, J=3.2, 6.8 Hz), 5.12-4.96 (m, 5H), 4.82 (d, 1H, J=10.6 Hz), 4.71-4.66 (m, 2H), 4.57 (d, 1H, J=11.3 Hz), 4.50 (d, 1H, J=12.1 Hz), 4.37 (d, 1H, J=12.1 Hz), 3.68-3.47 (m, 5H), 3.37 (dd, 1H, J=1.5, 11.0 Hz); ¹³C NMR (CDCl₃) 137.7, 137.6, 137.5, 135.7, 135.6, 128.5, 128.4, 128.3, 128.2, 128.2, 128.0, 127.9, 127.8, 127.7, 127.6, 127.5, 95.4, 78.9, 78.8, 75.6, 75.5, 73.4, 72.9, 71.5, 69.3, 69.2, 69.2, 67.9, 60.8; ³1P NMR (CDCl₃) 0.82; MS: calcd for 35 C₄1H₄2N₃O₈PNa 758.2. found m/z 758.0 (M+Na).

Dibenzyl-(4-acetamido-2,3,6-tri-O-benzyl-4-deoxy- α -D-glucopyranosyl) Phosphate. (FIG. 3(b) (95a))

Compound (FIG. 3(b) (95)) (430 mg, 0.80 mmol) gave 389 mg (65%) of the desired product. $[\alpha]_D$ =35° (c=1, CHCl₃); ¹H NMR (CDCl₃) 7.23-7.09 (m, 25H), 5.82 (dd, 1H, J=3.2, 6.8 Hz), 5.47 (d, 1H, J=8.5 Hz), 5.01-4.93 (m, 4H), 4.69 (d, 1H, J=11.7 Hz), 4.59 (d, 1H, J=11.1 Hz), 4.54 (m, 2H), 4.35 (d, 1H, J=11.9 Hz), 4.30 (d, 1H, J=11.9 Hz), 3.92 (m, 1H), 3.86 (m, 1H), 3.77 (t, 1H, J=9.7 Hz), 3.55 (m, 1H), 3.39 (m, 2H), 1.65 (s, 3H); ¹³C NMR (CDCl₃) 170.0, 138.2, 137.7, 137.3 135.7, 135.6, 135.5 128.3, 128.2, 128.1, 128.0, 127.9, 127.8, 50 127.7, 127.6, 127.4, 95.6, 79.4, 79.3, 77.2, 74.5, 73.3, 72.8, 72.2, 69.2, 50.8, 23.2; ³1P NMR (CDCl₃) 0.71; MS: calcd for C₄3H₄6NO₉PNa 774.3. found m/z 774.3 (M+Na).

Dibenzyl-(6-azide-2,3,4-tri-O-benzyl-6-deoxy-α-D-glucopyranosyl) Phosphate. (FIG. **3**(*b*) (89a))

Compound (FIG. 3(*b*) (89)) (430 mg, 0.76 mmol) gave 285 mg (51%) of the desired product and 160 mg the β isomer. $[\alpha]_D$ =41.5° (c=1, CHCl₃); ¹H NMR(CDCl₃) 7.28-7.16 (m, 60 25H), 5.87 (dd, 1H, J=3.2, 6.7 Hz), 5.05-4.92 (m, 4H), 4.85 (d, 1H, J=10.9 Hz), 4.81 (d, 1H, J=11.0 Hz), 4.71 (d, 1H, J=11.3 Hz), 4.70 (d, 1H, J=10.9 Hz), 4.60 (d, 1H, J=11.3 Hz), 4.50 (d, 1H, J=11.0 Hz), 3.81 (m, 2H), 3.54 (dt, 1H, J=9.5, 3.1 Hz), 3.47 (t, 1H, J=9.5 Hz), 3.18 (m, 2H); ¹³C NMR (CDCl₃) 65 138.2, 137.7, 137.3, 135.7, 135.6, 128.5, 128.4, 128.3, 128.0, 127.9, 127.8, 127.7, 127.6, 95.1, 80.7, 79.3, 79.2, 77.1, 75.5,

75.1, 73.0, 71.9, 69.3, 69.2, 69.2, 69.1, 50.7; $^3 1P$ NMR (CDCl $_3$) 0.75; MS: calcd for C $_4 1H_4 2N_3O_8 PNa$ 758.2. found m/z 758.1 (M+Na).

Dibenzyl-(6-acetamido-2,3,4-tri-O-benzyl-6-deoxyα-D-glucopyranosyl) Phosphate. (FIG. **3**(*b*) (90a))

Compound (FIG. 3(b) (90)) (430 mg, 0.80 mmol) gave 389 mg (65.0%) of the desired product. ¹H NMR (CDCl₃) 7.27-7.19 (m, 25H), 6.00 (br, 1H), 5.68 (dd, 1H, J=3.4, 5.5 Hz), 4.99-4.93 (m, 4H), 4.85 (d, 1H, J=11.9 Hz), 4.76 (d, 1H, J=10.6 Hz), 4.72 (d, 1H, J=10.5 Hz), 4.65 (d, 1H, J=11.5 Hz), 4.60 (d, 1H, J=11.5 Hz), 4.57 (d, 1H, J=10.5 Hz), 3.81 (m, 2H), 3.49 (dt, 1H, J=3.5, 9.4 Hz), 3.44 (m, 2H), 3.24 (t, 1H, J=9.5 Hz); ¹³C NMR (CDCl₃) 178.5, 138.3, 138.0, 137.9, 136.2, 136.1, 129.0, 128.9, 128.8, 128.5, 128.4, 128.3, 128.1, 95.5, 75.7, 75.6, 74.6, 73.9, 73.4, 72.9, 70.0, 69.9, 69.6, 69.5, 68.5, 54.0, 29.9; ³1P NMR (CDCl₃) 0.53; MS: calcd for $C_43H_46NO_9PNa$ 774.6. found m/z 774.3 (M+Na).

Dibenzyl-(4-azide-3-O-benzoyl-2-O-benzyl-4,6-dideoxy- α -D-glucopyranosyl) Phosphate. (FIG. 3(b) (106a))

Compound (FIG. 3(b) (106)) (323 mg, 0.76 mmol) gave 350 mg (72%) substantially pure product. $[\alpha]_D$ =100.1° (c=1, CHCl₃); ¹H NMR (CDCl₃) 8.16 (m, 2H), 7.62 (m, 1H), 7.49 (t, 2H, J=7.9 Hz), 7.34-714 (m, 11H), 5.98 (dd, 1H, J=3.2, 7.1 Hz), 5.68 (t, 1H, J=9.8 Hz), 5.13-5.05 (m, 4H), 4.68 (d, 1H, J=12.1 Hz), 4.50 (d, 1H, J=12.1 Hz), 3.81 (m, 1H), 3.68 (dt, 1H, J=3.0, 9.8 Hz), 3.28 (t, 1H, J=10.0 Hz), 1.24 (d, 3H, J=6.2 Hz); ¹³C NMR (CDCl₃) 185.8, 137.2, 136.1, 136.0, 133.8, 130.3, 129.9, 129.0, 128.9, 128.8, 128.6, 128.4, 128.1, 95.1, 77.7, 76.5, 72.7, 72.1, 70.1, 70.1, 69.7, 69.7, 68.7, 66.3, 18.6; ³1P NMR (CDCl₃) 0.52; MS: calcd for C₃4H₃4N₃O₈PNa 666.2 found m/z 666.2 (M+Na).

Strategy for Final Deprotection and Conversion to the Sodium Salt.

Set forth above is a general strategy for final deprotection and conversion to the sodium salt according to the present invention. The following method is another exemplary embodiment of such method used in accordance with the present invention.

In a typical reaction, the protected α-D-pyranosyl phosphate (0.5 mmol) was dissolved in 15 mL MeOH, 1.5 mL 1N NaHCO₃ solution and 150 mg 10% Pd/C were added. The mixture was stirred overnight at room temperature under hydrogen atmosphere after which the catalyst was removed by filtration and the filtrate concentrated and redissolved 10 mL water. The aqueous layer was extracted with EtOAc (10 mL), and then submitted to an anion exchange column (Dowex 1×8, 1.2×12 cm) eluted with 100 mL water, 100 mL 0.1 M NH₄HCO₃, 100 mL 0.2 M NH₄HCO₃ and 100 mL 0.3 M NH₄HCO₃. The product eluted with 0.2M NH₄HCO₃ and these fractions were pooled and co-evaporated with ethanol several times to remove excess NH₄HCO₃. The obtained sugar phosphate ammonium salt was subsequently dissolved in 5 mL water and applied to an AG-X8 cation-exchange column (Na+ type) eluted with 100 mL water. The product containing fractions were collected and lyophilized to give the desired product as the sodium salt.

Again, variations of this method may be made in accordance with the present invention, as would be apparent to one skilled in the art.

Disodium (3-amino-3-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (96))

Dibenzyl-(3-azide-2,4,6-tri-O-benzyl-3-deoxy-α-D-glu-copyranosyl) phosphate (250 mg, 0.34 mmol) gave 68 mg

(66%) of the title compound. $[\alpha]_D$ =68.1° (c=1, H₂O); ¹H NMR (D₂O) 5.46 (dd, 1H, J=3.0, 7.0 Hz), 3.93 (m, 1H), 3.85 (m, 1H), 3.74 (dd, 1H, J=4.5, 12.5 Hz), 3.69 (m, 1H), 3.58 (m, 1H), 3.45 (t, 1H, J=10.2 Hz); 13 C NMR (D₂O) 91.9, 71.2, 68.4, 65.3, 59.3, 54.8; ³1P NMR (D₂O) 2.85; HRMS: calcd ⁵ for C₆H₁3NO₉P 258.0379. found m/z 258.0372 (M+H).

Disodium-(3-acetamido-3-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (97))

Dibenzyl-(3-acetamido-2,4,6-tri-O-benzyl-3-deoxy-α-Dglucopyranosyl) phosphate (280 mg, 0.37 mmol) gave 59 mg (53%) of the desired product. $[\alpha]_D$ =93.6° (c=1, H₂O); 1 HNMR (D₂O) 5.43 (dd, 1H, J=2.9, 6.5 Hz), 4.07 (t, 1H, J=10.3 Hz), 3.88 (dd, 1H, J=2.5, 9.7 Hz), 3.80 (m, 1H), 3.71 (dd, 1H, J=4.7, 12.3 Hz), 3.57 (m, 1H), 3.40 (t, 1H, 10.1 Hz), 2.01 (s, 3H); ¹³C NMR (D₂O) 174.4, 92.8, 71.6, 69.5, 67.1, 59.8, 53.3, 21.5; ³1P NMR (D₂O) 2.07; HRMS: calcd for C₈H₁5NO₉P 300.0484. found m/z 300.0478 (M+H).

Disodium-(4-amino-4-deoxy- α -D-glucopyranosyl) Phosphate (FIG. 3(b) (91))

Dibenzyl-(4-azide-2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranosyl) phosphate (350 mg, 0.476 mmol) gave 77 mg (54%) the desired product. ¹H NMR (D₂O) 5.46 (dd, 1H, J=3.2, 7.1 Hz), 4.11 (m, 1H), 3.90-3.75 (m, 3H), 3.59 (m, 1H), 3.13 (t, 1H, J=10.2 Hz); ¹³C NMR (D₂O) 92.9, 71.3, 68.8, 68.2, 59.8, 51.7; ³1P NMR (D₂O) 2.80; HRMS: calcd for ³⁰ $C_6H_13NO_8P$ 258.0379. found m/z 258.0372 (M+H).

Disodium-(4-acetamido-4-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (92))

Dibenzyl-(4-acetamido-2,3,6-tri-O-benzyl-4-deoxy-α-Dglucopyranosyl) phosphate (370 mg, 0.50 mmol) gave 120 mg (71%) of the desired product. [α]_D=109.2° (c=1, H₂O); ¹H NMR (D₂O) 5.44 (dd, 1H, J=3.3, 7.2 Hz), 3.89 (m, 1H), 3.76 (m, 2H), 3.64 (dd, 1H, J12.4, 1.2 Hz), 3.53 (m, 2H), 1.99 (s, 3H); ¹³C NMR (D₂O) 173.8, 93.2, 71.4, 70.4, 69.8, 60.0, 50.6, 21.3; ³1P NMR (D₂O) 1.93; HRMS: calcd for C₈H₁5NO₉P 300.0484. found m/z 300.0499 (M+H).

Disodium-(6-amino-6-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (85))

Dibenzyl-(6-azide-2,3,4-tri-O-benzyl-6-deoxy-α-D-glucopyranosyl) phosphate (360 mg, 0.49 mmol) gave 85 mg 50 (57%) of the title compound. ¹H NMR (D₂O) 5.47 (dd, 1H, J=3.5, 6.8 Hz), 4.14 (dt, 1H, J=2.5, 12.6 Hz), 3.78 (t, 1H, J=9.5 Hz), 3.55 (m, 2H), 3.33 (t, 1H, J=9.3 Hz), 3.07 (dd, 1H, J=10.3, 12.9 Hz); ¹³C NMR (D_2O) 94.1, 73.4, 72.5, 72.4, 72.3, 68.6, 41.0; ³1P NMR (D₂O) 2.80; HRMS: calcd for 55 562.0839. found m/z 562.0835 (M+H). C₆H₁3NO₈P 258.0379. found m/z 258.0388 (M+H).

Disodium-(6-acetamido-6-deoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (86))

Dibenzyl-(6-acetamido-2,3,4-tri-O-benzyl-6-deoxy-α-Dglucopyranosyl) phosphate (340 mg, 0.45 mmol) gave 124 mg (79.4%) of the desired product. $[\alpha]_D = 60.5^{\circ}$ (c=1, H₂O); ¹H NMR (D₂O) 5.39 (dd, 1H, J=3.2, 6.5 Hz), 3.95 (t, 1H, J=7.1 Hz), 3.73 (t, 1H, J=9.4 Hz), 3.54 (m, 1H), 3.45 (m, 1H), 65 3.34 (dd, 1H, J=6.7, 14.1 Hz), 3.25 (t, 1H, J=9.5 Hz), 1.99 (s, 3H); ¹³C NMR (D₂O) 178.4, 97.4, 76.7, 75.9, 74.8, 73.8,

43.9, 25.6; ³1P NMR (D₂O) 2.98; HRMS: calcd for C₈H₁5NO_oP 300.0484. found m/z 300.0482 (M+H).

Disodium-(4-amino-4,6-dideoxy-α-D-glucopyranosyl) Phosphate (FIG. **3**(*b*) (102))

Dibenzyl-(4-azide-3-O-benzyl-2-O-benzyl-4,6dideoxy-α-D-glucopyranosyl) phosphate (300 mg, 0.466 mmol) was dissolved in a mixture of 10 mL of MeOH and 2 mL of toluene. To this solution was added 1.4 mL 1N NaOH and 100 mg of 10% Pd/C and the reaction stirred overnight under hydrogen atmosphere. The catalyst was removed by filtration, the filtrate concentrated to a volume of 4 mL, cooled to 0° C., and 0.7 mL 1N NaOH solution was added in a dropwise fashion. The mixture was stirred for 3 hr at 0° C., neutralized with 1N HOAc and the product purified via anion exchange as described in the general procedure above to give 86 mg (67%) of the substantially pure product. ¹H NMR $_{20}$ (D₂O) 5.44 (dd, 1H, J=3.2, 6.7 Hz), 4.24 (m, 1H), 3.88 (t, 1H, J=9.7 Hz), 3.56 (dd, 1H, J=1.3, 9.4 Hz), 2.94 (t, 1H, J=10.3 ${\rm Hz)}, 1.32 \, ({\rm d}, 3{\rm H}, {\rm J=}6.2 \, {\rm Hz}); {}^{13}{\rm C} \, {\rm NMR} \, ({\rm D_2O}) \, 92.9, 71.5, 68.3,$ 64.2, 56.6, 16.2; ³1P NMR (D₂O) 2.16. HRMS: calcd for $C_6H_13NO_7P$ 242.0429. found m/z 242.0441 (M+H).

E_n-Catalyzed Conversion.

A reaction containing 2.5 mM NTP, 5.0 mM sugar phosphate, 5.5 mM MgCl₂ and 10 U inorganic pyrophosphatase in a total volume of 50 µL 50 mM potassium phosphate buffer, pH 7.5 at 37° C. was initiated by the addition of 3.52 U E_n (1 U=the amount of protein needed to produce 1 μmol TDP-α-D-glucose min⁻¹) The reaction was incubated with slow agitation for 30 min at 37° C., quenched with MeOH (50 μL), centrifuged (5 min, 14,000×g) and the supernatant was stored at -20° C. until analysis by HPLC. Samples (30 μ L) were 35 resolved on a Sphereclone 5 u SAX column (250×4.6 mm) fitted with a guard column (30×4.6 mm) using a linear gradient (50-200 mM potassium phosphate buffer, pH 5.0, 1.5 mL min^{-1} , A₂75 nm).

The following compounds were prepared, preferably 40 according to the methods described herein:

- (109) Thymidine 5'-(6-amino-6-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0837 (M+H).
- (110) Uridine 5'-(6-amino-6-deoxy-α-D-glucopyranosyl 45 diphosphate). HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0640 (M+H).
 - (111) Thymidine 5'-(4-amino-4-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0848 (M+H).
 - (112) Uridine 5'-(4-amino-4-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0638 (M+H).
 - (113) Thymidine 5'-(3-amino-3-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂
 - (114) Uridine 5'-(3-amino-3-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0622 (M+H).
- (115) Thymidine 5'-(2-amino-2-deoxy-α-D-glucopyrano-60 syl diphosphate). HRMS (FAB): calc for C₁6H₂6O₁5N₃P₂ 562.0839. found m/z 562.0842 (M+H).
 - (116) Uridine 5'-(2-amino-2-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁5H₂4O₁6N₃P₂ 564.0632. found m/z 564.0630 (M+H).
 - (117) Thymidine 5'-(6-acetamido-6-deoxy-α-D-glucopydiphosphate). HRMS (FAB): $C_18H_28O_16N_3P_2$ 604.0945. found m/z 604.0953 (M+H).

(118) Uridine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 606.0737. found m/z 606.0732 (M+H).

(119) Thymidine 5'-(4-acetamido-4-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc C₁8H₂8O₁6N₃P₂ 604.0945. found m/z 604.0940 (M+H).

(120) Uridine 5'-(4-acetamido-4-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 606.0737. found m/z 606.0730 (M+H).

(121) Thymidine 5'-(3-acetamido-3-deoxy-α-D-glucopy- 10 diphosphate). HRMS (FAB): calc $C_18H_28O_16N_3P_2$ 604.0945. found m/z 604.0947 (M+H).

(122) Uridine 5'-(3-acetamido-3-deoxy-α-D-glucopyranosyl diphosphate). HRMS (FAB): calc for $C_17H_26O_17N_3P_2$ 606.0737. found m/z 606.0735 (M+H).

(123) Thymidine 5'-(2-acetamido-2-deoxy-α-D-glucopydiphosphate). HRMS (FAB): $C_18H_28O_16N_3P_2$ 604.0945. found m/z 604.0951 (M+H)

(124) Uridine 5'-(2-acetamido-2-deoxy-α-D-glucopyra-606.0737. found m/z 606.0738 (M+H).

(125) Thymidine 5'-(4-amino-4,6-dideoxy-α-D-glucopydiphosphate). HRMS (FAB): calc $C_16H_26O_14N_3P_2$ 546.0889. found m/z 546.0895 (M+H).

(126) Uridine 5'-(4-amino-4,6-dideoxy-α-D-glucopyra- 25 nosyl diphosphate). HRMS (FAB): calc for C₁5H₂4O₁5N₃P₂ 548.0682. found m/z 548.0673 (M+H).

Structure-Based Engineering of E_p

Expression, Purification and Mutagenesis of E_p.

 E_n may be modified in accordance with the present invention according to the following method: E_n and E_n mutants are expressed and purified by methods known in the art. For seleno-methionine-labeled protein, the expression vector was transformed into the methionine auxotroph E. coli B834 and grown, preferably overnight at a temperature of about 25° C. 35 to about 35° C., preferably about 30° C. in the presence of seleno-methionine. Seleno-methionine-labeled E_n is purified using the standard protocol but in the presence of DTT. All E_n mutant gene cassettes are generated by a two-step PCR approach. Mutant genes are subsequently characterized by 40 dsDNA sequencing of both strands.

According to a preferred method, expression and purification and E_p and E_p mutants were accomplished as described in Jiang, J., Biggins, J. B. & Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP- 45 and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000). For seleno-methionine-labeled protein, the expression vector was transformed into the methionine auxotroph E. coli B834 and grown overnight at 30° C. in the presence of 50 mg L⁻¹ seleno-methionine. Seleno-methion- 50 ine-labeled E_p was purified using the standard protocol but in the presence of 5 mM DTT. No additional proteolysis or modifications during this process were observed by mass spectrometry. All E_n mutant gene cassettes were generated by a two-step PCR approach. Mutant genes were subsequently 55 characterized by dsDNA sequencing of both strands.

Crystallization. A general crystallization technique that may be used in accordance with the present invention, is as follows: Purified E_p is concentrated in a buffer, and crystallized in a hanging drop by vapor diffusion at approximately 60 room temperature (20° C.). E_p -dTTP crystals are obtained against reservoir containing TTP, 2.0 M ammonium phosphate, 0.1 M Tris.HCl, pH 8.5, and 20 mM MgCl₂. Crystals grow with two monomers (half of the Ep tetramer) in the asymmetric unit. The E_p -UDP-Glc crystals were obtained against a reservoir containing 2 mM UDP-Glc, 1.9 M ammonium sulfate, isopropanol.

52

According to an exemplary method, the purified E_p was concentrated to 20 mg mL⁻¹ in a buffer containing 10 mM KCl, 2 mM MgCl₂ and 10 mM HEPES, pH 7.2, and crystallized in a hanging drop by vapor diffusion at room temperature (20° C.). The E_p -dTTP crystals were obtained against reservoir containing 2 mM TTP, 2.0 M ammonium phosphate, 0.1 M Tris.HCl, pH 8.5, and 20 mM MgCl₂. Crystals grow in the tetragonal space group $P4_32_{12}$ (a=b=120 Å, c=94 Å) with two monomers (half of the E_p tetramer) in the asymmetric unit. The E_n-UDP-Glc crystals were obtained against a reservoir containing 2 mM UDP-Glc, 1.9 M ammonium sulfate, and 7.5% isopropanol. These crystals grow in the orthorhombic space group P2₁₂₁₂₁ (a=93 Å, b=112 Å, c=132 Å) with four monomers (one tetramer) in the asymmetric unit.

Data Collection and Structure Determination.

Data may be collected and structure determination made according to methods that would be known to those skilled in the art, including, for example, x-ray crystallography.

According to an exemplary embodiment, crystals were nosyl diphosphate). HRMS (FAB): calc for C₁7H₂6O₁7N₃P₂ 20 harvested and flash frozen in the cold stream of an X-Stream cooling system (Rigaku) in the mother liquor with added 20-25% glycerol as a cryoprotectant. Data was collected either in house using a Rigaku RAXIS-IV imaging plate area detector, or at the NSLS Brookhaven beamline X9B. Oscillation photographs were integrated, scaled and merged using DENZO and SCALEPACK. (Otwinowski, Z. & Minor, W. Data Collection and Processing., Sawyer, L., Isaacs, N. & Bailey, S. Ed. SERC Daresbury Laboratory: Warrington, UK. 556-562 (1993).) Subsequent calculations were performed with the CCP4 program suite. (CCP4, The CCP4 suite: programs for X-ray crystallography. Acta Crystallogr. D, 50, 760-763 (1994).) The E_p -UDP-Glc structure was determined using the single wavelength anomalous diffraction phasing method. (Hendrickson, W. A., Determination of Macromolecular Structures from Anomalous Diffraction of Synchrotron Radiation. Science, 254, 51-58 (1991).) Only the dataset collected at the wavelength of the selenium absorption peak was processed. Peak wavelength anomalous data were input to the program SnB to identify the location of the Se atoms. Twenty peaks from the best solution were refined using MLPHARE (CCP4) employing only the peak wavelength anomalous differences in the resolution range 35 to 2 A. Additional Se sites were located using anomalous-difference fourier maps. The final round of MLPHARE consisted of 47 Se sites. Seven of these sites correspond to Se-methionines with dual sidechain conformation. The phases calculated from MLPHARE had a figure of merit of 0.34 which was improved to 0.72 by density modification with the program DM (CCP4). The resulting electron density map was clearly interpretable, indicating also the correct handedness of the Se substructure. The map was further improved using free atom refinement and the automatic chain tracing procedure of the wARP program. Out of the 1156 residues, the main chain of 1003 were automatically traced and very clear density could be seen for the rest of the structure. The unambiguous tracing and sequence assignment of the E_p tetramer was completed using the 0 program. (Jones, T. A., et al., Improved Methods for Building Protein Models in Electron Density Maps and the Location of Errors in these Models. Acta Crystallogr., A47, 110-119 (1991).) Refinement of the model by conventional least-squares algorithm was done with XPLOR. (Brunger, A. T., X-PLOR v. 3.1 Manual. New Haven: Yale University (1993).) The final refined E_p tetramer model at 2.0 A resolution had a free R-value (Brunger, A. T., Free R Value: A Novel Statistical Quantity for Assessing the Accuracy of Crystal Structures. Nature, 355, 472-475 (1992)) of 22.3% and included 9938 non-hydrogen atoms in 1156 well-ordered

residues (1-289 in each monomer) and 762 water molecules. In our determination, electron density was lacking for only the 3 C-terminal residues of each monomer. Restrained refinement of temperature factors was monitored throughout by the free R-factor criterion as set forth in Liu, H.-w. & Thorson, J. S. Pathways and Mechanisms in the Biogenesis of Novel Deoxysugars by Bacteria. Ann. Rev. Microbiol. 48, 223-256 (1994) and Johnson, D. A. & Liu, H.-w. Mechanisms and Pathways from Recent Deoxysugar Biosynthesis Research. Curr. Opin. Chem. Biol. 2, 642-649 (1998)). Stereochemical analysis of the refined model using PROCHECK (CCP4 suite) revealed main-chain and side-chain parameters better than, or within, the typical range of values for protein structures determined at 2.0 Å resolution (overall G-factor, $_{15}$ 2.2). None of the E_p residues fell in the disallowed region of the Ramachandran plot. (Ramachandran, G. N., Ramakrishnan, C. & Sasisekharan, V. Stereochemistry of Polypeptide Chain Configuration. J. Molec. Biol., 7, 95-99 (1963).) The E_n-dTTP structure was determined using the Molecular Replacement (MR) method, with our E_n-UDP-Glc structure as a search model and the program XPLOR. The final refined model (half of the E_p tetramer) at 2.1 Å resolution had a free

54

R value³5 of 23.5% and included 5017 non-hydrogen atoms in 578 well defined in the electron density map amino acids (1-289 for each monomer), and 387 water molecules. The PROCHECK overall G-factor is 2.5, and none of the $\rm E_p$ residues fell in the disallowed region of the Ramachandran plot.

Enzyme Assays and Determination of Steady State Kinetic Parameters.

Assays for product formation and steady state kinetics were accomplished using conditions similar to those described. in Jiang, J., Biggins, J. B. & Thorson, J. S. A General Enzymatic Method for the Synthesis of Natural and "Unnatural" UDP- and TDP-Nucleotide Sugars. J. Am. Chem. Soc. 122, 6803-6804 (2000). For the mutant pool assays, an aliquot which contained an eqimolar ratio of each mutant (60 µg) was utilized.

While the present invention is described with respect to particular examples and preferred embodiments, it is understood that the present invention is not limited to these examples and embodiments. The present invention as claimed therefore, includes variations from the particular examples and preferred embodiments described herein, as will be apparent to one of skill in the art.

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65					70					75					80
Glu	Glu	Pro	Leu	Gly 85	Thr	Ala	Gly	Pro	Leu 90	ГЛа	Leu	Ala	Glu	Glu 95	Val
Leu	ГЛа	Lys	Asp 100	Asp	Ser	Pro	Phe	Phe 105	Val	Leu	Asn	Ser	Asp 110	Val	Ile

Cys Asp Tyr Pro Phe Lys Glu Leu Ala Asp Phe His Lys Ala His Gly 115 120 125

Ala Ala Gly Thr Ile Val Ala Thr Lys Val Asp Glu Pro Ser Lys Tyr 130 140

Gly Val Ile Val His Asp Arg Asp Thr Pro Asn Leu Ile Asp Arg Phe 145 150 155 160

Val Glu Lys Pro Val Glu Phe Val Gly Asn Arg Ile Asn Ala Gly Leu 165 170 175

Tyr Ile Leu Asn Pro Ser Val Ile Asp Leu Ile Glu Met Arg Pro Thr \$180\$ \$190\$

Ser Ile Glu Lys Glu Thr Phe Pro Ile Leu Val Glu Gln Lys Gln Leu 195 \$200\$ 205

Asp Phe Leu Ser Gly Thr Cys Leu Tyr Leu Thr Ser Leu Ser Lys Lys 225 230 230 235

His Pro Glu Lys Leu Cys Lys Glu Lys Tyr Val His Gly Gly Asn Val 245 250 255

Leu Ile Asp Pro Thr Ala Lys Ile His Pro Ser Ala Leu Ile Gly Pro \$260\$

Asn Val Thr Ile Gly Pro Asn Val Val Val Gly Glu Gly Ala Arg Ile 275 280 285

Gln Arg Ser Val Leu Leu Ala Asn Ser Gln Val Lys Asp His Ala Trp 290 295 300

Val Lys Ser Thr Ile Val Gly Trp Asn Ser Arg Ile Gly Lys Trp Ala 305 310 315 320

Arg Thr Glu Gly Val Thr Val Leu Gly Asp Asp Val Glu Val Lys Asn

Glu Ile Tyr Val Asn Gly Ala Lys Val Leu Pro His Lys Ser Ile Ser 340 345 350

Ser Asn Val Glu Lys Glu Ser Ile Ile Met

<210> SEQ ID NO 9

<211> LENGTH: 445 <212> TYPE: PRT

<213 > ORGANISM: Thermotoga maritima

<400> SEQUENCE: 9

Met Arg Ala Leu Val Leu Ala Ala Gly Lys Gly Thr Arg Met Lys Ser

Lys Ile Pro Lys Val Leu His Pro Leu Ser Gly Arg Pro Met Ile Glu

Trp Val Ile Glu Thr Ala Gly Lys Val Ala Gln Lys Val Gly Val Val 35 40 45

Leu Gly Phe Glu Ala Glu Leu Val Arg Lys Ala Leu Pro Glu Trp Val

Asp Val Phe Val Gln Gly Glu Gln Leu Gly Thr Ala His Ala Val Met 65 70 75 80

Cys Ala Lys Asp Phe Ile Glu Pro Gly Asp Asp Val Leu Ile Leu Tyr

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											COII	CIII	ueu	
			85					90					95	
Gly Asp	Val	Pro 100	Leu	Ile	Ser	Glu	Asn 105	Thr	Leu	Lys	Arg	Met 110	Ile	Glu
Glu His	Arg 115	Lys	Gly	Ala	Asp	Val 120	Thr	Ile	Leu	Val	Ala 125	Asp	Leu	Glu
Asp Pro		Gly	Tyr	Gly	Arg 135	Val	Ile	Gln	Asp	Gly 140	Asp	Lys	Tyr	Arg
Ile Ile 145	Glu	Asp	Thr	Asp 150	Leu	Pro	Glu	Glu	Leu 155	Lys	Ser	Val	Thr	Thr 160
Ile Asn	Thr	Gly	Phe 165	Tyr	Val	Phe	Ser	Gly 170	Asp	Phe	Leu	Leu	Arg 175	Ala
Leu Pro	Glu	Ile 180	Lys	Asn	Glu	Asn	Ala 185	Lys	Gly	Glu	Tyr	Tyr 190	Leu	Thr
Asp Ala	Val 195	Asn	Phe	Ala	Glu	Lys 200	Val	Arg	Val	Val	Arg 205	Thr	Asp	Asp
Leu Leu 210		Ile	Thr	Gly	Val 215	Asn	Thr	Arg	Lys	Thr 220	Leu	Val	Trp	Leu
Glu Glu 225	Gln	Leu	Arg	Met 230	Arg	Lys	Ile	Glu	Glu 235	Leu	Leu	Glu	Asn	Gly 240
Val Thr	Ile	Leu	Asp 245	Pro	Ala	Thr	Thr	Tyr 250	Ile	His	Tyr	Ser	Val 255	Glu
Ile Gly	Met	Asp 260	Thr	Val	Ile	Tyr	Pro 265	Met	Thr	Phe	Ile	Glu 270	Gly	ГЛа
Ser Arg	Val 275	Gly	Glu	Asn	CAa	Glu 280	Ile	Gly	Pro	Met	Thr 285	Arg	Ile	Val
Аар Суа 290		Ile	Gly	Asn	Asn 295	Val	ГÀа	Ile	Thr	Arg 300	Ser	Glu	Cys	Phe
Lys Ser 305	Val	Ile	Glu	Asp 310	Asp	Val	Ser	Val	Gly 315	Pro	Phe	Ala	Arg	Leu 320
Arg Glu	Gly	Thr	Ile 325	Leu	ГÀЗ	ГÀа	Ser	Ser 330	ГЛа	Ile	Gly	Asn	Phe 335	Val
Glu Ile	Lys	Lys 340	Ser	Thr	Ile	Gly	Glu 345	Gly	Thr	ГÀа	Ala	Gln 350	His	Leu
Ser Tyr	Ile 355	Gly	Asp	Ala	Phe	Val 360	Gly	ГÀа	Asn	Val	Asn 365	Val	Gly	Ala
Gly Thr 370		Thr	CÀa	Asn	Tyr 375	Asp	Gly	ГÀа	ГЛа	380 TÀa	Asn	Pro	Thr	Phe
Ile Glu 385	Asp	Gly	Ala	Phe 390	Ile	Gly	Ser	Asn	Ser 395	Ser	Leu	Val	Ala	Pro 400
Val Arg	Ile	Gly	Lys 405	Gly	Ala	Leu	Ile	Gly 410	Ala	Gly	Ser	Val	Ile 415	Thr
Glu Asp	Val	Pro 420	Pro	Tyr	Ser	Leu	Gly 425	Leu	Gly	Arg	Ala	Arg 430	Gln	Val
Val Lys	Glu 435	Gly	Trp	Val	Leu	Lys 440	Lys	Arg	Lys	Glu	Glu 445			
<210> S <211> L <212> T <213> O	ENGT	H: 49	56	illu:	a sul	otil:	is							
<400> S	EQUE	NCE:	10											
Met Asp 1	Lys	Arg	Phe 5	Ala	Val	Val	Leu	Ala 10	Ala	Gly	Gln	Gly	Thr 15	Arg

Met Lys Ser Lys Leu Tyr Lys Val Leu His Pro Val Cys Gly Lys Pro

						71					US	7,9	06,	460	B2
						/1					-	con	tin	ued	
			20					25					30		
Met	Val	Glu 35	His	Val	Val	Asp	Glu 40	Ala	Leu	ГЛа	Leu	Ser 45	Leu	Ser	Lys
Leu	Val 50	Thr	Ile	Val	Gly	His 55	Gly	Ala	Glu	Glu	Val 60	Lys	ГÀа	Gln	Leu
Gly 65	Asp	Lys	Ser	Glu	Tyr 70	Arg	Val	Gln	Ala	Lys 75	Gln	Leu	Gly	Thr	Ala 80
His	Ala	Val	Lys	Gln 85	Ala	Gln	Pro	Phe	Leu 90	Ala	Asp	Glu	Lys	Gly 95	Val
Thr	Ile	Val	Ile 100	Cys	Gly	Asp	Thr	Pro 105	Leu	Leu	Thr	Ala	Glu 110	Thr	Met
Glu	Gln	Met 115	Leu	Lys	Glu	His	Thr 120	Gln	Arg	Glu	Ala	Lys 125	Arg	Thr	Ile
Leu	Thr 130	Ala	Val	Ala	Glu	Asp 135	Pro	Thr	Gly	Tyr	Gly 140	Arg	Ile	Ile	Arg
Ser 145	Glu	Asn	Gly	Ala	Val 150	Gln	Lys	Ile	Val	Glu 155	His	Lys	Asp	Ala	Ser 160
Glu	Glu	Glu	Arg	Leu 165	Val	Thr	Glu	Ile	Asn 170	Thr	Gly	Thr	Tyr	Cys 175	Phe
Asp	Asn	Glu	Ala 180	Leu	Phe	Arg	Ala	Ile 185	Asp	Gln	Val	Ser	Asn 190	Asp	Asn
Ala	Gln	Gly 195	Glu	Tyr	Tyr	Leu	Pro 200	Asp	Val	Ile	Glu	Ile 205	Leu	Lys	Asn
Glu	Gly 210	Glu	Thr	Val	Ala	Ala 215	Tyr	Gln	Thr	Gly	Asn 220	Phe	Gln	Glu	Thr
Leu 225	Gly	Val	Asn	Asp	Arg 230	Val	Ala	Leu	Ser	Gln 235	Ala	Glu	Gln	Phe	Met 240

Lys Glu Arg Ile Asn Lys Arg His Met Gln Asn Gly Val Thr Leu Ile

Asp Pro Met Asn Thr Tyr Ile Ser Pro Asp Ala Val Ile Gly Ser Asp 260 265 270

Thr Val Ile Tyr Pro Gly Thr Val Ile Lys Gly Glu Val Gln Ile Gly

Glu Asp Thr Ile Ile Gly Pro His Thr Glu Ile Met Asn Ser Ala Ile

Gly Ser Arg Thr Val Ile Lys Gln Ser Val Val Asn His Ser Lys Val 310

Val Ile Gly Asn Glu Val Lys Ile Gly Asn Phe Val Glu Ile Lys Lys 345

Thr Gln Phe Gly Asp Arg Ser Lys Ala Ser His Leu Ser Tyr Val Gly 355 360 365

Asp Ala Glu Val Gly Thr Asp Val Asn Leu Gly Cys Gly Ser Ile Thr $_{\rm 370}$ $_{\rm 375}$ $_{\rm 380}$

Val Asn Tyr Asp Gly Lys Asn Lys Tyr Leu Thr Lys Ile Glu Asp Gly

Ala Phe Ile Gly Cys Asn Ser Asn Leu Val Ala Pro Val Thr Val Gly 410

Glu Gly Ala Tyr Val Ala Ala Gly Ser Thr Val Thr Glu Asp Val Pro 425

Gly Lys Ala Leu Ala Ile Ala Arg Ala Arg Gln Val Asn Lys Asp Asp 440

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Tyr Val Lys Asn Ile His Lys Lys
<210> SEQ ID NO 11
<211> LENGTH: 257
<213> ORGANISM: Yersinia pseudotuberculosis
<400> SEQUENCE: 11
Val Lys Ala Val Ile Leu Ala Gly Gly Leu Gly Thr Arg Leu Ser Glu
Glu Thr Val Val Lys Pro Lys Pro Met Val Glu Ile Gly Gly Lys Pro
Ile Leu Trp His Ile Met Lys Leu Tyr Ser Ser Tyr Gly Ile Asn Asp 35 \  \  \, 40 \  \  \, 45
Phe Val Ile Cys Cys Gly Tyr Lys Gly Tyr Val Ile Lys Glu Tyr Phe
Ala Asn Tyr Phe Met His Met Ser Asp Ile Thr Phe Cys Met Arg Asp
Asn Glu Met Val Val His Gln Lys Arg Val Glu Pro Trp Asn Val Thr
Leu Val Asp Thr Gly Glu Asp Ser Met Thr Gly Gly Arg Leu Arg Arg
                               105
Val Lys Asp Tyr Val Lys Asp Asp Glu Ala Phe Cys Phe Thr Tyr Gly
Asp Gly Val Ser Asp Val Asn Ile Ala Glu Leu Ile Ala Phe His Lys
Ser His Gly Lys Gln Ala Thr Leu Thr Ala Thr Tyr Pro Pro Gly Arg
Phe Gly Ala Leu Asp Ile Lys Asp Lys Gln Val Arg Ser Phe Lys Glu
Lys Phe Lys Gly Asp Gly Ala Leu Ile Asn Gly Gly Tyr Phe Val Leu
Ser Pro Lys Val Ile Asp Leu Ile Asp Gly Asp Lys Ser Thr Trp Glu
Gln Glu Pro Leu Met Thr Leu Ala Ala Gln Gly Glu Leu Met Ala Phe
Glu His Ala Gly Phe Trp Gln Pro Met Asp Thr Leu Arg Asp Lys Ile
                  230
Tyr Leu His Glu Leu Trp Glu Glu Gly Arg Ala Pro Trp Lys Val Trp 245 \hspace{1cm} 250 \hspace{1cm} 255 \hspace{1cm}
Glu
<210> SEQ ID NO 12
<211> LENGTH: 257
<212> TYPE: PRT
<213 > ORGANISM: Salmonella typhimurium
<400> SEQUENCE: 12
Met Lys Ala Val Ile Leu Ala Gly Gly Leu Gly Thr Arg Leu Ser Glu
Glu Thr Ile Val Lys Pro Lys Pro Met Val Glu Ile Gly Gly Lys Pro
Ile Leu Trp His Ile Met Lys Met Tyr Ser Val His Gly Ile Lys Asp
                            40
Phe Ile Ile Cys Cys Gly Tyr Lys Gly Tyr Val Ile Lys Glu Tyr Phe
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What is claimed is:

- 1. A nucleotide sugar library comprising two or more nucleotide sugars produced by (a) combining $\alpha\text{-}D\text{-}hexopyranosyl phosphate and NTP} in the presence of at least one mutant Ep nucleotidylyltransferase, wherein the at least one mutant Ep nucleotidylyltransferase is mutated at one or more amino acids of SEQ ID NO:1 selected from the group consisting of V173, G147, W224, N122, G175, D111, E162, T201, I200, E199, R195, L89, L109, Y146, and Y177, and (b) recovering the resulting nucleotide sugars.$
- 2. A nucleotide sugar library comprising two or more nucleotide sugars produced by (a) combining α -D-hexopyranosyl phosphate other than Glc1P and NTP in the presence of at least one mutant Ep nucleotidylyltransferase, wherein the at least one mutant Ep nucleotidylyltransferase is mutated at 50 one or more amino acids of SEQ ID NO:1 selected from the group consisting of V173, G147, W224, N122, G175, D111, E162, T201, I200, E199, R195, L89, L109, Y146, and Y177, and (b) recovering the resulting nucleotide sugars.
- 3. A nucleotide sugar library comprising two or more 55 nucleotide sugars produced by (a) combining α -D-hexopyranosyl phosphate and NTP other than TTP in the presence of at least one mutant Ep nucleotidylyltransferase, wherein the at least one mutant Ep nucleotidylyltransferase is mutated at one or more amino acids of SEQ ID NO:1 selected from the 60 group consisting of V173, G147, W224, N122, G175, D111, E162, T201, I200, E199, R195, L89, L109, Y146, and Y177, and (b) recovering the resulting nucleotide sugars.
- 4. A nucleotide sugar library comprising two or more nuceleotide sugars selected from the group consisting of Thymidine 5'-(α -D-glucopyranosyl diphosphate) (58); Uridine 5'-(α -D-glucopyranosyl diphosphate) (59);

Thymidine 5'-(2-deoxy-α-D-glucopyranosyl diphosphate) (60):

Uridine 5'-(2-deoxy-α-D-glucopyranosyl diphosphate) (61):

Thymidine 5'-(3-deoxy- α -D-glucopyranosyl diphosphate) (62);

Uridine 5'-(3-deoxy-α-D-glucopyranosyl diphosphate) (63);

Thymidine 5'-(4-deoxy- α -D-glucopyranosyl diphosphate) (64);

Uridine 5'-(4-deoxy-α-D-glucopyranosyl diphosphate) (65);

Thymidine 5'-(6-deoxy-α-D-glucopyranosyl diphosphate) (66);

Uridine 5'-(6-deoxy- α -D-glucopyranosyl diphosphate) (67);

(67);
Thymidine 5'-(α-D-mannopyranosyl diphosphate) (68);
Uridine 5'-(α-D-mannopyranosyl diphosphate) (69);
Thymidine 5'-(α-D-galactopyranosyl diphosphate) (70);
Uridine 5'-(α-D-galactopyranosyl diphosphate) (71);
Thymidine 5'-(α-D-allopyranosyl diphosphate) (72);
Uridine 5'-(α-D-allopyranosyl diphosphate) (73);
Thymidine 5'-(α-D-altropyranosyl diphosphate) (74);
Uridine 5'-(α-D-altropyranosyl diphosphate) (75);
Thymidine 5'-(α-D-gulopyranosyl diphosphate) (76);
Uridine 5'-(α-D-gulopyranosyl diphosphate) (78);
Uridine 5'-(α-D-idopyranosyl diphosphate) (79);
Thymidine 5'-(α-D-talopyranosyl diphosphate) (80);
Uridine 5'-(α-D-talopyranosyl diphosphate) (81);
Thymidine 5'-(α-D-talopyranosyl diphosphate) (81);

Thymidine 5'-(6-amino-6-deoxy-α-D-glucopyranosyl diphosphate) (109);

60

65

Uridine 5'-(6-amino-6-deoxy-α-D-glucopyranosyl diphosphate) (110);

Thymidine 5'-(4-amino-4-deoxy- α -D-glucopyranosyl diphosphate) (111);

Uridine 5'-(4-amino-4-deoxy-α-D-glucopyranosyl diphosphate) (112);

Thymidine 5'-(3-amino-3-deoxy-α-D-glucopyranosyl diphosphate) (113);

Uridine 5'-(3-amino-3-deoxy-α-D-glucopyranosyl diphosphate) (114);

Thymidine 5'-(2-amino-2-deoxy- α -D-glucopyranosyl diphosphate) (115);

Uridine 5'-(2-amino-2-deoxy-α-D-glucopyranosyl ¹⁵ diphosphate) (116);

Thymidine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl diphosphate) (117);

Uridine 5'-(6-acetamido-6-deoxy-α-D-glucopyranosyl ²⁰ diphosphate) (118);

Thymidine 5'-(4-acetamido-4-deoxy- α -D-glucopyranosyl diphosphate) (119);

Uridine 5'-(4-acetamido-4-deoxy-α-D-glucopyranosyl 25 diphosphate) (120);

Thymidine 5'-(3-acetamido-3-deoxy-α-D-glucopyranosyl diphosphate) (121);

Uridine 5'-(3-acetamido-3-deoxy-α-D-glucopyranosyl ₃₀ diphosphate) (122);

Thymidine 5'-(2-acetamido-2-deoxy-α-D-glucopyranosyl diphosphate) (123);

Uridine 5'-(2-acetamido-2-deoxy- α -D-glucopyranosyl diphosphate) (124);

Thymidine 5'-(4-amino-4,6-dideoxy-α-D-glucopyranosyl diphosphate) (125);

Uridine 5'-(4-amino-4,6-dideoxy-α-D-glucopyranosyl diphosphate) (126);

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$$\begin{array}{c} \textbf{81} \\ \textbf{-continued} \\ \textbf{-continued} \\ \textbf{OH} \\ \textbf{OP} \\ \textbf{OH} \\$$

$$\begin{array}{c} H_2N \\ \\ OH \\ O-P \\ O-P \\ OH \\ \end{array}$$